

# **DIY Battery**

The League of Amazing Programmers EE Class – Level 1

#### **Materials**

- 3 dosage cups
- ½ cup water
- ½ tsp salt
- 30 cm long copper tape strip, 6 pcs
- 30 cm long aluminum tape strip, 6 pcs
- Small stones (for weight)
- Jumper wires with small alligator clips, 4 pcs
- Multimeter
- Digital clock with wires

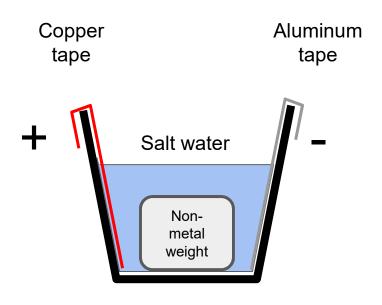


#### Instructions

This creates a simple battery using copper tape (often used for crafts), aluminum tape (often used for home heater ducting), and a salt water solution. It doesn't put out much current (about 1 mA max) but is enough to drive an LCD clock.

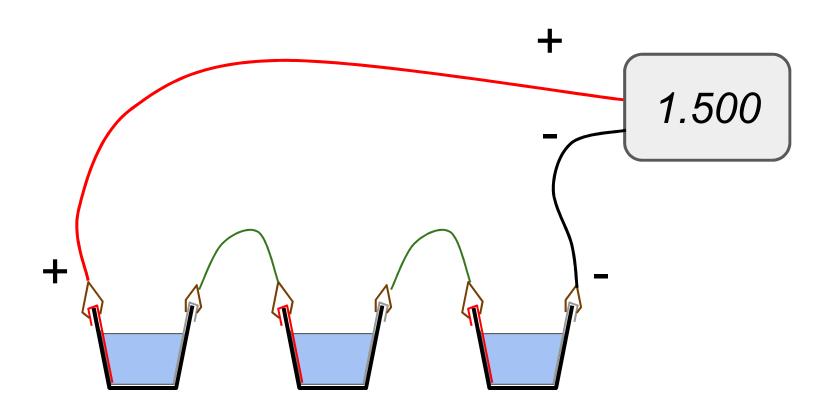
- Take a 30 cm length of each tape and tape on opposite sides of dosage cup
- Fold extra tape over the lip of the cup
- Wipe metal tape clean to remove oils and other contaminants
- Add a small stone or other non-metal weight to help keep it stable
- Attach alligator clips and wire together
- Make a salt water solution with  $\frac{1}{2}$  cup water and  $\frac{1}{2}$  to 1 teaspoon of salt
- Pour salt solution into cup. Fill ½ to ¾ to the top.

#### How to make one cell, about 0.5 Volts

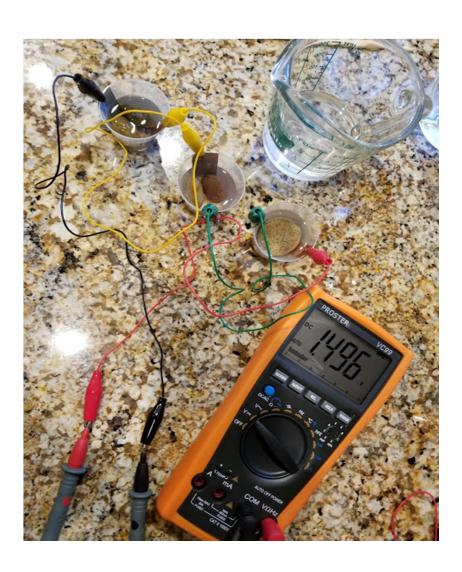




## Wire three cells together, about 1.5 Volts!



# Example Setup



# Appendix

### Electronegativity

Electronegativity is a chemical property which describes how well an atom can attract an electron to itself.

| Н   |     | Electronegativity values of the elements (Pauling scale) |     |     |     |     |     |     |     |     |     |     |     | He  |     |     |     |
|-----|-----|--|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|
| 2.1 | Be  | BCNOF  |     |     |     |     |     |     |     |     |     |     |     |     | Ne  |     |     |
| 1.0 | 1.5 |  |     |     |     |     |     |     |     |     |     | 2.0 | 2.5 | 3.0 | 3.5 | 4.0 | IVE |
| Na  | Mg  |  |     |     |     |     |     |     |     |     |     | Al  | Si  | Р   | S   | CI  | Ar  |
| 0.9 | 1.2 |  |     |     |     |     |     |     |     |     |     | 1.5 | 1.8 | 2.1 | 2.5 | 3.0 |     |
| K   | Ca  | Sc   | Ti  | ٧   | Cr  | Mn  | Fe  | Co  | Ni  | Cu  | Zn  | Ga  | Ge  | As  | Se  | Br  | Kr  |
| 8.0 | 1.0 | 1.3  | 1.5 | 1.6 | 1.6 | 1.5 | 1.8 | 1.8 | 1.8 | 1.9 | 1.6 | 1.6 | 1.8 | 2.0 | 2.4 | 2.8 | 3.0 |
| Rb  | Sr  | Υ  | Zr  | Nb  | Мо  | Tc  | Ru  | Rh  | Pd  | Ag  | Cd  | In  | Sn  | Sb  | Te  | 1   | Xe  |
| 8.0 | 1.0 | 1.2  | 1.4 | 1.6 | 1.8 | 1.9 | 2.2 | 2.2 | 2.2 | 1.9 | 1.7 | 1.7 | 1.8 | 1.9 | 2.1 | 2.5 | 2.6 |
| Cs  | Ва  | La   | Hf  | Ta  | W   | Re  | Os  | lr  | Pt  | Au  | Hg  | Ti  | Pb  | Bi  | Рο  | At  | Rn  |
| 0.7 | 0.9 | 1.1  | 1.3 | 1.5 | 1.7 | 1.9 | 2.2 | 2.2 | 2.2 | 2.4 | 1.9 | 1.8 | 1.8 | 1.9 | 2.0 | 2.2 | 2.4 |
| Fr  | Ra  | Ac   |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |
| 0.7 | 0.7 | 1.1  |     |     |     |     |     |     |     |     |     |     |     |     |     |     |     |

| Ce  | Pr  | Nd  | Pm  | Sm  | Eu  | Gd  | Tb  | Dy  | Нο  | Er  | Tm  | Yb  | Lu  |
|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|
| 1.1 | 1.1 | 1.1 | 1.1 | 1.1 | 1.1 | 1.1 | 1.1 | 1.1 | 1.1 | 1.1 | 1.1 | 1.1 | 1.2 |
| Th  | Pa  | U   | Np  | Pu  | Am  | Cm  | Bk  | Cf  | Es  | Fm  | Md  | No  | Lr  |
| 1.3 | 1.5 | 1.7 | 1.3 | 1.3 | 1.3 | 1.3 | 1.3 | 1.3 | 1.3 | 1.3 | 1.3 | 1.3 |     |

#### References

- List of electronegativity of the elments: <a href="https://sciencenotes.org/list-of-electronegativity-values-of-the-elements/">https://sciencenotes.org/list-of-electronegativity-values-of-the-elements/</a>
- How batteries work
  - https://www.youtube.com/watch?v=PyrWx4ExZE4
  - https://www.youtube.com/watch?v=9OVtk6G2TnQ
  - https://www.google.com/search?q=diy+lemon+battery&rlz=1C1NHXL\_enUS754US754&oq=diy+lemon+battery&aqs=chrome..69i57.2926j0j4&sourceid=chrome&ie=UTF-8#kpvalbx=1
  - https://en.wikipedia.org/wiki/Electric battery