

Design and Development of High-Performance Hydrogel Gas Sensors

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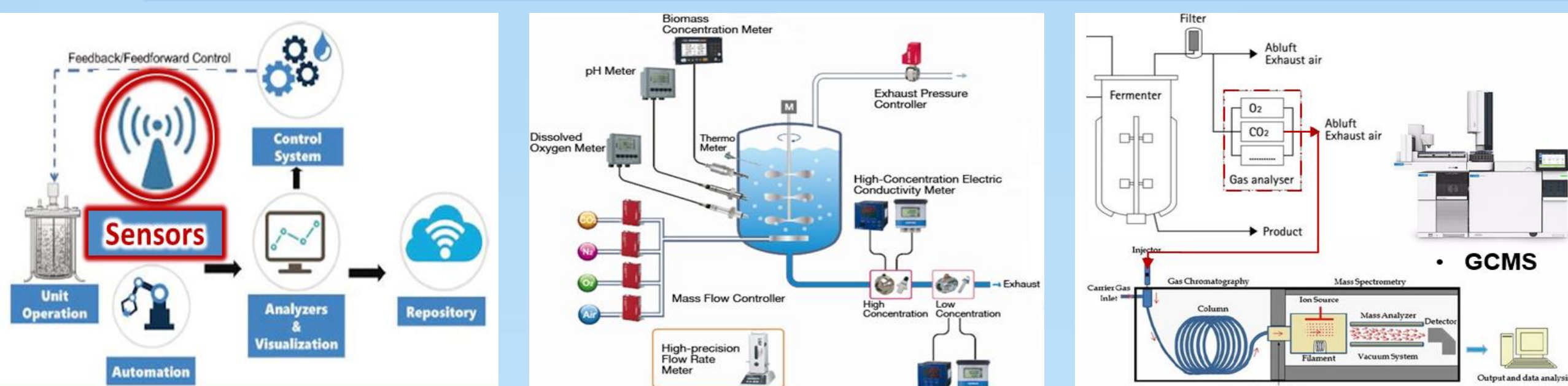
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Abstract

In order to achieve accurate monitoring of exhaust gas in bioreactor during biofabrication, this study presents a method for preparing a hydrogel containing dimethylamine and applies it to CO₂ detection at room temperature. By examining the structural characteristics and electrical response signals of the hydrogel, the CO₂ sensitivity mechanism of the dimethylamine-structured hydrogel is explored, providing guidance for the design and development of highly sensitive CO₂ sensors in room temperature environments. The results show that sensors made from the hydrogel material exhibit excellent CO₂ sensitivity at room temperature. Moreover, as the concentration of dimethylamine in the hydrogel solution increases, both the sensitivity and lifespan of the sensor are enhanced. This improvement is primarily due to the high selectivity of the dimethylamine structure in the hydrogel, which, with the assistance of water, can capture CO₂ gas molecules in the environment, thereby altering the hydrogel's conductivity. This study offers a novel approach for the design and development of CO₂ gas sensors with high sensitivity and long lifespan.

Introduction

Background Accurate monitoring of exhaust gas in bioreactor during biofabrication

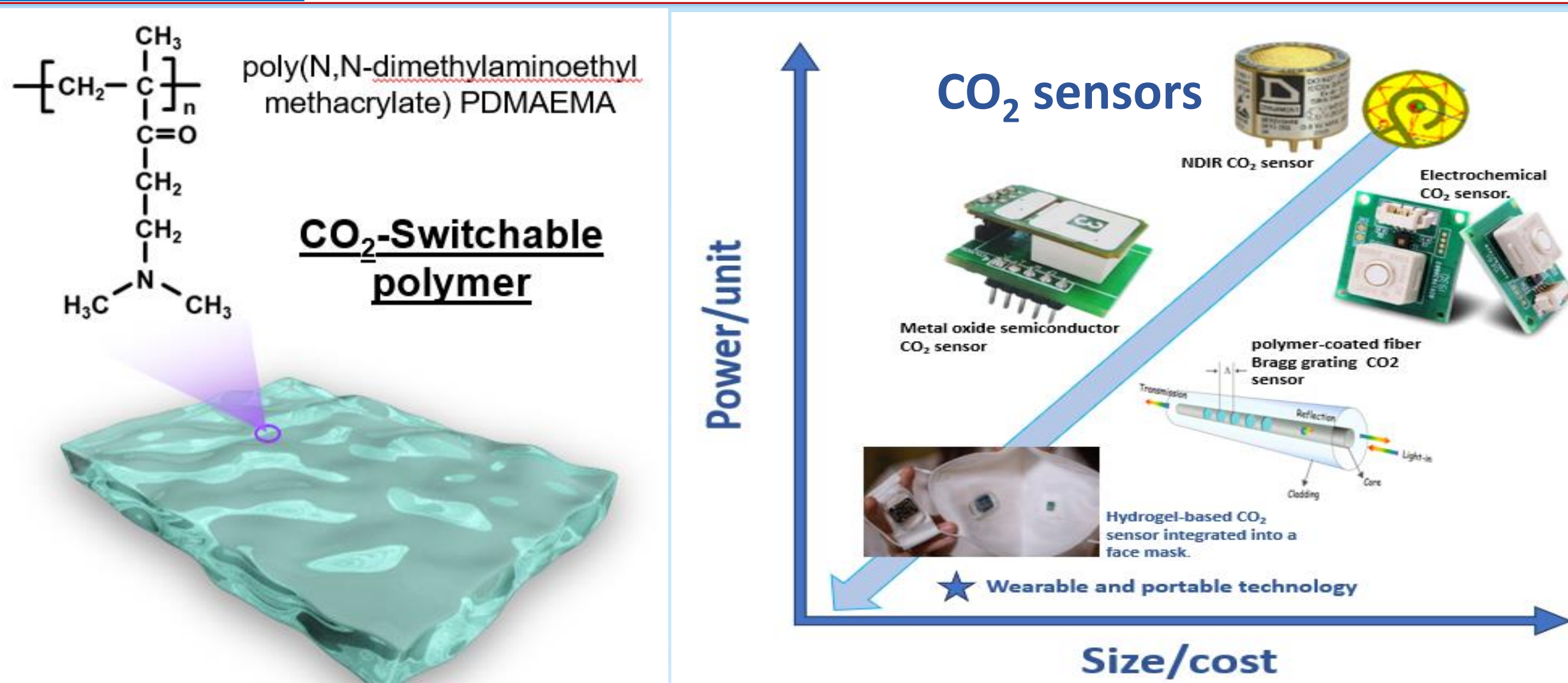


Bio-manufacturing

Process Monitoring

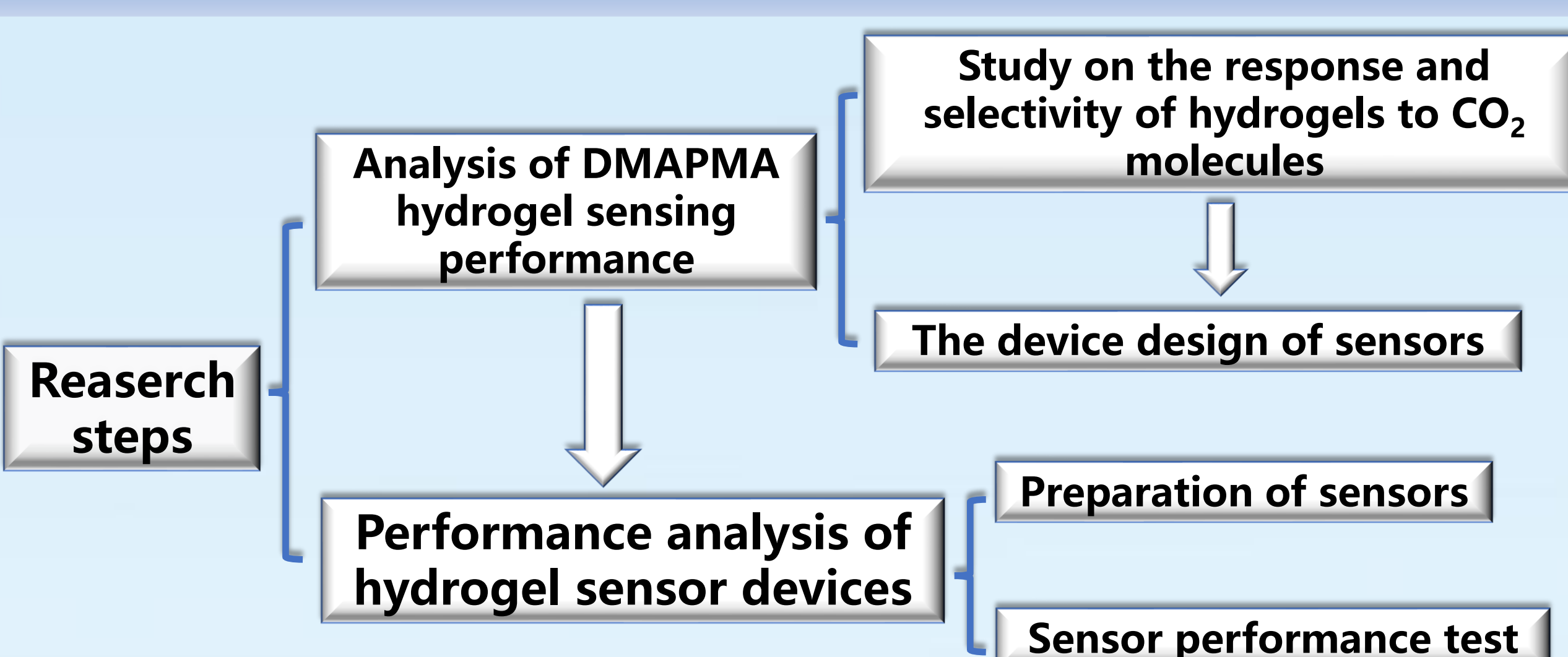
Emission Monitoring

Challenge Realize real-time online monitoring of exhaust gas in biofabrication process

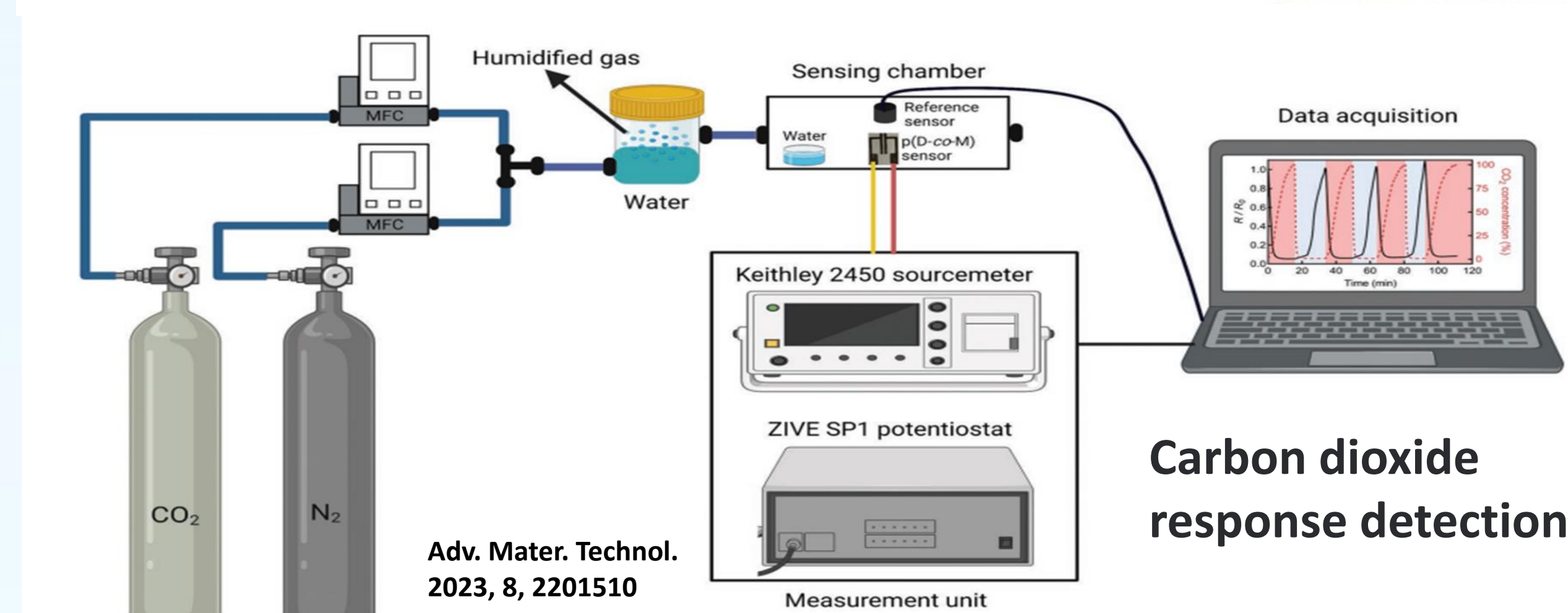
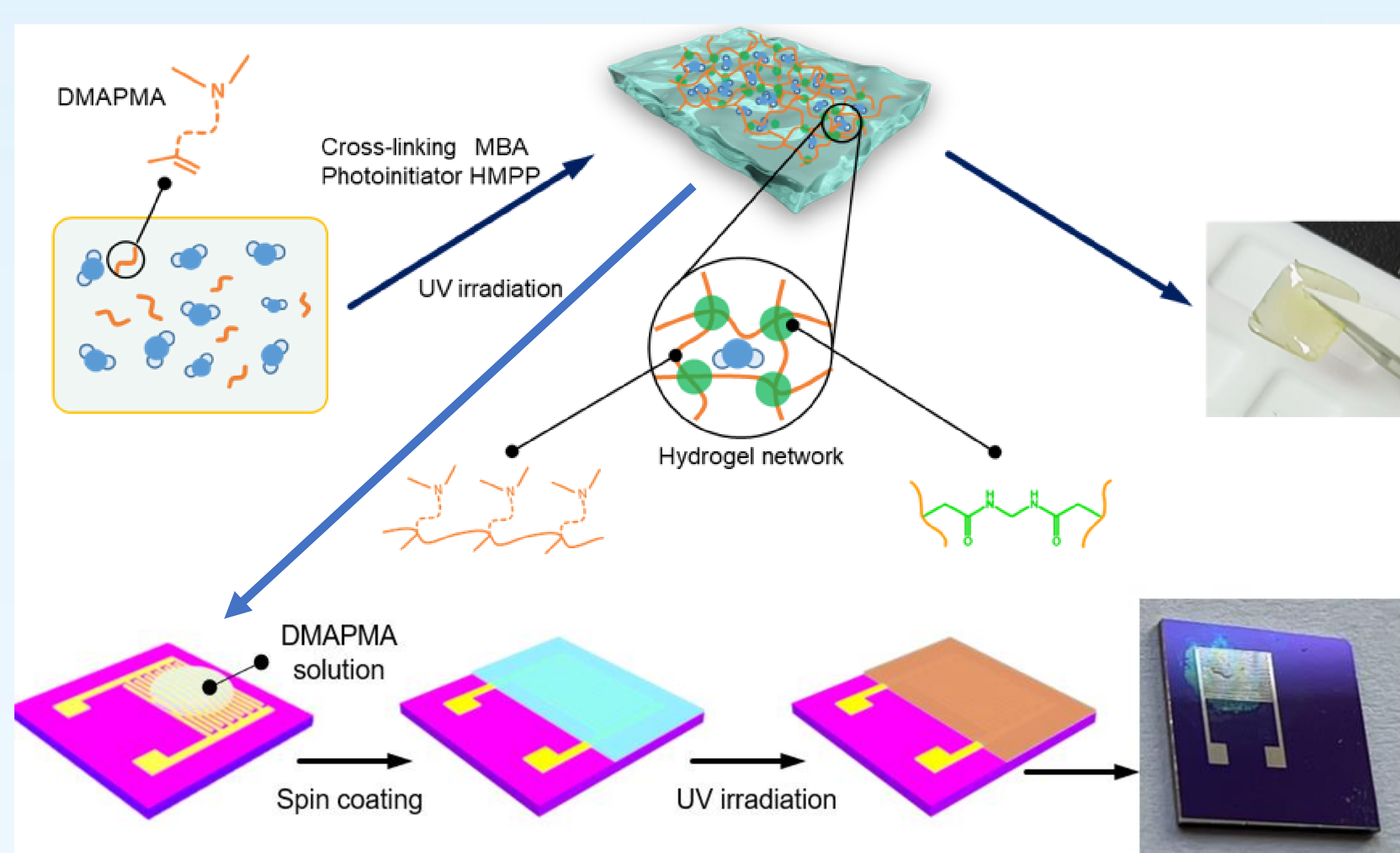


Compared to other CO₂ sensors, the hydrogel sensor with high-performance response has small device size and lower energy consumption, making it a promising choice for real-time online monitoring in biofabrication process.

Methods



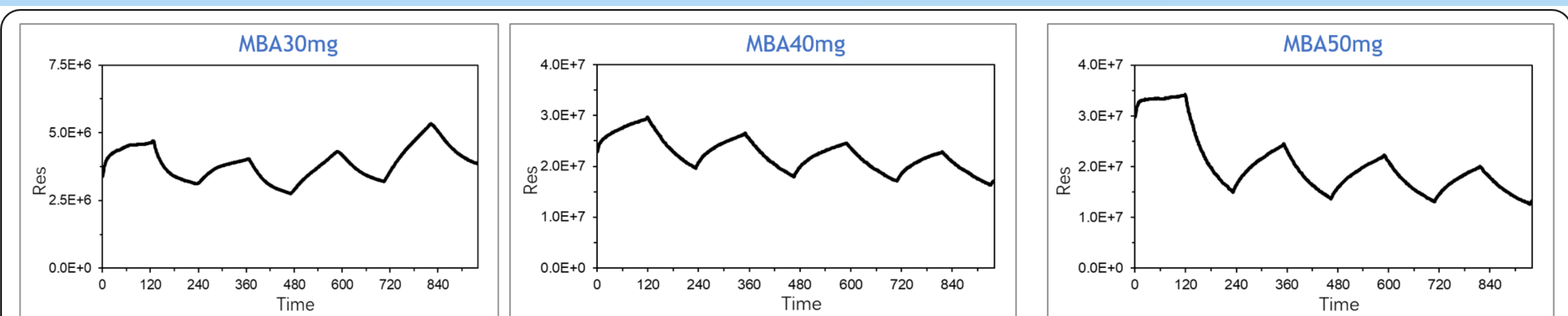
The materials for the preparation of hydrogel:
DMAPMA, MBA, HMPP, deionized water.



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Results

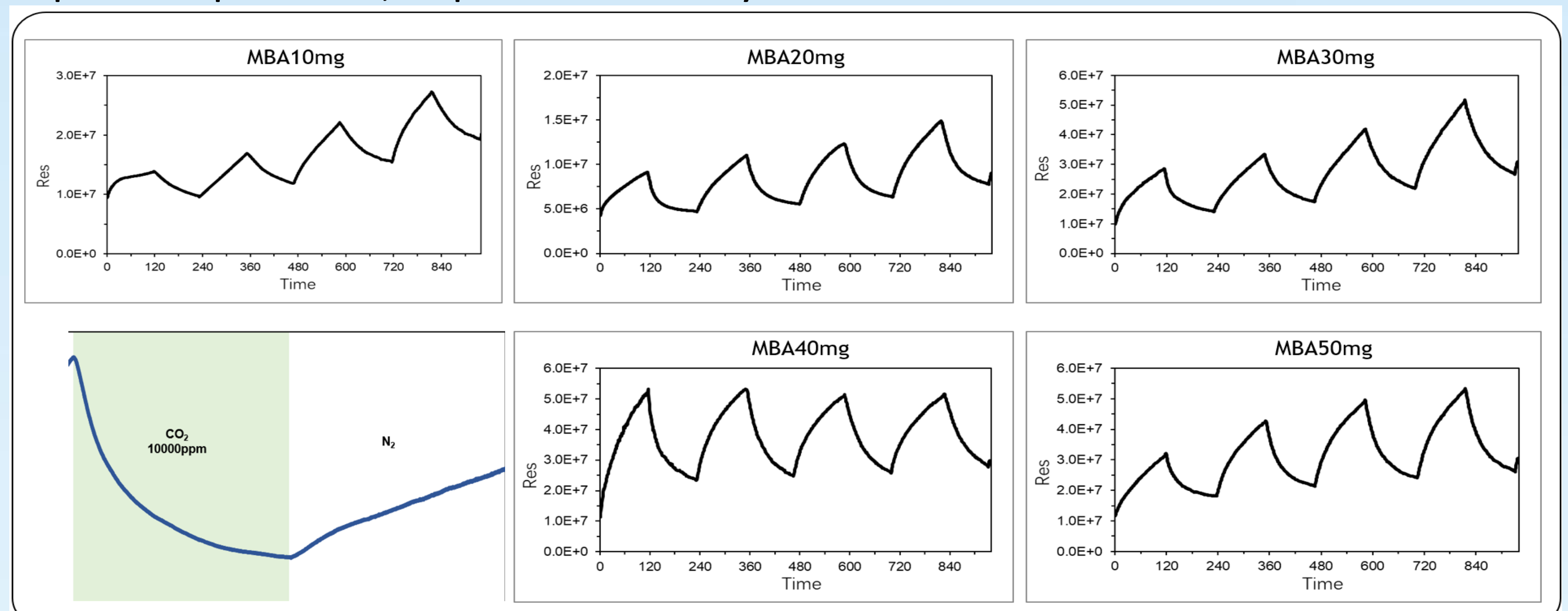
The analysis of hydrogel polymerization



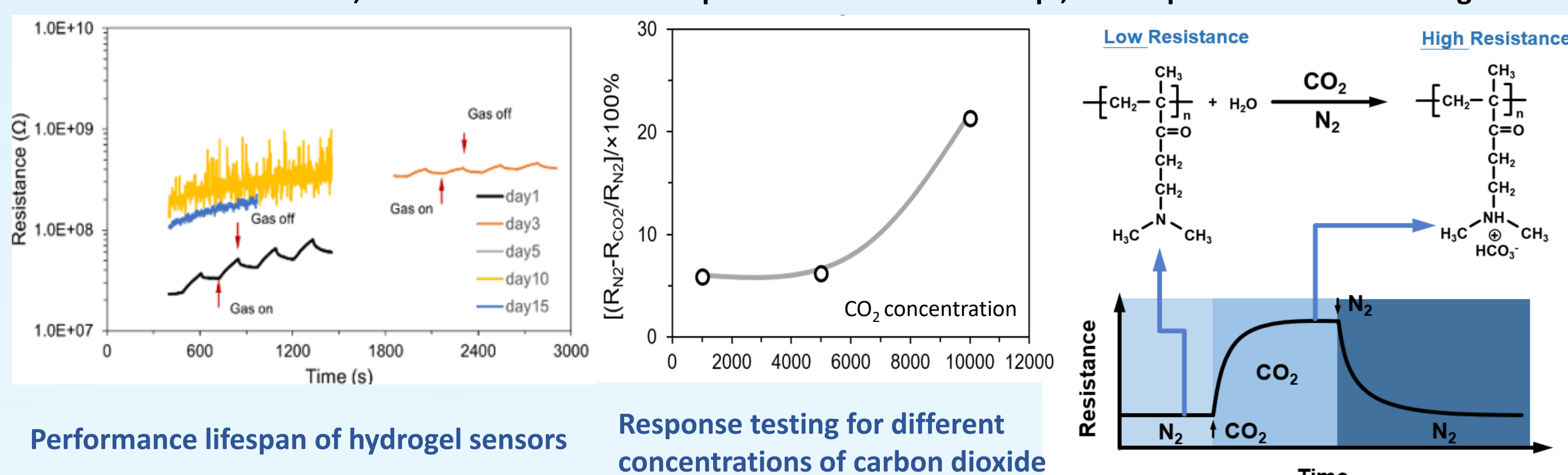
When the illumination time exceeds 10 minutes, the water content of the hydrogel decreases, the performance deteriorates, and the response curve does not return to the original peak

	DMAPMA(μl)	deionized water(μl)	MBA(mg)	HMPP(μl)	polymerization time(min)	
A	100		1	10	1	5
B	200		2	20	2	10
C	300	1000	3	30	3	15
D	500		4	40	4	20
E	1000		5	50	5	30

According to the experiments, when the volume of HMPP is ≥50μl, the precursor solution cannot be mixed uniformly, and when the mass of MBA is ≤10mg, the hydrogel does not polymerize. The most suitable illumination time is 10 minutes, less than 10 minutes of non-condensation, more than 10 minutes of hydrogel moisture reduction, resulting in poor device performance; The performance is directly related to the amount of DMAPMA and MBA.



Nitrogen and carbon dioxide were injected alternately for 120s. When MBA was 40mg, HMPP was 20μl, illumination was 10min, deionized water was 1000μl and DMAPMA was 500μl, the response was the most significant.



The increase in DMAPMA's concentration in the hydrogel solution leads to a higher concentration and total amount of polymers, which helps to improve the sensor's response to carbon dioxide and extends the sensor's lifespan. In the presence of water molecules, CO₂ molecules can undergo a reversible adsorption-desorption reaction with organic molecules containing a dimethylamine structure, demonstrating specific CO₂ molecular recognition capability.

Conclusion

1. When the volume of HMPP is below 50μl and the amounts of MBA are above 10mg, the hydrogel polymerization can achieve great performance under 10 minutes of ultraviolet light.
2. With the volume of HMPP and deionized water kept constant at 20μl and 1000μl, respectively, the sensitivity is most pronounced when MBA is 40mg and DMAPMA is 500μl.
3. As the concentration of DMAPMA increases, the lifespan of the hydrogel carbon dioxide sensor significantly improves.