# DFT Theory and Application to Defect Cluster Concentration

## Tigany Zarrouk

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		tart with the theory of DFT with the HK theorem and KS equation solve for the density.	ons

- Talk a bit about the approximations that are made.
- Alloy theory and the structure of Ti3Al
- Look at literature to do with all of this. Go through why solutes are incredibly important and why it is necessary do to this research.

- How can solutes lead to failure and why is oxygen in particular bad for the alloy.
- Is it wavy or planar slip
- Go into the calculations and what I have done for them
- Cite the research about the Fe and C vacancy concentration and what the implications are for alloy research
- What can I add to the research.

#### 1 Introduction

Since the discovery that quanta may accurately describe the phenomena of black body radiation, it has been known that atoms are governed by a different physical paradigm to that of matter on the macroscopic scale: quantum mechanics. To model the behaviour of atoms in a solid, we can use quantum mechanics.

The first postulate of of quantum mechanics states that a quantum mechanical system can be described by a wavefunction, a function of the positions of the quantum mechanical entities, and time, which satisfies the Schr\:odinger equation. Bonding in materials is heavily dependent on how electrons arrange themselves upon the assembly of atoms to make solid. By the first postulate we would expect upon assembly, the state of a the system can be fully described by a many body wavefunction: a wavefunction which is a function of the positions of each of the electrons and nuclei in the system.

Assuming that we have non-relativistic electrons, the Schr\:odinger equation one must solve is then

$$\Big(\sum_i -\frac{1}{2}\nabla_i^2 + \frac{1}{2}\sum_i \sum_i \frac{1}{|\mathbf{r}_i - \mathbf{r}_j|} + \sum_i \sum_I \frac{-Z_I}{|\mathbf{r}_i - \mathbf{R}_I|} \Big) \Psi(\{\mathbf{r}\};t) = E\Psi(\{\mathbf{r}\};t),$$

where  $\Psi(\{\mathbf{r}\};t)$  is an eigenstate and Hartree units have been used (  $e=m=\hbar=4\pi\epsilon_0=1$  ).

A time-independent observable is the expectation value of a given operator,

$$\left\langle \hat{O}\right\rangle =\frac{\langle \Psi|\hat{O}|\Psi\rangle}{\langle \Psi|\Psi\rangle},$$

this is an integral over all of the coordinates.

The electronic density operator is defined as

$$\hat{n}(\mathbf{r}) = \sum_{i=1}^{N} \delta(\mathbf{r} - \mathbf{r}_i)$$

$$n(\mathbf{r}) = \frac{\langle \mathbf{\Psi} | \hat{n}(\mathbf{r}) | \mathbf{\Psi} \rangle}{\langle \mathbf{\Psi} | \mathbf{\Psi} \rangle}.$$

The total energy is the expectation value of the Hamiltonian:

$$E = \frac{\langle \mathbf{\Psi} | \hat{H} | \mathbf{\Psi} \rangle}{\langle \mathbf{\Psi} | \mathbf{\Psi} \rangle} \tag{1}$$

$$= \left\langle \hat{T} \right\rangle + \left\langle \hat{V}_{\text{int}} \right\rangle + \int d^3 V_{\text{ext}} n(\mathbf{r}) + E_{II}, \tag{2}$$

where  $V_{\rm ext}$  is the external potential due to the coulomb interaction acting on the electrons from the nuclei—the expectation value has been explicitly written as an integral over the local electron density,  $E_{II}$  is the classical electrostatic nucleus-nucleus interaction energy,  $V_{\rm int}$  is the electron-electron interaction energy and  $\langle \hat{T} \rangle$  is the expectation value of the electronic kinetic energy.

Stationary points in the total energy correspond to eigenstates of the many-body Hamiltonian. One can vary the ratio in (2), or vary the numerator in (2) subject to the constraint of orthonormality ( $\langle \Psi | \Psi \rangle = 1$ ), which is possible with the use of Lagrange multipliers. One finds that upon variation of the bra  $\langle \Psi |$  that the ket must satisfy the time-independent Schr\:odinger equation:

$$\hat{H} | \mathbf{\Psi}_m \rangle = E | \mathbf{\Psi} \rangle,$$

where  $|\Psi_m\rangle$  is an eigenstate.

Quite often, the state we would most like to find is the lowest energy state (ground state) of the system, as this is the fundamental state from which other mechanisms return to or start from. This necessarily occurs at zero kelvin by the third law of thermodynamics. To find this state for the full system, one must minimise the energy with respect to the parameters of the many-body wavefunction that satisfies the Schr\:odinger equation and appropriate symmetry constraints (e.g. for electrons,  $\Psi$  must be antisymmetric). This quickly leads to a problem of computation: each electron has

four degrees of freedom (the quantum numbers n, m, l and s), and each nucleus will have  $n_I^d$  degrees of freedom; given a collection of just M atoms, we would have  $4M + Mn_I^d$  variables to minimise with respect to energy. To make the problem tractable, we must make a few approximations.

#### 1.1 Born-Oppenheimer Approximation

The first approximation that we make is the Born-Oppenheimer approximation, which allows us to separate nuclear and electronic motion. One must first make the assumption that the wavefunction which describes the system is a product state between the nuclear and electronic portions of the system.

$$\psi(\mathbf{R}, \mathbf{r}) = \Phi(\mathbf{R}) \cdot \psi_{\mathbf{R}}(\mathbf{r}),$$

where  $\psi_{\mathbf{R}}(\mathbf{r})$  is the electronic wavefunction and  $\Phi(\mathbf{R})$  is the nuclear portion of the wavefunction.

Due to the large disparity in mass between an electron and the nucleus  $(M_{\text{Nuc}} \sim 2000 m_e)$  we can neglect the contribution to the Hamiltonian that comes from the nuclear kinetic energy operator acting on the electronic wavefunction, as the resulting term is far smaller than the electronic kinetic energy operator acting on the electronic wavefunction.

$$T_{\text{Nuclear}}(\mathbf{R})\psi_{\mathbf{R}}(\mathbf{r}) = \sum_{I=1}^{M} -\frac{1}{2M_I}\nabla_{\mathbf{R}_I}^2\psi_{\mathbf{R}}(\mathbf{r})$$

$$T_{\mathrm{Electronic}}(\mathbf{r})\psi_{\mathbf{R}}(\mathbf{r}) = \sum_{i=1}^{M} -\frac{1}{2m_i} \nabla_{\mathbf{r}_i}^2 \psi_{\mathbf{R}}(\mathbf{r})$$

Due to this, we can say that the motion of the electrons is instantaneous with regard to the motion of the ions in the system and that the electrons relax into their ground state with respect to any configuration of the ions. This results in the total energy being a function of only the nuclear coordinates [1].

To obtain the total energy, we can solve for the electronic part of the wavefunction and then solve for the nuclear wavefunction,

$$(T_{\text{Nuclear}} + E_{\mathbf{R}})\Phi(\mathbf{R}) = E_{\text{Total}}\Phi(\mathbf{R}),$$

where  $E_{\mathbf{R}}$  contains all of the information of the configuration of the electrons in the system.

#### 1.2 Self-Consistent Mean-Field Theory

#### 1.3 Density Functional Theory

#### 1.3.1 Theory

Now that the electronic motion has been decoupled from that of the nucleus, we can now try to find the ground-state solution for a given configuration of ions. A computationally tractable way of doing this for a many-body solid describing all of the electrons is via the *Density Functional Theory*.

The main tenet of Density Functional Theory is: given a system that consists of N electrons, there exists an energy functional of the local electron density  $E[\rho(\mathbf{r})]$ , which corresponds to an antisymmetric wavefunction  $|\Psi\rangle$ ; upon minimising the energy with respect the changes in the electron density, subject to the constraint that the number of electrons is conserved,

$$\int_{\text{all space}} \rho(\mathbf{r}) = N,$$

the resulting density is unique and the value of the functional is the groundstate energy.

Assuming  $\rho(\mathbf{r})$  is the exact ground state density, and similarly that  $|\Psi\rangle$  is the ground-state, by the variational principle we can write,

$$\langle \Psi | \hat{\mathcal{T}} + \frac{1}{2} \sum_{i} \sum_{j} \frac{1}{|\mathbf{r}_{i} - \mathbf{r}_{j}|} |\Psi \rangle + \int \rho(\mathbf{r}) V_{\text{ext}}(\mathbf{r}) d\mathbf{r} \geq E_{0}$$

A functional is an entity that maps a function to a value, similar as how a function maps a variable to a value. The Hohenburg-Kohn theorem states that there exists a functional of the electron density which has the correct ground state energy upon minimising the energy with respect to the electron density [2]. Thus all the information of the system is contained in the electron density, which reduces the minimisation problem fron 4M variables to one of just 3 for any number of electrons in the system.

To actually find this density, one can use the Kohn-Sham equations to find a self-consistent solution for the electron density [3].

The process by which this happens are as follows: one solves for the Hartree potential first with a given input density (the solution of Poisson's equation), then one finds the total effective potential for the system which is the sum of the Hartree potential, the potential from the nuclei  $(V_{\text{ext}})$  and the exchange-correlation potential  $V_{\text{xc}}$ . The Schr\:odinger equation is subsequently solved, and a new electron density is found. This density can be

- 1.3.2 Practical steps towards accurate calculations
- 2 Defects in Materials
- 2.1 Vacancies and Solutes
- 2.2 Ti3Al Solutes and their effects.
- 2.3 Current research: Vacancy-Solute Complexes.
- 3 Bibliography

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