

test

Grade:

STUDENT ID



BEYER HIGH SCHOOL

2015-2016

For:

$$\frac{-1.65 \times 10^{-3} \text{ (molar)} \times 0.01 \text{ (L)} \times 0.01 \text{ (molar)}}{0.0001 \text{ (molar)}} \left(1 - \frac{1}{100}\right)$$

$$E = -1.55 \text{ mV}$$

For the higher molar mass due to the greater number of ions.

