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## DEGREES OF FREEDOM

$$L(q_1, q_2, \dots, q_n, \dot{q}_1, \dot{q}_2, \dots, \dot{q}_n)$$

GENERALIZED COORDS
GENERALIZED VELOCITIES

2N DEGREE OF FREEDOM

•  $L = L(q_1, \dots, \cancel{\dot{q}_n}, \dots, q_n, \dot{q}_1, \dots, \dot{q}_n)$

MISSING  $\dot{q}_n$

ALSO  $\dot{q}_n$

$$L(x, \dot{x}) = \frac{1}{2} m \dot{x}^2$$

$$\rightarrow m \ddot{x} = 0 \quad \frac{d}{dt}(m \dot{x}) = 0 \rightarrow \dot{p} = 0$$

$$\downarrow$$

$$\partial_{\dot{x}} L = p \quad p \text{ CONSTANT}$$

$$\frac{d}{dt} \partial_{\dot{x}} L = \partial_x L$$

$$\frac{d}{dt} p = 0 \Rightarrow p_k = \frac{\partial L}{\partial \dot{q}_k} \text{ IS A CONSERVED QUANTITY}$$

## EXERCISE of MOTION



$$\text{KINETIC} = \frac{1}{2} m \dot{x}^2 \quad \leftarrow v^2$$

$$\text{POTENTIAL} = \frac{1}{2} k x^2$$

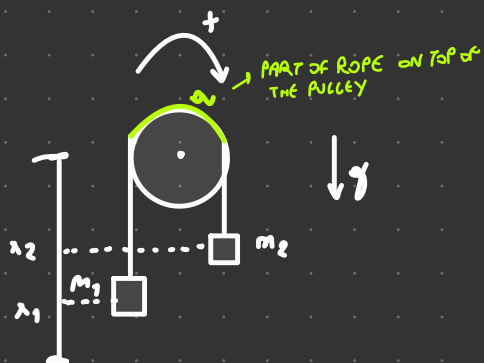
$$\frac{d}{dt} \partial_{\dot{x}} L = \partial_x L$$

$$L(x, \dot{x}) = \frac{1}{2} m \dot{x}^2 - \frac{1}{2} k x^2$$

KINETIC
POTENTIAL

$$\begin{cases} \partial_{\dot{x}} L = m \dot{x} \\ \partial_x L = -kx \end{cases} \quad \boxed{m \ddot{x} = -kx}$$

## EXERCISE ATWOOD MACHINE



$$\begin{aligned} (1) & L(x_1, x_2, \dot{x}_1, \dot{x}_2) \\ (2) & L(x_1, \dot{x}_1) \end{aligned}$$

$$\begin{aligned} \text{KINETIC } m_1 &: \frac{1}{2} m_1 \dot{x}_1^2 - \frac{1}{2} m_2 \dot{x}_2^2 \\ \text{POTENTIAL } m_1 &: m_1 g x_1 - m_2 g x_2 \\ \text{KINETIC } m_2 &: \frac{1}{2} m_1 \dot{x}_1^2 - \frac{1}{2} m_2 \dot{x}_2^2 \\ \text{POTENTIAL } m_2 &: m_2 g x_2 - m_1 g x_1 \end{aligned}$$

$$K = \frac{1}{2} m_1 \dot{x}_1^2 + \frac{1}{2} m_2 \dot{x}_2^2$$

$$U = -m_1 g x_1 - m_2 g x_2$$

$$L = \frac{1}{2} (m_1 \dot{x}_1^2 + m_2 \dot{x}_2^2) + g(m_1 x_1 + m_2 x_2)$$

$$l = x_1 + x_2 + a \rightarrow x_1 + x_2 = b \quad \leftarrow l - a$$

# THERMODYNAMICS

NOT A SINGLE ATOM  
A THERMODYNAMIC SYSTEM IS A MACROSCOPIC SYSTEM WITH A LARGE NUMBER OF DEGREES OF FREEDOM

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DOF OF SINGLE MOLECULES OF GAS



- MADE OF SMALL ELEMENTS THAT TRAVEL IN A SPACE
- EACH ONE HAVE ITS OWN VELOCITY
- THEY CAN INTERACT WITH EACH OTHER

- A LIQUID IS A THERMODYNAMIC SYSTEM TOO
- A SOLID IS A THERMODYNAMIC SYSTEM TOO

USE DYNAMICAL SYSTEMS AND ORDINARY DIFFERENTIAL EQUATIONS

LAGRANGIAN MECHANICS  
TO GET ODE STARTING  
FROM ITS ENERGY

LAGRANGIAN  
SUM OF KINETIC ENERGY  
OF ALL THE PARTICLES MINUS  
THE POTENTIAL ENERGY DUE  
TO INTERACTION

## PROBLEMS

- SMALL NUMBER OF DOF. THE EQUATION OF MOTION FOR LARGE SYSTEM WITH MANY DOF IT WOULD BE **INTRACTABLE**
- EVEN IF THE MOTION EQUATION WERE TRACTABLE. TO BE SOLVE IT NEEDS THE INITIAL CONDITION

→ EVEN IF I CAN WRITE IT DOWN  
NOBODY CAN SOLVE THEM

→ TO KNOW  $\ddot{x}=0$  NEEDS

- $x(0)$
- $\dot{x}(0)$

⇒ USE REDUCTION APPROACH

DON'T CARE ABOUT THE  
SINGLE MOLECULE BUT THE  
GENERAL APPROACH

- GAS
- BOILING WATER
- CHEMICAL REACTION
- RESISTORS

- USE MACROSCOPIC DOF INSTEAD OF MICROSCOPIC
- LOOK FAR AWAY THE SYSTEM

• FOR A GAS: TEMPERATURE, PRESSURE, VOLUME, NUMBER OF PARTICLES

• RESISTOR: CURRENT, POTENTIAL DIFFERENCE, TEMPERATURE, RESISTANCE

↓  
CURRENT INSIDE

• SPRING/RUBBER BAND: TENSION,  $\Delta x$  ELONGATION, TEMPERATURE, ELASTIC CONSTANT

GOAL OF THERMODYNAMICS: DESCRIBE SYSTEM IN TERMS OF THEIR  
MACROSCOPIC DEGREES OF FREEDOM  
(AKA THERMODYNAMICS VARIABLE)