1D AND 2D POROUS MEDIA FIXED BED REACTOR SIMULATIONS WITH DUO

18th OpenFOAM Workshop (Genoa, Italy, 2023)

Eric Daymo (Tonkomo, LLC)
Anna Lee Tonkovich (Tonkomo, LLC)
Matthias Hettel (Karlsruhe Institute of Technology)
Akash Shirsath (Karlsruhe Institute of Technology)





Tonkomo, LLC

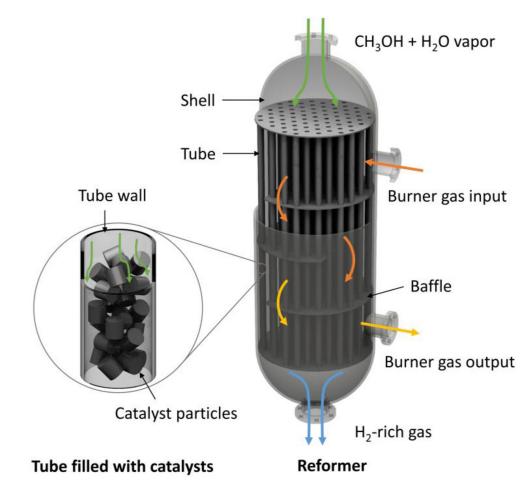
- Owners: Lee Tonkovich, PhD & Eric Daymo, PE
 - Reaction engineering expertise with 30+ years of experience (>150 patents)
 - Co-founders of Velocys (microchannel reactors)
 - Experienced with algal bioreactors (Heliae)
 - Now, owners of a small business dedicated to developing innovative solutions for *climate* – *energy* – *food* – *water*
- Motivation: cost-effective and reliable tools for reaction engineering
 - Commercial is CFD expensive for small businesses
 - OpenFOAM is missing required features for reaction engineers
- Vision: Affordable tools that liberate creativity, accelerate commercial product development, and enhance entrepreneurism





Catalytic Fixed Bed Reactors

- Tubular fixed bed catalytic reactors are the "workhorse" of the chemical industry
- CFD is one tool used to analyze fixed-bed reactor performance
- OpenFOAM does not have extensive built-in tools for such packed bed reactor calculations
 - Surface reactions
 - Packed bed heat transfer



Zhu, J., Cui, X., Araya, S.S. Comparison between 1D and 2D numerical models of a multi-tubular packed-bed reactor for methanol steam reforming, International Journal of Hydrogen Energy 47 issue 54 (2022), 22704-22719



DETCHEM and DUO



www.detchem.com

- DETCHEM stands for DETailed CHEMistry
 - Software package for reacting flow simulation
 - www.detchem.com

Reference: Deutschmann et al. 2022

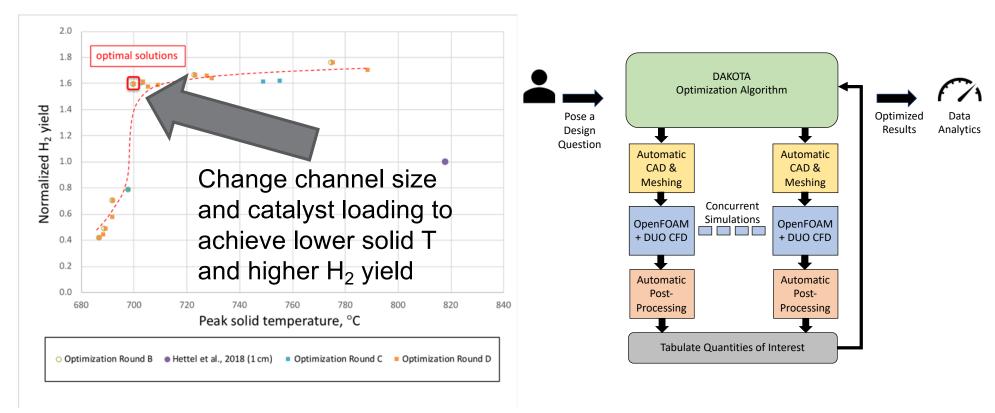
- DUO: DETCHEM und OpenFOAM coupling
 - Surface reactions
 - Kinetic theory transport properties
 - Mixture/multicomponent diffusion

Product	Description
Manual	Manual for DETCHEM
EQUIL	Equilibrium and adiabatic reactor temperature calculation.
GASPROBE	Tool for evaluating the transport and kinetic properties of the chemical species participating in a gas-phase reaction mechanism.
SURFPROBE	Tool for evaluating the transient and steady state surface coverages of various chemical species participating in a surface reaction mechanism.
ВАТСН	Reactor model for simulating the behavior of an ideal batch reactor for gas-phase as well as for surface reactions.
MPTR	Reactor model for simulating the behaviour of a tank reactor with gas-phase and multiple condensed phases.
CSTR	Reactor model for simulating behavior of an continuous stirred tank reactor for gas-phase as well as surface reactions.
PLUG	Reactor model for simulating the behavior of an ideal plug flow reactor for gas-phase as well as surface reactions.
STAG	Transient one-dimensional model for the simulation of stagnation flow reactors with models for catalytic surface reactions, external and internal diffusion, and gas-phase reactions.
PBR	Simulates a packed-bed reactor including a full-featured internal pellet resistance model (e.g., calculate effectiveness factor as a function of reactor length, or calculate internal pellet composition as a function of pore length). PBR is validated against experimental data and accurately captures radial heat transfer effects in a packed bed.
CHANNEL	Two dimensional model based on boundary layer equations for simulating the behavior of tubular reactor for gas-phase as well as for surface reactions.
MONOLITH	Transient code for simulating catalytic monoliths in 2D and 3D.
SOFC	Two-dimensional solid-oxide fuel cell model for the simulation of button cell, planar cells as well as tubular cells for co- flow configuration.
RESERVOIR	Transient single channel model.
DUO	Interface to OpenFOAM to calculate coupled fluid and solid regions with additional detailed gas-phase or surface chemistry (e.g. coated foams). For the modeling of monoliths the flow in the channels can be handled in 1D or 2D optionally.





Why do we need simpler CFD? One reason: Optimization and Discovery



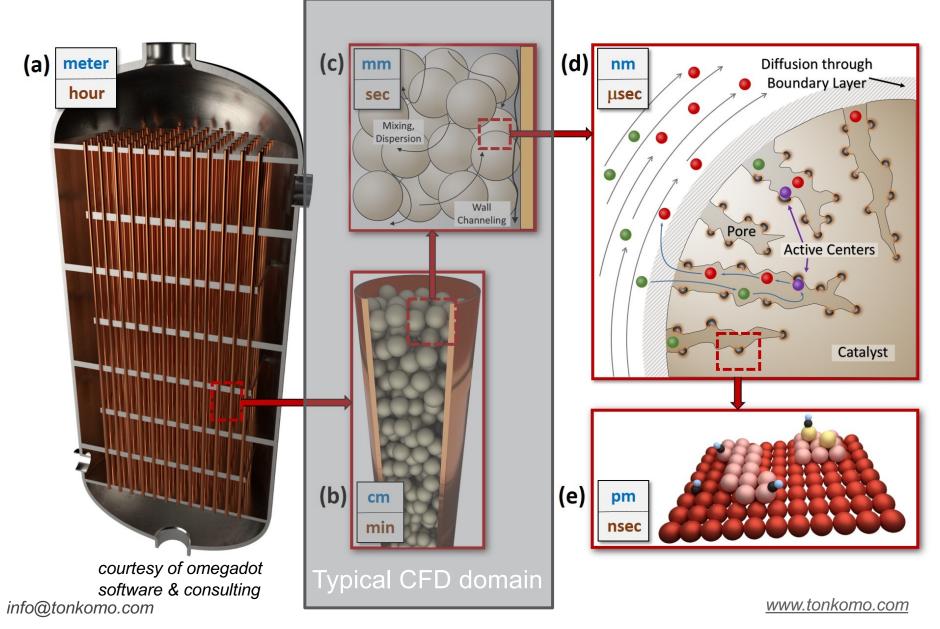
Optimization: often 10,000's of cases run ... need fast results

Daymo, E., Tonkovich, A., Hettel, M., Guerrero, J. Accelerating Reactor Development with Accessible Simulation and Automated Optimization Tools, Chemical Engineering and Processing - Process Intensification, 2019





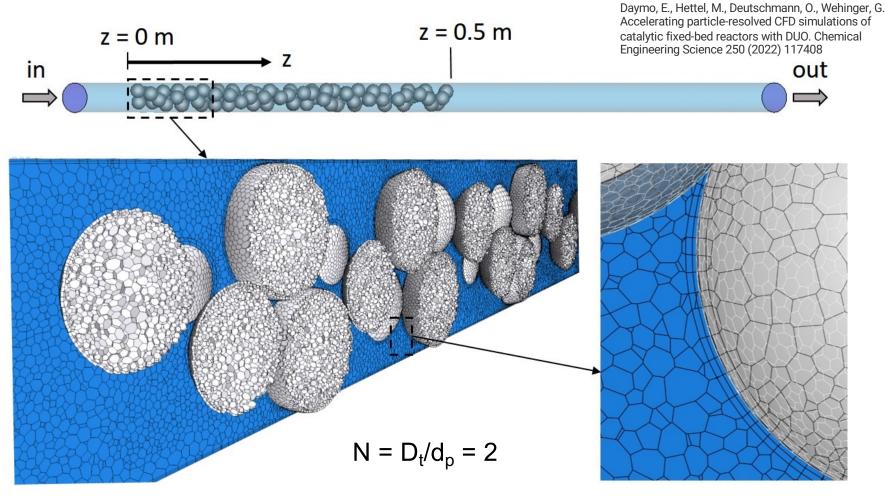
Inside a fixed bed reactor







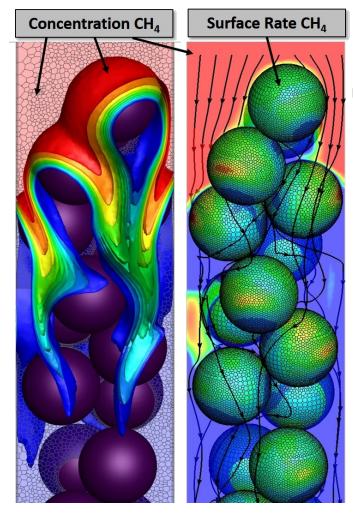
Particle Resolved CFD (PRCFD)



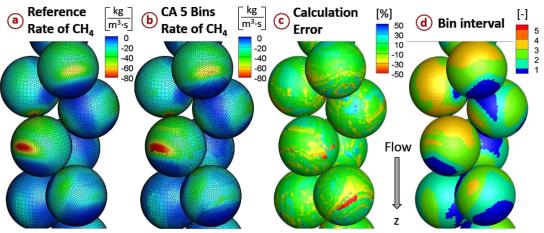


1,559,544 fluid cells, 774,386 cells solid cells 217,810 reactive surface faces

Particle resolved CFD

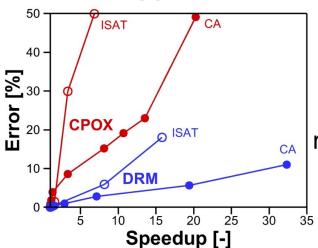


Daymo, E., Hettel, M., Deutschmann, O., Wehinger, G. Accelerating particle-resolved CFD simulations of catalytic fixed-bed reactors with DUO. Chemical Engineering Science 250 (2022) 117408



surface chemistry on individual particles simulated

cell agglomeration speeds-up chemistry



yet even with speedup, calculation of 86 particles required 100's of core hours

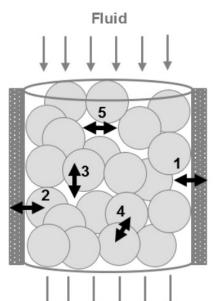
nfo@tonkomo.com <u>www.tonkomo.com</u>



Continuum Model Options: Fixed Bed Reactor Modeling

results presented here use a single energy equation

 Single energy equation for solid and fluid



- 1: convection wall fluid
- 2: radiation, conduction wall particle
- 3: convection, radiation particle fluid
- 4: radiation, conduction particle particle
- 5: convection fluid fluid

Figure 1. Heat transport mechanisms in fixed-bed reactors through which gas flows (modified according to [4]).

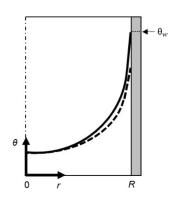
Stegehake, C., Riese, J., Grünewald. Modeling and validating fixed-bed reactors: A state-of-the-art review. ChemBioEng Rev 2019, 6, No. 2, 28–44

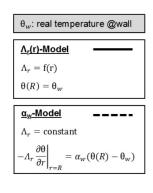
- Separate fluid and solid energy equations
 - Heat transfer resistance between solid and fluid phases
 - OpenFOAM has this capability, but with a fixed fluid-solid heat transfer coefficient



Continuum Model Options: Fixed Bed Reactor Modeling

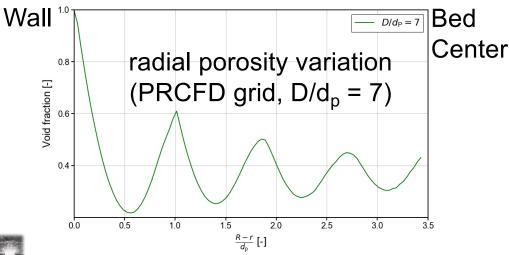
results presented here use the α_w model





Stegehake, C., Riese, J., Grünewald. M., Modeling and validating fixed-bed reactors: A state-of-the-art review. ChemBioEng Rev 2019, 6, No. 2, 28–44

- α_w model
 - BC T(r=R) \neq T_{wall}
 - $\varepsilon(\mathbf{r})$ = constant
 - $u(r) = u_{superficial}$



• Λ_R model

• BC
$$T(r=R) = T_{wall}$$

• ε(r) varies

•
$$u(r) = \frac{u_{superficial}}{\varepsilon(r)}$$

Effective thermal diffusivity

steady state energy balance:

$$\nabla \cdot \left(\rho \cdot u \cdot h_{s,t}\right) - \nabla \cdot \left(\alpha_{eff} \cdot \nabla h_{s}\right) = -\sum_{i}^{n} h_{i}^{o} \cdot R_{i} - \nabla \cdot \sum_{i}^{n} h_{i} \cdot \left(\overrightarrow{j_{i}} + \rho \cdot Y_{i} \cdot u_{c}\right) + \alpha_{eff} \cdot \nabla Y_{i}))$$

$$\uparrow$$

$$k_{eff} = k_{rf} + k_{rs} \qquad cp_{eff} = \frac{C\rho_{f}\rho_{f}\varepsilon + (1 - \varepsilon)C\rho_{s}\rho_{s}}{\rho_{f}\varepsilon + (1 - \varepsilon)\rho_{s}}$$

$$k_{rs} = k_f \left(\left(1 - \sqrt{1 - \epsilon} \right) + \sqrt{1 - \epsilon} \left\{ \frac{2}{1 - \frac{k_f}{k_s} B} \left[\frac{\left(1 - \frac{k_f}{k_s} \right) B}{\left(1 - \frac{k_f}{k_s} B \right)^2} ln \frac{k_s}{k_f B} - \frac{B + 1}{2} - \frac{B - 1}{1 - \frac{k_f}{k_s} B} \right] \right\} \right)$$
Solid

Zehner and Schlünder (1978)

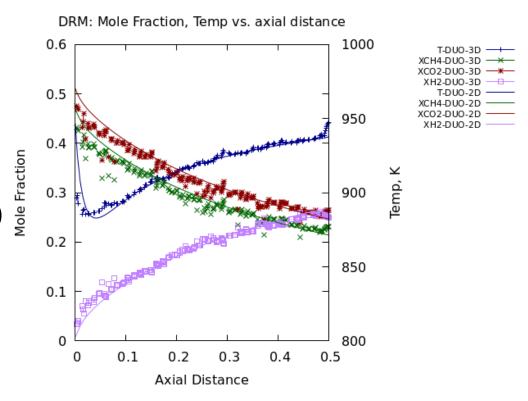
$$k_{rf} = k_f \left[\frac{RePr}{8.65 \left(1 + 19.4 \left(\frac{d_p}{d_t} \right)^2 \right)} \right]$$
 Fluid

Specchia et al. (1980)

Porous media: Dry Reforming of Methane CFD 2D vs 3D validation test

- Dry reforming of methane example
 - CO_2 + $CH_4 \Leftrightarrow 2CO + 2H_2$
- 3D DUO PRCFD vs. 2D by DUO porous media:

 Excellent fit
 (Wedge geometry, 1000 cells)
- Note the low N (D/d_p)
 value of 2...



100's - 1000's of core hours for 3D PRCFD, to anywhere from "a few" core hours to minutes for 2D porous media



Porous media: Comparison against data Steam Methane Reforming (SMR)

Hoang, D.L., Chan, S.H., Ding. O.L. Kinetic and modelling study of methane steam reforming over sulfide nickel catalyst on a gamma alumina support. Chemical Engineering Journal, 112 issues 1-3 (2005), 1-11.

1.
$$CH_4 + H_2O \Leftrightarrow CO + 3H_2$$

2.
$$CO + H_2O \Leftrightarrow CO_2 + H_2$$

$$3. \quad CH_4 + 2H_2O \iff CO_2 + 4H_2$$

$$Q_{SMR} = 1 + K_{CH_4} P_{CH_4} + K_{H_2O} P_{H_2O}$$

$$K_{H_2O} = K_{0,H_2O} \cdot e^{\left(\frac{\Delta H_{H_2O}}{RT}\right)}$$

$$R_1 = \frac{k_1}{P_{H_2}^{2.5}} \left(P_{CH_4} P_{H_2O} - \frac{P_{H_2}^3 P_{CO}}{K_{e1}} \right) \cdot \frac{1}{Q_{SMR}^2}$$

$$R_2 = \frac{k_2}{P_{H_2}} \left(P_{CO} P_{H_2O} - \frac{P_{H_2} P_{CO_2}}{K_{e2}} \right)$$

$$R_3 = \frac{k_3}{P_{H_2}^{3.5}} \left(P_{CH_4} P_{H_2O}^2 - \frac{P_{H_2}^4 P_{CO_2}}{K_{e3}} \right) \cdot \frac{1}{Q_{SMR}^2}$$

$$k_i = k_{0,i} \cdot e^{\left(\frac{-EA_i}{RT}\right)}$$

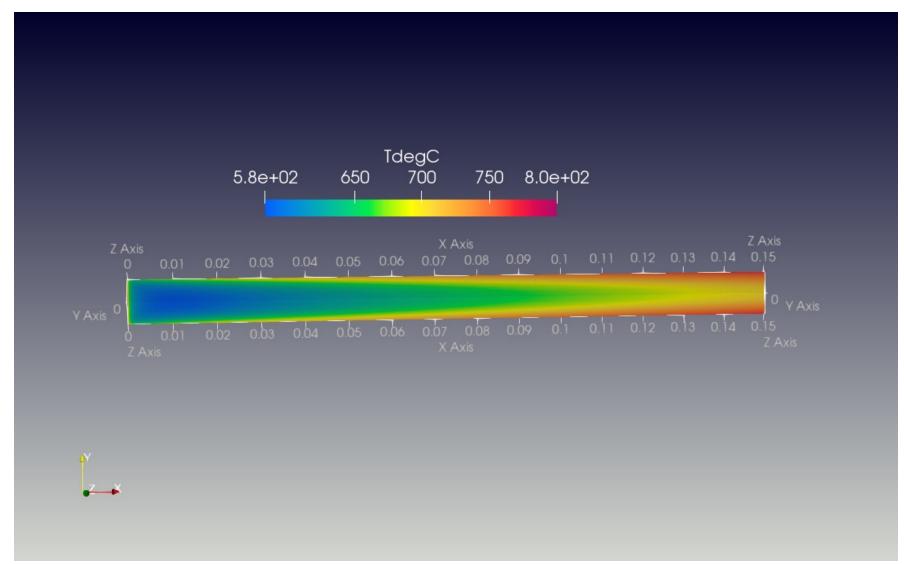
$$K_{e_1} = e^{\left(\frac{-26830}{T} + 30.114\right)}$$

$$K_{CH_4} = K_{0,CH_4} \cdot e^{\left(\frac{\Delta H_{CH_4}}{RT}\right)}$$

$$K_{e_2} = e^{\left(\frac{4400}{T} - 4.036\right)}$$

$$K_{e_3} = K_{e_1} K_{e_2}$$

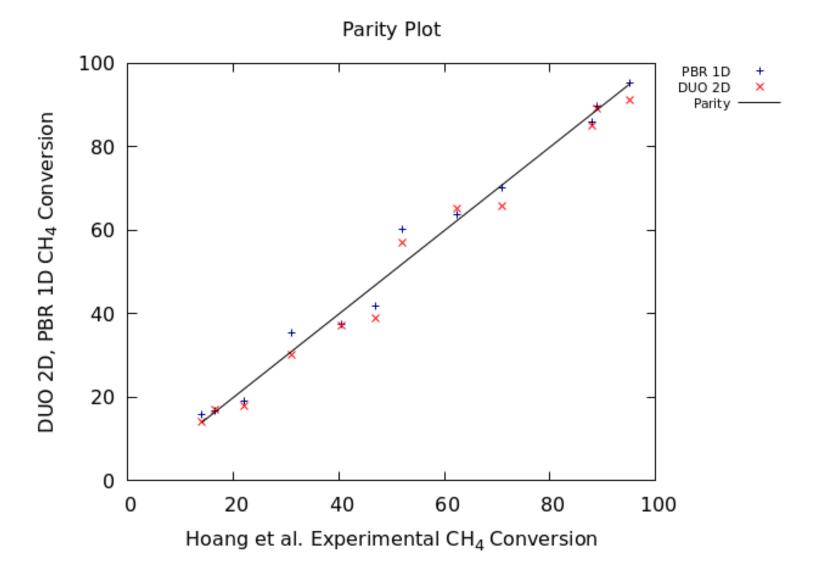
Radial temperature gradients are > 200°C





<u>www.tonkomo.com</u>

Parity plot: Porous media vs. Expt.





inlet

Further acceleration: 1D

1D grid

 OpenFOAM can simulate isothermal or adiabatic reactor

 What about nonisothermal, nonadiabatic? e.g.,

- Constant wall T
- Temperature inside reactor can vary

slip empty

slip





outlet

1D Packed Bed Reactor

Energy balance with overall heat transfer coefficient term

$$\nabla \cdot (\rho \cdot u \cdot h_{s,t}) - \nabla \cdot (\alpha_{eff} \cdot \nabla h_s)$$

$$= \frac{\partial p}{\partial t} - \sum_{i}^{n} h_i^o \cdot R_i - \nabla \cdot \sum_{i}^{n} h_i \cdot ((\overrightarrow{j_i} + \rho \cdot Y_i \cdot u_c) + \alpha_{eff} \cdot \nabla Y_i)) + U \frac{4}{d_t} (T - T_w)$$

EEqn.H

```
if (useOverallU) {
EEqn += - overallU*Tdiff*4/(dtube);}
```

 T_w = wall temperature d_t = tube diameter

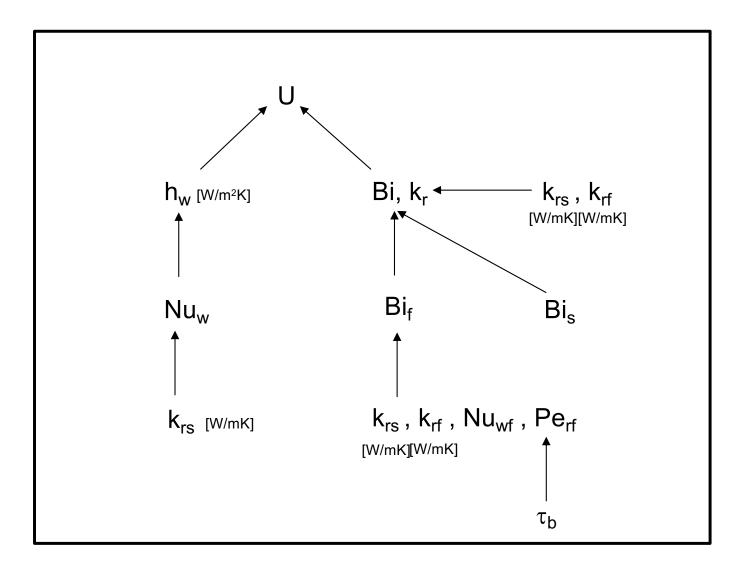
Many correlations are involved... including...

$$\frac{1}{U} = \frac{1}{h_w} + \frac{R_t}{3k_{eff}} \frac{Bi + 3}{Bi + 4}$$

$$Nu_w = \left(1.3 + \frac{5}{N}\right) \frac{k_{rs}}{k_f} + 0.19Pr^{1/3}Re^{3/4}$$

Martin & Nilles (1993)

Additional Equations







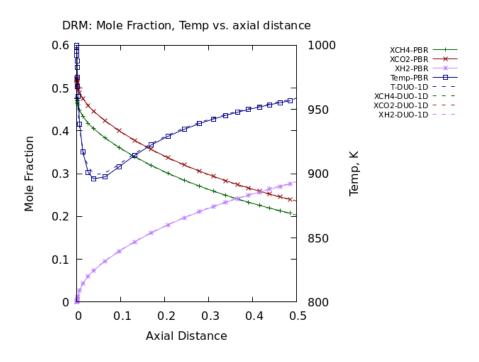
Validation test: 1D DUO vs 1D DETCHEMPBR Dry Reforming of Methane

Microkinetic surface model

 1D DUO vs 1D DETCHEMPBR

 Perfect match with DETCHEMPBR

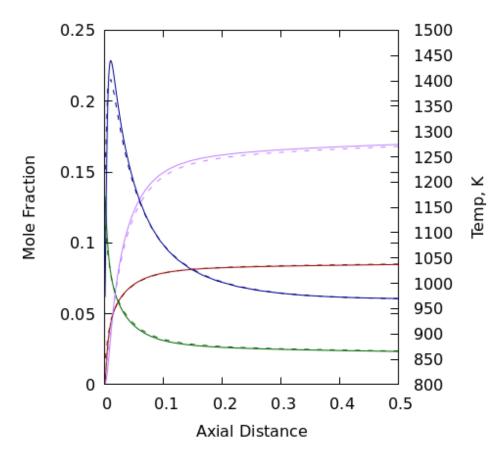
$$CH_4 + CO_2 = 2CO + 2H_2$$

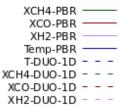


6 gas phase species 13 surface species 52 surface reactions (200 cells)

Validation test: 1D DUO vs 1D DETCHEMPBR Catalytic Partial Oxidation of Methane

$$CH_4 + 0.5 O_2 = CO + 2H_2$$





7 gas phase species11 surface species38 reactions (microkinetics)

Dramatic time reduction...

3D PRCFD 100's/1000's of core hours

2D Porous Media < 1 to "a few" core hours

1D Porous Media Core **seconds** to **minutes**





Example code (OpenFOAM 10)

https://github.com/TonkomoLLC/OFW18

```
forAll(cellsZone, cI)
    if (cellsZone[cI] == celli)
       // Effective radial solid thermal conductivity, Zhener and Schlunder
       scalar kfks = kappaCells[celli] / kappaSolid:
       scalar kskf = kappaSolid / kappaCells[celli];
       scalar Cterm = 1.25: // setup for spheres!
       scalar Bterminner = (1-porosity)/porosity;
       scalar Bterm = Cterm*pow(Bterminner, 10./9.);
       scalar Bterm2 = 1 - kfks*Bterm:
       scalar Bterm3 = (1 - kfks)*Bterm;
       scalar term0 = 1 - sqrt(1-porosity);
       scalar term1 = sort(1-porosity):
       scalar term2 = 2.0/Bterm2;
       scalar term3 = Bterm3 / pow(Bterm2,2.0);
       scalar term4 = log(kskf/Bterm);
       scalar term5 = (Bterm+1)/2;
       scalar term6 = (Bterm - 1) / Bterm2;
       scalar kappaEffSolid = kappaCells[celli]*(term0 + term1*(term2*(term3*term4-term5-term6)));
       // Effective radial fluid thermal conductivity, Specchia et al.
       scalar ReNum = thermoMixture.rho(pCells[celli], TCells[celli])*USuperficial*diamParticle/muCells[celli];
       scalar PrNum = CpCells[celli] * muCells[celli]/kappaCells[celli];
       scalar Re a = ReNum:
       scalar Nnum = diamParticle/diamTube;
       scalar B_Specchia = 1 + 19.4 * pow(Nnum, 2.0);
       scalar Pe H = 8.65*B Specchia:
       scalar kappaEffConvection = kappaCells[celli]*(Re_a*PrNum/Pe_H);
       kappaCells[celli] = kappaEffSolid + kappaEffConvection;
       // Effective heat capacity
       scalar rhoCelli = thermoMixture.rho(pCells[celli], TCells[celli]);
       scalar rhoAverageCelli = porosity*rhoCelli + (1 - porosity)*rhoSolid;
       scalar cPCelli = ((CpCells[celli] * porosity * rhoCelli)
                   + ((1 - porosity)*cPSolid*rhoSolid))/rhoAverageCelli;
       scalar cPFactor = CpCells[celli] / cPCelli;
       kappaCells[celli] *= cPFactor;
```

heRhoThermoPorous Suitable for 2D porous heat transfer

```
kappaEff.correctBoundaryConditions();
// Calculate wall heat transfer coefficient
// H. Martin and M. Nilles. Radiale Warmeleitung in durchstromten Schuttungsrohren. Chem. Ing. Tech. 65 (1993), 14681477.
volScalarField Nuw = (1.3 + 5./Nfactor)*(kappaEffSolid/thermo.kappa()) + 0.19*pow(PrNum, 0.33333)*pow(Rep, 0.75);
volScalarField hw = Nuw * thermo.kappa() / (diamParticle*UnitLength);
// Bi_s = solid Biot number
// A.G. Dixon. Wall and particle-shape effects on heat transfer in packed beds.
// Chemical Engineering Communications, 71:1 (1988), 217-237.
scalar Bi_s_power = (diamTube/diamParticle)-1.0;
scalar Bi_s = 2.41+(0.156*(std::pow(Bi_s_power,2.0)));
// Bi f = fluid Biot number
// A.G. Dixon. Wall and particle-shape effects on heat transfer in packed beds.
// Chemical Engineering Communications, 71:1 (1988), 217-237.
scalar Pe_rf_infinity = 12.0; // sphere
volScalarField Nu_wf = 0.523*(1.0-diamParticle/diamTube)*pow(PrNum,1./3.)*pow(Rep,0.738);
scalar tortuosity = 1.5-(0.5*porosity);
volScalarField Pe_rf = (1.0/(((porosity*tortuosity)/(PrNum*Rep))+(1./Pe_rf_infinity)));
volScalarField Bi_f = Nu_wf*diamTube/(2.0*diamParticle)*(Pe_rf/(Rep*PrNum));
// Overall tube Biot number
// D.L. Cresswell and A. G. Dixon, Theoretical prediction of effective heat transfer parameters in packed beds.
// AICHE J. 25 (1979), 663.
// Overall heat transfer coefficient
// A.G. Dixon. An improved equation for the overall heat transfer coefficient in packed beds.
// Chemical Engineering and Processing 35 (1996), 323-331.
forAll(rho, celli)
    forAll(cellsZone, cI)
        if (cellsZone[cI] == celli)
            if (Rep[celli] < 10.)
                //alphaEff[celli] = kappaEff[celli]/CpEff[celli]:
```

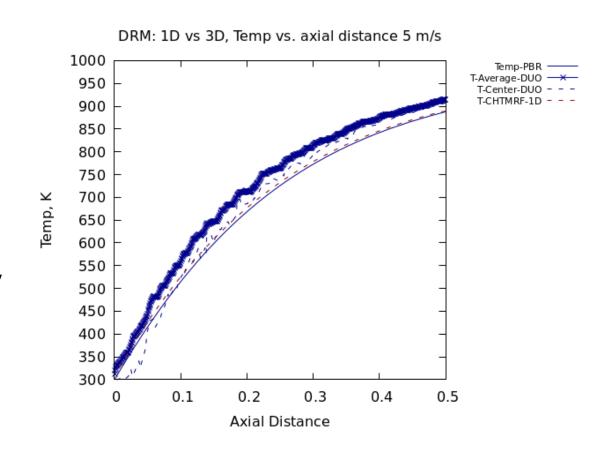
chtMultiRegionFoam-porous1D Includes overall U calculation

nfo@tonkomo.com <u>www.tonkomo.com</u>



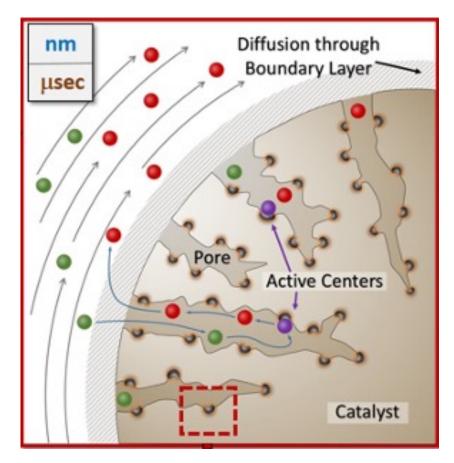
Example Case

- 1D DUO
- 1D DETCHEMPBR
- 3D DUO(86 particles)
- Heat Transfer Only (DRM feed)





Intra-particle diffusion



courtesy of omegadot software & consulting

- DETCHEM has a builtin "washcoat Models"
 - 1. Solve for effectiveness factor, η

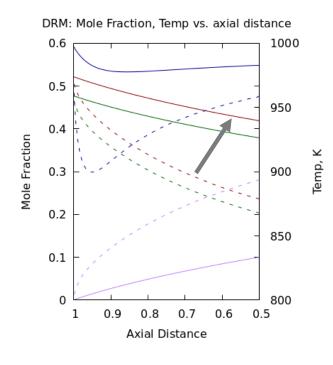
$$\eta = \frac{\tanh(\phi)}{\phi} \phi = L \sqrt{\frac{k}{D_{eff,i}}}$$

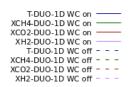
Solve reaction diffusion equation

$$-\nabla \cdot \left(\rho \cdot D_{i,m} \cdot \nabla Y_i\right) = R_i$$



Dry Reforming Pore Diffusion Effect





- thickness = $d_p/2$ = 6.325 mm
- internal porosity= 0.5
- tortuosity = 3
- pore diameter = 1 μm

Pore diffusion: here, substantially lowers conversion 38% conversion without pore diffusion 12% conversion with pore diffusion

Why use DUO for 1D PBR models?

OpenFOAM brings...

- OpenFOAM handles FVM calculus for axial conduction, diffusion effects
- Wide range of numerical options
- Supporting infrastructure

DETCHEM brings...

 Kinetic theory transport properties (via DETCHEM)

- Microkinetic surface reaction models
- Intraparticle diffusion models
- "UDF" format for complex reaction kinetics expressions



Summary

- Faster simulations: 3D → 2D → 1D
- 1D and 2D reactor simulations oftentimes adequately model tubular reactor performance
- Porous media simulations of packed bed reactors
 - Combined solid and fluid energy balance
 - Effective thermal conductivity combines solid and fluid effects
- 1D non-isothermal, non-adiabatic reactors can be simulated implementing an overall heat transfer coefficient





Thank you for your attention!



Eric Daymo, PE Tonkomo, LLC Gilbert, Arizona, United States

Email: eadaymo@tonkomo.com

Twitter: @TonkomoLLC



Lee Tonkovich



Matthias Hettel



Akash Shirsath

Accelerate solutions to real problems with accessible reactive CFD, how can we help?

