



Office Hours!

Instructor:

Trevor Klar, trevorklar@math.ucsb.edu

Office Hours:

Mondays 2–3PM

Tuesdays 10:30–11:30AM

Thursdays 1–2PM

or by appointment

Office:

South Hall 6431X (Grad Tower, 6th floor, blue side, first door on the right)

© 2017 Daryl Cooper, Trevor Klar

§7.9: Population Growth

Assume each generation of bunnies has 3 times as many bunnies as previous one. Initially have 100 bunnies. How many bunnies after n generations?

$$\begin{array}{lll} A = 100 \times 3n & B = 100 + 3n & C = 100(1 + 3n) \\ D = 100^{3n} & E = 100 \times 3^n & \end{array}$$

§7.9: Population Growth

Assume each generation of bunnies has 3 times as many bunnies as previous one. Initially have 100 bunnies. How many bunnies after n generations?

$$\begin{array}{lll} A = 100 \times 3n & B = 100 + 3n & C = 100(1 + 3n) \\ D = 100^{3n} & E = 100 \times 3^n & \end{array}$$

Answer: E

§7.9: Population Growth

Assume each generation of bunnies has 3 times as many bunnies as previous one. Initially have 100 bunnies. How many bunnies after n generations?

$$\begin{array}{lll} A = 100 \times 3n & B = 100 + 3n & C = 100(1 + 3n) \\ D = 100^{3n} & E = 100 \times 3^n & \end{array}$$

Answer: E

Start with 100

After 1 generation have 100×3 bunnies

§7.9: Population Growth

Assume each generation of bunnies has 3 times as many bunnies as previous one. Initially have 100 bunnies. How many bunnies after n generations?

$$\begin{array}{lll} A = 100 \times 3n & B = 100 + 3n & C = 100(1 + 3n) \\ D = 100^{3n} & E = 100 \times 3^n & \end{array}$$

Answer: E

Start with 100

After 1 generation have 100×3 bunnies

After 2 generations have $100 \times 3 \times 3$ bunnies

§7.9: Population Growth

Assume each generation of bunnies has 3 times as many bunnies as previous one. Initially have 100 bunnies. How many bunnies after n generations?

$$\begin{array}{lll} A = 100 \times 3n & B = 100 + 3n & C = 100(1 + 3n) \\ D = 100^{3n} & E = 100 \times 3^n & \end{array}$$

Answer: E

Start with 100

After 1 generation have 100×3 bunnies

After 2 generations have $100 \times 3 \times 3$ bunnies

After 3 generations have $100 \times 3 \times 3 \times 3$ bunnies

§7.9: Population Growth

Assume each generation of bunnies has 3 times as many bunnies as previous one. Initially have 100 bunnies. How many bunnies after n generations?

$$\begin{array}{lll} A = 100 \times 3n & B = 100 + 3n & C = 100(1 + 3n) \\ D = 100^{3n} & E = 100 \times 3^n & \end{array}$$

Answer: E

Start with 100

After 1 generation have 100×3 bunnies

After 2 generations have $100 \times 3 \times 3$ bunnies

After 3 generations have $100 \times 3 \times 3 \times 3$ bunnies

So...after n generations have

$$100 \times \underbrace{3 \times 3 \times \cdots \times 3}_{n \text{ times}} = 100 \times 3^n \text{ bunnies.}$$

More Bunnies

We saw that:

- if we start with 100 bunnies, and
- the bunny population triples every generation,

then we have 100×3^n bunnies after n generations.

1. How many generations until there are $10^7 = 10,000,000$ bunnies?

$$\begin{array}{lll} A = \log(5/3) & B = 5 - \log(3) & C = 5 / \log(3) \\ D = 5/3 & E = 10^5/3 & \end{array}$$

More Bunnies

We saw that:

- if we start with 100 bunnies, and
- the bunny population triples every generation,

then we have 100×3^n bunnies after n generations.

1. How many generations until there are $10^7 = 10,000,000$ bunnies?

$$\begin{array}{lll} A = \log(5/3) & B = 5 - \log(3) & C = 5 / \log(3) \\ D = 5/3 & E = 10^5/3 & \end{array}$$

$$\begin{array}{lll} A \approx 0.22 & B \approx 4.52 & C \approx 10.48 \\ D \approx 1.67 & E \approx 3,333 & \end{array}$$

More Bunnies

We saw that:

- if we start with 100 bunnies, and
- the bunny population triples every generation,

then we have 100×3^n bunnies after n generations.

1. How many generations until there are $10^7 = 10,000,000$ bunnies?

$$\begin{array}{lll} A = \log(5/3) & B = 5 - \log(3) & C = 5 / \log(3) \\ D = 5/3 & E = 10^5/3 & \end{array}$$

$$\begin{array}{lll} A \approx 0.22 & B \approx 4.52 & C \approx 10.48 \\ D \approx 1.67 & E \approx 3,333 & \boxed{C} \end{array}$$

Flu Outbreak

- 2.** At the start of an outbreak of H1N1 flu in a large herd of cattle, there were 5 infected individuals. The numbers doubles every 3 days. How many days until there are 80 infected cows?

$$A = \log(16)/\log(2) \quad B = \log(16/2) \quad C = 16/\log(2)$$

$$D = 3 \log(16)/\log(2) \quad E = \log(48/2)$$

Flu Outbreak

- 2.** At the start of an outbreak of H1N1 flu in a large herd of cattle, there were 5 infected individuals. The numbers doubles every 3 days. How many days until there are 80 infected cows?

$$A = \log(16)/\log(2) \quad B = \log(16/2) \quad C = 16/\log(2)$$

$$D = 3 \log(16)/\log(2) \quad E = \log(48/2)$$

- 2.** At the start of an outbreak of H1N1 flu in a large class of students, there were 5 infected individuals. The numbers doubles every 3 days. How many days until there are 80 infected students?

$$A = \log(16)/\log(2) \quad B = \log(16/2) \quad C = 16/\log(2)$$

$$D = 3 \log(16)/\log(2) \quad E = \log(48/2)$$

Flu Outbreak

- 2.** At the start of an outbreak of H1N1 flu in a large herd of cattle, there were 5 infected individuals. The numbers doubles every 3 days. How many days until there are 80 infected cows?

$$A = \log(16)/\log(2) \quad B = \log(16/2) \quad C = 16/\log(2)$$

$$D = 3 \log(16)/\log(2) \quad E = \log(48/2)$$

- 2.** At the start of an outbreak of H1N1 flu in a large class of students, there were 5 infected individuals. The numbers doubles every 3 days. How many days until there are 80 infected students?

$$A = \log(16)/\log(2) \quad B = \log(16/2) \quad C = 16/\log(2)$$

$$D = 3 \log(16)/\log(2) \quad E = \log(48/2) \quad \boxed{D}$$

Doubling Time Formula

Suppose something doubles every K minutes*. If there is a mass of A at time $t = 0$, how much is there at time t minutes?

*Any time unit will work, not just minutes. Just be consistent!

Doubling Time Formula

Suppose something doubles every K minutes*. If there is a mass of A at time $t = 0$, how much is there at time t minutes?

$$\text{mass after } t \text{ minutes} = A \times 2^{(t/K)}$$

Idea: t/K is number of doubling periods in t minutes.

*Any time unit will work, not just minutes. Just be consistent!

Doubling Time Formula

Suppose something doubles every K minutes*. If there is a mass of A at time $t = 0$, how much is there at time t minutes?

$$\text{mass after } t \text{ minutes} = A \times 2^{(t/K)}$$

Idea: t/K is number of doubling periods in t minutes.

- 3.** A disease spreads through a community. On March 1 there were 100 infected people. The number of people doubles in a 3 days. How many infected people are there t days after March 1?

$$A = 2^t \quad B = 3 \times 2^{t/100} \quad C = 100 \times 2^t \quad D = 100 \times 2^{t/3}$$

*Any time unit will work, not just minutes. Just be consistent!

Doubling Time Formula

Suppose something doubles every K minutes*. If there is a mass of A at time $t = 0$, how much is there at time t minutes?

$$\text{mass after } t \text{ minutes} = A \times 2^{(t/K)}$$

Idea: t/K is number of doubling periods in t minutes.

- 3.** A disease spreads through a community. On March 1 there were 100 infected people. The number of people doubles in a 3 days. How many infected people are there t days after March 1?

$$A = 2^t \quad B = 3 \times 2^{t/100} \quad C = 100 \times 2^t \quad D = 100 \times 2^{t/3} \quad \boxed{D}$$

How many days until there are 1,000 infected people?

$$A = \log(10)/\log(2) \quad B = 3 \log(10)/\log(2) \quad C = 3 \log(5) \\ D = 3(\log(10) - \log(2)) \quad E = 3 \log(20)$$

*Any time unit will work, not just minutes. Just be consistent!

Doubling Time Formula

Suppose something doubles every K minutes*. If there is a mass of A at time $t = 0$, how much is there at time t minutes?

$$\text{mass after } t \text{ minutes} = A \times 2^{(t/K)}$$

Idea: t/K is number of doubling periods in t minutes.

- 3.** A disease spreads through a community. On March 1 there were 100 infected people. The number of people doubles in a 3 days. How many infected people are there t days after March 1?

$$A = 2^t \quad B = 3 \times 2^{t/100} \quad C = 100 \times 2^t \quad D = 100 \times 2^{t/3} \quad \boxed{D}$$

How many days until there are 1,000 infected people?

$$A = \log(10)/\log(2) \quad B = 3 \log(10)/\log(2) \quad C = 3 \log(5) \\ D = 3(\log(10) - \log(2)) \quad E = 3 \log(20) \quad \boxed{B}$$

*Any time unit will work, not just minutes. Just be consistent!

A More Complicated Example

$$\text{mass after } t \text{ minutes} = A \times 2^{(t/K)}$$

where

- K is the doubling time, and
- t/K is the number of doubling periods in t minutes.

4. A colony of mold is growing on a cheeseburger in the back of a dorm refrigerator. When discovered it has a mass of 10 mg. One week later it was found to have a mass of 30 mg. What is the doubling time measured in days?

$$\begin{aligned} A &= \log(2)/\log(3) & B &= 7 \log(2)/\log(3) & C &= 7 \log(2/3) \\ D &= 7 \log(3/2) \end{aligned}$$

Hint: We know A and the mass t days after discovery (for some t).

A More Complicated Example

$$\text{mass after } t \text{ minutes} = A \times 2^{(t/K)}$$

where

- K is the doubling time, and
- t/K is the number of doubling periods in t minutes.

4. A colony of mold is growing on a cheeseburger in the back of a dorm refrigerator. When discovered it has a mass of 10 mg. One week later it was found to have a mass of 30 mg. What is the doubling time measured in days?

$$A = \log(2)/\log(3) \quad B = 7 \log(2)/\log(3) \quad C = 7 \log(2/3) \\ D = 7 \log(3/2)$$

Hint: We know A and the mass t days after discovery (for some t).

Solving $30 = 10 \times 2^{7/K}$ gives \boxed{B}

§7.11: Half-Life, Doubling Time

The half-life of a radioactive isotope is the time it takes for **half** of the isotope to decay.

Example: Isotope W has a **half-life** of **10** years. How much remains after **20** years?

§7.11: Half-Life, Doubling Time

The half-life of a radioactive isotope is the time it takes for **half** of the isotope to decay.

Example: Isotope W has a **half-life** of **10** years. How much remains after **20** years? **None?**

§7.11: Half-Life, Doubling Time

The half-life of a radioactive isotope is the time it takes for **half** of the isotope to decay.

Example: Isotope W has a **half-life** of **10** years. How much remains after **20** years? **None?**

$$\frac{1}{2} \times \frac{1}{2} \times (\text{amount you start with})$$

§7.11: Half-Life, Doubling Time

The half-life of a radioactive isotope is the time it takes for **half** of the isotope to decay.

Example: Isotope W has a **half-life** of **10** years. How much remains after **20** years? **None?**

$$\frac{1}{2} \times \frac{1}{2} \times (\text{amount you start with})$$

Idea: In half-life problems, convert time into **half-lives**.

§7.11: Half-Life, Doubling Time

The half-life of a radioactive isotope is the time it takes for **half** of the isotope to decay.

Example: Isotope W has a **half-life** of **10** years. How much remains after **20** years? **None?**

$$\frac{1}{2} \times \frac{1}{2} \times (\text{amount you start with})$$

Idea: In half-life problems, convert time into **half-lives**.

In this problem, the half-life is **10 years**. Therefore, **20 years** is **two half-lives**.

§7.11: Half-Life, Doubling Time

The half-life of a radioactive isotope is the time it takes for **half** of the isotope to decay.

Example: Isotope W has a **half-life** of **10** years. How much remains after **20** years? **None?**

$$\frac{1}{2} \times \frac{1}{2} \times (\text{amount you start with})$$

Idea: In half-life problems, convert time into **half-lives**.

In this problem, the half-life is **10 years**. Therefore, **20 years** is **two half-lives**.

In general: After **n** half-lives,

$$\text{remaining amount} = \left(\frac{1}{2}\right)^n \times (\text{amount started with})$$

§7.11: Half-Life, Doubling Time

The half-life of a radioactive isotope is the time it takes for **half** of the isotope to decay.

Example: Isotope W has a **half-life** of **10** years. How much remains after **20** years? **None?**

$$\frac{1}{2} \times \frac{1}{2} \times (\text{amount you start with})$$

Idea: In half-life problems, convert time into **half-lives**.

In this problem, the half-life is **10 years**. Therefore, **20 years** is **two half-lives**.

In general: After **n** half-lives,

$$\text{remaining amount} = \left(\frac{1}{2}\right)^n \times (\text{amount started with})$$

- 5.** Start with **120** grams of an isotope with a half-life of **12** years. How many grams remains after **36** years?

$$A = 0 \quad B = 10 \quad C = 15 \quad D = 20 \quad E = 40$$

§7.11: Half-Life, Doubling Time

The half-life of a radioactive isotope is the time it takes for half of the isotope to decay.

Example: Isotope W has a half-life of 10 years. How much remains after 20 years? **None?**

$$\frac{1}{2} \times \frac{1}{2} \times (\text{amount you start with})$$

Idea: In half-life problems, convert time into half-lives.

In this problem, the half-life is 10 years. Therefore, 20 years is two half-lives.

In general: After n half-lives,

$$\text{remaining amount} = \left(\frac{1}{2}\right)^n \times (\text{amount started with})$$

- 5.** Start with 120 grams of an isotope with a half-life of 12 years. How many grams remains after 36 years?

A= 0

B= 10

C= 15

D= 20

E= 40

C

Another Example

In general: After n half-lives,

$$\text{remaining amount} = \left(\frac{1}{2}\right)^n \times (\text{amount started with})$$

Another Example

In general: After n half-lives,

$$\text{remaining amount} = \left(\frac{1}{2}\right)^n \times (\text{amount started with})$$

6. An isotope has a half-life of 5 years.

(a) If we start with 70 grams, how many grams will be left after t years?

$$\begin{aligned} A &= 70 \left(\frac{1}{2}\right)^t & B &= 5 \left(\frac{1}{2}\right)^{70t} & C &= 70 \left(\frac{1}{2}\right)^{5t} \\ D &= 70 \left(\frac{1}{2}\right)^{t/5} & E &= 0 \end{aligned}$$

Another Example

In general: After n half-lives,

$$\text{remaining amount} = \left(\frac{1}{2}\right)^n \times (\text{amount started with})$$

6. An isotope has a half-life of 5 years.

(a) If we start with 70 grams, how many grams will be left after t years?

$$\begin{aligned} A &= 70 \left(\frac{1}{2}\right)^t & B &= 5 \left(\frac{1}{2}\right)^{70t} & C &= 70 \left(\frac{1}{2}\right)^{5t} \\ D &= 70 \left(\frac{1}{2}\right)^{t/5} & E &= 0 & \boxed{D} \end{aligned}$$

(b) How many years until 10 grams remain?

$$A = 5(\log(7) - \log(2)) \quad B = \log(7)/\log(2) \quad C = 5 \log(7/2)$$

$$D = 5 \log(7)/\log(2) \quad E = \log(7)/(5 \log(2))$$

Another Example

In general: After n half-lives,

$$\text{remaining amount} = \left(\frac{1}{2}\right)^n \times (\text{amount started with})$$

6. An isotope has a half-life of 5 years.

(a) If we start with 70 grams, how many grams will be left after t years?

$$\begin{aligned} A &= 70 \left(\frac{1}{2}\right)^t & B &= 5 \left(\frac{1}{2}\right)^{70t} & C &= 70 \left(\frac{1}{2}\right)^{5t} \\ D &= 70 \left(\frac{1}{2}\right)^{t/5} & E &= 0 & \boxed{D} \end{aligned}$$

(b) How many years until 10 grams remain?

$$A = 5(\log(7) - \log(2)) \quad B = \log(7)/\log(2) \quad C = 5 \log(7/2)$$

$$D = 5 \log(7)/\log(2) \quad E = \log(7)/(5 \log(2)) \quad \boxed{D}$$

Half-Life Formula

Suppose something has a half-life of K years[†]. If there is a mass of A at time $t = 0$, how much is there at time t years?

[†]Any time unit will work, not just years. Just be consistent!

Half-Life Formula

Suppose something has a half-life of K years[†]. If there is a mass of A at time $t = 0$, how much is there at time t years?

$$\text{mass after } t \text{ years} = A \times \left(\frac{1}{2}\right)^{(t/K)}$$

Idea: t/K is number of half-lives in t years.

[†]Any time unit will work, not just years. Just be consistent!

Half-Life Formula

Suppose something has a half-life of K years[†]. If there is a mass of A at time $t = 0$, how much is there at time t years?

$$\text{mass after } t \text{ years} = A \times \left(\frac{1}{2}\right)^{(t/K)}$$

Idea: t/K is number of half-lives in t years.

7. (Radiocarbon Dating) A bone is found with 2% of the usual amount of carbon-14 in it. The half-life of carbon-14 is 5730 years. How old (in years) is the bone?

$$A = 5730 \log(.01) / \log(2) \quad B = 5730 \log(50) / \log(2)$$

$$C = 5730 \times 50 \quad D = \text{wicked old}$$

[†]Any time unit will work, not just years. Just be consistent!

Half-Life Formula

Suppose something has a half-life of K years[†]. If there is a mass of A at time $t = 0$, how much is there at time t years?

$$\text{mass after } t \text{ years} = A \times \left(\frac{1}{2}\right)^{(t/K)}$$

Idea: t/K is number of half-lives in t years.

7. (Radiocarbon Dating) A bone is found with 2% of the usual amount of carbon-14 in it. The half-life of carbon-14 is 5730 years. How old (in years) is the bone?

$$A = 5730 \log(.01) / \log(2) \quad B = 5730 \log(50) / \log(2)$$

$$C = 5730 \times 50 \quad D = \text{wicked old}$$

Answer: $\boxed{B} \approx 32,000$ years

[†]Any time unit will work, not just years. Just be consistent!