# A variationally computed IR line list for the methyl radical CH<sub>3</sub>

Ahmad Y. Adam,<sup>†</sup> Andrey Yachmenev,<sup>‡</sup> Sergei N. Yurchenko,<sup>¶</sup> and Per Jensen\*,<sup>†</sup>

†Fakultät für Mathematik und Naturwissenschaften, Physikalische und Theoretische
Chemie, Bergische Universität Wuppertal, D-42097 Wuppertal, Germany

‡Center for Free-Electron Laser Science, Deutsches Elektronen-Synchrotron DESY,

Notkestraße 85, D-22607 Hamburg, Germany, The Hamburg Center for Ultrafast Imaging,

Universität Hamburg, Luruper Chaussee 149, D-22761 Hamburg, Germany

¶Department of Physics and Astronomy, University College London, Gower Street, London

WC1E 6BT, United Kingdom

E-mail: jensen@uni-wuppertal.de

2 Abstract

3

8

9

10

11

We present the first variational calculation of a room hot temperature ab initio line list for the CH<sub>3</sub> radical. It is based on a high level ab initio potential energy surface and dipole moment surface of CH<sub>3</sub> in the ground electronic state. The ro-vibrational energy levels and Einstein A coefficients were calculated using the general-molecule variational approach implemented in the computer program TROVE. Vibrational energies and vibrational intensities are found to be in very good agreement with the available experimental data. The line list comprises 9,127,123 ro-vibrational states ( $J \le 40$ ) and 2,058,655,166 transitions covering the wavenumber range up to 10000 cm<sup>-1</sup> and should be suitable for temperatures up to T = 1500 K.

#### 2 Introduction

13

carbon combustion processes? in atmospheric chemistry,? in the chemistry of semiconductor production,? in the chemical vapor deposition of diamond,? and in many chemical processes 15 of current industrial and environmental interest. It is also present in exo-planetary atmospheres,? in the atmospheres of Saturn? and Neptune,? and in the interstellar medium,? 17 where it is thought to be one of the most abundant free radicals.? Because of its central role in this variety of situations, its structural and spectroscopic parameters have been the subject of numerous studies. A number of different spectroscopic techniques have been used to determine its absolute concentration in the gas phase, including UV/visible,? infrared,? and Raman spectroscopies. ??????? In addition, CH<sub>3</sub> is an example of a molecule with large vibrational contribution to the hyperfine coupling constant, accounting for up to about 41% of the total value (see Ref. ? and references therein). Owing to the importance of CH<sub>3</sub> in various contexts, in particular in astrophysics and 25 -chemistry, its concentrations or column densities in remote environments such as interstellar space, the terrestrial atmosphere, exo-planetary atmospheres, and the outer layers 27 of cool stars are of interest and it is desirable to determine these by remote-sensing spectroscopic methods. A prerequisite for such determinations is the knowledge of the transition moments for the observed transitions, and these must often be obtained in theoretical calculations as done, for example, in the ExoMol project?? by Yurchenko and coworkers. ????????????????? This project aims at providing theoretically computed transition moments and simulated spectra for (small to medium-sized) general polyatomic molecules of astrophysical and/or -chemical interest. 34 At equilibrium, the three protons of electronic-ground-state  $\tilde{X}^{-2}A_2''$  CH<sub>3</sub> form an equi-35 lateral triangle with the C nucleus at the centre-of-mass of the planar structure with  $D_{3h}$ point group symmetry (see Table A-10 of Ref. ? ). There is no permanent dipole moment, and so the pure rotational transitions are dipole forbidden and very weak. Also, the

The methyl radical is one of the most important free radicals and plays a central role in hydro-

planar ground-state equilibrium structure precludes most one-photon transitions to excited electronic states. Owing to the extremely weak rotational spectrum, determinations of concentrations and column densities for CH<sub>3</sub> must be made with rovibrational transitions in the infrared region. The most suitable transitions are those in the intense  $\nu_2$  fundamental band at 606 cm<sup>-1</sup> (where  $\nu_2$  is the out-of-plane bending mode). This band provides convenient transitions for concentration measurements and has been used extensively for this purpose. As mentioned above, the corresponding transition moments must be known in order that concentrations can be determined, and the present work can be viewed as a first step towards providing extensive catalogues of theoretical transition moments for CH<sub>3</sub>, so-called line lists, of use in astrophysical studies. In the present work we apply a high level ab initio potential energy surface, refined by means of experimental spectroscopic data, and ab initio dipole moment surfaces to compute, with the TROVE program, ???? sufficient energies and transition moments for generating a room-temperature hot (T = 1500 K) IR line list for CH<sub>3</sub>.

# 53 Potential energy and dipole moment

# 54 Potential energy surface

The potential energy surface (PES) employed for the electronic ground state of CH<sub>3</sub> in the present work is based on the *ab initio* surface reported in Ref. ? , which we denote as PES-1. The PES-1 electronic energies were computed on a grid of 24 000 symmetry-unique molecular geometries employing the open-shell RCCSD(T)-F12b<sup>?</sup>? level of theory (explicitly correlated F12 restricted coupled cluster included single and double excitations with a noniterative correction for triples) and the F12-optimized correlation consistent polarized valence basis set cc-pVQZ-F12. In correlated calculations the carbon inner-shell electron pair was treated as frozen core. The analytical representation for the PES was obtained in a least-squares fitting procedure using the functional form from Lin *et al.*?:

$$V(\xi_{1}, \xi_{2}, \xi_{3}, \xi_{4a}, \xi_{4b}; \sin \bar{\rho}) = V_{e} + V_{0}(\sin \bar{\rho}) + \sum_{j} F_{j}(\sin \bar{\rho}) \, \xi_{j}$$

$$+ \sum_{j \leqslant k} F_{jk}(\sin \bar{\rho}) \, \xi_{j} \, \xi_{k} + \sum_{j \leqslant k \leqslant l} F_{jkl}(\sin \bar{\rho}) \, \xi_{j} \, \xi_{k} \, \xi_{l}$$

$$+ \sum_{j \leqslant k \leqslant l \leqslant m} F_{jklm}(\sin \bar{\rho}) \, \xi_{j} \, \xi_{k} \, \xi_{l} \, \xi_{m} + \dots;$$
(1)

this function is expressed in terms of the stretching variables

$$\xi_k = 1 - \exp\left[-a(r_k - r_e)\right], \quad k = 1, 2, 3,$$
 (2)

where  $r_k$  is the instantaneous value of the distance between the C nucleus and the proton  $H_k$  labeled k = 1, 2, or 3;  $r_e$  is the common equilibrium value of the three  $r_k$  bond lengths, and a is a Morse parameter. Furthermore, the symmetrized bending variables  $(\xi_{4a}, \xi_{4b})$  are defined as

$$(\xi_{4a}, \xi_{4b}) = \left(\frac{1}{\sqrt{6}} [2\alpha_1 - \alpha_2 - \alpha_3], \frac{1}{\sqrt{2}} [\alpha_2 - \alpha_3]\right)$$
(3)

with  $\alpha_i$  as the bond angle  $\angle(H_jXH_k)$  where (i, j, k) is a permutation of the numbers (1,2,3). Finally, the variable

$$\sin \bar{\rho} = \frac{2}{\sqrt{3}} \sin[(\alpha_1 + \alpha_2 + \alpha_3)/6] \tag{4}$$

describes the out-of-plane bending. At the planar equilibrium configuration, we have  $\alpha_1$  +

 $\alpha_2 + \alpha_3 = 360^{\circ}$  and so  $\sin \bar{\rho} = \sin \bar{\rho}_e = 1$ . The functions  $V_0(\sin \bar{\rho})$  and  $F_{jk...}(\sin \bar{\rho})$  in Eq. (1)

66 are defined as

$$V_0(\sin \bar{\rho}) = \sum_{s=1}^4 f_0^{(s)} (1 - \sin \bar{\rho})^s, \tag{5}$$

$$F_{jk...}(\sin \bar{\rho}) = \sum_{s=0}^{N} f_{jk...}^{(s)} (1 - \sin \bar{\rho})^{s},$$
 (6)

where the quantities  $f_0^{(s)}$  and  $f_{jk...}^{(s)}$  in Eqs. (5) and (6) are expansion coefficients. The optimized values of the parameters a,  $r_{\rm e}$ ,  $f_0^{(s)}$ , and  $f_{jk...}^{(s)}$  are given in the supplementary material to Ref. ? together with Fortran 90 routine for calculating PES values.

The analytical form of PES-1? is given in terms of the ab initio cc-pVQZ-F12 values of the equilibrium structural parameters,  $r_{\rm e}=1.0774$  Å and  $\alpha_{\rm e}=120.0^{\circ}$ ,? for the electronic ground state of CH<sub>3</sub>. In the present work, we optimized the values  $r_{\rm e}$  and  $\alpha_{\rm e}$  in a least-squares fitting to experimentally derived to rotational energy spacings within the vibrational states of CH<sub>3</sub>. The fitting produced the values of  $r_{\rm e}=1.0763$  Å and  $\alpha_{\rm e}=120.0$ . We adopt these optimized values of the equilibrium structural parameters. All results presented in the remainder of the present work are based on the analytical potential energy function called PES-2, obtained from PES-1? by replacing the ab initio cc-pVQZ-F12 values of  $r_{\rm e}$  and  $\alpha_{\rm e}$  by the adjusted values given here. The remaining PES-2 parameter values are identical to those of PES-1 and can be obtained from the supplementary material to Ref. ? .

#### Dipole moment surface

The dipole moment surface (DMS) for the electronic ground state of CH<sub>3</sub> was computed using the MOLPRO? program package. Frozen-core calculations were carried out for 19 361 symmetry-unique geometries (15 600 below 30 000 cm<sup>-1</sup>) using the spin-restricted open-shell coupled cluster theory RCCSD(T)? and the augmented correlation consistent valence basis set aug-cc-pVTZ,?? employing the two-point stencil central finite differences with the electric field strength of 0.002 a.u.

We employ the so-called symmetrized molecular-bond (SMB) representation? [which is an extension of the molecular-bond (MB) representation? ] to formulate analytical functions describing the molecular dipole moment components. The SMB representation makes use of the projections  $\bar{\mu} \cdot \mathbf{e}_k$  of the dipole moment on the molecular bonds, where  $\mathbf{e}_k$  is the unit vector along the C-H<sub>k</sub> bond,

$$\mathbf{e}_k = \frac{\mathbf{r}_k - \mathbf{r}_4}{|\mathbf{r}_k - \mathbf{r}_4|} \tag{7}$$

with  $\mathbf{r}_k$ , k=1, 2, 3, as the position vector of proton k and  $\mathbf{r}_4$  as the position vector of the C nucleus.

We form symmetry-adapted linear combinations of the MB projections  $\bar{\mu} \cdot \mathbf{e}_i$ :

$$\bar{\mu}_{A_2''}^{\text{SMB}} = (\bar{\boldsymbol{\mu}} \cdot \mathbf{e}_{\text{N}})$$
 (8)

$$\bar{\mu}_{E'_a}^{\text{SMB}} = \frac{1}{\sqrt{6}} \left[ 2 \left( \bar{\boldsymbol{\mu}} \cdot \mathbf{e}_1 \right) - \left( \bar{\boldsymbol{\mu}} \cdot \mathbf{e}_2 \right) - \left( \bar{\boldsymbol{\mu}} \cdot \mathbf{e}_3 \right) \right]$$
(9)

$$\bar{\mu}_{E_b'}^{\text{SMB}} = \frac{1}{\sqrt{2}} \left[ (\bar{\boldsymbol{\mu}} \cdot \mathbf{e}_2) - (\bar{\boldsymbol{\mu}} \cdot \mathbf{e}_3) \right],$$
 (10)

where, in addition to the vectors  $\mathbf{e}_k$ , we have introduced  $\mathbf{e}_N = \mathbf{q}_N/|\mathbf{q}_N|$  with  $\mathbf{q}_N$  as the 'trisector'

$$\mathbf{q}_{N} = (\mathbf{e}_{1} \times \mathbf{e}_{2}) + (\mathbf{e}_{2} \times \mathbf{e}_{3}) + (\mathbf{e}_{3} \times \mathbf{e}_{1}). \tag{11}$$

The subscripts  $\Gamma = A_2''$ ,  $E_a'$ , and  $E_b'$  of the quantities  $\bar{\mu}_{\Gamma}^{\text{SMB}}$  in Eqs. (8)–(10) refer to the irreducible representations (Table A-10 of Ref. ? ) of the CH<sub>3</sub> molecular symmetry group  $D_{3h}(M)$ ; the electronically averaged dipole moment  $\bar{\mu}$  generates the representation  $A_2'' \oplus E'$ . The quantity  $\bar{\mu}_{A_2''}^{\text{SMB}}$  is antisymmetric under the inversion operation?  $E^*$  and vanishes at planarity, so that  $\bar{\mu}$  has only two non-vanishing, linearly independent components at planarity. These two components vanish at planar configurations with  $D_{3h}$  point group symmetry.

The three components of the SMB dipole moment in Eqs. (8)–(10) are represented by  $4^{\text{th}}$  order polynomial expansions

$$\bar{\mu}_{\Gamma}^{\text{SMB}}(\chi_{1}, \chi_{2}, \chi_{3}, \chi_{4a}, \chi_{4b}; \rho) = \mu_{0}^{\Gamma}(\sin \bar{\rho}) + \sum_{i} \mu_{i}^{\Gamma}(\sin \bar{\rho})\chi_{i} + \sum_{i \leq j} \mu_{ij}^{\Gamma}(\sin \bar{\rho})\chi_{i}\chi_{j}$$

$$+ \sum_{i \leq j \leq k} \mu_{ijk}^{\Gamma}(\sin \bar{\rho})\chi_{i}\chi_{j}\chi_{k} + \sum_{i \leq j \leq k \leq l} \mu_{ijkl}^{\Gamma}(\sin \bar{\rho})\chi_{i}\chi_{j}\chi_{k}\chi_{l},$$

$$(12)$$

in terms of the variables

89

$$\chi_k = \Delta r_k (1 - \exp(-\Delta r_k))^2, \quad (k = 1, 2, 3)$$
 (13)

with  $\Delta r_k = r_k - r_e$  and

$$(\chi_4, \chi_5) = (\xi_{4a}, \xi_{4b}) \tag{14}$$

where  $(\xi_{4a}, \xi_{4b})$  are defined in Eq. (3). The expansion coefficients  $\mu_{ij...}^{\Gamma}(\sin \bar{\rho})$  are defined as

$$\mu_{ij\dots}^{\Gamma}(\sin\bar{\rho}) = \sum_{s=0}^{N} \mu_{ij\dots}^{\Gamma(s)} (1 - \sin\bar{\rho})^{s}, \tag{15}$$

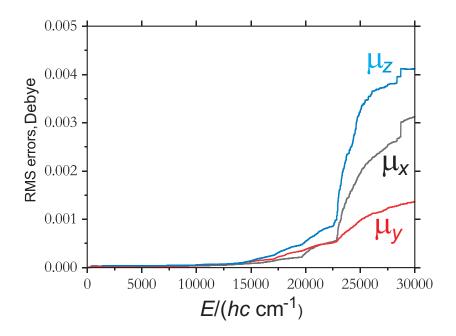
where  $\sin \bar{\rho}$  is given by Eq. (4) and the maximal order of the polynomial is N=8. For more details the reader is referred to Ref.? . The final fit of the 15600 geometries required a total number of 218 parameters (131 for  $\bar{\mu}_{A_2''}^{\rm SMB}$  and 87 for  $\bar{\mu}_{E_a'}^{\rm SMB}$  and  $\bar{\mu}_{E_b'}^{\rm SMB}$ ) and reproduced the ab initio data with a root-mean-square (RMS) differences of 0.003 D, 0.001 D and 0.004 D 100 for the x, y and z components respectively and energies up to  $30\,000~{\rm cm}^{-1}$ , see 1. A series 101 of fittings to the x, y and z ab initio dipole-moment components have been carried out, 102 including in the data set for each fitting the ab initio points with electronic energy (relative 103 to the potential energy minimum)  $V \leq E$ , and increasing E. 1 shows the RMS deviation 104 for each dipole-moment component as a function of E. The DMS expansion parameter set 105 and the Fortran 90 functions are included in the supplementary material. 106

# Intensity simulations with TROVE

### 108 General expressions

The 'readiness' of the molecule to make an absorption or emission transition from an initial ro-vibrational state i to a final ro-vibrational state f is expressed by the line strength???  $S(f \leftarrow i)$ , a quantity with units of [dipole moment]² (typically Debye²). For  $S(f \leftarrow i) = 0$ , the transition does not take place and need not be considered. As discussed in Ref.? transitions with  $S(f \leftarrow i) \neq 0$  are said to satisfy selection rules which we can derive from symmetry considerations before we do quantitative, numerical calculations of  $S(f \leftarrow i)$ . Thus, these calculations need only be done for transitions satisfying the selection rules, and

Figure 1: Root-mean-square (RMS) errors of the fittings to the *ab initio* dipole moment values. The results of a series of fittings are shown. In each fitting, the data set includes the *ab initio* points with electronic energy (relative to the potential energy minimum)  $V \leq E$  (see text).



after obtaining values of  $S(f \leftarrow i)$  for these transitions, we can compute the corresponding Einstein coefficients and absorption intensities.

The initial(final) state i(f) has the rotation-vibration wavefunction  $|\Phi_{\rm rv}^{(i)}\rangle(|\Phi_{\rm rv}^{(f)}\rangle)$ . The line strength???  $S(f \leftarrow i)$  of the ro-vibrational transition  $f \leftarrow i$  is

$$S(f \leftarrow i) = g_{\text{ns}} \sum_{M_f, M_i} \sum_{A = X, Y, Z} \left| \left\langle \Phi_{\text{rv}}^{(f)} | \bar{\mu}_A | \Phi_{\text{rv}}^{(i)} \right\rangle \right|^2, \tag{16}$$

where  $g_{ns}$  is the nuclear spin statistical weight factor? and  $\bar{\mu}_A$  is the electronically averaged component of the molecular dipole moment along the space-fixed axis? A = X, Y, or Z. The quantum numbers  $M_i$  and  $M_f$  are the projections of the total angular momentum  $\hat{\mathbf{J}}$  on the Z axis for the initial and final states, respectively. Assuming that the molecules considered are in thermal equilibrium at the absolute temperature T, the intensity of a spectral line is determined as

Here  $\tilde{\nu}$  is the absorption wavenumber, and Eq. (17) refers to the transition from the state

122

$$I(f \leftarrow i) = \frac{8\pi^3 N_A \tilde{\nu}_{if}}{(4\pi\epsilon_0)3hc} \frac{e^{-E_i/kT}}{Q} \left[ 1 - \exp(-hc\tilde{\nu}_{if}/kT) \right] S(f \leftarrow i). \tag{17}$$

i with energy  $E_i$  to the state f with energy  $E_f$ , where  $hc\tilde{\nu}_{if} = E_f - E_i$ . Q is the partition 123 function defined as  $Q = \sum_{j} g_{j} \exp(-E_{j}/kT)$ , where  $g_{j}$  is the total degeneracy of the state 124 with energy  $E_j$  and the sum runs over all energy levels of the molecule, and other symbols 125 have their usual meanings. The total degeneracy  $g_j$  is given by (2J+1) times the spin 126 degeneracy (M=2) and times the nuclear spin degeneracy which is 4, 0, 2, 4, 0, 2 for 127  $A'_1, A'_2, E', A''_1, A''_2$ , and E'' symmetries respectively. The ground electronic state of CH<sub>3</sub> is 128 a doublet  $(\tilde{X}^2A_2'')$  with a small splitting?? in the rovibrational energy levels due to spin-129 rotation interactions, around 0.01 cm<sup>-1</sup>, which we therefore chose to ignore for this work— 130 by those exactly by the spin-multiplicity factor of 2. 131 A detailed expression for the line strength of an individual ro-vibrational transition 132 within an isolated electronic state of an XY<sub>3</sub> pyramidal molecule is given in Eq. (21) of 133 Yurchenko et al.? Provided that the population of the lower (initial) state is defined by 134 the Boltzmann distribution, it is sufficient to we consider only transitions starting from the levels below  $E_i^{\text{max}}/hc = \frac{39}{200000} \, \text{cm}^{-1}$ , which corresponds to Boltzmann factors of  $\exp(-E_i/kT) > 2 \times 10^{-7} \exp(-E_i/kT) > 2 \times 10^{-4}$  entering into Eq. (17) for T = 300T = 1500 K. 137 It is common to use the partition function for estimating the completeness of the line list 138 for a given temperature ?. By considering the ratio  $Q_{9000\,\mathrm{cm}^{-1}}/Q_{\mathrm{total}}$ , where  $Q_{\mathrm{total}}$  is the 139 converged partition function value calculated by explicitly summing over all energy levels 140 and  $Q_{9000 \text{ cm}^{-1}}$  is the partition function calculated using only those levels with energies up to 141 the threshold of  $9000 \text{ cm}^{-1}$ . This ratio gives 95 % completeness at temperatures below 1500142 K, which we estimate as the maximal temperature for our line list. For similar reasons, the

range of the rotational excitations can be safely limited at J = 26J = 40. The wavenumber range selected is 0 to  $810\,000$  cm<sup>-1</sup>; the upper energy limit (i.e., the maximal energy for the final state) corresponds to a term value of  $E^{\text{max}}/hc = 12\,19\,000$  cm<sup>-1</sup>.

#### 147 Computational details

We use a symmetry-adapted basis set in the variational nuclear motion calculations. The
Hamiltonian matrix is factorized into size-independent blocks according to the irreducible
representations of the  $D_{3h}(M)$  molecular symmetry group:  $A'_1$ ,  $A'_2$ ,  $A''_1$ ,  $A''_2$ , E', and E''.

The  $A'_2$  and  $A''_2$  matrices are of no interest for CH<sub>3</sub> as the corresponding states have zero
nuclear spin statistical weights and do not exist in Nature. The E' and E'' matrices each
split into two sub-blocks, of which only one must be diagonalized.

The calculation of the matrix elements  $\langle \Phi_{\rm rv}^{(f)} | \bar{\mu}_A | \Phi_{\rm rv}^{(i)} \rangle$  in Eq. (16) is the bottle-neck in the spectrum simulations. Here, the wavefunctions  $\Phi_{\rm rv}^{(w)}$  are expressed as symmetry-adapted linear combinations of basis functions (see Eq. (65) of Yurchenko *et al.*?):

$$|\Phi_{\text{rv}}^{(w)}\rangle = \sum_{VK\tau_{\text{rot}}} C_{VK\tau_{\text{rot}}}^{(w)} |J_w K m_w \tau_{\text{rot}}\rangle |V\rangle, \quad w = i \text{ or } f,$$
 (18)

where the  $C_{VK\tau_{\text{rot}}}^{(w)}$  are expansion coefficients,  $|J_w K m_w \tau_{\text{rot}}\rangle$  is a symmetrized rotational basis function,  $\tau_{\text{rot}}$  (= 0 or 1) is associated with the rotational parity, and  $|V\rangle$  is a vibrational basis function. In order to speed up this part of the calculations, we applied a pre-screening procedure to the expansion coefficients  $C_{VK\tau_{\text{rot}}}^{(f)}$ . All terms with coefficients less than the threshold value of  $\frac{10^{-22}}{10^{-13}}$  were excluded from the integration.

A speedup in the evaluation of the dipole moment matrix elements  $\langle \Phi_{\rm rv}^{(f)} | \bar{\mu}_A | \Phi_{\rm rv}^{(i)} \rangle$  can be done in two steps. First, for a given lower state i the following effective line strength is evaluated:

$$S_{i,VK}^A = \langle \Phi_{\rm rv}^{(i)} | \bar{\mu}_A | \phi_{VK} \rangle, \tag{19}$$

where we introduced a short-hand notation  $\phi_{VK}$  for the primitive basis function  $|J_wK m_w \tau_{\rm rot}\rangle$ 

 $\times |V\rangle$ . Once all  $S_{i,VK}$  are computed, in the second step the line strength  $S(f \leftarrow i)$  is evaluated as

$$S(f \leftarrow i) = g_{\rm ns} \sum_{m_i, m_f} \sum_{A = X, Y, Z} \left| \sum_{V, K} C_{VK\tau_{\rm rot}}^{(f)} S_{i, VK}^A \right|^2.$$
 (20)

We had to compute a very large number of transitions satisfying the selection rule  $|J_f - J_i|$ 159  $\leq 1$ , where  $J_i$  and  $J_f$  are the values of the angular momentum quantum number J for 160 the initial and final state, respectively. Consequently, we saved memory by organizing the 161 calculation of the ro-vibrational eigenstates and the  $S(f \leftarrow i)$ -values such that at a given 162 time, only eigenvectors for states with two consecutive J-values, J and J+1, are available for 163 the computation of  $S(f \leftarrow i)$ -values. This algorithm is implemented in the GPU GAIN-MPI program?. 165 The vibrational basis set  $|V\rangle$  is obtained in TROVE using a multi-step contraction and 166 symmetrization procedure, starting from local primitive basis set functions, each depending 167 on one variable only (see Refs. ??? and references therein). Thus, a compact representa-168 tion of the vibrational basis set is obtained in a form optimized for the molecule of interest. 169 The final vibrational basis set is represented by the eigenfunctions of the purely vibrational 170 part of the Hamiltonian; we call these eigenfunctions the 'J=0 basis'.

#### Results

171

#### Basis set convergence and empirical adjustment of the vibrational 173 band centers

The dimensions of the Hamiltonian matrix blocks to be diagonalized are important in determining the accuracy with which highly excited ro-vibrational energies and wavefunctions can 176 be computed, and consequently it is important to derive by empirical methods the smallest basis set that is consistent with the required eigenvalue accuracy (that is to say, the optimum size for 'convergence').

TROVE employs polyad number truncation??? to control the size of the vibrational basis set, with the polyad number P given by:

$$P = 2(n_1 + n_2 + n_3) + n_4 + n_5 + n_6, (21)$$

where  $n_i$  are the quantum numbers connected with the primitive functions  $\phi_{n_i}(\xi_i)$ . That is, we include in the primitive basis set only those functions  $\phi_{n_i}(\xi_i)$  for which  $P \leq P_{\text{max}}$ .

182

The vibrational term values could be converged even more tightly by extrapolating the

energies calculated with different-size basis sets  $P_{\text{max}}$  to the complete vibrational basis set 183 limit. However, this is not considered necessary for the purpose of generating a line list 184 since the corrections from such an extrapolation will be small compared with the inherent 185 errors in the term values that are caused by the imperfection of the underlying potential 186 energy surface. Instead, if we aim for higher accuracy in a pragmatic manner, we can resort 187 to a more empirical approach where the theoretical vibrational term values are replaced by 188 accurate experimental vibrational band centres, whenever these are available in the literature. 189 In this case, we adjust the vibrational band centers 'by hand', and by doing so, we shift the 190 rotational energy structure towards better agreement with experiment. This procedure can 191 be regarded as an empirical basis set correction scheme and will be denoted as EBSC scheme. 192 We adopt the EBSC scheme for the vibrational bands  $2\nu_2$ ,  $\nu_1$ ,  $\nu_4^1$ , and  $\nu_3^1$ , for which 193 accurate experimental data is available, in combination with PES-2, where we have adjusted 194 the equilibrium structure of the molecule to fit the experimentally derived pure rotational 195 term values. The vibrational basis set was truncated at the polyad number  $P_{\rm max}=32$ . The 196 incorporation of experimental information in the EBSC scheme is obviously a departure from 197 a purely ab initio approach which is considered to be justified by the gain in accuracy that 198 can be achieved when computing an extensive ro-vibrational line list. 199

To achieve higher accuracy for the vibrational band centers, another more thorough refinement of our PES would be needed. However, considering a severe lack of the accurate experimental data on the vibrationally excited states of  $CH_3$ , we decided for a pragmatic, alternative procedure based on the EBSC substitution scheme in conjunction with the  $r_{\rm e}$ refinement. Even though we cannot avoid imperfections for all bands which where not EBSC-corrected, a more accurate rotational structure should be useful for assignments of future experimental works on  $CH_3$ .

1 lists the vibrational band centers of methyl radical up to 5000 cm<sup>-1</sup>, as derived from experimental data and from variational calculations of the present work computed using PES 2 in conjunction 2. These experimental band centers were used to replace the theoretical values  $P_{\text{max}} = 32$  values in all J > 0 TROVE calculations according with the EBSC substitution. This table also shows the effect of the polyad-number on the vibrational energy.

2 shows a comparison of the pure rotational energies  $(J \leq 5)$  of CH<sub>3</sub> before and after refinement of  $r_{\rm e}$  illustrating the importance of this step.

The vibrational transition moments are defined as

$$\mu_{V'V} = \sqrt{\sum_{\alpha = x, y, z} |\langle V' \mid \bar{\mu}_{\alpha} \mid V \rangle|^2}$$
 (22)

where  $|V'\rangle$  and  $|V\rangle$  denote J=0 vibrational wavefunctions and  $\bar{\mu}_{\alpha}$  is the electronicallyaveraged dipole moment in the molecular frame (see the section entitled 'Dipole moment
surface' above). For calculation of vibrational transition moments we used our *ab initio*PES-1 and truncated the vibrational basis set at polyad number  $P_{\text{max}}=32$ . A number of
computed transition moments for the strongest lower lying bands are listed in Table 3 where
they are compared with the available experimental data The complete list of theoretical
transition moments is given as Supporting Information and can be also found at www.exomol.

222

Table 1: Vibrational band centers (cm $^{-1}$ ) of  $^{12}\mathrm{CH}_3$  from variational calculations.

Γ	State	Ref.	Obs.a	$P_{\text{max}} = 24^b$	$P_{\text{max}} = 32^c$
$A_1'$	$2\nu_2$	?	1288.1	1279.77	1281.24
	$2\nu_4$			2737.63	2739.64
	$4\nu_2$			2773.65	2776.86
	$ u_1$	?	3004.42	3002.71	3002.76
	$3\nu_4^3$			4118.59	4120.58
	$\nu_1 + 2\nu_2$			4258.97	4260.53
	$6\nu_2$			4391.99	4397.00
	$ u_3^1 +  u_4^1 $			4537.94	4538.93
	$4\nu_4$			5371.39	5364.56
	$2\nu_2 + 3\nu_4^3$			5475.84	5480.07
	$4\nu_2 + 2\nu_4$			5601.91	5607.20
E'	$ u_4^1$	?	1397.0	1385.99	1387.26
	$2\nu_2 + \nu_4^1$			2688.80	2691.61
	$2\nu_4^2$			2759.77	2762.05
	$ u_3^1$	?	3160.8	3158.88	3158.83
	$3\nu_{4}^{1}$			4074.69	4075.46
	$2\nu_2 + 2\nu_4^2$			4087.92	4091.72
$A_2''$	$ u_2$	?	606.453	602.43	602.43
	$3\nu_2$			2010.09	2010.09
	$\nu_2 + 2\nu_4^0$			3372.27	3371.59
	$5\nu_2$			3569.96	3569.95
	$\nu_1 + \nu_2$			3596.35	3596.30
	$3\nu_4^3$			4768.70	4767.06
	$\nu_2 + 2\nu_4^0$			4823.32	4822.79
	$\nu_1 + 3\nu_2$			4981.58	4981.52
E''	$ u_2 +  u_4^1$			2000.24	2002.22
	$\nu_2 + 2\nu_4^2$			3388.24	3391.11
	$3\nu_2 + \nu_4^1$			3426.45	3430.06
	$ u_2 +  u_3^1 $			3736.40	3736.97
	$\nu_2 + 3\nu_4^1$			4726.62	4728.62
	$3\nu_2 + 2\nu_4^2$			4835.22	4839.85
	$\nu_1 + \nu_2 + \nu_4^1$			4980.92	4983.16

<sup>&</sup>lt;sup>a</sup> Experimental values of band centers used to replace the theoretical values  $P_{\text{max}} = 32$ , see

text.  $^b$  Computed using the  $P_{\rm max}=24$  basis set in conjunction with PES-2.?  $^c$  Computed using the  $P_{\rm max}=32$  basis set in conjunction with PES-2.?  $^14$ 

Table 2: Theoretical rotational term values  $(N \leq 5, \text{ in cm}^{-1})$  of CH<sub>3</sub>, computed with TROVE using different equilibrium structure parameters.

States			Term values				
$\overline{N}$	K	$ au_{ m rot}$	Obs.	ObsCalc. <sup>a</sup>	ObsCalc. $^b$		
1	1	0	14.3189	0.032377	0.004027		
2	0	1	57.4396	0.112005	-0.002023		
2	2	0	38.1186	0.092340	0.017004		
2	1	0	52.6112	0.106875	0.002511		
3	3	0	71.3965	0.179934	0.038989		
3	2	0	95.5353	0.203902	0.014649		
3	1	0	110.0032	0.219365	0.001200		
4	0	0	191.2473	0.375024	-0.004034		
4	4	0	114.1491	0.295456	0.070301		
4	2	0	172.0038	0.353500	0.012772		
4	3	0	147.9203	0.327970	0.035289		

 $<sup>^</sup>a$  Calculated using  $r_{\rm e}=1.07736927$  Å and  $\alpha_{\rm e}=120.0^\circ$  (PES-1, see text).

#### 223 Intensity simulations

In order to simulate absorption spectra at a given temperature T and within a particular 224 wavenumber range, the upper and lower energies and the Einstein coefficients  $A(f \leftarrow i)$  [or 225 the line strengths  $S(f \leftarrow i)$ ; the relationship between  $A(f \leftarrow i)$  and  $S(f \leftarrow i)$  is described 226 in Ref. ? ] of all transitions in this range must be known; in practice, only the transitions 227 above a certain minimum intensity are included. The simplest way to present the spectral 228 data is a 'stick' diagram where the height gives the integrated absorption coefficient from 229 Eq. (17). In this section we report such simulations for the CH<sub>3</sub> absorption bands covering 230 the frequency range 600–1200 cm<sup>-1</sup> for the out-of-plane bending mode  $\nu_2$ . The line strengths 231 in Eq. (17) are computed from Eq. (16) with the spin statistical weights  $g_{\rm ns}$  from Ref. ? . 232 The simulations are carried out using PES-2 and the computed DMS described above. We 233 used a partition-function value of  $Q = \frac{724.6306732.734}{2}$ , obtained at 300 K by summing over 234 all variational term values  $(J \le 40)$  below 36871.73 cm<sup>-1</sup>. With the limits defined above, 235 we computed  $\frac{22\,805\,378\ \text{transitions}}{2,058,655,166\ \text{transitions}}$  using the GPU GAIN-MPI

 $<sup>^</sup>b$  Calculated using  $r_{\rm e}=1.0762977119$  Å and  $\alpha_{\rm e}=120.0^\circ$  (PES-2, see text).

Table 3: Band Centers  $\nu_{fi}$  and Vibrational Transition Moments  $\mu_{fi}$  for CH<sub>3</sub>.

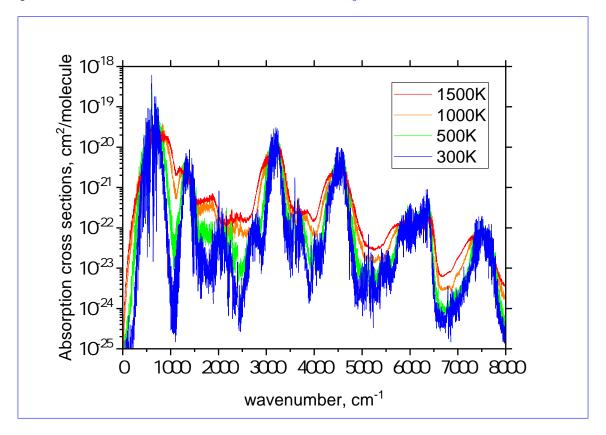
States		$\nu_{fi}/\mathrm{cm}^{-1}$	Calc. $\mu_{fi}/D$	Obs. $\mu_{fi}/D$	Ref.
$\overline{f}$	i				
$2\nu_2$	$\nu_2$	678.81	0.25684	0.31(6)	[?]
$ u_2$	0	602.43	0.20403	0.215(25)	[ ? ? ? ]
$ u_3^1$	0	3158.83	0.03999	0.03(27)	[??]
$\overline{ u_4^1}$	0	1387.26	0.02931		
$2\nu_3 + \nu_4$	0	4529.74	0.02049		
$ u_1 +  u_4^1$	0	4383.56	0.00866		
$2\nu_2 + \nu_3^1$	0	4396.18	0.00486		
$2\nu_3^2$	0	6294.76	0.00462		
$ u_1 +  u_3^1$	0	6076.68	0.00321		
$2\nu_4^2$	0	2762.05	0.00313		
$\nu_3^1 + 2\nu_4$	0	5864.94	0.00242		
$3\nu_4^1$	0	4075.46	0.00186		
$2\nu_2 + \nu_3^1 + \nu_4^1$	0	5789.16	0.00130		
$4\nu_2$	0	5856.39	0.00116		

<sup>&</sup>lt;sup>a</sup> The transitions originate in the vibrational ground state (i=0) with the exception of the hot band  $2\nu_2 \leftarrow \nu_2$ .

#### 237 program.?

2 gives an overview (log-scale) of the absorption spectrum of CH<sub>3</sub> at T = 300 K different temperatures produced using the line list (log-scale) by means of the ExoCross program. <sup>?</sup> 3 shows four regions with the strongest, dipole-allowed bands  $\nu_2$ ,  $\nu_4$ ,  $\nu_3$  and  $\nu_3 + \nu_4$ .

Figure 2: An overview of the absorption spectrum (cross sections) of CH<sub>3</sub> at T = 300different temperatures T = 300, 500, 1000 and 1500 K generated using our line list and the Gaussian line profile with the full-width-at-half-maximum of 21 cm<sup>-1</sup>.

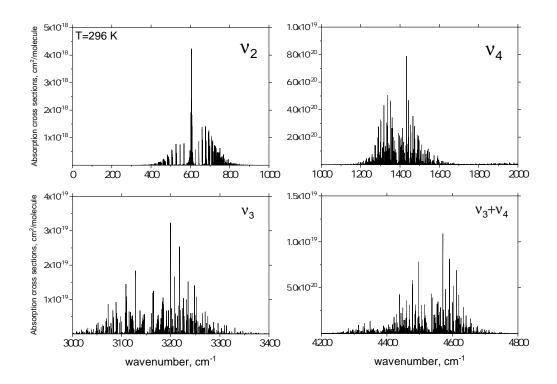


4 shows the emission spectrum of CH<sub>3</sub> in the wavenumber interval 600-1200 cm<sup>-1</sup>, simulated with TROVE at two different levels of theory, rovibrational and purely vibrational.

The simulated spectra are compared to an experimental spectrum recorded by Hermann and Leone?? (see Fig. 4 of Ref. ?).

4(a) shows a simulation of the CH<sub>3</sub> emission spectrum, computed with TROVE at a temperature of T = 300 K, taking into account all vibrational transitions in the 600-1200 cm<sup>-1</sup> interval that are calculated with the  $P_{\text{max}} = 32$  basis set. The simulation involves the con-

Figure 3: A selection of the strongest absorption bands of  $\mathrm{CH_3}$  at  $T=296~\mathrm{K}$  generated using the line list. A Gaussian line profile with the half-width-half-maximum of 0.08 cm<sup>-1</sup> was used in production of the cross sections shown.



volution of the calculated intensities with a Gaussian line shape function with a full width at half maximum (FWHM) of 17 cm<sup>-1</sup>.

The simulation is compared with the experimental spectrum obtained by Hermann and Leone, <sup>?</sup> shown in 4(b).\* In the experiment, the CH<sub>3</sub> radicals were produced the in a photo-fragmentation process of methyl iodide CH<sub>3</sub>I. Hermann and Leone <sup>?</sup> suggested that after the breaking of the C–I bond, the CH<sub>3</sub> radicals are predominantly produced in excited states of the out-of-plane bending mode  $\nu_2$ . The CH<sub>3</sub> fragment of a dissociated CH<sub>3</sub>I molecule will typically have a pyramidal structure close to that of the methyl group in CH<sub>3</sub>I. For a CH<sub>3</sub> radical, however, which is planar at equilibrium, such structures are associated with high excitations of the  $\nu_2$  vibrational mode. Excitations with  $\nu_2 \leq 10$  have been observed in the

<sup>\*</sup>Reproduced from Hermann, H. W.; Leone, S. R. The Journal of Chemical Physics 1982, 76, 4759–4765, with the permission of AIP Publishing.

emission experiment of Refs. ??.

The low-temperature line list generated for CH<sub>3</sub> in the present work does not cover 259 an energy interval large enough to allow an accurate simulation of In order to simulate 260 vibrationally very hot transitions (hotter than 1500 K) corresponding to the experimental 261 spectrum of Refs. ? which involves out-of-plane bending states  $v_2 \nu_2$  with  $v_2 \leqslant 10$ . We have 262 circumvented this problem by carrying out simulations of the experimental spectrum 10, we 263 have carried out simulations employing a so-called one-band model (with the one band being 264 the  $\nu_2$  fundamental band here). In the one-band model, we use the  $\nu_2$ -band data from the 265 'parent', 300 K ro-vibrational line list also for the hot bands accompanying the  $\nu_2$  band. 266 The procedure employed is detailed in Ref. ? and we outline it briefly here: We initially 267 produce the 300 K,  $\nu_2$ -band cross sections by means of the ExoCross program? and convolve 268 them with a Gaussian profile of FWHM =  $17 \text{ cm}^{-1}$  between 0 and  $1200 \text{ cm}^{-1}$ , generating 269 1201 data points. Then the wavenumbers of the computed cross sections are shifted by 270 -606.4531 cm<sup>-1</sup>, positioning the band center at zero, and the cross sections are normalized. 271 A local version of ExoCross now obtains, from the vibrational transition moments computed 272 with TROVE, the vibrational band intensities for the hot bands  $(v_2 + 1) \nu_2 \leftarrow v_2 \nu_2$ ,  $v_2 \leqslant$ 273 9. Finally, the simulated spectrum is generated by placing, for each hot band, the  $\nu_2$ -band profile at the band center  $\nu_{fi}$  of the hot band in question, scaled by its vibrational band 275 intensity. Such simulations have been carried out for temperatures T of 1000, 2000, and 276 3000 K, respectively, and the results are shown in 4(c), where they can be compared to the 277 experimental results?? in 4(b). 278 4(c) shows that the 'standard' T = 300 K rovibrational simulation of the  $CH_3$  emission 279 spectrum [4(a)] has little resemblance to the experimental spectrum from Ref. ? 280 Obviously in the experiment, the CH<sub>3</sub> molecules populate states of much higher energies 281 than those accessed in thermal equilibrium at T = 300 K. However, among the 'vibrational 282 simulations' in 4(c), the curve obtained for T=2000 K has a very substantial similarity to 283 the experimental curve. This confirms the suggestion by Hermann and Leone?? that dis-284

sociation of CH<sub>3</sub>I produces CH<sub>3</sub> radicals in highly excited states of the out-of-plane bending mode  $\nu_2$ . The successful simulation of the emission spectrum of Refs. ? ? lends credibility 286 to the ab initio DMS of the present work; the intensities based on this DMS are in very good 287 qualitative agreement with experiment. 288 Our complete T = 300T = 1500 K CH<sub>3</sub> line list is given as Supporting Information 289 accessed via the repository www.zenodo.org, see Ref. ? . It details the transition energies, 290 line strengths, and Einstein coefficients  $A(f \leftarrow i)$ , together with the absorption intensities 291 estimated for T = 300 and the temperature dependent partition function Q(T). The line list 292 should be applicable for the temperatures below 1500 K. However, the simulated spectra 293 will become increasingly inaccurate with increasing temperature. The line list is given using 294 the ExoMol format? which can be used together with the ExoCross program? to generate 295 spectra of  $CH_3$ . 296

#### 97 Conclusion

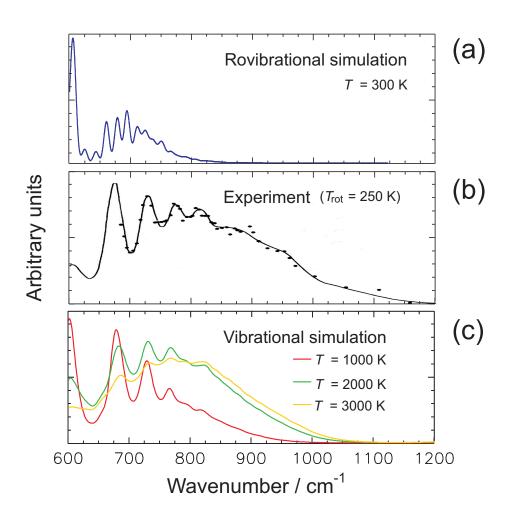
We have presented calculated spectra for the methyl radical covering a large part of the infrared region. Detailed comparisons with observed room-temperature spectra show excellent agreement for the position and intensity of the transitions.

Our ultimate aim is the construction of a CH<sub>3</sub> line list capable of reproducing and predicting observed spectra at temperatures much higher than 300 K. This will facilitate detections
of the methyl radical in space. The work presented here represents the first step towards
this goal, which involves the generation, refinement, and validation of the required potential
energy and dipole moment surfaces, as well as establishing the level of accuracy that can
be achieved in the variational nuclear-motion calculations with our computational resources.
In this initial phase, we have produced an methyl radical line list consisting of 22.8 million
transitions between 3,671,465–2 billion transitions between 9,127,123 energy levels for rovibrational states up to  $J_{\text{max}} = 25$ –40 and energies up to 2019 000 cm<sup>-1</sup>. Even though this

 $_{310}$  list is less complete and less accurate than ultimately desired, it can be used to produce

simulated CH<sub>3</sub> spectra that agree well with observation at room temperature.

Figure 4: Emission spectra of CH<sub>3</sub>. (a) Rovibrational simulation assuming the CH<sub>3</sub> radicals to be in thermal equilibrium at T=300 K. (b) CH<sub>3</sub> ( $\nu_2$ ) out-of-plane bending mode emission spectrum (dots) obtained? after dissociation of CH<sub>3</sub>I  $\rightarrow$  CH<sub>3</sub>( $\nu_2$ ) + I\*. The continuous curve is the best fit? involving the hot bands ( $v_2+1$ )  $\nu_2 \leftarrow v_2 \nu_2$  with  $v_2 \leq 9$ . The spectrum is convolved with a spectrometer slit function with a FWHM of 33 cm<sup>-1</sup> and a 19 cm<sup>-1</sup> bandwidth ascribed to the breadth of the  $\Delta K=0$  manifold of transitions with varying J values. Reproduced from Ref. ? with the permission of AIP Publishing. (c) Vibrational simulations at temperatures T=1000, 2000, and 3000 K, respectively, taking into account the vibrational transitions ( $v_2+1$ )  $\nu_2 \leftarrow v_2 \nu_2$  with  $v_2 \leq 9$  (see text).



# 312 Acknowledgement

A.Y. acknowledges support from DESY (HGF IVF) and from the Cluster of Excellence 'The
Hamburg Centre for Ultrafast Imaging' of the Deutsche Forschungsgemeinschaft (DFG) EXC 1074 - project ID 194651731. S.Y. is grateful for support from the UK Science and
Technology Research Council (STFC) ST/R000476/1. This work made extensive use of
UCL's Legion high performance (HPC) computing facilities as well as of HPC provided by
DiRAC supported by STFC and BIS. It was further supported in part by grant JE 144/25-1
from the DFG.

# Supporting Information Available

The Supporting Information for this work includes the: (i) dipole moment parameters  $\mu_{k,l,m,\dots}^{(\Gamma)}$  together with the Fortran routine, (ii) potential energy parameters  $f_{jk}^{(s)}$ ; (iii) Fortran routines for calculating the dipole moment components; the complete cool and potential energy values for a given geometry. Our complete  $T=1500 \mathrm{K}$  CH<sub>3</sub> line list with transition energies, line strengths, and Einstein coefficients  $A(f \leftarrow i)$ , together with the partition function can be accessed via the Zenodo repository, www.zenodo.org? as well as absorption intensities estimated for  $T=300 \mathrm{K}$ ); a Fortran program to generate synthetic spectra. at

#### FIGURE CAPTIONS

- 1 Root-mean-square (RMS) errors of the fittings to the *ab initio* dipole moment values. The results of a series of fittings are shown. In each fitting, the data set includes the *ab initio* points with electronic energy (relative to the potential energy minimum)  $V \leq E$  (see text).
  - An overview of the absorption spectrum (cross sections) of CH<sub>3</sub> at T = 300 different temperatures T = 300, 500, 1000 and 1500 K generated using our line list and the Gaussian line profile with the full-width-at-half-maximum of  $\frac{2 \text{ cm}^{-1}}{2 \text{ cm}^{-1}}$ ...
  - 3 A selection of the strongest absorption bands of CH<sub>3</sub> at T=296 K generated using the line list. A Gaussian line profile with the half-width-half-maximum of 0.08 cm<sup>-1</sup> was used in production of the cross sections shown.
  - Emission spectra of CH<sub>3</sub>. (a) Rovibrational simulation assuming the CH<sub>3</sub> radicals to be in thermal equilibrium at T=300 K. (b) CH<sub>3</sub> ( $\nu_2$ ) out-of-plane bending mode emission spectrum (dots) obtained? after dissociation of CH<sub>3</sub>I  $\rightarrow$  CH<sub>3</sub>( $\nu_2$ ) + I\*. The continuous curve is the best fit? involving the hot bands ( $v_2+1$ )  $v_2 \leftarrow v_2 \nu_2$  with  $v_2 \leq 9$ . The spectrum is convolved with a spectrometer slit function with a FWHM of 33 cm<sup>-1</sup> and a 19 cm<sup>-1</sup> bandwidth ascribed to the breadth of the  $\Delta K=0$  manifold of transitions with varying J values. Reproduced from Fig. 4 of Ref. ? with the permission of AIP Publishing. (c) Vibrational simulations at temperatures T=1000, 2000, and 3000 K, respectively, taking into account the vibrational transitions ( $v_2+1$ )  $\nu_2 \leftarrow v_2 \nu_2$  with  $v_2 \leq 9$  (see text).