# **Chapter 20 Tutorial**

# Thermodynamics-IV

# **Chapter 20: The Second Law of Thermodynamics**

### **Question 1:**

A room air conditioner has a coefficient of performance of 2.9 on a hot day and uses 850 W of electrical power.

- (a) How many joules of heat does the air conditioner remove from the room in one minute?
- (b) How many joules of heat does the air conditioner deliver to the hot outside air in one minute?
- (c) Explain why your answers to parts (a) and (b) are not the same.

**IDENTIFY:** 
$$K = \frac{|Q_{\rm C}|}{|W|}$$
 and  $|Q_{\rm H}| = |Q_{\rm C}| + |W|$ .

**SET UP:** The heat removed from the room is  $|Q_{\rm C}|$  and the heat delivered to the hot outside is  $|Q_{\rm H}|$ .  $|W| = (850 \text{ J/s})(60.0 \text{ s}) = 5.10 \times 10^4 \text{ J}.$ 

**EXECUTE:** (a) 
$$|Q_C| = K|W| = (2.9)(5.10 \times 10^4 \text{ J}) = 1.48 \times 10^5 \text{ J}$$

**(b)** 
$$|Q_{\rm H}| = |Q_{\rm C}| + |W| = 1.48 \times 10^5 \text{ J} + 5.10 \times 10^4 \text{ J} = 1.99 \times 10^5 \text{ J}.$$

**EVALUATE:** (c) 
$$|Q_H| = |Q_C| + |W|$$
, so  $|Q_H| > |Q_C|$ .

#### Question 2:

An aircraft engine takes in 9000 J of heat and discards 6400 J each cycle.

- (a) What is the mechanical work output of the engine during one cycle?
- (b) What is the thermal efficiency of the engine?

**IDENTIFY:** For a heat engine, 
$$W = |Q_{\rm H}| - |Q_{\rm C}|$$
.  $e = \frac{W}{Q_{\rm H}}$ .  $Q_{\rm H} > 0$ ,  $Q_{\rm C} < 0$ .

**SET UP:** 
$$|Q_{\rm H}| = 9000 \text{ J. } |Q_{\rm C}| = 6400 \text{ J.}$$

**EXECUTE:** (a) 
$$W = 9000 \text{ J} - 6400 \text{ J} = 2600 \text{ J}.$$

**(b)** 
$$e = \frac{W}{O_{\text{H}}} = \frac{2600 \text{ J}}{9000 \text{ J}} = 0.29 = 29\%.$$

**EVALUATE:** Since the engine operates on a cycle, the net Q equal the net W. But to calculate the efficiency we use the heat energy input,  $Q_H$ .

### Question 3:

The Otto-cycle engine in a Mercedes-Benz SLK230 has a compression ratio of 8.8.

- (a) What is the ideal efficiency of the engine? Use Y = 1.40.
- (b) The engine in a Dodge Viper GT2 has a slightly higher compression ratio of 9.6. How much increase in the ideal efficiency results from this increase in the compression ratio?

**IDENTIFY:**  $e = 1 - r^{1-\gamma}$ 

**SET UP:** r is the compression ratio.

**EXECUTE:** (a)  $e = 1 - (8.8)^{-0.40} = 0.581$ , which rounds to 58%.

**(b)**  $e = 1 - (9.6)^{-0.40} = 0.595$  an increase of 1.4%.

**EVALUATE:** An increase in r gives an increase in e.