SINGLE DISPLACEMENT REACTION-1



MATERIAL REQUIRED



ADD 2 SPATULA OF COPPER SULPHATE IN WATER AND DISSOLVE IT



PUT FEW GRANULES OF ZINC IN THE SOLUTION



KEEP IT FOR SOME TIME. COLOUR OF SOLUTION CHANGES TO GREENISH AND COATING OF COPPER IS DEPOSITED ON ZINC GRANULES

More reactive metal displaces less reactive metal from its salt solution $CuSO_4 + Zn \rightarrow ZnSO_4 + Cu$

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