## **NEUTRALISATION REACTION**



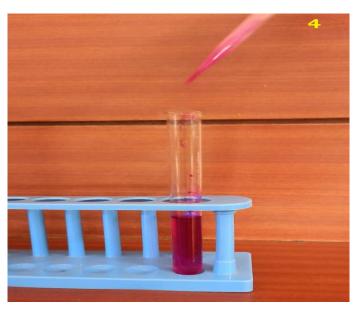
MATERIAL REQUIRED



POUR SOME SODIUM HYDROXIDE IN ONE TESTTUBE



POUR SULPHURIC ACID IN ANOTHER TESTTUBE



POUR FEW DROPS OF PHENOLPHTHALEIN IN THE TESTTUBE CONTAINING SODIUM HYDROXIDE AND COLOUR OF SOLUTION STARTS CHANGING TO PINK.

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POUR FEW DROPS OF PH. IN THE TESTTUBE CONTAINING SULPHURIC ACID. THERE IS NO CHANGE IN COLOUR



NOW POUR SODIUM HYDROXIDE IN THE TESTTUBE CONTAINING SULPHURIC ACID.



THE PINK COLOUR DISAPPEARS

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When acid reacts with base it becomes neutral. This is called neutralisation reaction.