## **SINGLE DISPLACEMENT REACTION -2**



MATERIAL REQUIRED



ADD 2 SPATULA OF COPPER SULPHATE IN BEAKER AND DISSOLVE IT.



PUT TWO IRON NAILS IN THE SOLUTION



COATING OF COPPER IS DEPOSITED ON THE IRON NAIL

More reactive metal displaces less reactive metal from its salt solution

 $CuSO_4 + Fe \rightarrow FeSO_4 + Cu$ 

**ARSHA SARAF'S LAB**