

Test 1, PHYSIOLOGY 1

1./ When solutes are in motion to fill all of the available volume is called:

- A. filtration
- B. diffusion**
- C. osmosis
- D. electrical conduction
- ___ E. regulation

2./ pH

- A. indicates the negative concentration of logarithm
- B. indicates positive logarithm
- C. indicates the negative logarithm of hydrogen ion concentration**
- D. indicates the logarithm of hydroxyl ion concentration
- ___ E. none of the above

3./ 40 percent of body weight

- A. bones
- B. extracellular fluid compartment
- C. blood plasma
- D. intracellular fluid compartment**
- ___ E. water content of newborns

4./ Filtration and osmosis

- A. are accompanied with motion of solutes in a solution
- B. occur through a semipermeable membrane**
- C. expansion to fill all of the available volume
- D. are only pathological processes
- ___ E. occur within one compartments

5./ Which of the following is true?

- A. Interstitial compartment is part of the intracellular compartment
- B. EC and IC compartments make up the total body fluid compartment**
- C. Blood plasma is approx..15 percent of body weight
- D. Interstitial compartment is located within the blood vessels
- ___ E. Cytoplasmic water is part of the extracellular compartment.

6./ In error, a patient is infused with large volumes of a solution that causes lysis of his red blood cells. The solution was most likely

- A. isotonic NaCl
- B. isotonic mannitol
- C. hypertonic NaCl
- D. hypotonic NaCl**
- ___ E. hypertonic urea

7./ Assuming complete dissociation of all solutes, which of the following solutions would be hyperosmotic to 1 mM NaCl?

- A. 1 mM CaCl_2**
- B. 1.5 mM glucose
- C. 1 mM glucose
- D. 1 mM fructose
- ___ E. 1 mM KCl

8./ Buffers

- A. are strong acids
- B. are isosmotic solutions
- C. are formed exclusively in the blood
- D. are colored solutions
- E. prevent changes in pH when H^+ ions are added to or removed from the solution**

9./ Increase of interstitial osmotic concentration results in

- A. increased blood volume
- B. decreased Na^+ concentration
- C. increased intracellular volume
- D. edema**
- E. none of the above

10./ When exsiccosis (dehydration) occurs,

- A. the EC volume is decreasing.**
- B. the EC volume is increasing.
- C. concentration of ions is decreasing in the EC
- D. edema develops.
- E. intracellular volume increases.

11./ Which of the following characteristics is shared by simple and facilitated diffusion of glucose

- A. requires Na^+ gradient
- B. is saturable
- C. requires metabolic energy
- D. is inhibited by the presence of galactose
- E. occurs down on electrochemical gradient**

12./ Buffers

- A. are strong bases
- B. are isosmotic solutions
- C. prevent changes in pH when H^+ ions are added to or removed from the solution**
- D. are organic solutions
- E. are mixtures of metal ions

13./ The permeability of a solute in a lipid bilayer will be increased by an increase in the

- A. molecular radius of the solute
- B. thickness of the bilayer
- C. lipophilicity of the solute**
- D. hydrophilicity of the solute
- E. concentration difference of the solute across the bilayer

14./ Acidosis

- A. increase in blood pCO_2 level**
- B. increase in hydroxyl ion concentration
- C. decrease in hydrogen ion concentration
- D. increase in bicarbonate ion concentration in the blood
- E. decrease in blood pCO_2 level

15./ Body temperature in humans

- A. is independent from the environmental temperature**
- B. lower in newborns
- C. higher in the elderly
- D. is stable only in pathological conditions
- E. is continuously increasing because of overheating**

16 Comparative physiology

- A. describes lineage...
- B. Deals with physical properties of various living organisms**

C D E

17./ A women runs marathon in 28°C weather and replaces all volume lost in sweat by drinking distilled water. After the marathon she will have

- A. decreased total body weight
- B. decreased hematocrit
- C. decreased intracellular fluid(ICF)
- D. decreased plasma osmolarity**
- ___ E. increased intracellular osmolarity

18) In error a patient is infused with large volumes of a solution that causes shrinking of his red blood cells. The solution is most likely:

- A. Hypotonic Urea
- B. Isotonic Mannitol
- C. Hypertonic Saline**
- D. Hypotonic NaCl
- E. Isotonic NaCl

19) Etiology of a disease could be:

- A. Abdominal
- B. Pulmonological
- C. Confidential
- D. Acquired**
- E. Impaired

20) Findings:

- A. Reflect the subjective experience of the patient
- B. Might be determined by instruments**
- C. Can be detected in the course of physical examination
- D. Are obvious reasons of death
- E. None of the above

21) Water intake from the food is more or less similar amount to that of:

- A. Water intake by cellular metabolism
- B. Water intake by drinking
- C. Water loss by expiration**
- D. Water in feces
- E. None of the above

22) Typical Sign of hyperkalemia on ECG

- A. Narrow QRS
- B. Low ST and low T
- C. Prominent U
- D. ST Elevation
- E. ST depression and peaked T**

23) 5% of our body weigh is

- A. inorganic ions
- B. extracellular fluid compartement
- C. blood plasma**
- D. itracellular fluid compartement
- E. water content of newborns

24) Which of the followin statement is true:

- A. interstitial fluid is located within the blood vessels
- B. interstitial comp. Is part of the intracellular comp.
- C. blood plasma is the apex 15% of our body weigt
- D. increasing fat content is parallel with deacrising water content**
- E. cytoplasmatic water is part of the extracellular comp.

25) After use pipets ups should remove

- A. by hand
- B. built up tip remover**

- C. both above
- D. using clean cloth
- E. cutting

26) Pressured interstitial osmotic concentration is parallel with

- A. increased blood volume
- B. increased blood Na⁺ concentration
- C. decreases intracellular volume
- D. dehydration
- E. none of above

27) equation incorrect

- A. Intracellular comp. + plasma = EC
- B. Extra comp + plasma = Intr comp.
- C. inter comp < plasma
- D. extra comp. - plasma = intr comp
- E. EC < IC

28) Cells of unicellular organisms

- A. are in indirect contact with external environment
- B. exchange material with the external environment
- C. independent from each other
- D. dependent on the external environment
- E. might have cellular organisms

29) Microscopy immersion oil is used :

- to generate more detailed picture

Select the single incorrect answer (126-170).

10) Normal intracellular sodium ion concentration

- A. Is lower than extracellular sodium ion concentration
- B. Is about 10 times higher than extracellular potassium ions...
- C. Is 15 mmol/L
- D. Is about one tenth of intracellular potassium ion concentration
- E. Is lower than extracellular chloride ion concentration

126./ Parts of regulatory circuits

- A. set point
- B. control mechanisms
- C. actual value
- D. overcompensation
- E. concentration units (return point)

127./ Intracellular concentrations of ions

- A. 142 mM of Na⁺ ion
- B. 15 mM of Na⁺ ion
- C. 3 mM of Cl⁻ ion
- D. 150 mM of K⁺ ion
- E. variable mM of Ca⁺⁺

128/ The cells of multicellular organisms

- A. are in indirect contact with the external environment
- B. are in indirect contact with the internal environment
- C. are in direct contact with the internal environment
- D. are functioning relatively independently from the external environment
- E. are mainly depending on the internal environment

129./ The major elements of homeostasis are

- A. osmosis

- B. isothermia
- C. isohydria
- D. isomeria
- E. isovolemia

130./ Physiological concentrations of plasma ingredients

- A. 142 mM of Na^+ ion
- B. 102 mM of Na^+ ion
- C. 105 mM of Cl^- ion
- D. 4.4 mM of K^+ ion
- E. 2.2 mM of Ca^{++}

131./

- A. The pH of IC compartment is 6.8.
- B. The pH of blood is 7.35-7.45.
- C. The pH of EC compartment is 7.35-7.45.
- D. Acidic metabolic products decrease the pH of IC compartment
- E. The pH of cytoplasm is above 7.45.

132./ Buffers of the blood

- A. carbonic acid/bicarbonate
- B. TRIS/HCl
- C. protein/protein anion
- D. dihydrogen phosphate/hydrogen phosphate
- E. aium ion/ammonia

133/ When body temperature rises above 42 °C,

- A. many of the proteins of human body are precipitated.
- B. the speed of enzyme reactions will be decreased
- C. thermoregulation is set to heat loss
- D. hyperventilation is occurring
- E. it is a life threatening situation.

134./ Typical membrane molecules

- A. receptors
- B. ion channels
- C. surface markers
- D. nucleic acids
- E. phospholipids

135./ Active transport could be

- A. symport
- B. antiport
- C. filtration
- D. electroneutral
- E. electrogenic

136./ Erythropoiesis takes place in

- A. fetal spleen.
- B. fetal kidney.
- C. fetal liver.
- D. fetal bone marrow
- E. yolk sac

137) Typical analytical error

- A. Mislabeling
- B. Instrumental error
- C. Human Error

- D. Incorrectly prepared solutions
- E. Dilution and pipetting errors

138) What are the signs of depleted body water?

- A. Confusion
- B. Dryness of mucous membranes
- C. Weight loss
- D. Decreased saliva secretion
- E. Decreased Urine Volume

Establish the appropriate association

- A. Adverse consequence of severe acidemia
 - B. Adverse consequence of severe alkalemia
 - C. Both of the Them
 - D. None of them
- 16) Hyperventilation **A**
- 17) Hypokalemia **B**
- 18) Cerebral signs and symptoms **C**
- 19) Reduction of threshold in ventricular fibrillation **A**
- 20) Cirrhosis **D**

- A – mmol/L
- B – mOsmol/L
- C – both
- D – none

- A-- 176. concentration unit for one solute
- D-- 177. gram-molecular weight /liter
- D-- 178. units of volume
- B-- 179. concentration unit for a mixture of solutes
- B-- 180. the concentration of all osmotically active particles

- A – pH
- B – pOH
- C – both
- D – none

- C-- 181. negative logarithm of ion concentration
- D-- 182. chemical formula
- D-- 183. partial concentration
- A-- 184. negative logarithm of hydrogen ion concentration
- B-- 185. negative logarithm of hydroxyl ion concentration

- A – buffer
 - B – aqueous solution
 - C – both
 - D – none
- C-- 186. consists of solvent and solute(s)
- D-- 187. mixture of O_2 and N_2
- A-- 188. regulates hydrogen-ion concentration
- A-- 189. H_2CO_3/HCO_3^-
- B-- 190. 0.5 M HCl

- A – active transport
- B – facilitated diffusion

C – both

D – none

--C-- 191 . transmembrane process

--B-- 192. occurs down on chemical potential

--A-- 193. occurs up on chemical potential

--C-- 194. requires carrier molecule

--D-- 195. no carrier protein is required

A – behavioral adaptation

B – physiological thermoregulation

C – both

D – none

--B-- 196. thermoreceptors

--B-- 197. increased basic metabolism

--D-- 198. decreased glucose oxidation

--C-- 199. typical to humans

--B-- 200. hyperventillation

