

CHAPTER 10

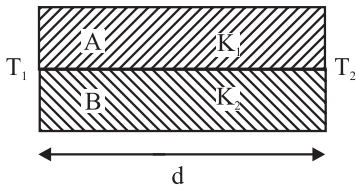
Thermal Properties of Matter

Thermal Expansion

- A copper rod of 88 cm and an aluminium rod of unknown length have their increase in length independent of increase in temperature. The length of aluminium rod is : (2019)
($\alpha_{\text{Cu}} = 1.7 \times 10^{-5} \text{ K}^{-1}$ and $\alpha_{\text{Al}} = 2.2 \times 10^{-5} \text{ K}^{-1}$)
 - 6.8 cm
 - 113.9 cm
 - 88 cm
 - 68 cm
- Coefficient of linear expansion of brass and steel rods are α_1 and α_2 . Lengths of brass and steel rods are l_1 and l_2 respectively. If $(l_2 - l_1)$ is maintained same at all temperatures, which one of the following relations holds good? (2016 - I)
 - $\alpha_1 l_2 = \alpha_2 l_1$
 - $\alpha_1 l_2^2 = \alpha_2 l_1^2$
 - $\alpha_1^2 l_2 = \alpha_2^2 l_1$
 - $\alpha_1 l_1 = \alpha_2 l_2$
- The value of coefficient of volume expansion of glycerin is $5 \times 10^{-4}/\text{K}$. The fractional change in the density of glycerin for a rise of 40°C in its temperature, is: (2015 Re)
 - 0.010
 - 0.015
 - 0.020
 - 0.025
- Two identical bodies are made of a material for which the heat capacity increases with temperature. One of these is at 100°C , while the other one is at 0°C . If the two bodies are brought into contact, then, assuming no heat loss, the final common temperature is: (2016 - II)
 - Less than 50°C but greater than 0°C
 - 0°C
 - 50°C
 - More than 50°C
- Steam at 100°C is passed into 20 g of water at 10°C . When water acquires a temperature of 80°C , the mass of water present will be: [Take specific heat of water = $1 \text{ cal/g}^\circ\text{C}$ and latent heat of steam = $540 \text{ cal/g}^\circ\text{C}$]: (2014)
 - 24 g
 - 31.5 g
 - 42.5 g
 - 22.5 g

Heat Transfer and Thermal Conductivity

Specific Heat, Latent Heat and Calorimetry

- The quantities of heat required to raise the temperature of two solid copper spheres of radii r_1 and r_2 ($r_1 = 1.5 r_2$) through 1 K are in the ratio: (2020)
 - $\frac{9}{4}$
 - $\frac{3}{2}$
 - $\frac{5}{3}$
 - $\frac{27}{8}$
- A piece of ice falls from a height h so that it melts completely. Only one-quarter of the heat produced is absorbed by the ice and all energy of ice gets converted into heat during its fall. The value of h is [Latent heat of ice is $3.4 \times 10^5 \text{ J/kg}$ and $g = 10 \text{ N/kg}$]: (2016 - I)
 - 34 km
 - 544 km
 - 136 km
 - 68 km
- The unit of thermal conductivity is : (2019)
 - J m K^{-1}
 - $\text{J m}^{-1} \text{K}^{-1}$
 - W m K^{-1}
 - $\text{W m}^{-1} \text{K}^{-1}$
- Two rods A and B of different material are welded together as shown in figure. Their thermal conductivities are K_1 and K_2 . The thermal conductivity of the composite rod will be: (2017-Delhi)
 
 - $\frac{3(K_1 + K_2)}{2}$
 - $K_1 + K_2$
 - $2(K_1 + K_2)$
 - $\frac{K_1 + K_2}{2}$
- The two ends of a metal rod are maintained at temperatures 100°C and 110°C . The rate of heat flow in the rod is found to be 4.0 J/s . If the ends are maintained at temperatures 200°C and 210°C , the rate of heat flow will be: (2015)
 - 16.8 J/s
 - 8.0 J/s
 - 4.0 J/s
 - 44.0 J/s

