## FINAL REVIEW SOLUTIONS

1. a. Assume there are n links pointing down, N-n links pointing up, then The partition function is

$$Z = \sum_{n=0}^{N} {N \choose n} e^{-\beta(-wl)(2n-N)}$$
$$= e^{-\beta wNl} (1 + e^{2\beta wl})^{N}$$
$$= (e^{\beta wl} + e^{-\beta wl})^{N}$$

The free energy is

$$-\beta F = \log Z = N \log \left( e^{\beta wl} + e^{-\beta wl} \right)$$

Energy

$$E = -\frac{\partial}{\partial \beta} \log Z = -Nwl \tanh(\beta wl)$$

So the equilibrium length will be

$$L = -\frac{E}{w} = Nl \tanh(\beta wl)$$

- b. From the expression of L, we see that as we heat up the band, the length will decrease, so the bucket will move up.
  - From an entropic point of view, as heat comes into the system, the entropy should increase, and we know that the entropy of the band increases when it gets shorter.
- c. See Fig. 1 on next page.

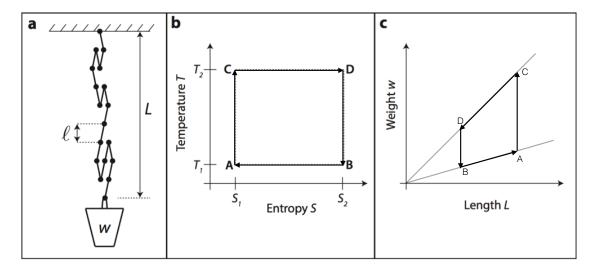


FIG. 1. Rubber band and Carnot cycle.

2. a. At zero temperature,

$$N = \frac{2A}{(2\pi\hbar)^2} \int_0^{p_F} dp 2\pi p = \frac{2A}{(2\pi\hbar)^2} 2\pi m \int_0^{\epsilon_F} d\epsilon = \frac{A}{\pi\hbar^2} m\epsilon_F$$

So the Fermi energy is

$$\epsilon_F = \frac{n\pi\hbar^2}{m}$$

b. A spin-up particle with energy  $\epsilon$  will have kinetic energy  $\epsilon + \gamma B$ , while a spin-down particle with energy  $\epsilon$  will have kinetic energy  $\epsilon - \gamma B$ 

$$N_{+} = \frac{A}{2\pi\hbar^{2}}m(\epsilon_{F} + \gamma B)$$
$$N_{-} = \frac{A}{2\pi\hbar^{2}}m(\epsilon_{F} - \gamma B)$$

i. The magnetization is

$$M = \gamma (N_+ - N_-) = \frac{Am}{\pi \hbar^2} \gamma^2 B$$

ii. The magnetic susceptibility is

$$\chi = \frac{\partial M}{\partial B} = \frac{Am}{\pi \hbar^2} \gamma^2$$

c. At temperature T,

$$\begin{split} N = & \frac{Am}{\pi\hbar^2} \int_0^\infty d\epsilon \frac{1}{e^{\beta(\epsilon - \mu)} + 1} \\ = & \frac{Amk_BT}{\pi\hbar^2} \log \frac{e^{-\beta\mu} + 1}{e^{-\beta\mu}} \\ = & \frac{Am\epsilon_F}{\pi\hbar^2} + \mathcal{O}(e^{-\frac{T_F}{T}}) \\ E = & \frac{Am}{\pi\hbar^2} \int_0^\infty d\epsilon \frac{\epsilon}{e^{\beta(\epsilon - \mu)} + 1} \\ = & \frac{Am(k_BT)^2}{\pi\hbar^2} \left( \frac{\pi^2}{6} + \frac{1}{2} \left( \frac{T_F}{T} \right)^2 + \mathcal{O}(e^{-\frac{T_F}{T}}) \right) \\ = & \frac{Am\epsilon_F^2}{2\pi\hbar^2} + \frac{\pi^2}{6} \frac{Am}{\pi\hbar^2} (k_BT)^2 \end{split}$$

The energy per particle is

$$\varepsilon = \frac{E}{N} = \frac{\epsilon_F}{2} + \frac{\pi^2}{6} \frac{(k_B T)^2}{k_B T_F}$$

The specific heat per particle is

$$C_V = \frac{\partial E}{\partial T} = k_B \frac{\pi^2}{3} \frac{T}{T_F}$$

For two dimensional ideal gas,

$$C_V = k_B$$

We see that the specific heat per electron is parametrically small by a factor of  $\frac{T}{T_F}$ .

## 4.10) HEAT CAPACITY OF INTERGALACTIC SPACE

For the atoms,  $C_{\rm H}/V$  = (3/2)nk<sub>B</sub>. For photons, we first re-express the proportionality factor in the expression (20) by inserting the Stefan-Boltzmann constant,  $\sigma_{\rm B} = \pi^2 k_{\rm B}^4/60 \mu^3 c^2$ :

$$U/V = 4\sigma_B T^4/c$$
 ,  $C_{rad}/V = 16\sigma_B T^3/c$  ;  $C_{H}/C_{rad} = \frac{3ck_B}{32\sigma_B} \frac{N/V}{T^3} \approx 2.84 \times 10^{-10}$  .

<u>Comment</u>. It usually simplifies numerical calculations if one expresses U/V in the form used here, involving  $\sigma_{\rm B}/c$  and T, rather than the form (20) directly.

# 6.3) DISTRIBUTION FUNCTION FOR DOUBLE OCCUPANCY STATISTICS

(a) Occupancy: 0 1 2

Energy: 0 
$$\varepsilon$$
  $2\varepsilon$ 

$$\mathcal{J}_{DO} = 1 + \lambda e^{-\varepsilon/\tau} + \lambda^2 e^{-2\varepsilon/\tau}$$

$$\langle N \rangle_{DO} = \frac{1}{2} [0 + \lambda e^{-\varepsilon/\tau} + 2\lambda^2 e^{-2\varepsilon/\tau}].$$

(b) Occupancy: 0,0 0,1 1,0 1,1 
Energy: 0 
$$\varepsilon$$
  $\varepsilon$   $2\varepsilon$   $2\varepsilon$   $3=$  1 +  $2\lambda e^{-\varepsilon/\tau}$  +  $\lambda^2 e^{-2\varepsilon\tau}$  =  $[1+\lambda e^{-\varepsilon/\tau}]^2$ ; 
 $< N> = \frac{1}{3}[$  0 +  $2\lambda e^{-\varepsilon/\tau}$  +  $2\lambda^2 e^{-2\varepsilon/\tau}]$ .

The last line may be written

 = 
$$2\lambda e^{-\varepsilon/\tau} \frac{1+\lambda e^{-\varepsilon/\tau}}{[1+\lambda e^{-\varepsilon/\tau}]^2} = \frac{2}{\exp[(\varepsilon-\mu)/\tau]+1}$$
.

The Gibbs sum is square of a single-orbital Gibbs sum, as one would expect for two independent single-orbital systems. See Problem 3.9 for comparison. The occupancy is just twice the Fermi-Dirac distribution function.

## 6.6) ENTROPY OF MIXING

For a single gas of N atoms with the concentration n=N/V, from (34):

$$\sigma = N[\log(n_Q/n) + 5/2]$$

For two distinguishable independent gases, each state of gas A may be combined with each state of gas B to generate a distinct state of the combined system. If  $g_A$  and  $g_B$  are the numbers of states accessible to each gas,  $g=g_Ag_B$  is the number of states accessible to the two-gas system, and the two-gas entropy is the sum of the one-gas entropies:  $g=g_A+g_B$ .

This is true both before and after diffusive contact. Before contact, each of the two concentrations is  $n=n_1=N/V$ , hence

$$\sigma_{i} = N[\log(n_{QA}/n_{i}) + \log(n_{QB}/n_{i}) + 5] ,$$

where  $n_{QA}$  and  $n_{QB}$  will in general be different. After diffusive equilibrium,  $n=n_f=N/2V=n_1/2$ ; therefore

$$\sigma_{f} = N[\log(n_{QA}/n_{f}) + \log(n_{QB}/n_{f}) + 5]$$

$$= N[\log(2n_{QA}/n_{i}) + \log(2n_{QB}/n_{i}) + 5]$$

$$= \sigma_{i} + 2N \log 2 .$$

The additional term 2N log 2 may be understood as follows. For every A-atom orbital in volume  $V_A$  there is an equivalent A-atom orbital in  $V_B$ , and for every B-atom orbital in  $V_B$  there is an equivalent B-atom orbital in  $V_A$ . States in which these equivalent orbitals are occupied are initially inaccessible, but they become accessible upon diffusive contact. Every accessible state of the two-gas system with diffusive contact may be viewed as having been generated from one of the initially accessible states by interchanging the occupancies of an arbitrary number (from o to 2N) of equivalent orbital pairs. For 2N occupied

orbitals, there are  $2^{2N}$  distinguishable combinations of interchanges, leading to  $2^{2N}$  distinguishable accessible states for every initially accessible state. The entropy of mixing is the logarithm of this multiplicity of mixing:  $\sigma_{M} = \log 2^{2N} = 2N \log 2$ .

If the particles are indistinguishable, these occupancy interchanges do not lead to new distinguishable states but merely re-arrange the accessible states amongst each other. The two gases in diffusive contact are then simply a single gas of 2N atoms with the unchanged concentration  $\mathbf{n_f} = 2N/2V = \mathbf{n_i}$ , and with the entropy

$$\sigma_f = 2N[\log(n_Q/n_1) + 5/2]$$
.

For indistinguishable gases  $n_{QA} = n_{QB}$ ; therefore  $\sigma_f = \sigma_1$ .

Comment. Students often ask whether particles can be "nearly" identical. The answer is: no! Particles differ either by a finite discrete amount or not at all. For example, the different isotopes of an element differ in the (discrete) number of neutrons in their nuclei.

## 7.2) ENERGY OF RELATIVISTIC FERMI GAS

(a) With the help of (6) --  $n_F = (3N/\pi)^{1/3}$ , which remains valid -- we obtain, with  $n = N/L^3$ :

$$\varepsilon_{\rm F} = p_{\rm F}c = (M\pi c/L)n_{\rm F} = M\pi c(3n/\pi)^{1/3}$$

(b) With (9):

$$U(0) = 2\sum_{\epsilon} (n) = \pi \int_{0}^{n_{F}} n^{2} \epsilon(n) dn = \pi (M\pi c/L) \int_{0}^{n_{F}} n^{3} dn$$
$$= \pi (M\pi c/L) n_{F}^{4} / 4 = \pi \epsilon_{F} n_{F}^{3} / 4 = (3N/4) \epsilon_{F}.$$

## 7.10) RELATIVISTIC WHITE DWARF STARS

(a) We take over the result (88,89) for the energy of the extreme relativistic electron Fermi gas. If we write  $n = N/V = 3N/4\pi R^3$ , this result becomes

$$U_0 = \frac{3}{4} N \varepsilon_F = \frac{3}{4} N M \pi c (9N/4\pi^2 R^3)^{1/3}$$
$$= \frac{3}{4} \left(\frac{9\pi}{4}\right)^{1/3} \frac{M c N^{4/3}}{R} .$$

We equate this to the magnitude of the gravitational potential energy, assuming a uniform density:

$$\langle P.E. \rangle = \frac{3}{5} \frac{GM^2}{R} = \frac{3}{5} GM_H^2 \frac{N^{6/3}}{R}$$

(See Problems 4.1 and 7.6). The radius cancels out and we obtain the relation for N

$$N^{2/3} = \frac{5}{4} \left(\frac{9\pi}{4}\right)^{1/3} \frac{\text{Mc}}{\text{GM}_{H}^{2}} .$$

(b) N =  $8.4 \times 10^{57}$ . If the Sun were all hydrogen: N = M/M<sub>H</sub> =  $2 \times 10^{33}$  g/M<sub>H</sub> =  $1.3 \times 10^{57}$ .