

* A Cold Storage is to be maintained at -5°C while the surroundings are at 35°C . The heat leakage from the surroundings into the Cold Storage is estimated to be 20 kW . The actual C.O.P of the refrigeration plant is one third of an ideal plant working between the same temperatures. Find the power required to drive the plant [Page No - 105, Example: 2.4]...

* A vapour compression refrigerator uses methyl chloride (R-40) and operates between temperature limits of -10°C and 45°C . At entry to the compressor the refrigerant is dry saturated and after compression it acquires a temperature of 60°C . Find the C.O.P of the refrigerator. The relevant properties of methyl chloride are as follows [Page No - 119 Example - 4.6]

* In an ammonia vapour compression system the pressure in the evaporator is 2 bar . Ammonia at exit is 0.85 dry and at entry its dryness fraction is 0.19 . During compression the work done per kg of ammonia is 150 kJ . Calculate the C.O.P and the volume of vapour entering the compressor per minute if the rate of ammonia circulation is 4.5 kg/s . The latent heat and specific volume at 2 bar are 1325 kJ/kg and $0.18\text{ m}^3/\text{kg}$ respectively [Page - 113, ex: 4.1]

* An aircraft refrigeration plant has to handle a cabin load of 30 tonnes . The atmospheric temperature is 17°C . The atmosphere air is compressed to a pressure of 0.9 bar and temperature of 30°C due to ram action. This air is then further compressed in a compressor at 4.75 bar . Cooled in a heat exchanger to 67°C expanded in turbine to 1 bar pressure supplied to cabin. The air leaves the cabin at a temperature of 27°C . The isentropic efficiency of both compressor and turbine are 0.9 . Calculate the mass of air circulated per minute and C.O.P for air. $c_p = 1.004\text{ kJ/kg K}$ and $c_p/c_v = 1.4$ [Page - 74, ex: 3.2]