## FE | I | 19-12-18 Paper / Subject Code: 58604 / Applied Chemistry - I.

## Time: 2 Hours

N.B.



Marks: 60

- 1. Question number 1 is compulsory.
- 2. Attempt any three questions from Q.2 to Q.6.
- 3. Draw neat diagrams and write chemical equations where necessary.
- 4. Figures to right indicate full marks.

Atomic Weight: H=1, C=12, O=16, Ca=40, Na=23, Mg=24, S=32, Cl=35.5, N=14, A1=27, K=39

1.	Solve any five.	
(a)	Explain the principle of EDTA method.	3
(b)	What is glass transition temperature. Write its significance.	3
(c)	Write the significance of the following properties of lubricants: i) Emulsification ii) Cloud point iii) Fire point	3
(d)	What is RCC? What are the advantages of RCC over concrete?	3
(e)	Explain the reduced phase rule.	3
(f)	Distinguish between thermoplastic and thermosetting polymer.	3
(g)	20 ml sample of waste water was refluxed with 30 ml of potassium dichromate solution and after refluxing the excess unreacted dichromate required 11 ml of 0.1 N FAS solution. Blank of 20 ml of distilled water on refluxing with 30 ml of dichromate solution required 14 ml of 0.1 N FAS solution. Calculate the COD value of wastewater.	3
?. (a)	A sample of water contains following impurities:  Mg(HCO <sub>3</sub> ) <sub>2</sub> =73mg/lit, MgSO <sub>4</sub> = 120 mg/lit, CaCl <sub>2</sub> =222 mg/l and Ca (NO <sub>3</sub> ) <sub>2</sub> =164 mg/lit. The purity of lime is 74% and soda is 90%. Calculate the quantity of lime and soda needed for softening of 50,000 litres of water.	6
(b)	<ul><li>i) Write a brief note on polymers used in medical field.</li><li>ii) Name two additives added in blended oils. Give one example of each.</li></ul>	3 2
(c)	Explain with the help of chemical reactions "Setting and Hardening" of cement.	4
3.(a)	What is fabrication of plastic? Explain injection moulding process with a neat diagram.	6
<b>(b)</b>	i)Discuss the advantages and limitations of phase rule. ii)Differentiate between SWNT and MWNT	3 2
	지수도 수 있는 것이다. 전 10 전 1	

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(c)	A zeolite softener was completely exhausted and was regenerated by passing 1000 litres of NaCl solution, containing 100mg/lit of NaCl. How many litres of a sample water of hardness 500ppm can be softened by this softener?	4
4. (a)	Draw the diagram for demineralization process and write suitable reactions involved in the process. What are the advantages and disadvantages of the method.	6
(b)	i) Find the acid value of the given oil whose 20ml required 2.8ml of N/10 KOH during titration. (Density of oil = 0.86g/ml)	
	ii)Write a short note on decay of concrete.	2
(c)	Natural rubber requires vulcanization. Give reasons. With appropriate reactions explain how the drawbacks are overcome?	4
5. (a)	Write preparation, properties and uses of following polymers: (Any two)	6
	i) Kevlar ii) Silicone rubber iii) Buna S	
(b)	i) Explain Activated sludge method with the help of diagram.	3
	ii) What is grease? What are the conditions in which greases are used?	2
(c)	Draw the phase diagram of one component system and find out the number of degree of freedom along the curves and areas.	4
6. (a)	What are lubricants? Define Lubrication. Explain Hydrodynamic lubrication mechanism with neat diagram.	6
(b)	i) Define a) Phase b) Component c) Degree of freedom	3
	ii) Write a short note on Reverse Osmosis.	2
(c)	Explain laser ablation method for production of CNTs.	4