

**Time: 2 hours      Activity Sheet – March 2020      Marks: 40**

- Note:**
- (i) All questions are compulsory.
  - (ii) Use of calculator is not allowed.
  - (iii) The numbers to the right of the questions indicate full marks.
  - (iv) In case of MCQs Q.1. (A) only the first attempt will be evaluated and will be given credit.
  - (v) For every MCQ, the correct alternative a, b, c, or d with sub-question number is to be written as an answer.  
For e.g. i. a, ii. b, iii. c.
  - (vi) Scientifically correct, labelled diagrams should be drawn wherever necessary.

**Q.1. (A) Choose the correct alternative. [5]**

- i. According to Mendeleev's periodic law, properties of elements are periodic function of their \_\_\_\_\_.
  - a. atomic numbers
  - b. atomic masses
  - c. densities
  - d. boiling points
- ii. The vapour content in the air is measured using a physical quantity called \_\_\_\_\_.
  - a. absolute humidity
  - b. relative humidity
  - c. dew point
  - d. humidity
- iii. For the normal human eye, the near point is at \_\_\_\_\_ cm.
  - a. 10
  - b. 20
  - c. 25
  - d. 30
- iv. The astronomical object closest to us in our galaxy is \_\_\_\_\_.
  - a. Mars
  - b. Venus
  - c. Jupiter
  - d. Moon
- v. In the Wilfley table method, the particles of gangue are separated by \_\_\_\_\_ separation method.
  - a. magnetic
  - b. froth floatation
  - c. hydraulic
  - d. gravitational

**(B) Answer the following.**

**[5]**

i. **Find the odd one out.**

Voltmeter, Ammeter, Thermometer, Galvanometer

ii. **Find the correlation and write in one sentence.**

Alkene :  $C = C$  : : Alkyne : \_\_\_\_\_

iii. **State true or false.**

The frequency of AC is 50 Hz.

iv. **Make pairs.**

Column A	Column B
1. The wavelength of red light	a. 600 nm
	b. 700 nm
	c. 500 nm

v. Name the first artificial satellite sent by Russia in space.

**Q.2. (A) Give scientific reasons. (Any two)**

**[4]**

- The weight of an object changes from place to place though its mass is constant.
- Stars twinkle but we do not see the twinkling of planets.
- Elements belonging to the same group have the same valency.

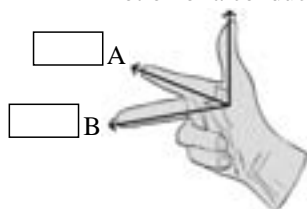
**(B) Answer the following. (Any three)**

**[6]**

i. How much heat energy is necessary to raise the temperature of 5 kg of water from  $20^{\circ}\text{C}$  to  $100^{\circ}\text{C}$ ?

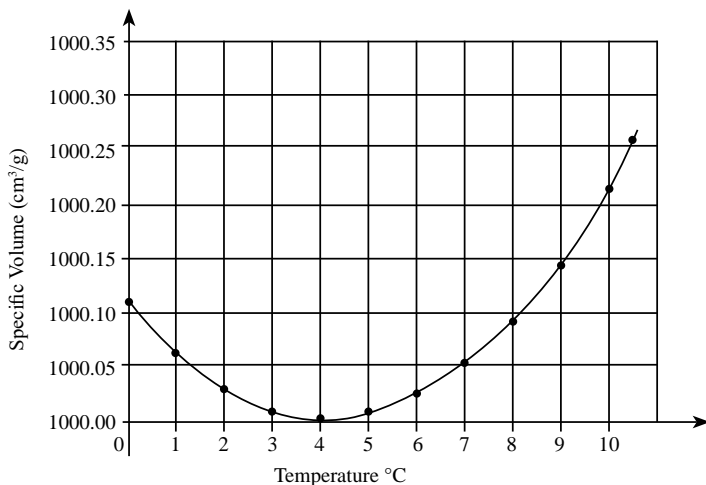
ii. Observe the given figure of Fleming's Right Hand Rule and write the labels of A and B correctly.

Motion of a conductor

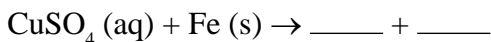


iii. **Observe the given graph and answer the following questions.**

- Name the process represented in the figure.
- At what temperature does this process take place?



iv. Complete the given chemical reaction.

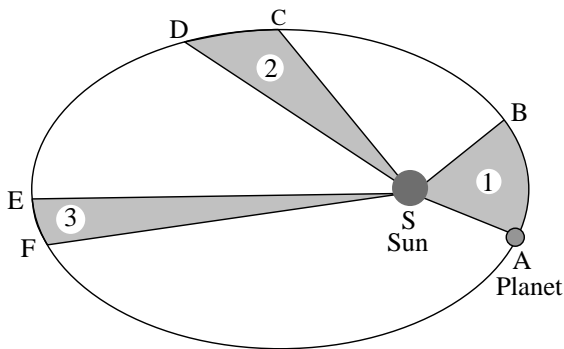


Name the type of reaction.

v. Write a short note on alloying.

**Q.3. Answer the following questions. (Any five) [15]**

- i. An element has its electronic configuration as 2, 8, 2. Now, answer the following questions.
  - a. What is the atomic number of this element?
  - b. What is the group of this element?
  - c. To which period does this element belong?
- ii. Observe the given figure showing the orbit of a planet moving around the Sun and write the three laws related to it.



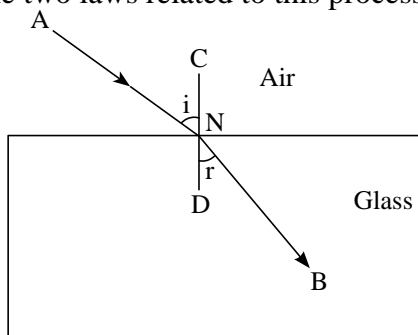
**iii. Read the given passage and answer the following questions.**

The home electrical connection consists of 'live', 'neutral' and 'earth' wires. The 'live' and the 'neutral' wires have potential difference of 220 V. The 'earth' is connected to the ground. Due to a fault in equipment, or if the plastic coating on the 'live' and the 'neutral' wires gives away, the two wires come in contact with each other and a large current flows through it producing heat. If any inflammable material (such as wood, cloth, plastic, etc.) exists around that place, then it can catch fire. Therefore, a fuse wire is used as a precautionary measure.

- Name the two wires having potential difference of 220 V.
- What is short circuit?
- Write the function of a fuse.

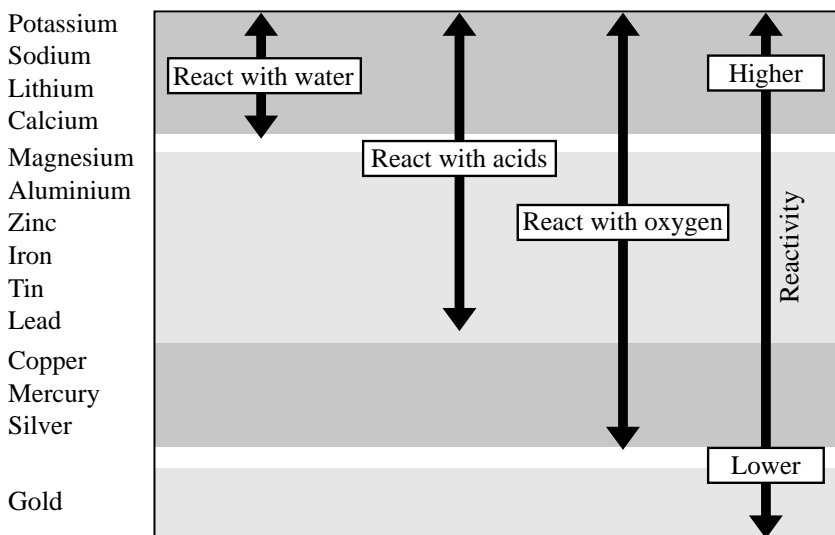
**iv. Observe the given figure and answer the following questions.**

- Name the process represented in the figure.
- State the two laws related to this process.



- v. What is an artificial satellite? Name any two types of artificial satellites and state their functions.**
- vi. Answer the following questions.**
  - Define hydrocarbons.
  - Name the types of hydrocarbons.
  - Name two carbon compounds used in day-to-day life.

vii. **Observe the given figure of reactivity series of metals and answer the following questions.**



**Reactivity series of metals**

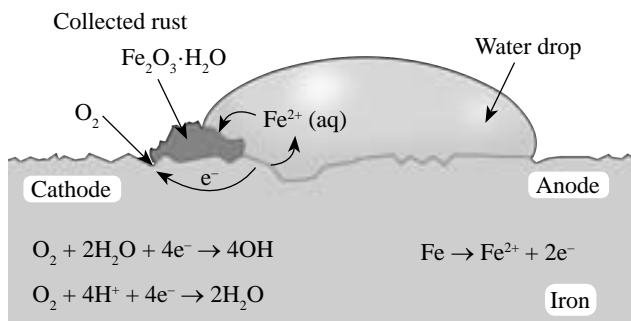
- Name two metals which react with water.
- Name two moderately reactive metals.
- Name the most highly reactive metal and the least reactive metal.

viii. **Complete the following table.**

Straight chain of carbon compounds	Structural formula	Molecular formula	Name
C	$  \begin{array}{c}  \text{H} \\    \\  \text{H} - \text{C} - \text{H} \\    \\  \text{H}  \end{array}  $	CH <sub>4</sub>	Methane
C – C			Ethane
C – C – C		C <sub>3</sub> H <sub>8</sub>	
C – C – C – C	$  \begin{array}{cccc}  \text{H} & \text{H} & \text{H} & \text{H} \\    &   &   &   \\  \text{H} - \text{C} & - \text{C} & - \text{C} & - \text{C} - \text{H} \\    &   &   &   \\  \text{H} & \text{H} & \text{H} & \text{H}  \end{array}  $		

**Q.4. Answer any one of the following.****[5]**

- i. Draw a scientifically correct labelled diagram of a human eye and answer the questions based on it.
- Name the type of lens in the human eye.
  - Name the screen at which the maximum amount of incident light is refracted.
  - State the nature of the image formed of the object on the screen inside the eye.
- ii. Observe the following picture and answer the questions.
- What is rust?
  - Write the chemical formula of rust.
  - Write the reaction of oxidation of iron at anode.
  - Write the reaction of oxidation of iron at cathode.
  - What is corrosion?



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