

Time: 2 hours Activity Sheet – December 2020 Marks: 40

- v. Lithium (Li), _____ and Potassium (K) is Dobereiner's triad.
- a. Magnesium (Mg) b. Aluminium (Al)
c. Sodium (Na) d. Calcium (Ca)

(B) Solve the following subquestions. [5]

- i. **State true or false.**

The refractive index depends upon the velocity of light in medium.

- ii. **Write the correlated answer.**

Torch : Concave lens :: Camera : _____

- iii. **Find the odd one out.**

Zinc, Iron, Phosphorus, Sodium

- iv. Draw the structural formula of C_3H_8 .

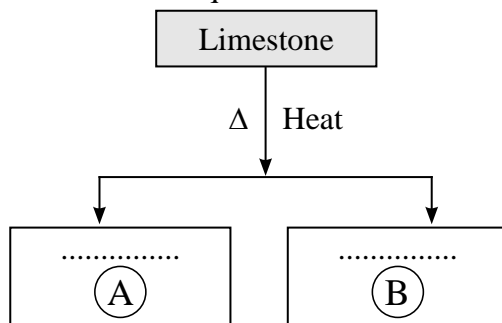
- v. Which satellite is used in educational field among INSAT and GSAT series?

Q.2. (A) Give scientific reasons. (Any two) [4]

- i. Stars appear to be twinkling at night.
ii. Simple microscope is used for watch repairs.
iii. Copper vessels turn greenish and silver articles turn blackish when kept open in air for a long time.

(B) Answer the following questions. (Any three) [6]

- i. An object takes 5 seconds to reach the ground from a height of 5 m on a planet. What is the value of 'g' on that planet?
ii. Identify 'A' & 'B' from the following table and complete the table. Write the chemical equation.



- iii. Write the modern periodic law and also give the names of ‘blocks’ in the modern periodic table.
- iv. Distinguish between ‘alternating current’ and ‘direct current’.
- v. Define specific heat capacity. Write its S.I. unit.

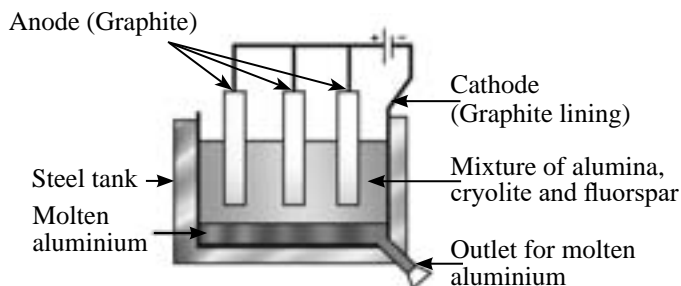
Q.3. Answer the following. (Any five) [15]

- i. An iron ball of mass 3 kg is released from a height of 125 m and falls freely to the ground. Assuming that the value of ‘g’ is 10 m/s^2 , calculate:
 - a. time taken by the ball to reach the ground
 - b. velocity of the ball on reaching the ground.
- ii. An element has its electron configuration as (2, 8, 2). Answer the following.
 - a. What is the ‘atomic number’ of this element?
 - b. What is the ‘group’ of this element?
 - c. To which period does this element belong?
- iii. a. Write the ‘endothermic’ or ‘exothermic’ nature of the reaction.
$$2\text{KClO}_3(\text{s}) \xrightarrow{\Delta} 2\text{KCl}(\text{s}) + 3\text{O}_2 \uparrow$$
 - b. Balance the given chemical equation.
$$\text{NaOH}(\text{aq}) + \text{H}_2\text{SO}_4(\text{aq}) \rightarrow \text{Na}_2\text{SO}_4(\text{aq}) + \text{H}_2\text{O}(\text{l})$$
 - c. From the given reaction, identify ‘oxidant’ and ‘reductant’.
$$\text{CuO} + \text{H}_2 \rightarrow \text{Cu} + \text{H}_2\text{O}$$
- iv. A copper sphere of 100 g mass is heated to raise its temperature to 100°C and is released in water of mass 195 g and temperature 20°C in a copper calorimeter. If the mass of the calorimeter is 50 g, what will be the maximum temperature of water? (Specific heat of copper = $0.1 \text{ cal/g}^\circ\text{C}$)
- v. a. Draw a neat labelled diagram of ‘dispersion of white light through glass prism’.
 - b. Which coloured ray is the least deviated?
 - c. Which coloured ray is the most deviated?

vi. Complete the following table for convex lens.

	Position of object	Position of image	Size of image	Nature of image
(a)	At focus F_2	Point image	Real and inverted
(b)	At $2F_1$	At $2F_2$	Real and inverted
(c)	Between F_1 & O (within focal length)	On the same side (object side)	Very large

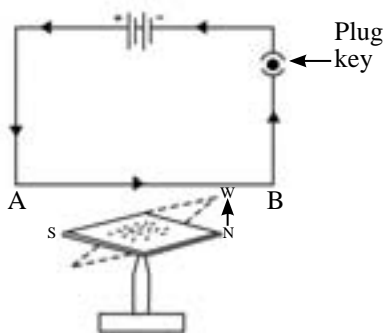
vii. Observe the following diagram and answer the questions.



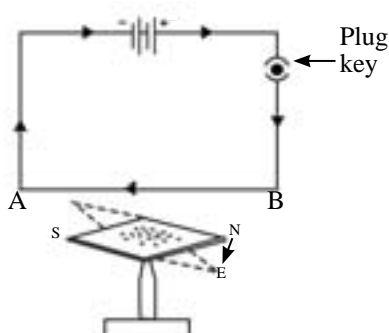
- Write the 'anode reaction'.
 - Write the 'cathode reaction'.
 - What is the purpose of mixing 'cryolite' and 'fluorspar' with 'alumina' in the electrolytic reduction of alumina?
- viii.
 - What is the principle behind the working of a satellite launch vehicle?
 - Write the formula for 'escape velocity'.
 - Write the long form of 'ISRO'.

Q.4. Solve the following questions. (Any one)**[5]**

i. Observe the diagrams and answer the questions.



Magnetic needle



Magnetic needle

- a. Which effect of electric current is shown in the above figures?
 - b. What will happen if the number of electric cells is increased on the magnetic needle?
 - c. If the distance between the conductor and magnetic needle is increased, what will be the effect on the intensity of the magnetic field?
 - d. If the ends of electric cell are interchanged, what will be the effect on the magnetic needle?
 - e. Write the names of any *two* instruments which work on magnetic effect of electric current.
- ii. Answer the following.
- a. Draw the electron-dot structure of Methane.
 - b. Define Homologous series.
 - c. Write the IUPAC names of the following.
 - i. $\text{CH}_3\text{--CH}_2\text{--COOH}$
 - ii. $\text{CH}_3\text{--CHOH--CH}_3$
 - iii. $\text{CH}_3\text{--CO--CH}_2\text{--CH}_3$