Time	e: 2 hours Acti	vity Sheet –	Ma	rch 202	23]	Marks	: 40
Q.1. (i)	(A) Choose the correction The device used				currer	ıt is	s cal	[5]
	(a) a voltmeter (c) a galvanometer		` ′	an an				
(ii)	If a ray of light passes from a denser medium to a rarer medium in a straight line, the angle of incidence must be							
	(a) 0° (b) 3	0°	(c)	60°		(d)	90°	
(iii)	The power of con-	vex lens	of	focal	lengt	th 20	0 cm	is
	(a) +5.0 D (b) 0	0.20 D	(c)	–5.0 I)	(d)	0.5 D)
(iv)	Good conductor of el	ectricity is						
` /	(a) Bromine (b) I					(d)	Sulph	nur
(v)	The height of medium earth orbit above the surface of the earth is:							
	(a) 1,500 km	((b)	250 k	m			
	(c) 45,000 km	((d)	25,00	0 km			
(B) (i)	Answer the following questions. [5 Find the odd one out: Loudspeaker, Microphone, Electric motor, Magnet							[5]
(ii)	Complete the co-relation: CuI ₂ : Brown:: AgCl:							
(iii)	Match the pair.							
	Group 'A'	Gr	Group 'B'					
	Substance	Refra	Refractive index		X			
	Air	(a) 1.33	3					

(b) 1.46 (c) 1.0003

- (iv) State true or false.
 - Wavelength of red light is close to 700 nm.
- (v) Write the name of small satellite made by a group of students from COEP (College of Engineering, Pune) sent to the space through ISRO in 2016.

Q.2. (A) Give scientific reasons. (Any two) [4]

- (i) For electric power transmission, copper or aluminium wire is used.
- (ii) Lemon or tamarind is used for cleaning copper vessels turned greenish.
- (iii) Elements belonging to the same group have the same valency.

(B) Answer the following questions. (Any three) [6]

- (i) How do we feel about air in each of the following conditions?(a) Relative humidity is more than 60%.
 - (b) Relative humidity is less than 60%.
- (ii) Complete the following reaction: $C_{12}H_{22}O_{11} \xrightarrow{\text{heat}} \dots + \dots + \dots$
- (iii) Distinguish between Mass and Weight.
- (iv) Complete the following table:

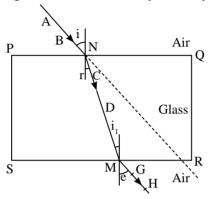
Type of Satellite	The names of Indian Satellite and launcher		
(a) Navigational	Satellite:		
Satellite	Launcher:		
(b) Earth observation	Satellite:		
Satellite	Launcher:		

(v) Define periods and groups of modern periodic table.

Q.3. Answer the following questions. (Any five) [15]

(i) Calculate the escape velocity on the surface of the moon given the mass and radius of the moon to be 7.34×10^{22} kg and 1.74×10^6 m respectively. (Given: $G=6.67\times10^{-11}$ Nm²/kg²).

- (ii) An element has its electron configuration as 2, 8, 1. Now answer the following questions:
 - (a) What is the atomic number of this element?
 - (b) What is the group of this element?
 - (c) To which period does the element belong?
- (iii) Observe the figure and name the ray AB, ray CD, ray GH:



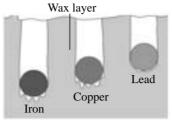
(iv) Read the following sentence and answer the questions:

NaCl is an ionic compound.

- (a) Why is NaCl an ionic compound?
- (b) State any two properties of ionic compounds.
- (v) Identify the physical and chemical changes from the following phenomena:
 - (a) Transformation of ice into water
 - (b) Ripening of fruit
 - (c) Milk turned into curd
 - (d) Evaporation of water
 - (e) Digestion of food in the stomach
 - (f) Iron fillings get attracted towards the magnet

[5]

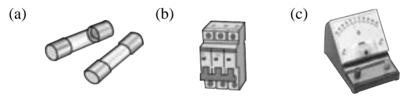
(vi) Observe the following figure and answer the questions.



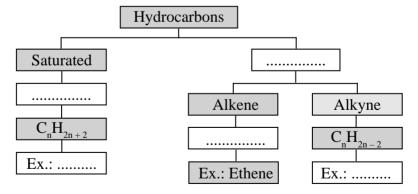
Specific heat capacity of metals

- (a) Which element has maximum specific heat capacity? Justify.
- (b) Which element has minimum specific heat capacity? Justify.
- (c) Define specific heat of object.

(vii) Identify figures A, B, C and give their uses:



(viii) Complete the following flowchart:



Q.4. Answer any *one* of the following questions.

(i) Observe the figure and answer the following questions:



- (a) Name the defect of vision represented in above figure.
- (b) State the reasons for this defect.
- (c) How is it corrected?
- (d) Draw the diagram to show the correction of this defect.

(ii) Complete the following table:

S.N.	Common Name	Structural Formula	IUPAC Name	
(a)	Ethylene	$CH_2 = CH_2$	•••••	
(b)	Acetylene		Ethyne	
(c)	Acetic acid	CH ₃ -COOH		
(d)	Methyl alcohol		Methanol	
(e)		CH ₃ -CO-CH ₃	Propane-2-one	

