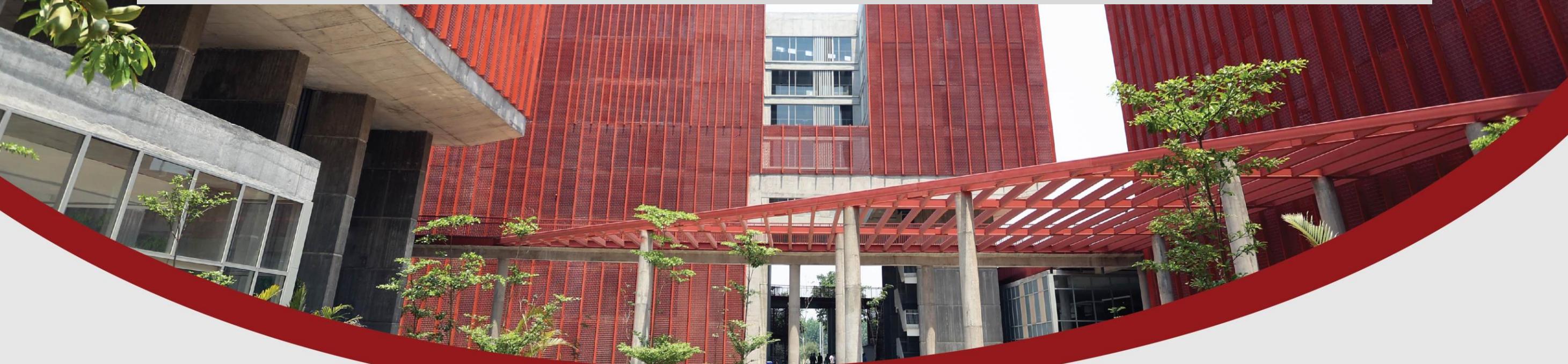


Classification of Materials



Thapar Institute of Engineering & Technology
(Deemed to be University)

Bhadson Road, Patiala, Punjab, Pin-147004

Contact No. : +91-175-2393201

Email : info@thapar.edu



THAPAR INSTITUTE
OF ENGINEERING & TECHNOLOGY
(Deemed to be University)

Syllabus

Structure of solids: Classification of engineering materials, Structure-property relationship in engineering materials, Crystalline and non crystalline materials, Miller Indices, Crystal planes and directions, Determination of crystal structure using X-rays, Inorganic solids, Silicate structures and their applications. Defects; Point, line and surface defects.

CLO: Classify engineering materials based on its structure.

CLO: Draw crystallographic planes and directions.

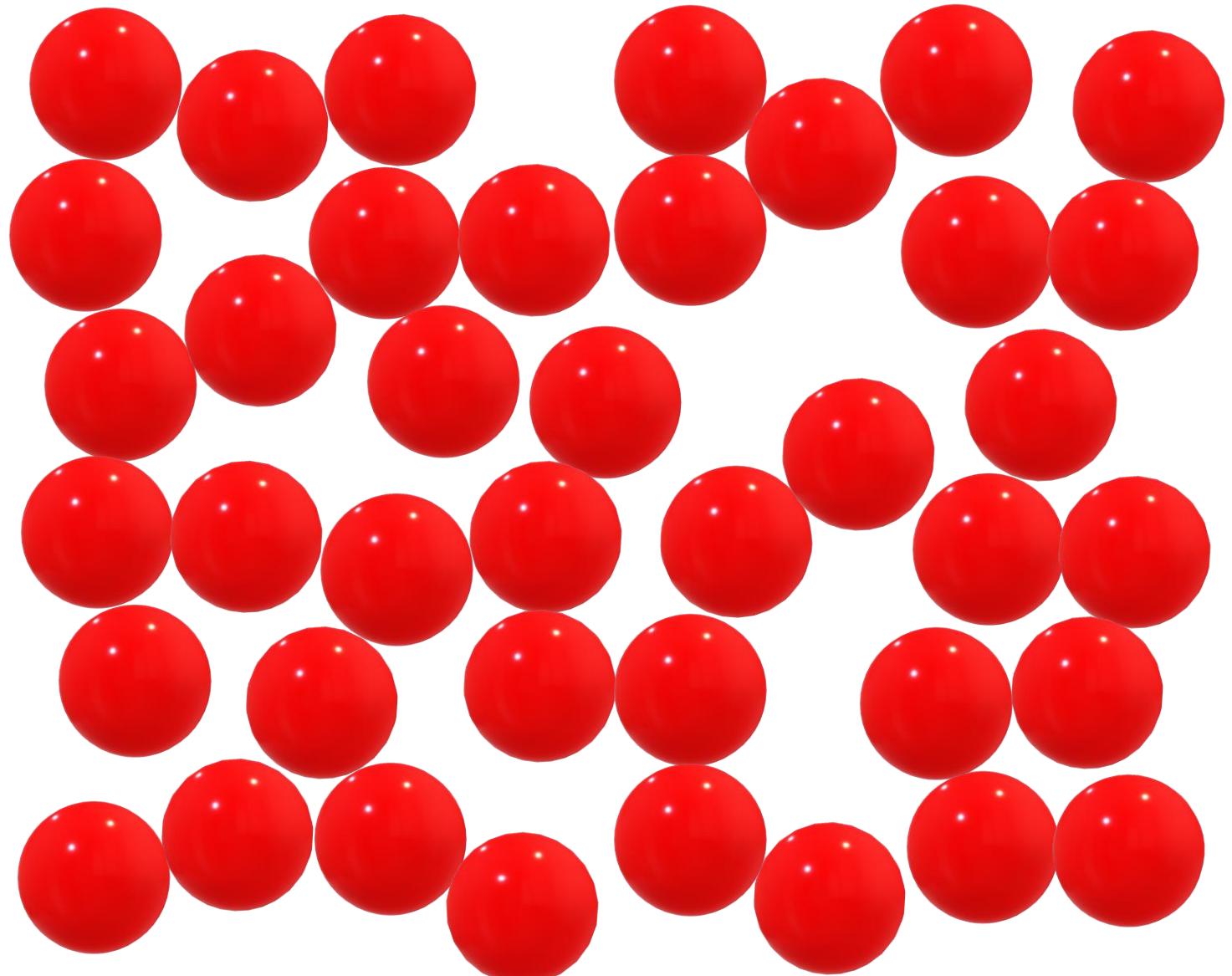
Metals
and
Alloys

Ceramics
and
Glasses

Polymers

Amorphous Solid

Particles



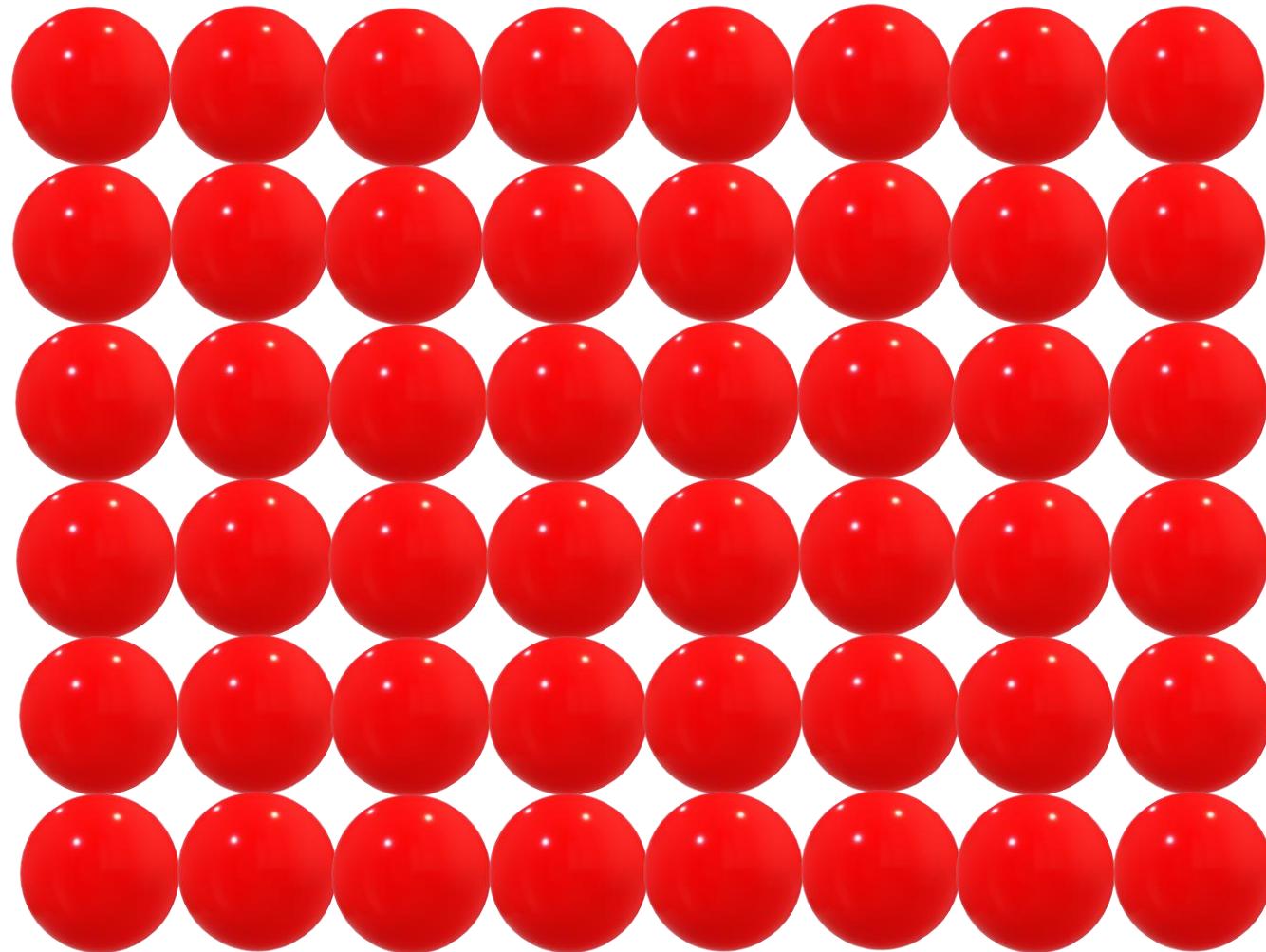
Example: Cotton candy



Generally glasses are amorphous

Crystalline Solid

Particles



Example: Table Salt



Generally metals are crystalline

Properties

Metals

- Ductile
- Opaque
- Good mechanical strength
- Good conductor of heat and electricity

Ceramics

- Hard and Brittle
- Glasses are transparent/translucent
- High Melting point
- Good mechanical strength
- Bad conductor of heat and electricity

Polymers

- Ductile
- Poor tolerance to heat
- Non corrosive
- Light weight
- Low mechanical strength
- Bad conductor of heat and electricity

Metals



Metals: Cu, Ni, Fe, Au, Si, Al, Brass (Cu-Zn alloy)

Ceramics

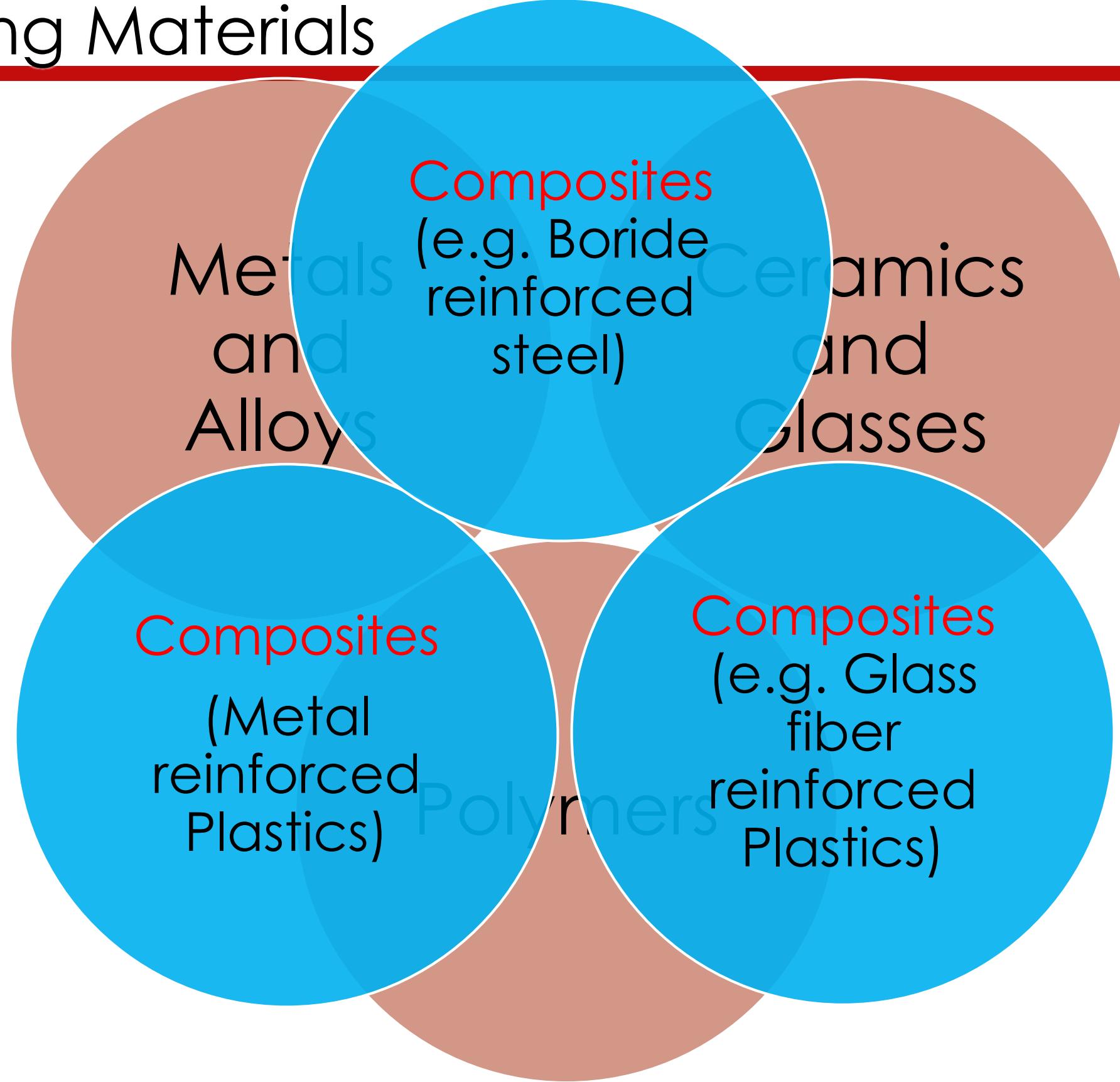


Ceramics: Usually oxides, nitrides, carbides) Alumina (Al_2O_3), Zirconia (Zr_2O_3)

Polymers



Polymers: Thermoplasts, Elastomers, Polythene, Polyvinyl Chloride (PVC), Polypropylene, Nylon



Remember

- Alloys have two or more than two metals. At least one of the elements should be a metal.
- Composites have two different material types.

Property is the response to the stimulus

Mechanical

- Tensile strength
- Hardness
- Ductility
- Brittleness

Electrical

- Resistivity
- Electrical Conductivity

Thermal

- Thermal conductivity
- Specific Heat

Physical

- Density
- Porosity

Optical

- Refraction
- Reflection
- Absorption
- Transmission

Magnetic

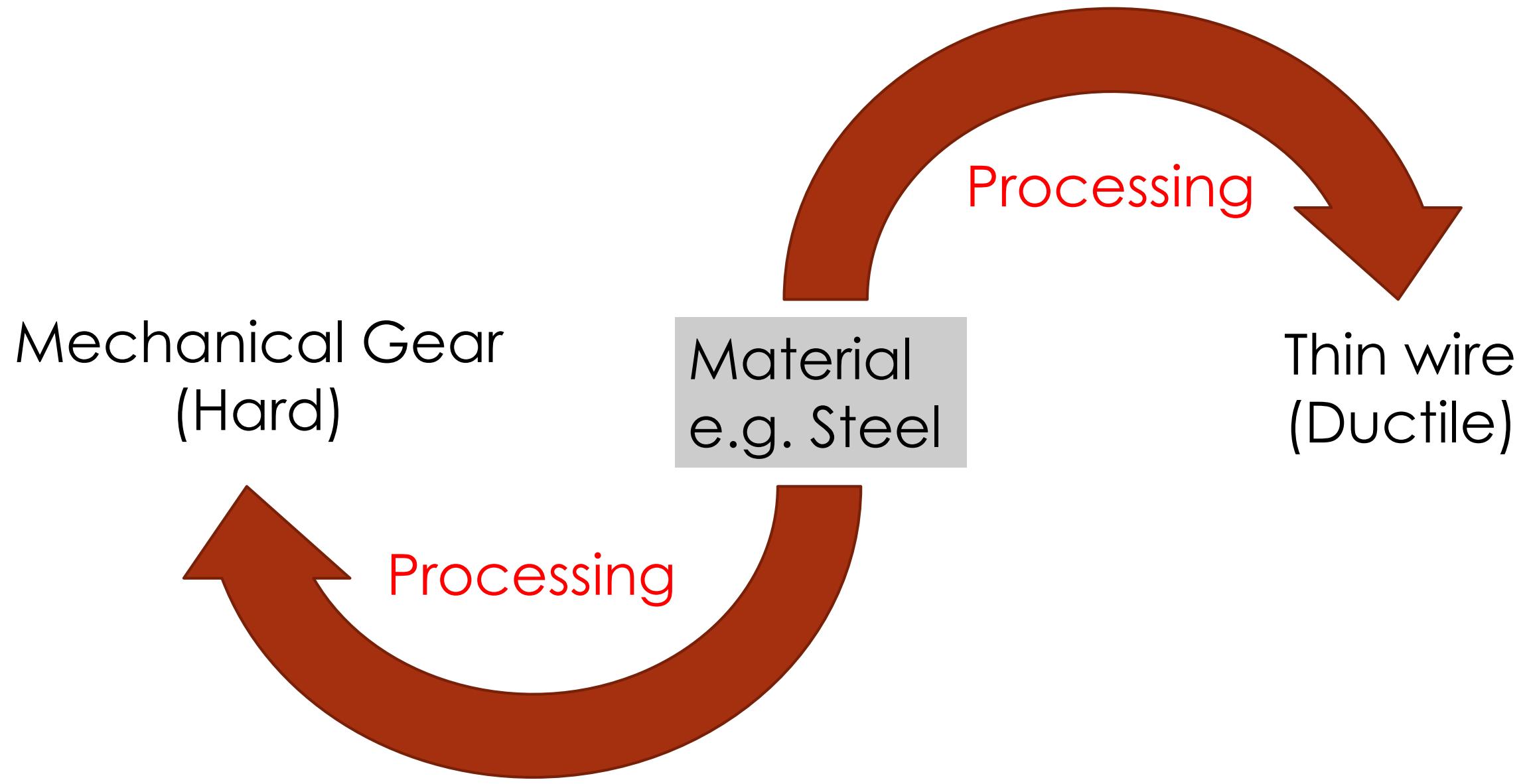
- Ferromagnetism
- Diamagnetism
- paramagnetism

Dielectric

- Polarizability
- Dielectric constant
- Ferroelectric
- Piezoelectric

Materials design approach

- Processing of material is very important to get desired properties.
- Properties are depend on the structure of the material.



Summary

1. Alloys have two or more than two metals. At least one of the elements should be a metal.
2. Composites have two different material types.
3. Response to the stimulus is called as properties.
4. Properties need specific structure for optimum performance.