

1   **Expression and engineering of unexplored PET degrading enzymes from *Microbispora*,**

2   *Nonomuraea, Micromonospora* genus

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23 **KEYWORDS**

24 PETase, PET degradation, *Microbispora*, *Nonomuraea*, *Micromonospora*, enzymes

25

26 **Abbreviations:**

27 lcPET – Low crystallinity poly(ethylene terephthalate)

28 hcPET – High crystallinity poly(ethylene terephthalate)

29 PET – Poly(ethylene terephthalate)

30 TPA – Terephthalic acid

31 BHET – Bis(2-hydroxyethyl) terephthalic acid

32 MHET – Mono(2-hydroxyethyl) terephthalic acid

33 LCC – Leaf-branch compost cutinase

34 IsPETase – *Ideonella sakaiensis* PETase

35 SUMO – Small Ubiquitin-like Modifier

36 MBP – Maltose Binding Protein

37 **ABSTRACT**

38 Low recycling rates have resulted in the alarming rate of accumulation of a widely used  
39 plastic material, polyethylene terephthalate (PET). With the build-up of plastics in our  
40 environment, there is an urgent need to source for more sustainable solutions to process them.  
41 Biological methods such as enzyme-catalyzed PET recycling or bioprocessing are seen as a  
42 potential solution to this problem. Actinobacteria, known for producing enzymes involved in  
43 the degradation of complex organic molecules, are of particular interest due to their potential  
44 to produce PET degrading enzymes. The highly thermostable enzyme, leaf-branch compost  
45 cutinase (LCC) found in Actinobacteria is one such example. This work expands on the  
46 discovery and characterization of new PET degrading enzymes from *Microbispora*,  
47 *Nonomuraea*, and *Micromonospora* genus. Within this genus, we analyzed enzymes from the  
48 polyesterase-lipase-cutinase family, which have ~60% similarity to LCC, where one of the  
49 enzymes was found to be capable of breaking down PET and BHET at 45-50 °C. Moreover,  
50 we were able to enhance the enzyme's depolymerization rate through further engineering,  
51 resulting in a two-fold increase in activity.

52

53 **IMPORTANCE** The proliferation of PET plastic waste poses a significant threat to human  
54 and environmental health, making it an issue of increasing concern. In response to this  
55 challenge, scientists are investigating eco-friendly approaches, such as bioprocessing and  
56 microbial factories, to sustainably manage the growing quantity of plastic waste in our  
57 ecosystem. Despite the existence of enzymes capable of degrading PET, their scarcity in  
58 nature limits their applicability. The objective of this study is to enhance our understanding of  
59 this group of enzymes by identifying and characterizing novel ones that can facilitate the  
60 breakdown of PET waste. This data will expand the enzymatic repertoire and provide  
61 valuable insights into the prerequisites for successful PET degradation.

## 62 INTRODUCTION

63 Polyethylene terephthalate (PET) is amongst the most widely manufactured and utilized  
64 plastic material for consumer and industrial applications due to its favorable physiochemical  
65 properties and durability (1). With an estimated production of at least 1 million PET bottles  
66 every minute (2) and a global recycling rate of 9% (3), the rapid and large accumulation of  
67 non-biodegradable post-consumer PET waste either end up in landfills, terrestrial or aquatic  
68 environment. These PET that end up in the terrestrial or aquatic environment not only have a  
69 slow rate of decomposition, but can also subsequently enter the marine ecosystem as  
70 microplastics - posing risks to the environment and the health of both animals and humans (4,  
71 5). Consequently, addressing the PET waste crisis is imperative for both environmental and  
72 health sustainability. Traditional solid waste treatment methods, such as landfill and  
73 incineration, have limitations in terms of secondary pollution and limited land resources. The  
74 main thermomechanical and chemical recycling methods are energy-intensive, requiring high  
75 temperatures, and can result in alterations to the properties of the PET plastic (6). Thus, there  
76 is a growing interest in less energy-intensive or "natural" methods for processing this  
77 material.

78

79 The discovery of the first PET degrading enzyme PETase from *Ideonella sakaiensis*  
80 (IsPETase) (7, 8) in 2016, gave hopes to a biological solution through enzymes and microbial  
81 factories for processing this inert plastic. Enzyme-catalyzed PET recycling or bioprocessing  
82 can proceed under mild reaction conditions, with minimal energy and chemical usage. This  
83 method of recycling is thus a more environmentally responsible alternative to petroleum-  
84 derived production processes. To improve PETase towards practical application, there has  
85 been numerous studies reporting more thermostable and active versions of the enzyme (9).  
86 Since then, various PET degradation enzymes have also been identified, including CALB

87 lipase from *Candida antarctica* (10), cutinases from *Fusarium solani*, *Humicola insolens*, and  
88 *Thermobifida fusca* (11). According to the Plastics-Active Enzymes Database (PAZy), there  
89 are ~3000 homologs of PET-active enzymes by profile hidden Markov models (12).

90

91 Actinobacteria is a diverse group of bacteria known to produce a wide range of enzymes that  
92 are utilized in various industrial and medical applications, such as antibiotic production and  
93 degradation of environmental contaminants (13, 14). In particular, the actinobacteria are of  
94 interest due to their ability to produce cutinases and hydrolases, enzymes involved in the  
95 degradation of complex organic molecules and synthesis of essential nutrients (14). One of  
96 the most outstanding and highly thermostable PET degrading enzymes found in the  
97 actinobacteria is the leaf-branch compost cutinase (LCC), which was mined from  
98 metagenomes (15). A recent report by Erickson et.al on another PET degrading enzyme  
99 within the actinobacteria family, further underscores the potential of actinobacteria as a  
100 source for PET depolymerization. Given the capabilities of actinobacteria in PET  
101 depolymerization, this study aim to explore other actinobacteria for new PET degrading  
102 enzymes (16).

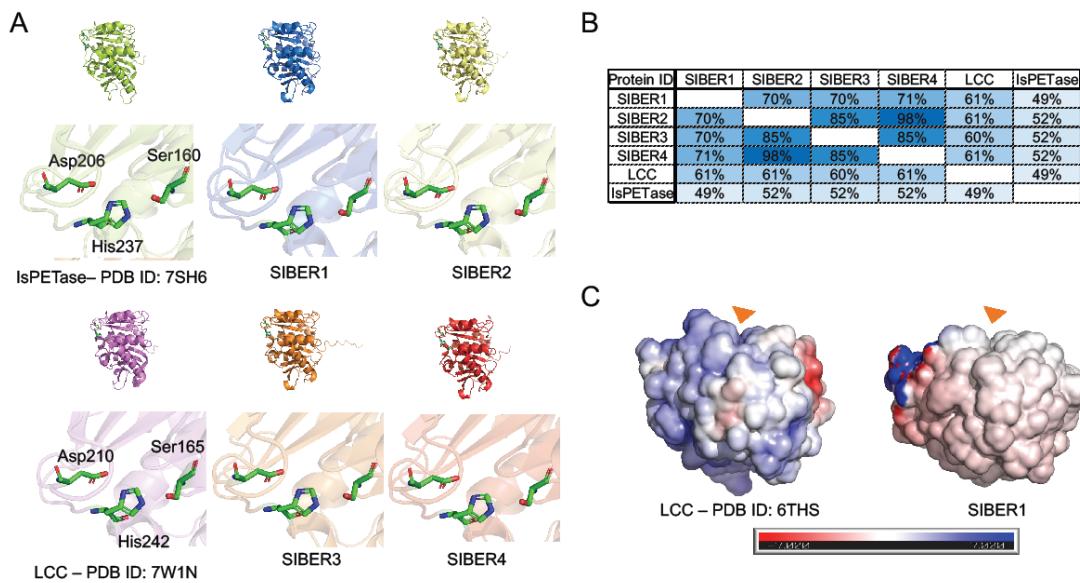
103

104 In this work, we expand the enzymes for PET breakdown from actinobacteria through the  
105 discovery, engineering, and characterization of new and unexplored PET degrading enzymes  
106 from *Microbispora*, *Nonomuraea*, *Micromonospora* genus.

107 **RESULTS**

108 **Genomic mining and *in silico* analysis of distant cutinase homologs**

109 With reference to the reported LCC, potential hydrolases were obtained from the genera  
110 *Microbispora*, *Nonomuraea*, and *Micromonospora* through mining actinobacterial genomic  
111 sequences in public databases such as Genbank and Natural Products Discovery Center  
112 (NPDC) (17), as well as the Singapore-based Natural Organism Library (NOL) (18). Within  
113 these hits, we selected four representative sequences from NOL. Sequence alignment analysis  
114 between the four cutinases indicate sequence similarity (~60-61 % homology to LCC, Fig.  
115 1B), including the presence of the three active site triad residues for PET breakdown (Asp,  
116 Ser and His, Fig. 1A). Cysteine residues for disulfide bridge formation were also observed  
117 (Fig. S1). Search within the ESTHER database (19) also shows high similarity hits of these  
118 enzymes as part of the polyesterase-lipase-cutinase family; similar to many previously  
119 reported PET hydrolases. (Fig. S2). Absence of an extended loop and extra disulfide bond  
120 near the active site also classify these enzymes to be Type I (LCC) rather than Type II  
121 (IsPETase) (20). Structural alignment based on *in silico* AlphaFold models also proposed that  
122 their general structure is closely aligned to that of the LCC, including positions of the active  
123 site residues (Table S1, Fig. S1). Although the core active sites and general structures are  
124 conserved, electrostatic prediction of the surface indicated sequence divergence are mainly  
125 surface residues. This observation is consistent to conclusions drawn from an extensive study  
126 on PET breakdown enzymes (16).



127

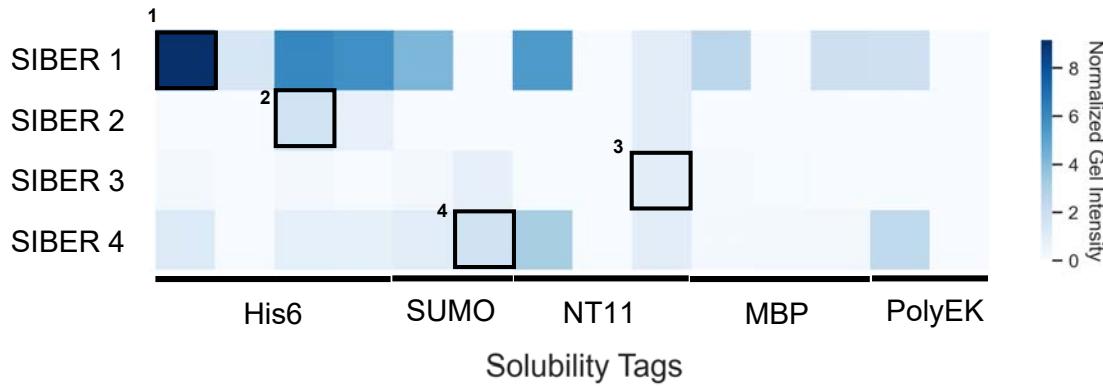
128 **Figure 1. Genomic mining and *in silico* analysis of distant cutinase homologs.** (A)  
129 Structural alignment of LCC and IsPETase against AlphaFold models of SIBER 1-4. Active  
130 sites residues are shown in inset below. RSMD of the *in-silico* models against 6THS range  
131 from 0.54 to 0.68 Å (Table S1). (B) Similarity scores between SIBER 1-4, LCC and  
132 IsPETase using BLASTP. (C) Electrostatic representative of solvent accessible surfaces of  
133 LCC and SIBER1 at pH7 ( $\pm 7$  kT/e, APBS-PDB2PQR (21))

134

### 135 Optimizing heterologous expression

136 To investigate if the shortlisted sequences encode functional enzymes, there is a need to  
137 identify a suitable expression system. For heterologous expression of these enzymes in  
138 *Escherichia Coli* (*E. coli*), optimization was first performed for the protein expression  
139 constructs of SIBER 1-4. These were screened with various combinations of solubility tags  
140 for optimal expression (Fig. 2). Solubility aids which involved the Small Ubiquitin-like  
141 Modifier (SUMO (22)) and a 11 amino acid tag, NT11 (23), and Maltose Binding Protein  
142 (MBP) were investigated. Our optimization data showed varied expression levels across  
143 enzymes and tag combinations (Fig. 2). Overall, SIBER 1 constructs had significantly higher

144 expression levels. The most productive constructs obtained from the distinct sequences were  
145 further scaled up and the resulting proteins were purified for characterization (SIBER 1-  
146 SIBER 171, SIBER 2- SIBER 196, SIBER 3-SIBER 207, SIBER 4- SIBER 228, Table S2-3,  
147 Fig. S3-S4). Protein yields ranged from 3-8 mg/L.



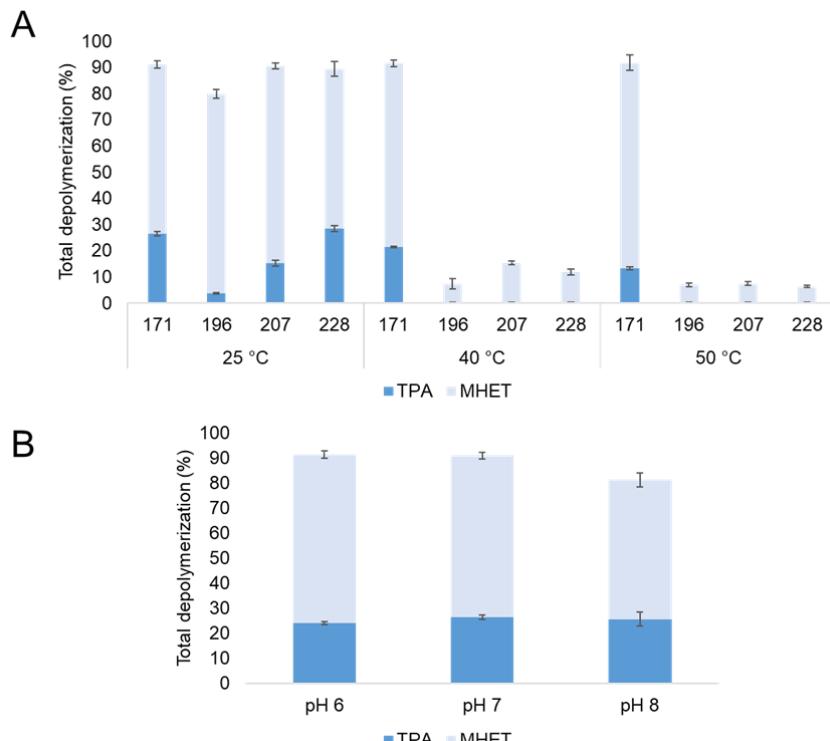
148  
149 **Figure 2. Protein expression optimization.** This figure shows a heat map depicting  
150 expression levels of SIBER 1-4 genes fused with different solubility tags in *E. coli*. The final  
151 constructs were scaled up for characterization are annotated on the heat map; 1: SIBER 171,  
152 2: SIBER 196, 3: SIBER 207, 4: SIBER 228. The proteins are His-tagged purified and ran on  
153 a protein gel. Quantification was made using ImageJ and proteins were normalized to  
154 proteins expressed with polyEK tags (last column).

155

#### 156 **BHET and PET breakdown assays, characterization of SIBER 171, 196, 207, 228**

157 To assess the functional activity of the enzymes, the purified enzymes (SIBER 171, SIBER  
158 196, SIBER 207, SIBER 228) were subjected to various assays of different substrates (e.g.,  
159 BHET, hcPET, lcPET). Under ambient temperature (25 °C) at pH 7, all the enzymes were  
160 able to hydrolyze BHET to various extents (Fig 3). The best mutant is SIBER 171 which  
161 resulted in ~95% depolymerization by the end of 24 h. SIBER 171 was also functional up to  
162 50 °C, whereas SIBER 196, SIBER 207 and SIBER 228 were only functional at 25 °C.

163 Further characterization of SIBER 171 also showed that their activities are maintained across  
164 pH 6-8.



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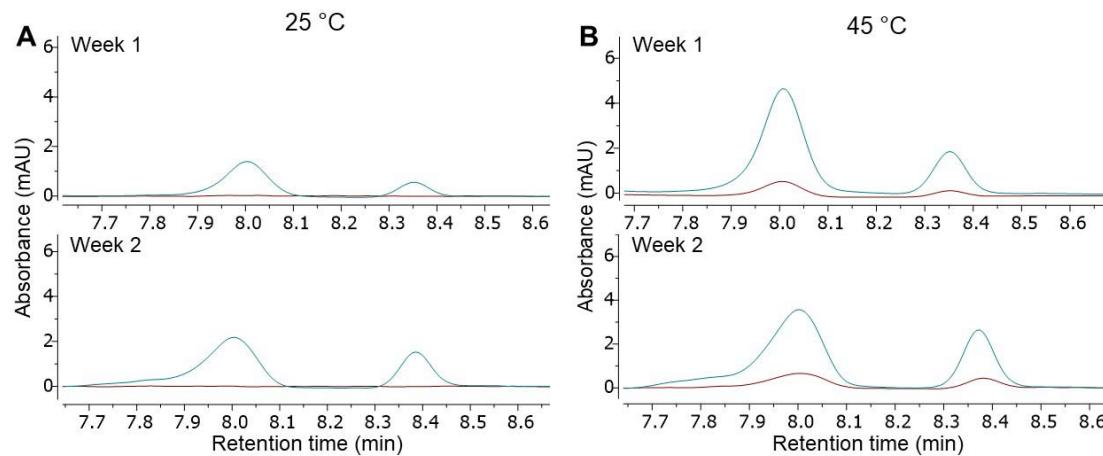
167 **Figure 3.** Characterization of SIBER 171, 196, 207 and 228. (A) Depolymerization of 5 mM  
168 BHET for 24 hours at various temperatures, pH 7. X axis are annotated as mutant numbering  
169 SIBER-. (B) Depolymerization of 5 mM BHET for 24 hours at various pH, 25°C, for SIBER  
170 171.

171

172 Applying the reported optimized LCC conditions (70 °C, 48 h) for PET depolymerization to  
173 these enzymes did not show any PET breakdown, suggesting that the native enzymes are not  
174 tolerable to such high temperatures. Re-assaying these enzymes at lower temperatures (25 °C  
175 or 45 °C) over longer time (up to 3 weeks), we found that all five enzymes were able to  
176 assimilate high crystallinity PET (hcPET), though at a very slow rate (Fig. 4, S6). The best  
177 performing enzyme herein is again SIBER 171. Conversions mostly doubled in the second

178 week when compared to the first week, indicating that the enzyme was still active. None of  
179 the variants are sufficiently thermostable to effectively break down high crystallinity (>35%)  
180 PET at 45 °C, though small amount of PET conversion was observed at 45 °C after 1 week  
181 for the enzymes SIBER 171, there was no further conversion after one week, suggesting that  
182 SIBER 171 might have been inactivated (Fig. 4, S6).

183



184

185 **Figure 4.** Depolymerization of high crystallinity PET (Goodfellow biaxially oriented film)  
186 with SIBER 171 (blue) and control (red) at (A) 25 °C over 2 weeks and (B) 45 °C over 2  
187 weeks. The LC spectra extracted indicate peaks corresponding to TPA (RT 8 min and 8.3 –  
188 8.4 min, Fig. S8). Control reaction is the equivalent reaction setup in the absence of enzymes.  
189

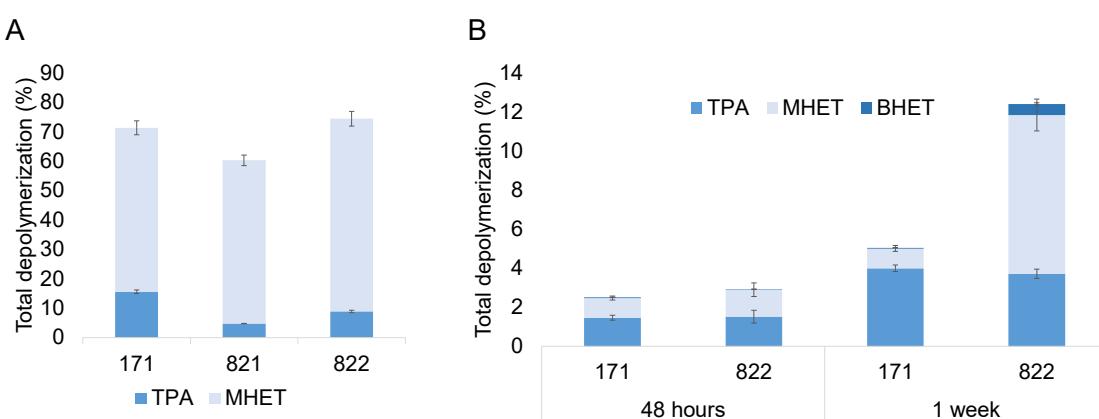
## 190 **Engineering of SIBER 171**

191 To determine if similar optimizations made to LCC can be applied to improve SIBER 171,  
192 additional mutations were introduced, including the insertion of a disulfide bond for  
193 increased thermal stability (D238C/S283C, as numbered in (9)), F243I mutation to restore  
194 activity and N246M mutation for further optimization. Here, we made similar mutations for  
195 SIBER 171, resulting in SIBER 821 (SIBER 1-ICC) and SIBER 822 (SIBER 1-ICCM). As a  
196 result of these mutations, we were also able to obtain significantly higher yielding enzymes

197 (30-40 mg/L, Fig. S5). Subsequently these purified enzymes were characterized for BHET  
198 breakdown at 50°C and PET breakdown at both 45 °C and 65 °C.

199

200 During BHET hydrolysis (50 °C, pH 7), we observed detrimental effects of ICC mutations  
201 which are however restored with N246M mutation (Fig. 5a). Under these conditions, there  
202 are minimal differences between native and engineered enzymes. However, using low  
203 crystallinity PET (lcPET) substrates at 45 °C, there appears to be significant improvement of  
204 SIBER 822 over the native enzymes over one week (Fig. 5). At 65 °C, no PET  
205 depolymerization was observed for both enzymes. Our observations suggest that, despite not  
206 being enough to increase its working temperature, the addition of a di-sulfide bridge to  
207 SIBER 1 resulted in potentially better stability at 45°C over 1 week when compared to its  
208 native enzyme. Although this is lower than optimized LCC (9) which yield PET  
209 depolymerization of 40% after 48 hours (Fig. S7), it is still encouraging to note that SIBER  
210 822 is significantly more active than IsPETase double mutant (24) (~0.08%-0.09% at 48  
211 hours, Table S4). We anticipate that the further improvement of the enzyme towards  
212 practical applications will necessitate the implementation of additional engineering  
213 techniques, such as directed evolution and computational design.



214  
215 **Figure 5.** Engineering SIBER 171. (A) Depolymerization of 5 mM BHET for 50 °C, pH 7,  
216 24 hours. X axis are annotated as mutant numbering SIBER-. (B) Depolymerization of low

217 crystallinity PET (Goodfellow amorphous film) at 45 °C over 48 hours and 1 week. X axis

218 (mutant numbering SIBER-). X axis are annotated as mutant numbering SIBER-.

219 **DISCUSSION**

220 This study aims to deepen the understanding of PET-degrading enzymes by identifying and  
221 describing new ones that can aid in PET waste biodegradation. Although some PET-  
222 degrading enzymes are known, their rarity restricts the utilization of it. Even with the addition  
223 of 34 more recently characterized PET degrading enzymes from the actinobacteria family  
224 (16), this still adds up to only a few dozen of verified PET-active enzymes (12). Exploring a  
225 new genus space, we found hydrolases that have varying specificities towards substrates  
226 BHET, MHET, and even PET. Within representative four sequences, we have uncovered and  
227 characterized a new thermostable PETase within the genus of *Microbispora*, *Nonomuraea*,  
228 and *Micromonospora* genus. Furthermore, despite sharing only 60% homology with LCC, we  
229 discovered that incorporating similar mutations previously used in the engineering of LCC  
230 (9) could improve SIBER 171's activity and stability. The engineered enzyme (SIBER 822)  
231 exhibited over two-fold improvement for lcPET depolymerization compared to its wild type.

232

233 To date, there are over 2000 known PET homologs and several enzyme engineering studies  
234 have proposed targeted mutations to improve thermostability, activity, and accessibility to  
235 PET. The results of this work raise questions about the factors that determine a PETase and  
236 the relationship between thermostability and PET degradation efficiency, especially in light  
237 of the impact of the substrate's physical properties (25), such as crystallinity. Among the four  
238 enzymes, there were notable differences and correlations in solubility and expression yields  
239 and activity. Here, among the SIBER 1-4 genes, SIBER 1 constructs had consistent higher  
240 protein expression yields. After engineering of SIBER 171, we also observed significant  
241 expression yield increase in engineered SIBER 822. This is consistent with the discussions of  
242 correlations between protein folding and protein activity, where the thermodynamics of the  
243 sequences may play a role in proper and accurate folding (26). Beyond stability, further

244 investigation on the mechanism of PET degradation is required to gain a deeper  
245 understanding on the potential to design and develop new and highly efficient PETases. The  
246 understanding on enzyme-plastic interactions would allow PETases to be applied to other  
247 plastics and feedstocks.

248 **MATERIALS AND METHODS**

249 All solvents used in product analysis (acetonitrile, formic acid, dimethylformamide) were  
250 purchased from commercial suppliers. Bis (2-hydroxyethyl) terephthalate (BHET),  
251 terephthalic acid (TPA), potassium phosphate ( $\text{NaH}_2\text{PO}_4$  and  $\text{Na}_2\text{HPO}_4$ ), and 4-bromobenzoic  
252 acid were obtained from Sigma-Aldrich. High crystallinity PET film (ES301250) and low  
253 crystallinity PET film (ES301445) were obtained from Goodfellow and was ground into  
254 powder.

255

256 **Construction of plasmids optimized for *E. coli* expression**

257 Codon-optimized DNA sequences obtained from *E. coli* were synthesized from Twist  
258 Biosciences. These inserts were assembled via golden gate into pET28a (+), cloned with *E.*  
259 *coli* Omnimax competent cells. *E. coli* XJb (DE3) autolysis competent cells (Zymo Research)  
260 was subsequently used for the transformation of these 6xHis tag constructs for protein  
261 expression. Table S2 shows the amino acid sequences of the gene used and the solubility tags  
262 used are shown in Table S3.

263

264 **High throughput screening of heterologous expression constructs**

265 1 mL overnight autoinduction media (Merck) was used for the inoculation of single colonies  
266 obtained from the transformation, in which 50  $\mu\text{g}/\text{mL}$  kanamycin, 1.5 M L-arabinose and 0.5  
267 M magnesium chloride was also added. A single freeze thaw cycle in lysis buffer (50 mM  
268 sodium phosphate buffer, 300 mM sodium chloride, 10 mM Imidazole and 0.03 % TritonX-  
269 100), from the cell pellet harvested allowed the cell to release its embedded protein. To  
270 capture these His-tagged proteins, Ni resin (PureCube, Cube Biotech) was used. These  
271 proteins were subsequently eluted in 50 mM sodium phosphate buffer pH 7.0, 300 mM  
272 sodium chloride and 500 mM imidazole. Quantification of the protein elutes on a protein gel

273 was performed by ImageJ to compare protein yields across the different constructs. Intensity  
274 of each band obtained was normalized to the intensity of the reference band on the standard  
275 ladder used, as well as molecular weight of each protein. This densitometric comparison was  
276 used to determine the best yielding constructs which were then scaled up for purification and  
277 subsequent characterization.

278

279 **Protein expression and purification**

280 A starter culture of 5 mL grown in LB Broth containing 50 µg/mL kanamycin was prepared,  
281 where a single colony from the transformation was inoculated overnight at 37 °C. The starter  
282 culture was diluted 200-fold in a fresh 1 L LB Broth containing 50 µg/mL kanamycin, 1.5 M  
283 L-arabinose and 0.5 M Magnesium chloride. Subcultures were grown at 37 °C until the  
284 optical density at 600 nm (OD600) reached ~0.4 – 0.5. 100 µM isopropyl β-D-1-  
285 thiogalactopyranoside (IPTG) was then added to induce the expression of proteins. Following  
286 which, the cultures were incubated overnight at 16 °C for 18 to 20 hours. Cells were  
287 harvested from this culture using a centrifuge (15 minutes, 8000 g) maintained at 4 °C, before  
288 resuspending and freezing the resulting pellet at -80 °C in 10 mL of lysis buffer containing 50  
289 mM sodium phosphate buffer pH 7.0, 300 mM sodium chloride, 10 mM Imidazole and 0.03  
290 % TritonX-100. To ensure protein stability, subsequent purification steps were also carried  
291 out at 4 °C.

292

293 The frozen pellet was thawed with the addition of 10mL lysis buffer and sonicated to release  
294 proteins from the cells. The supernatant obtained from centrifugation of lysates at 13,500xg  
295 was loaded on Ni resin and incubated for 1 hour to capture the His-tagged proteins. 20mL of  
296 50mM sodium phosphate buffer, 300mM sodium chloride and 50mM imidazole was then  
297 used to wash the resin, before eluting the bound proteins with 50mM sodium phosphate

298 buffer pH 7.0, 300mM sodium chloride and 500mM imidazole. Lastly, the proteins were  
299 exchanged into 50 mM sodium phosphate pH 7.0 with 10% glycerol to allow for long-term  
300 storage at -80 °C. Subsequent BHET and PET degradation studies were carried out with these  
301 purified proteins.

302

### 303 **BHET depolymerization assay**

304 Dimethylformamide was used to dissolve 1 M BHET stock solution. BHET stock solution  
305 (2.5 µL, 5 mM) was pipetted into a 2 mL glass vial containing 500 µL of 100 mM potassium  
306 phosphate buffer (pH 7) and 1.67 µM purified protein. The glass vial was tightly capped, and  
307 the reaction mixture was incubated at 25 °C for 24 hours in an Eppendorf ThermoMixer®.  
308 After 24 hours, the reaction is quenched with 500 µL of methanol. The mixture was  
309 transferred into a 10 mL centrifuge tube, the vial was washed with 4 mL of buffer/methanol  
310 (1:1, v/v) and the contents were transferred to the centrifuge tube. This is followed by the  
311 addition of 5 mL 0.5 mM 4-bromobenzoic acid in buffer/methanol (1:1, v/v) as an internal  
312 standard. The reaction mixture was sonicated, and an aliquot was filtered with a 0.2 µm  
313 syringe filter and analyzed via UHPLC-MS.

314

### 315 **Calibration of BHET and TPA**

316 BHET and TPA stock solutions (1 mM) were prepared by dissolving the solids in 5 mL of in  
317 buffer/methanol (1:1, v/v), followed by the addition of 5 mL 0.5 mM 4-bromobenzoic acid in  
318 buffer/methanol (1:1, v/v) as an internal standard. 6 concentrations ranging from 0.05 mM to  
319 1 mM were prepared from the stock solution and analyzed via UPHLC-MS. A calibration  
320 was plotted with the molar ratio against area ratio. LC spectrum of TPA (Fig. S8) revealed 2  
321 peaks (RT 8 min and RT 8.4 min) and the LC-MS spectrum extracted from both peaks

322 correspond to TPA [M-H]<sup>-</sup> ion m/z 165 (Fig. S8). The area of both peaks was taking into  
323 consideration in the plotting of calibration curve of TPA (Fig. S8).

324

### 325 **Calibration of MHET**

326 MHET (1 M) was prepared by dissolving the solids in dimethylformamide. 6 concentrations  
327 ranging from 0.03 mM to 0.18 mM were prepared from the MHET stock solution, topped up  
328 with 0.5 mM 4-bromobenzoic acid internal standard and analyzed via UHPLC-MS. A  
329 calibration was plotted with the molar ratio against area ratio.

330

### 331 **PET depolymerization assay**

332 PET powder (2 mg, 20 mM) was weighed into a 2 mL glass vial and fully submerged in 500  
333 µL of 100 mM potassium phosphate buffer (pH 7) with 1.67 µM purified protein. The glass  
334 vial was tightly capped, and the reaction mixture was incubated at 500 rpm and 25-50°C for  
335 48 hours to up to 28 days in an Eppendorf ThermoMixer<sup>®</sup>. The reaction is quenched with 500  
336 µL of methanol. The mixture was transferred into a 10 mL centrifuge tube, the vial was  
337 washed with 4 mL of buffer/methanol (1:1, v/v) and the contents were transferred to the  
338 centrifuge tube. This is followed by the addition of 5 mL 0.5 mM 4-bromobenzoic acid in  
339 buffer/methanol (1:1, v/v) as an internal standard. The reaction mixture was sonicated, and an  
340 aliquot was filtered with a 0.2 µm syringe filter and analyzed via UHPLC-MS.

341

### 342 **ACKNOWLEDGMENTS**

343 This research is supported by Agency for Science, Technology and Research, Singapore,  
344 A\*STAR <C211917006> and <C211917003>. J.B. acknowledges A\*STAR Graduate  
345 Academy (A\*GA) for his scholarship funding. Sequences derived for this work were mined  
346 from the in-house A\*STAR National Organism Library (NOL) and public Genbank database.

347 We gratefully acknowledge Dr. Siew Bee Ng (SIFBI), her team and Elena Heng for their  
348 contribution to the sequencing of the strains from NOL. A patent application has been filed  
349 for this work.

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