



ENGINEERING PHYSICS

R Vasudevan Iyer, Ph.D.

Department of Science and Humanities



Class #21

- The kinetic energy and potential energy operators are $\hat{K} = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2}$ and \hat{V}
- Since sum of kinetic and potential energies is the total energy E we have $E = K + V$
- If K and V are treated as operators then the sum of them is also an operator. Thus $\hat{E} = \hat{K} + \hat{V}$
- The total energy operator is referred to as the Hamiltonian operator and is written as \hat{H} . Thus $\hat{H} = \hat{K} + \hat{V} = -\frac{\hbar^2}{2m} \frac{\partial^2}{\partial x^2} + \hat{V}$
- The Hamiltonian can also be written as the time operator, $\hat{H} = i\hbar \frac{\partial}{\partial t}$

One dimensional Schrodinger's time dependent equation

- ❖ We know that quantum systems are described by the wavefunction, ψ
- ❖ Quantum systems would undergo changes as time progresses. For example the electron in an atom changes its position, momentum, energy etc with time as it interacts with the environment.
- ❖ This change must manifest itself as changes in the wavefunction and thus ψ must change with time
- ❖ How do we monitor such a change?
- ❖ This was first shown by Erwin Schrodinger and the equation showing time evolution of a quantum system is known as the Schrodinger's time dependent equation.



One dimensional Schrodinger's time dependent equation

- ❑ We start with the equation $E = K + V$
- ❑ The physical quantities are now treated as operators and we write an operator equation, $\hat{E} = \hat{K} + \hat{V}$
- ❑ We apply the operators to the wavefunction of the quantum system as follows: $\hat{E}\Psi = \hat{K}\Psi + \hat{V}\Psi$
- ❑ The operators are now written in their mathematical form which then leads to $i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + \hat{V}\Psi$. Note that we still have not defined the operator corresponding to the potential energy. We will elucidate this point later
- ❑ The equation shown is the famous Schrodinger's equation and is written in the concise form as $\hat{H}\Psi = i\hbar \frac{\partial \Psi}{\partial t}$



One dimensional Schrodinger's time dependent equation

$$i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + \hat{V}\Psi$$

- we note that the Schrodinger's time dependent equation has a first order derivative in time and second order derivative in space
- The unique feature of this equation is the presence of i. If not for i this equation would resemble the heat equation. Due to the presence of i it becomes an equation representing the flow of probability
- It is a partial differential equation.
- As it is linear if ψ_1 is a solution to this equation and ψ_2 another independent solution then $a\psi_1 + b\psi_2$ is also a solution. This has deep significance



One dimensional Schrodinger's time dependent equation

- We will come across quantum systems where the potential energy is a function of space only ($V(x)$).
- Also in many quantum systems the total energy for a given state remains constant ($E = \text{constant}$)
- When we consider these two aspects we ask the question – how does the Schrodinger's equation change.
- To see what changes will happen we take the time dependent equation try to solve it



- To solve a partial differential equation we make use of a technique called **SEPARATION OF VARIABLES**
- We assume that the wavefunction, Ψ , which is a function of space and time, can be written as $\Psi(x, t) = \psi(x)f(t)$

- We now substitute this in the equation $i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + \hat{V}\Psi$

to get $i\hbar \frac{\partial \psi f}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \psi f}{\partial x^2} + \hat{V}\psi f$

- In the LHS we see that the derivative is w.r.t time and hence we write it as

$$\psi \left(i\hbar \frac{df}{dt} \right) = -\frac{\hbar^2}{2m} \frac{\partial^2 \psi f}{\partial x^2} + \hat{V}\psi f$$



One dimensional Schrodinger's time dependent equation

- In RHS we note that the first term is a derivative of space. In the second term we see that then potential energy operator acts on the wave function. We have chosen systems where potential energy is a function of space only and hence any mathematical form representing the potential energy operator must act on functions of space only.
- Thus we have
$$\psi \left(i\hbar \frac{df}{dt} \right) = f \left(-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} + \hat{V}\psi \right)$$
- Also note that we change from **partial derivatives** to **total derivatives**



➤ Continuing further we make the following changes $\frac{1}{f} \left(i\hbar \frac{df}{dt} \right) =$

$$\frac{1}{\psi} \left(-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} + \hat{V}\psi \right)$$

➤ Now note that LHS is a function of **time** only whereas the RHS is a function of **space** only. If they have to be equated then both have to be equal to some constant. Let us call the constant as **G (referred to as separation constant)**

➤ Thus $\frac{1}{f} \left(i\hbar \frac{df}{dt} \right) = \mathbf{G}$ and $\frac{1}{\psi} \left(-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} + \hat{V}\psi \right) = G$



- Solving for LHS we get $f = Ae^{-\frac{i}{\hbar}Gt}$. Generally one form of the wave function is $\Psi = Ae^{\frac{i}{\hbar}(px-Et)}$. This can be written as $\Psi = Ae^{\frac{i}{\hbar}(px)}e^{-\frac{i}{\hbar}Et}$
- We see a pattern. The time behaviour is always has the form $e^{-\frac{i}{\hbar}Et}$. The space behaviour in this case is $\psi = e^{\frac{i}{\hbar}(px)}$ whereas in a general case is denoted by $\psi(x)$. Hence for behaviour of quantum system is always represented by $\Psi = \psi(x)e^{-\frac{i}{\hbar}Et}$. We immediately see that the separation constant is the total energy E which we have taken to be constant.



One dimensional Schrodinger's time dependent equation

- Thus we have $\frac{1}{\psi} \left(-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} + \hat{V}\psi \right) = E$
- The obvious question now is what mathematical representation can we assign to \hat{V} ?
- In all the situations we will come across we have the following: $V = 0$ or $V = V_0$ or $V = V(x)$. Therefore our potential energy operator is simply V . We cannot ascribe any mathematical form to it.
- Hence we have $\frac{1}{\psi} \left(-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} + V\psi \right) = E$ which can be written as $-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} + V\psi = E\psi$. This is one dimensional **Schrodinger's time independent** equation.



ENGINEERING PHYSICS

One dimensional Schrodinger's time dependent equation



- The equation can be written as $\hat{H}\psi = E\psi$ the **eigen value equation**.
- Therefore when we solve the Schrodinger's time independent equation we are solving the eigen value equation to obtain the eigen value **E** and the eigen function **ψ**
- Most of the time the Schrodinger's equation is presented as $\frac{d^2\psi}{dx^2} + \frac{2m}{\hbar^2} (E - V)\psi = 0$
- The 1D Schrodinger's time independent is an ordinary linear differential equation and as before



- If ψ_1 is a solution to the time independent Schrodinger's equation and ψ_2 another independent solution then which of the following is true?.
- $a\psi_1 + b\psi_2$ is also a solution
- $a\psi_1 + b\psi_2$ is not a solution
- $ab\psi_1\psi_2$ is also a solution
- $a\psi_1 / b\psi_2$ is also a solution





THANK YOU

R Vasudevan Iyer, Ph.D.

Professor, Department of Science and Humanities

rviyer@pes.edu

