



## ENGINEERING PHYSICS

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### Class #42

#### Carbon dioxide laser

- Gas laser
- Molecular laser
- CO<sub>2</sub> molecule: Modes
- Construction
- Energy level diagram

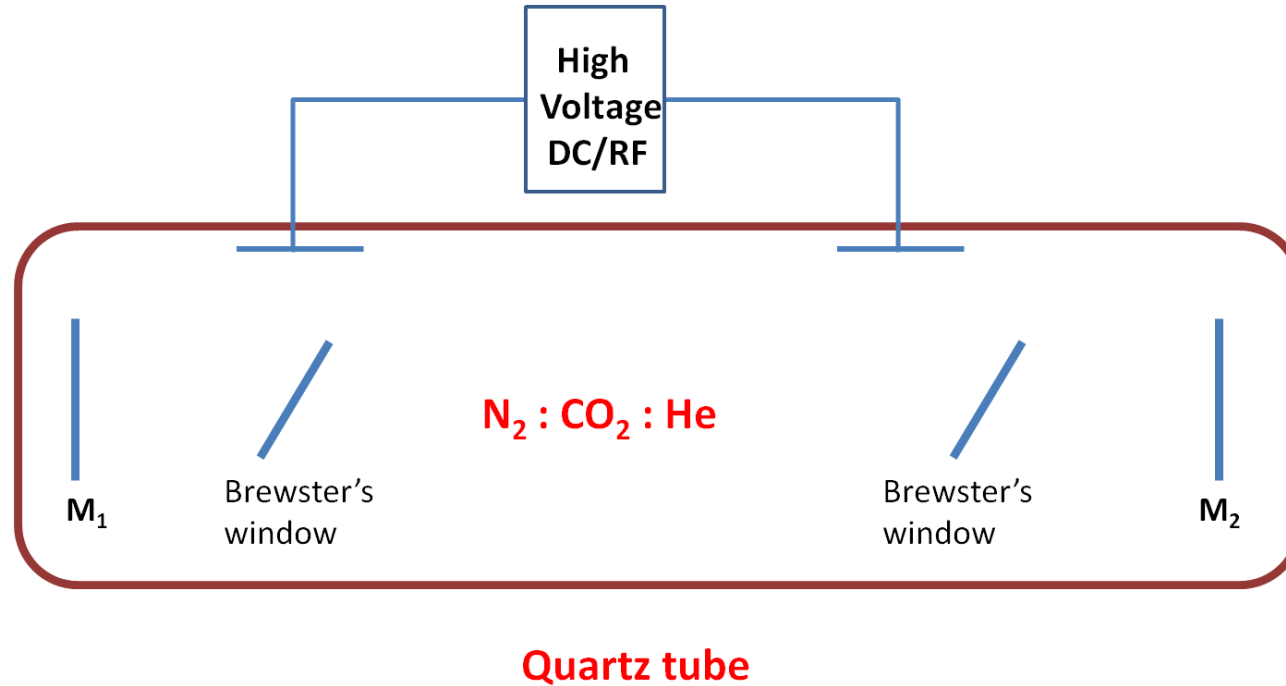
### ➤ *Suggested Reading*

1. *Optical Electronics, A. Yariv*
2. *Course material developed by the Department*

### ➤ *Reference Videos*

<https://ocw.mit.edu/resources/res-6-005-understanding-lasers-and-fiberoptics-spring-2008/laser-fundamentals-i/>

- Molecular laser
- Transitions in molecular energies
- IR laser
- High power laser (kW)



- Evacuated quartz tube (long and narrow)
- Gas mixture of  $N_2 : CO_2 : He$
- A DC or RF supply for electron discharge through the gas mixture
- The mirrors and Brewster windows are made of semiconducting materials such as Ge to avoid IR absorption.

CO<sub>2</sub> molecule has three modes of vibration:

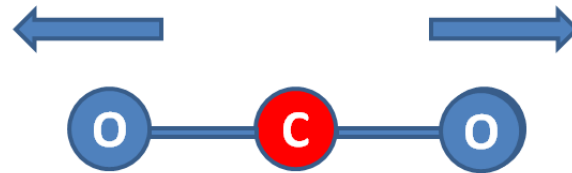
1. Symmetric stretching
2. Asymmetric stretching and
3. Bending mode.

- The quantized energies of the symmetric stretching are denoted as  $(n00)$
- The quantized energies of the asymmetric stretching are denoted as  $(00n)$
- The quantized energies of the bending mode are denoted as  $(0n0)$   
where  $n$  is a positive integer.

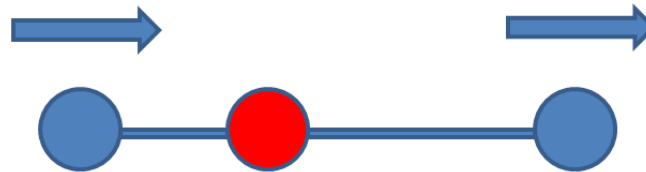
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## CO<sub>2</sub> Molecule: Modes

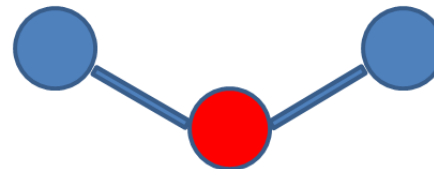
Symmetric stretching  
Bond lengths always equal



Asymmetric stretching  
Bond lengths Unequal



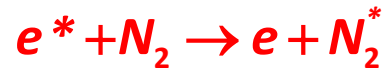
Bending mode



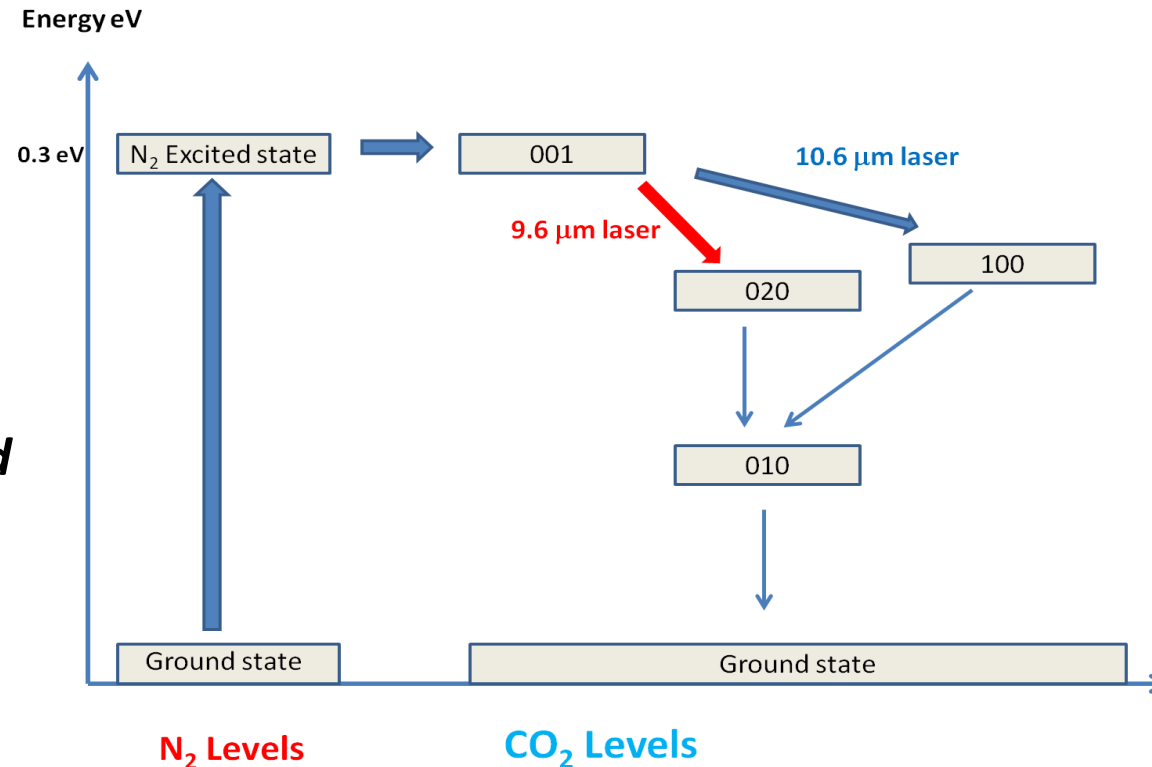
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## CO<sub>2</sub> laser Energy level diagram

*Collision of the first kind*



*Collision of the second kind*



- Fast moving electrons from the discharge collide with N<sub>2</sub> molecules and excite them to their first excited state at 0.3 eV.
- These excited N<sub>2</sub> molecules then collide with CO<sub>2</sub> molecules and selectively excite them to the asymmetric 001 state.



### *Check Your Understanding (Yes/No)*

- 1. The energy difference between states of molecular vibrations correspond to UV wavelength*
- 2. Nitrogen gas is a buffer gas in this system*
- 3. He gas is used to depopulate the (010) state*
- 4. Carbon dioxide laser produces a powerful IR beam*



# THANK YOU

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