



ENGINEERING CHEMISTRY

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Energy storage devices – Supercapacitors



Class content:

- *Supercapacitors*
 - *Principle*
 - *Construction and working*
 - *Advantages*
 - *Disadvantages*
 - *Applications*
- *Ragone Plot*

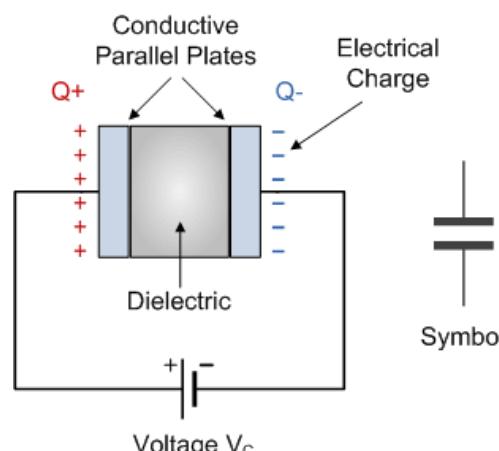
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Capacitor

- A two-terminal electrical component that has the **ability to store energy** in the form of electrical charge
- Has two conducting plates separated by a dielectric
- When a DC voltage is connected across the capacitor, one plate becomes positive and the other negative
- The **charge accumulation on the plates** causes a voltage or potential difference across the capacitor



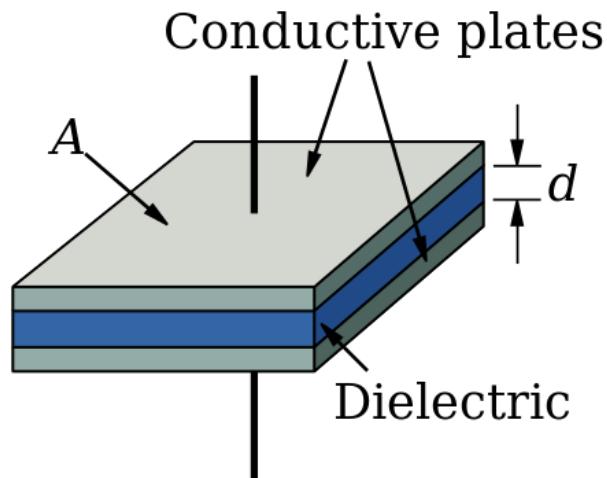
Source:https://www.electronics-tutorials.ws/capacitor/cap_1.html

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- **Capacitance** is the charge accumulation capability of a capacitor
- Charge (Q) stored in a capacitor is directly proportional to the voltage (V) developed across the capacitor; $Q \propto V$
- $Q = C \times V$, where C is the capacitance measured in **farads**
- Capacitance is directly proportional to **area of plates (A)** and inversely proportional to the **distance between the two plates (D)**; $C \propto \frac{A}{d}$



Source:[https://www.wikiwand.com
/en/Capacitor_types](https://www.wikiwand.com/en/Capacitor_types)

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Supercapacitors

- Also known as **ultra capacitors** or **electrochemical double-layer capacitors(EDLC)**
- Are energy storage devices that have **high capacitance** and are used to store large amounts of electrical charge
- **Charge and discharge** very quickly
- Capacitance is several thousand times that of a Capacitor
 - Capacitor ----- μF
 - Supercapacitor ----- **thousands of Farads**

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Construction:

Electrodes: made of high surface area materials such as porous carbon, graphene, carbon nanotubes and certain conducting polymers or carbon aerogel

Electrolyte: KOH, H_2SO_4 , Na_2SO_4

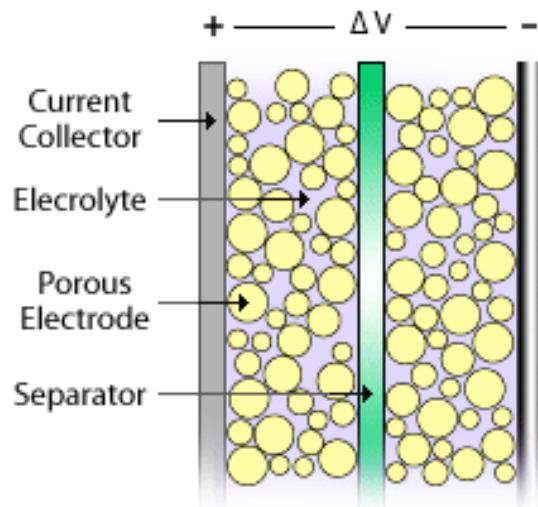
Separator: an ion permeable separator is placed between the electrodes in order prevent electrical contact, but still allows ions from electrolyte to pass through e.g., porous polypropylene

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Separator is sandwiched between the electrodes which are soaked with **electrolyte** and sealed inside a case.
The electrodes are flanked with **current collectors**



Source:<https://wwwazonano.com/article.aspx?ArticleID=3044>

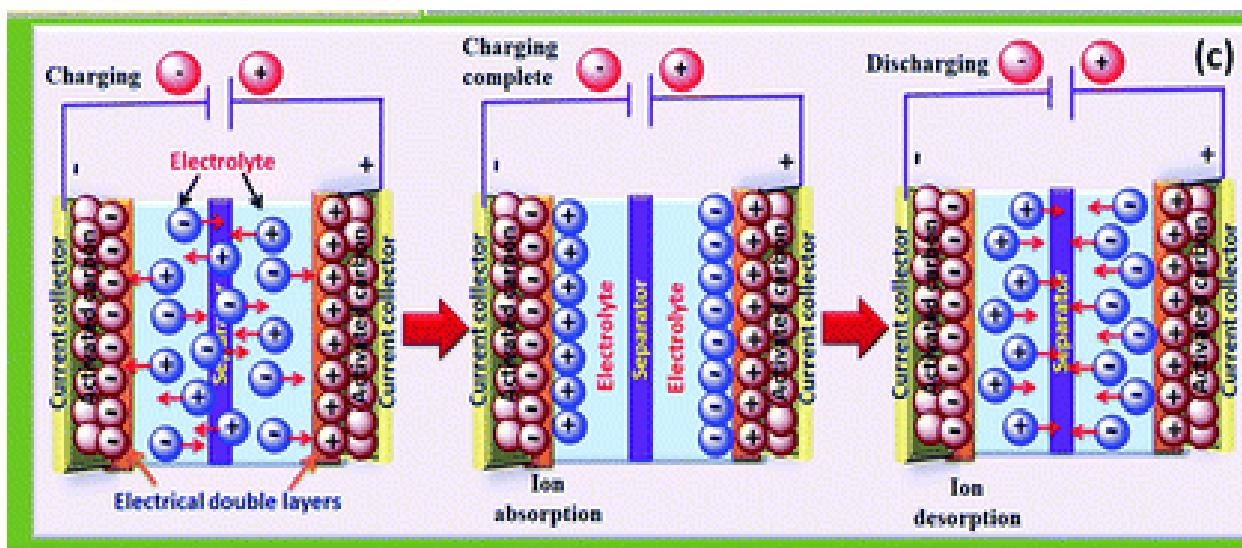
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Working:

- When a potential is applied, the positive electrode attracts negative ions in the electrolyte, while negative electrode attracts the positive ions
- Formation of electrical double layer at entire electrode /electrolyte interface with a charge separation in nanometer scale



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- Only **absorption and desorption of ions** takes place at the electrode during charging and discharging and there are **no redox reactions**
- Distance between the charged layers in nanorange, **d is very small** and use of porous electrodes gives very high surface area , **A is large**

$$C \propto \frac{A}{d} \quad ; \text{Capacitance is very high}$$

- Formation of the electrical double-layer at each electrode/electrolyte

interface, overall capacitance (C_T) is expressed as:

$$\frac{1}{C_T} = \frac{1}{C_+} + \frac{1}{C_-}$$

where C_+ is the capacitance of cathode/electrolyte interface and C_- is the capacitance of anode/electrolyte interface

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Advantages:

- **Rapid charging**; charge in a few seconds
- **High power density** as they discharge very fast
- **High cycle life**, can be cycled millions of time
- **Safe** as extremely low internal resistance and extremely low heating rates

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Disadvantages:

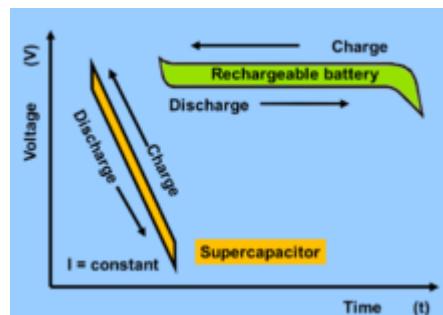
Low energy density

High self discharge

Linear discharge voltage

High cost

Power available for a short duration



[Source:https://store.chipkin.com/articles/beyond-the-lithium-ion-battery-a-look-at-supercapacitors-and-other-batteries](https://store.chipkin.com/articles/beyond-the-lithium-ion-battery-a-look-at-supercapacitors-and-other-batteries)

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Applications:

- Memory back-up
- Hybrid cars for start-stop application
- Flash photography devices in digital cameras, flash lights, portable media players
- As an intermediate energy storage for FM radios, cell phones, and emergency kits

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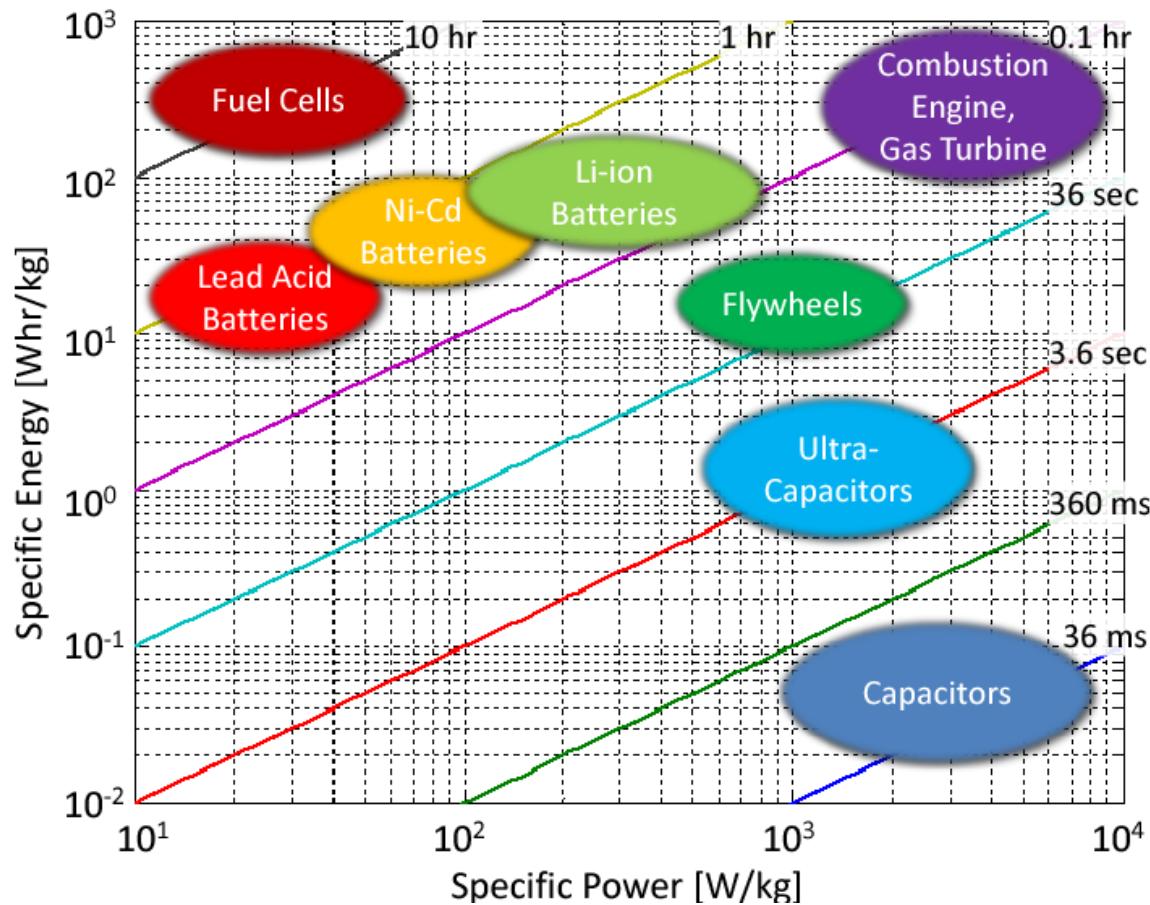


Ragone Plot

- Energy density (Wh/kg) is plotted against Power density (W/kg)
- To compare performance of various energy storage devices
- Since it uses a double-logarithmic chart, storage technologies with very different storage properties can be compared in one plot

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Source: <https://www.semanticscholar.org/paper/Education-on-vehicle-electrification%3ABattery-Fuel-Moura-Siegel/7e83f03396b55f2a894461480dda7e9c154f5721/figure/0>

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- **Fuel cells** have high energy density as the electroactive species can be continuously supplied but the power density is low due to slow kinetics of redox reactions at electrodes
- **Battery systems** offer moderate energy density and power density.
- **Ultracapacitors** (supercapacitors) can deliver very high power density as they can discharge a large amount of charge quickly (because no redox reaction is involved) but energy density is very limited because the charge cannot be stored for a long time



THANK YOU
