



# ENGINEERING CHEMISTRY

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## Energy storage devices - Batteries

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### *Class content:*

- *Classification of Batteries*

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**Based on working, two types of batteries:**

### Primary battery

- Electroactive material cannot be regenerated
- Discarded after electroactive species is consumed
- Galvanic cell
- Dry cell,  $\text{Li-MnO}_2$

### Secondary battery

- Electroactive material can be regenerated
- Can be used several times
- Galvanic as well as electrolytic cell
- Storage battery
- Li-ion, Pb-acid, Ni-Cd

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### Reserve batteries

- One of the **key component is kept separated** from the rest of the battery allowed to come in contact with other components when required
- Usually the electrolyte is the component that is isolated
- When one of the key components of the cell is separated from the rest of the cell, chemical reaction between the cell components (self discharge) is prevented
- Has **unlimited shelf life**
- Usually designed for **emergency use** e.g mid-air crisis
- **Examples:** **Mg-AgCl(water activated) , Zn-Ag<sub>2</sub>O(KOH activated)**
- Applications: Sonobuoys, Weather balloons, Emergency lights etc.,

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### Salient features:

#### Quick activation

- Deliver high power for a short time - many kW for several minutes
- No self-discharge
- Unlimited shelf life
- High reliability
- Examples : Mg-AgCl(water activated) , Zn-Ag<sub>2</sub>O(KOH activated)



# THANK YOU

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