



# ENGINEERING CHEMISTRY

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## Energy storage devices – Supercapacitors

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### *Class content:*

- ***Supercapacitors***
  - *Principle*
  - *Construction and working*
  - *Advantages*
  - *Disadvantages*
  - *Applications*
- ***Ragone Plot***

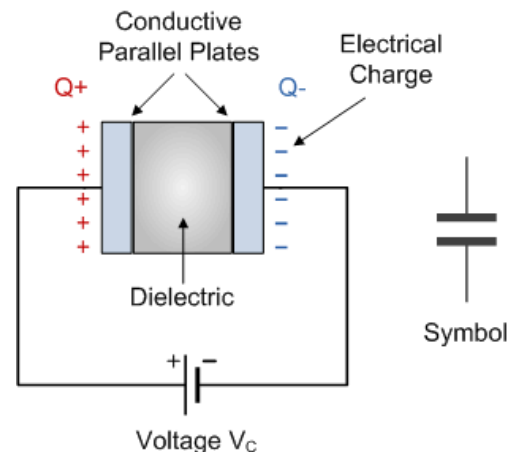
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## Energy Storage devices- Supercapacitors

### Capacitor

- A two- terminal electrical component that has the **ability to store energy** in the form of electrical charge
- Has two conducting plates separated by a dielectric
- When a DC voltage is connected across the capacitor, one plate becomes positive and the other negative
- The **charge accumulation on the plates** causes a voltage or potential difference across the capacitor

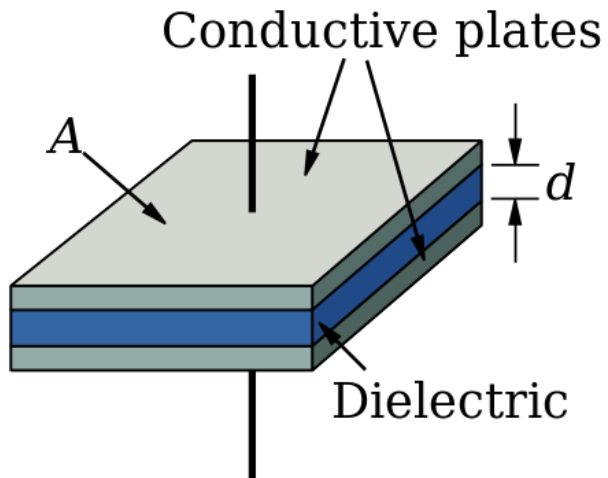
Source:[https://www.electronics-tutorials.ws/capacitor/cap\\_1.html](https://www.electronics-tutorials.ws/capacitor/cap_1.html)



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- **Capacitance** is the charge accumulation capability of a capacitor
- Charge (Q) stored in a capacitor is directly proportional to the voltage (V) developed across the capacitor;  $Q \propto V$
- $Q = C \times V$ , where C is the capacitance measured in **farads**
- Capacitance is directly proportional to **area of plates (A)** and inversely proportional to the **distance between the two plates (D)**;  $C \propto \frac{A}{d}$



Source: [https://www.wikiwand.com/en/Capacitor\\_typeS](https://www.wikiwand.com/en/Capacitor_typeS)

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### Supercapacitors

- Also known as **ultra capacitors** or **electrochemical double-layer capacitors(EDLC)**
- Are energy storage devices that have **high capacitance** and are used to store large amounts of electrical charge
- **Charge and discharge** very quickly
- Capacitance is several thousand times that of a Capacitor
  - **Capacitor -----  $\mu\text{F}$**
  - **Supercapacitor ----- thousands of Farads**

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### Construction:

**Electrodes:** made of high surface area materials such as porous carbon, graphene, carbon nanotubes and certain conducting polymers or carbon aerogel

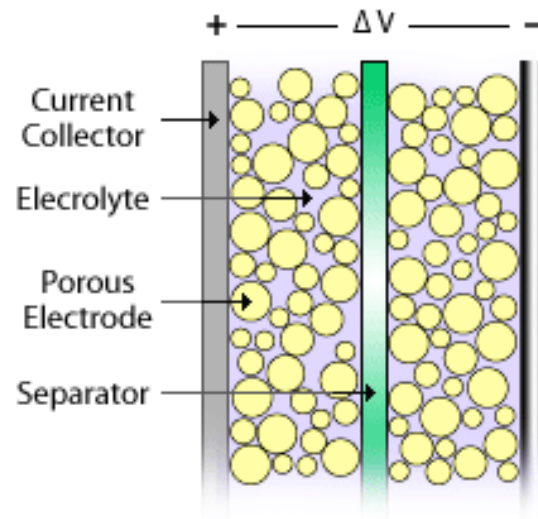
**Electrolyte:** KOH,  $\text{H}_2\text{SO}_4$ ,  $\text{Na}_2\text{SO}_4$

**Separator:** an ion permeable separator is placed between the electrodes in order prevent electrical contact, but still allows ions from electrolyte to pass through e.g., porous polypropylene

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**Separator** is sandwiched between the electrodes which are soaked with **electrolyte** and sealed inside a case. The electrodes are flanked with **current collectors**



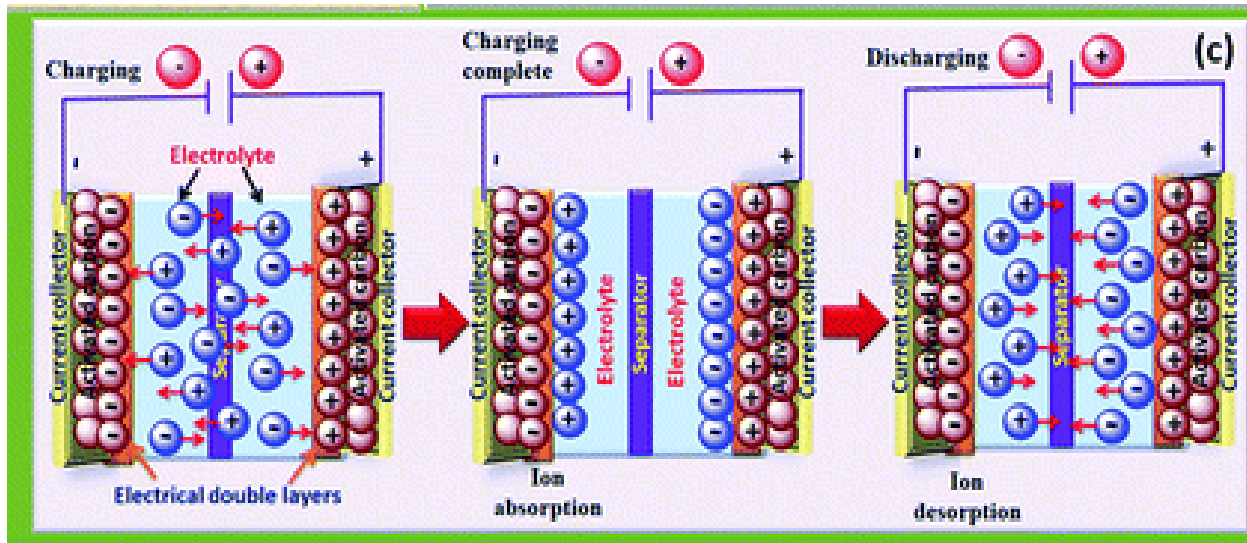
Source: <https://www.azonano.com/article.aspx?ArticleID=3044>

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### Working:

- When a potential is applied, the positive electrode attracts negative ions in the electrolyte, while negative electrode attracts the positive ions
- Formation of electrical double layer at entire electrode /electrolyte interface with a charge separation in nanometer scale





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- Only **absorption and desorption of ions** takes place at the electrode during charging and discharging and there are **no redox reactions**
- Distance between the charged layers in nanorange, **d is very small** and use of porous electrodes gives very high surface area , **A is large**

$$C \propto \frac{A}{d} ; \text{Capacitance is very high}$$

- Formation of the electrical double-layer at each electrode/electrolyte

interface, overall capacitance ( $C_T$ ) is expressed as: 
$$\frac{1}{C_T} = \frac{1}{C_+} + \frac{1}{C_-}$$

where  $C_+$  is the capacitance of cathode/electrolyte interface and  $C_-$  is the capacitance of anode/electrolyte interface

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### Advantages:

- **Rapid charging**; charge in a few seconds
- **High power density** as they discharge very fast
- **High cycle life**, can be cycled millions of time
- **Safe** as extremely low internal resistance and extremely low heating rates



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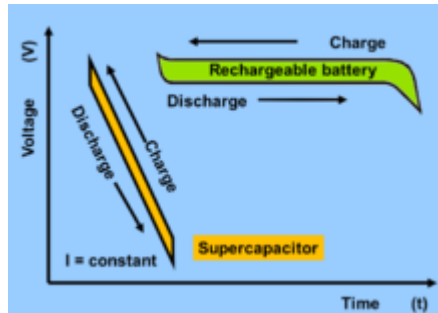
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### Disadvantages:

Low energy density

High self discharge

Linear discharge voltage



Source: <https://store.chipkin.com/articles/beyond-the-lithium-ion-battery-a-look-at-supercapacitors-and-other-batteries>

High cost

Power available for a short duration

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### Applications:

- Memory back-up
- Hybrid cars for start-stop application
- Flash photography devices in digital cameras, flash lights, portable media players
- As an intermediate energy storage for FM radios, cell phones, and emergency kits

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### Ragone Plot

- **Energy density** (Wh/kg) is plotted against **Power density** (W/kg)
- To **compare performance** of various energy storage devices
- Since it uses a double-logarithmic chart, storage technologies with very different storage properties can be compared in one plot

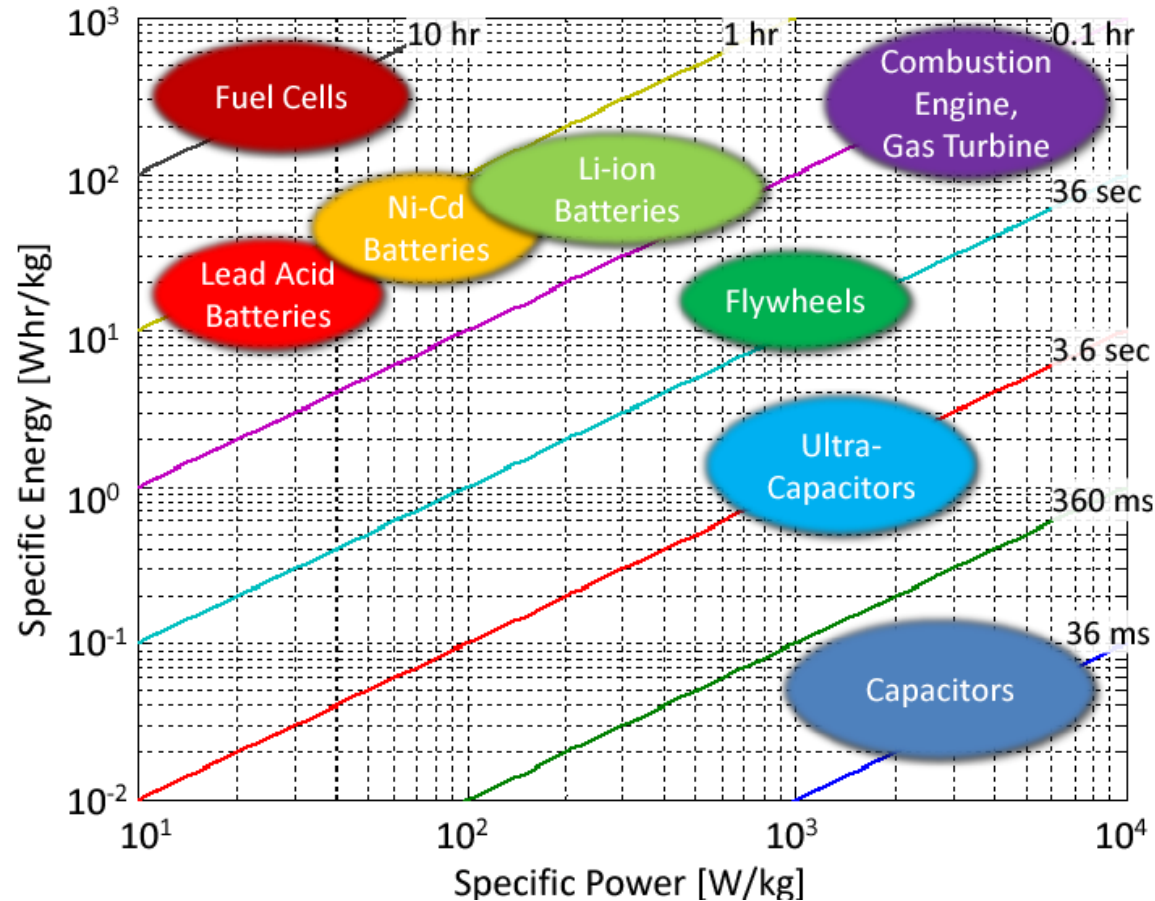
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Source: <https://www.semanticscholar.org/paper/Education-on-vehicle-electrification%3A-Battery-Fuel-Moura-Siegel/7e83f03396b55f2a894461480dda7e9c154f5721/figure/0>

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- **Fuel cells** have high energy density as the electroactive species can be continuously supplied but the power density is low due to slow kinetics of redox reactions at electrodes
- **Battery systems** offer moderate energy density and power density.
- **Ultracapacitors** (supercapacitors) can deliver very high power density as they can discharge a large amount of charge quickly (because no redox reaction is involved) but energy density is very limited because the charge cannot be stored for a long time





# THANK YOU

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