



# ENGINEERING CHEMISTRY

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Department of Science and Humanities

# ENGINEERING CHEMISTRY

## Electrochemical Sensors-Biosensor- Glucose Sensor

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*Class Content :*

**BIOSENSORS**

**GLUCOSE SENSOR**

*Components of glucose biosensor*

*Working of glucose biosensor*

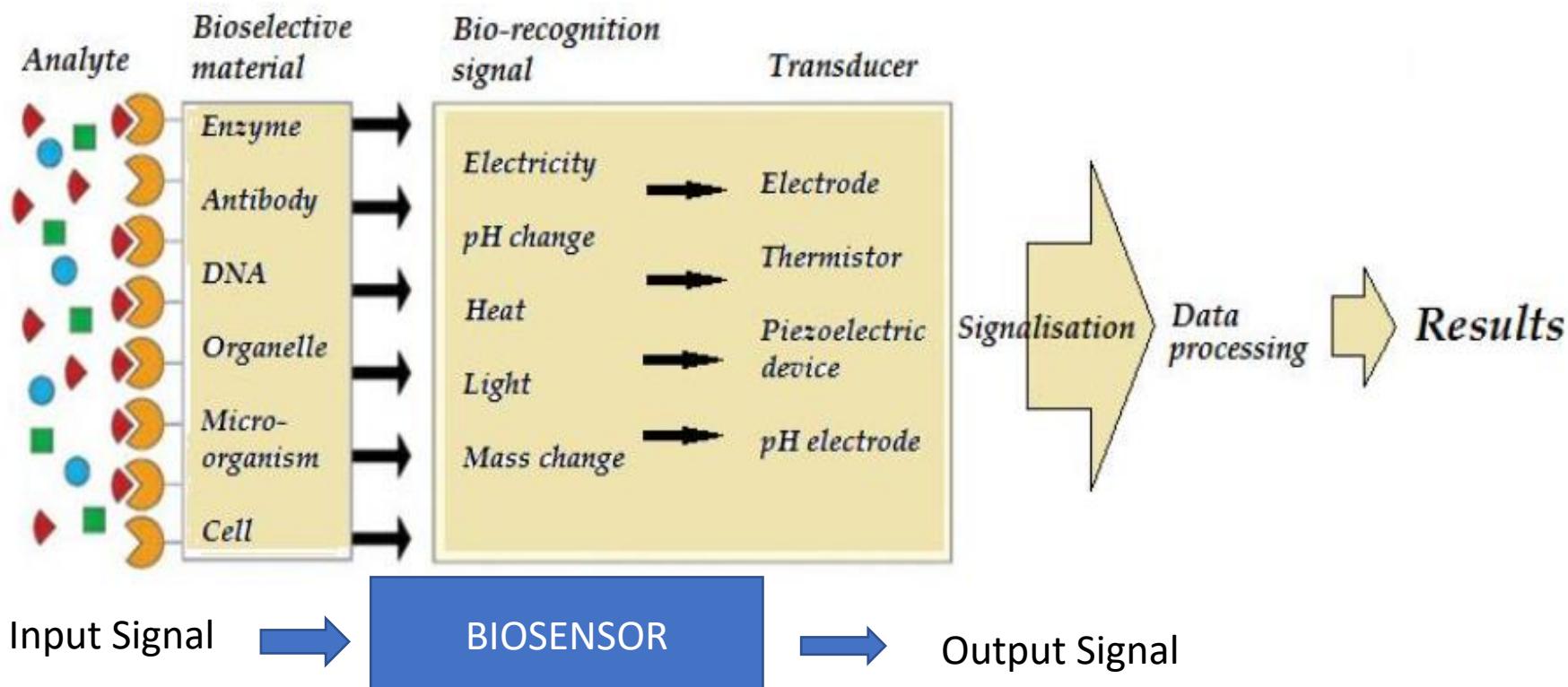
*Limitations*

# ENGINEERING CHEMISTRY

## Electrochemical Sensors-Biosensor- Glucose Sensor

### Biosensors

A biosensor is an analytical device which is used to determine the presence and concentration of a specific substance in a biological analyte.



## Blood Glucose Monitoring

- Blood Glucose Monitoring is a way of checking the concentration of glucose in the blood using a glucometer.
- Provides quick response to tell if the sugar is high or low indicating a change in diet, exercise or insulin.
- Over time, it reveals individual of blood glucose changes.



### Why Monitor Blood Glucose?

- Reduces risk of developing complications with diabetes.
- Allows diabetics to see if the insulin and other medications they are taking are working.
- Gives diabetics an idea as to how exercise and food affect their blood sugar.
- May prevent hypoglycemia or hyperglycemia



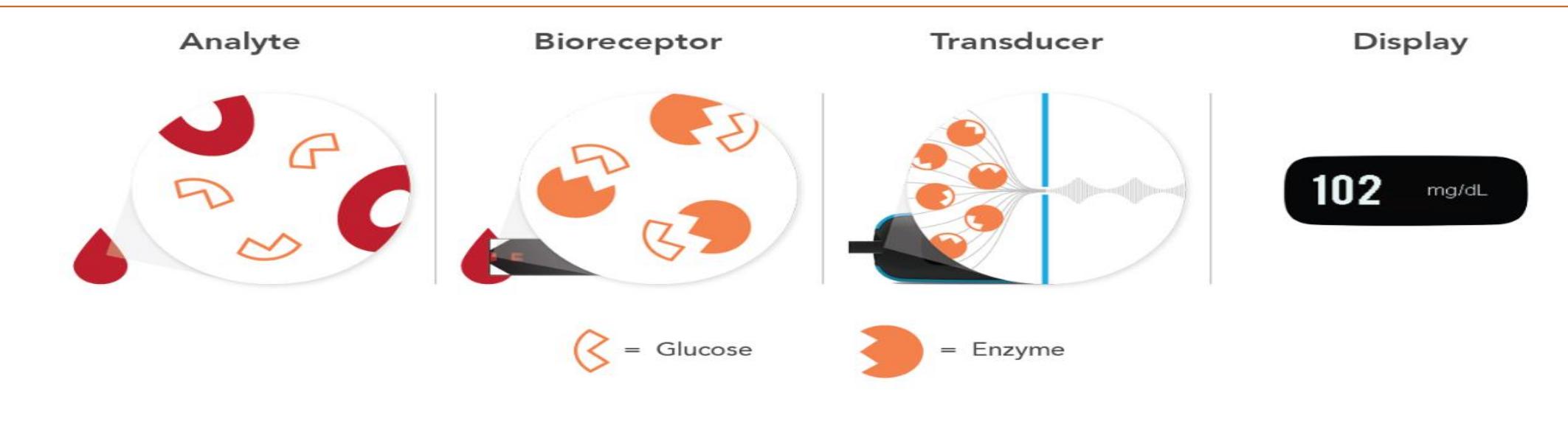
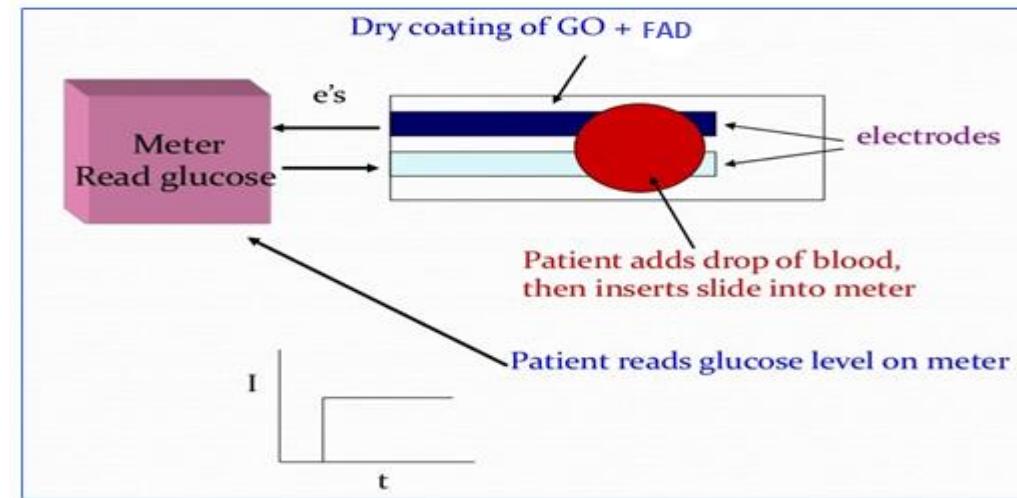
BLOOD SUGAR LEVEL CHART			
	FASTING	JUST ATE	3 HOURS AFTER EATING
NORMAL	80-100	170-200	120-140
PRE-DIABETIC	101-125	190-230	140-160
DIABETIC	126+	220-300	200+

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## Electrochemical Sensors-Biosensor- Glucose Sensor

### Components of Glucose Sensor

- Analyte
- Bioreceptor
- Transducer
- Electronics and display



### Working of Glucose Biosensor

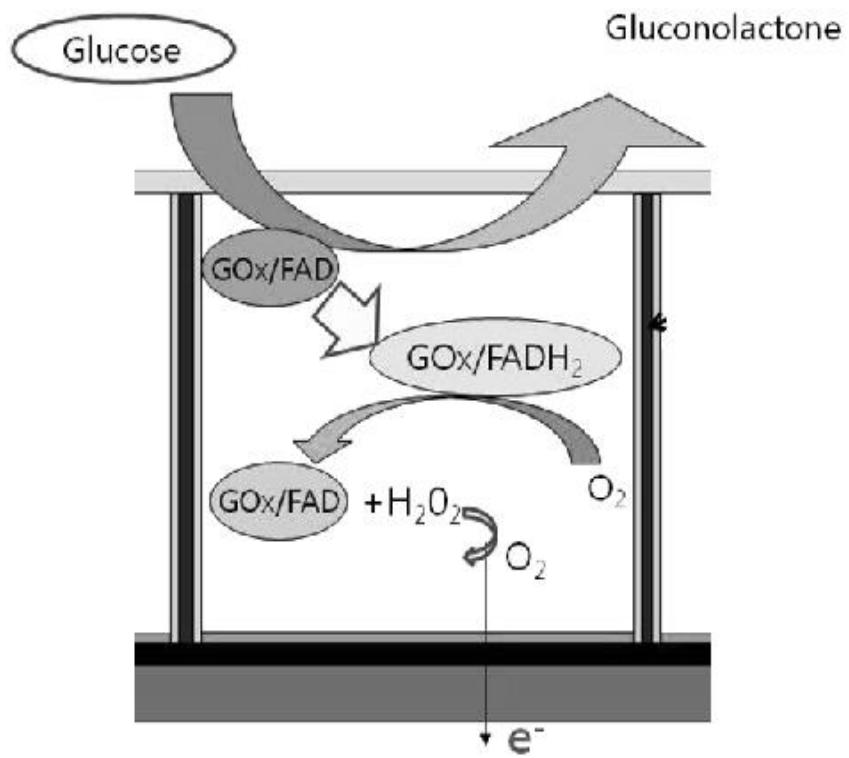
- Developed by Updike and Hicks
- Enzyme Glucose oxidase(GOx) catalyze the oxidation of glucose by molecular oxygen producing glucolactone and hydrogen peroxide.
- In order to work as a catalyst, GOx requires a redox cofactor –flavin adenine dinucleotide (FAD), works as an initial electron acceptor and is reduced to FADH<sub>2</sub>.
- $\text{Glucose} + \text{GOx} - \text{FAD} \rightarrow \text{Glucolactone} + \text{GOx} - \text{FADH}_2$

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### Working of Glucose Biosensor

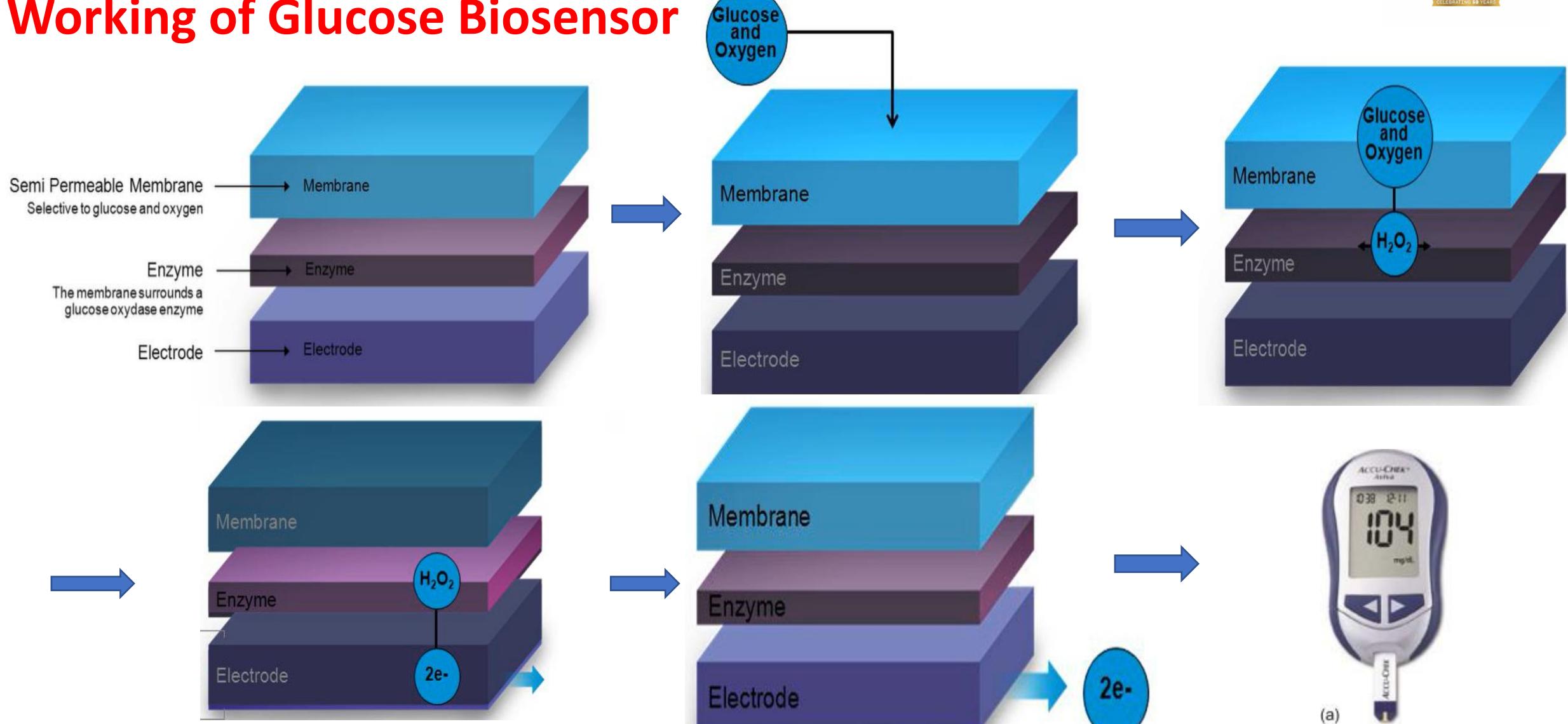
- The cofactor is regenerated by reacting with oxygen, leading to the formation of hydrogen peroxide
- $\text{GOx} - \text{FADH}_2 + \text{O}_2 \rightarrow \text{GOx} - \text{FAD} + \text{H}_2\text{O}_2$
- Hydrogen peroxide is oxidized at a platinum electrode.
- The number of electron transfers, at electrode surface is directly proportional to the number of glucose molecules present in the blood.
- $\text{H}_2\text{O}_2 \rightarrow 2\text{H}^+ + \text{O}_2 + 2\text{e}^-$



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### Working of Glucose Biosensor



### Limitations of Glucose Sensors

- Extreme environmental conditions like haematocrit values, or medication interferences may potentially falsify blood glucose readings.
- **Limited selectivity-** interference from ascorbic acid and uric acid



**THANK YOU**

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