

05 FRIDAY

095-270 • WK 18

APRIL

2 Thousand 19

— After Auto Zero put the Potassium Dichromate solution labeled as solution (A) and Press Start Key

— Now take absorbance at 430 nm for solution (B)

— Tolerance is given in below table

S.No	Wavelength	Absorbance	maximum tolerance
1	235	124.5	122.9 to 126.2
2	257	144.0	142.8 to 145.7
3	313	48.6	47.0 to 50.3
4	350	106.6	104.9 to 108.2
5	430	15.9	15.7 to 16.1

⑤ Limit of Stray Light  $\Rightarrow$

1- Dry of the Potassium Chloride by heating to constant weight at  $130^{\circ}\text{C}$

$\Rightarrow$  1.20 gm of dried Potassium Chloride — 100ml  
 — 100ml D.K. makeup to distilled water

$\Rightarrow$  Select the method file of Limit of Stray Light in the instrument

$\Rightarrow$  Selecting the file Press Reference button for Baseline correction

APR-2019

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1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 . . . . .



S.No	Wavelength nm
1	198
2	199
3	200
4	201
5	202

to be used  
 Absorbance  
 Greater than 2  
 SATURDAY  
 APRIL 096-263 • WK. 14

06

⇒ check the absorbance of above solution using water as a blank at 200 nm

#### ④ Resolution Power ⇒

⇒ Prepare 0.02% solution of toluene in Hexane UV

⇒ Select the method file of Resolution Power in the instrument

⇒ After selecting the file Press Reference button for baseline correction

⇒ measure the absorbance of above solution at 266 nm and 269 nm using Hexane UV as Blank solution

⇒ The Ratio of absorbance maxima at 269 nm to that of 266 nm minima should be more than 1.5

⇒ Note down the Report to the internal calibration certificate and in Instrument log book.

SUNDAY 07

2020-09-14 00:00:00

MAY

JUNE

T F S S M T W T F S S

08

MONDAY

098-267 • WK 15

APRIL

2 Thousand 19

① Control of wavelength  $\Rightarrow$  1.0 gm of Helium oxide  
 1.4 m Perchloric Acid  $\rightarrow$  25 ml v.f.

② Control of absorbance  $\Rightarrow$  Dry 130°C Potassium dichromate.  
 60 mg  $\rightarrow$  1000 ml v.f (A), 60 mg  $\rightarrow$  100 ml v.f (B)  
 0.005 M Sulfuric Acid makeup.

③ Limit of stray light  $\Rightarrow$  Dry 130°C Potassium chloride  
 1.20 gm  $\rightarrow$  100 ml v.f makeup distilled water

④ Resolution Power  $\Rightarrow$  0.02% solution of Toluene in Hexane!



1. Pencil 2.34

The Permitted tolerance is given in below table

S.No	maxima wavelength (nm)	tolerance (nm)
1	241.15 nm <del>241.15 nm</del>	240.15 to 242.15 nm
2	207.15 nm	206.15 nm to 208.15 nm
3	361.5 nm	360.50 to 362.50 nm
4	536.3 nm	533.30 to 539.30 nm

## ① Control of Absorbance

- Dry a quantity of the Potassium dichromate by heating to constant weight at  $130^{\circ}\text{C}$ .
- 60 mg of dried Potassium dichromate - 1000 ml 100%  $\text{H}_2\text{SO}_4$  make up to 0.005 M Sulfuric Acid Solution. (A)
- 60 mg of dried Potassium dichromate - 1000 ml 100%  $\text{H}_2\text{SO}_4$  make up to 0.005 M  $\text{H}_2\text{SO}_4$  Solution (B)
- Blank - 0.005 M  $\text{H}_2\text{SO}_4$  Solution in Cuvette and Press Reference to Zero.



03

WEDNESDAY

APRIL

093-272 • WK 14

2 Thousand 19

U.V. Spectro Photo meter CUV Poros 2.34

U.V. Principal  $\rightarrow$  Beer Lambert Law

U.V. Ray = 200  $\rightarrow$  400 nm (near visible)  
 Visible Ray = 400  $\rightarrow$  800 nm

$$A \propto CL$$

A = Absorbance

C = Concentration

L = Optical path length

Calibration  $\Rightarrow$ 

- ① Control of wavelength - 1.0 gm Holmium oxide
- ② Control of Absorbance
- ③ Limit of stray light
- ④ Resolution Power

① Control of wavelength  $\Rightarrow$ 1.0 gm of Holmium oxide  $\rightarrow$  25ml v.v

makeup to 1.4m Perchloric Acid

- Select the method file of Control of wavelength in the instrument
- After selecting the file Press Reference button for Baseline Calibration

APR 2019

M T W T F S S M T W T F S S M T W T F S S M T W T F S S  
 1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 . . . . .

SATURDAY  
MARCH

082-283 • WK. 12

Dissolution (DR)

इसमें  
100% है  
एक यूनान है जिसे हमारी तराई में  
समय में Dissolve हो रही है

$\Rightarrow T_{\text{temp.}} = 37^{\circ}\text{C} \pm 0.5^{\circ}\text{C}$

⇒ Rpm Check = Digital tachometer

⇒ tolerance 4µm = C or D (80%)

Note  $\Rightarrow$  45 minutes  $\xrightarrow{\text{or}}$  45/25 Realize  $\Rightarrow$  100% of it

$\Rightarrow$  Prednisone  $\rightarrow$  Kortikosteroid  $\rightarrow$   $10\text{ mg}$

⇒ RPM Collection = 4%

IP  $\Rightarrow$  ① Parallel ② Basket

USP  $\Rightarrow$  ① Basket ② Paddle

$\Rightarrow$  tablets  $\lim_{x \rightarrow \infty} \Rightarrow 75\%$

Calibration DP

Buffer - water (500 ml)

Time - 30 min Basket and Paddle  
Cost = Production

24.5.2017  $\Sigma$  = Produktische

Yaselski - Predictions is my subject

Unit Paddis  $\Rightarrow$  28 to 42

Basket  $\Rightarrow$  58 to 86



09 SATURDAY

2 Thousand 19

068-297 • WK 10 MARCH

molarity  $\Rightarrow$  molarity is defined as the number of moles of solute per liter of solution.

$$\text{molarity } m = \frac{\text{moles of solute}}{\text{liter of solution}}$$

molality  $\Rightarrow$  molality is defined as the number of moles of solute per kilogram of solvent.

$$\text{molality } m = \frac{\text{moles of solute}}{\text{kilogram of solvent}}$$

Normality  $\Rightarrow$  Normality is another section that

Normality  $\Rightarrow$  Normality is defined as the number of equivalents per liter of solution.

$$\text{Normality} = \frac{\text{Number of equivalent}}{\text{liter of solution}}$$

10 SUNDAY

# Tablets Types -

- ① Uncoated (Dispersible) Tablets
- ② Coated Tablets
- ③ Enteric Coated Tablets.

## Uncoated -

- ① Description -
- ② Average weight (20 Tablets)
- ③ Uniformity of weight (20 Tablets)

Not more than  $< 80\text{mg}$  - 10%  
 $+ 80\text{mg}$  to  $250\text{mg}$  - 7.5%  
 Above  $250\text{mg}$  - 5%

- ④ Dimensions - length  
 - width  
 - thickness

- ⑤ Hardness -  $\text{Kg/cm}^2$

- ⑥ Disintegration time -  
 Uncoated - NMT 15 minutes  
 Film coated - NMT 30 minutes  
 Sugar coated - NMT 60 minutes

Enteric coated - 2 Hrs 0.1 M HCl + (Buffer mix)

(11.45 gm  $\text{K}_2\text{H}_2\text{PO}_4$  + 19.28 gm  $\text{Na}_2\text{HPO}_4$ ) mix Phosphate  
 Buffer PH - 6.8 for 1 Hrs 60 minutes.

## ⑦ Dissolution -

$\text{K}_2\text{H}_2\text{PO}_4$  - potassium dihydrogen phosphate  
 $\text{Na}_2\text{HPO}_4$  - Disodium hydrogen phosphate

2020-09-14 00:00:00

APRIL

MAY

JUNE

ACCOUNT OPENED

S S M T W T F S S  
 27 28 29 30 . . . . .



12

TUESDAY

2 Thousand 19

071-294 • WK. 11

MARCH

Capsule -

Size - '0' and '2'

① Description

② Average weight

③ Uniformity of weight

Less than 300mg -  $\pm 10\%$ Above 300mg -  $\pm 7.5\%$ 

④ Disintegration time - NMT 30 minutes

⑤ Dissolution.

Liquid (Syrup - Suspension)

① Description

② pH

③ Weight per ml

④ Assay.

MAR-2019

M	T	W	T	F	S	S	M	T	W	T	F	S	S	M	T	W	T	F	S	S	M	T	W	T	F	S	S			
1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31

# ALCOA

- A - Attributable
  - L - Legible
  - C - contemporaneous
  - O - Original
  - A - Accurate
- when  
हैं  
लिखित  
कनटेम्पोरनीयस  
ऑरिजनल  
who performed

# PLUS

- C - Complete
- C - consistent
- E - Enduring
- A - Available



TOC  $\Rightarrow$  River water, cleaning validation  
Pw, WFI,

Limit  $\Rightarrow$  500 PPB NMT

Calibration  $\Rightarrow$  Potassium hydrogen phthalate (KHP)

System Suit  $\Rightarrow$  Every week / every day

(Sucrose / 1-4 Benzoin)

(Limit 85 to 115%)

1 billion =  $10^9$

1 million =  $10^6$

1 trillion =  $10^{12}$

1 quadrillion =  $10^{15}$

(mg) solution 1000 mg solution 1000

1000 mg solution 1000 mg solution 1000

1000 mg solution 1000 mg solution 1000

1000 mg solution 1000 mg solution 1000

1000 mg solution 1000 mg solution 1000

17

THURSDAY

017-348 • WK 03

JANUARY

2 Thousand 19

L-lysine

Std

TEST

50mg — 100 ml volumetric

10ml rest/ml — 100 ml v.f.

2 ml — 100 ml v.f.

10 ml — 100 ml v.f.

makeup with water

CD

5 ml Std Solution — 25 ml v.f.

5 ml sample — 25 ml v.f.

3 ml — Dye

3 ml — Dye

(150mg Nistigobin + 5 ml methanol + 5 ml pyridine)

(v.f. water bath 30 min)

cool + makeup with water

d = 570

N-2019

T	W	T	F	S	S	M	T	W	T	F	S	S	M	T	W	T	F	S	S	M	T	W	T	F	S	S							
1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	.	.	.



29

TUESDAY  
JANUARY

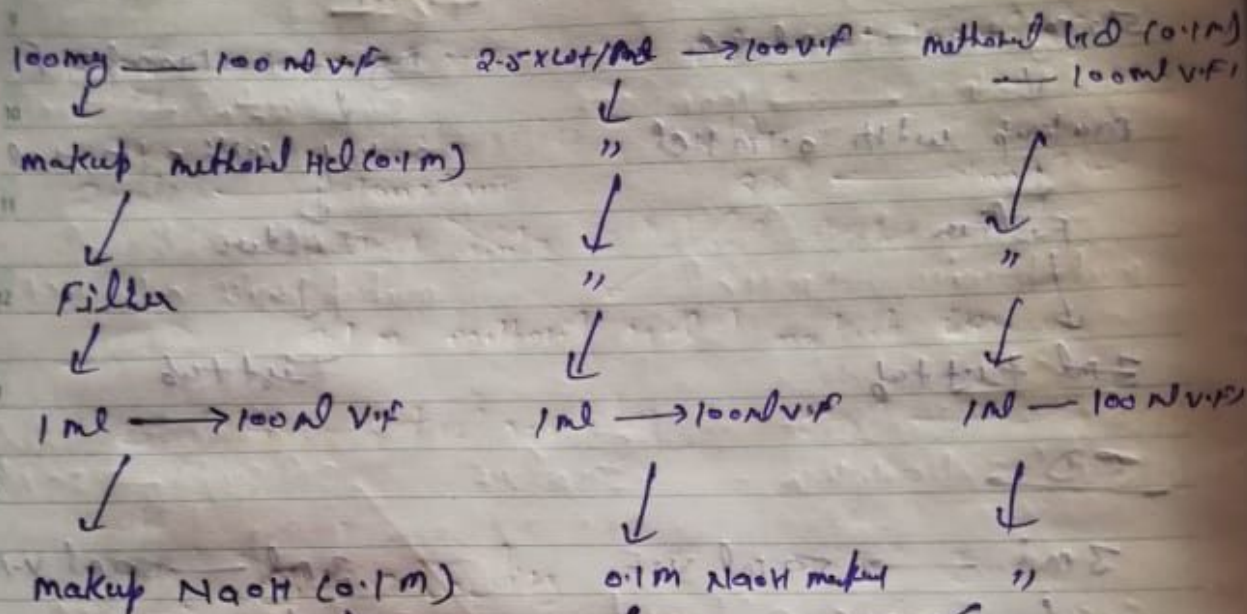
2 Thousand 19

Albendazole

Std

Sample

Blank



methanol HCl 0.1M = (4.25 ml  $\rightarrow$  500 ml v/v)  
HCl + methanol makeup

NaOH 0.1M = (2.1 gm  $\rightarrow$  500 ml v/v)  
water makeup.

JAN-2019

M T W T F S S M T W T F S S M T W T F S S M T W T F S S