



The research of the relation between the dielectric permittivity and the polarizability

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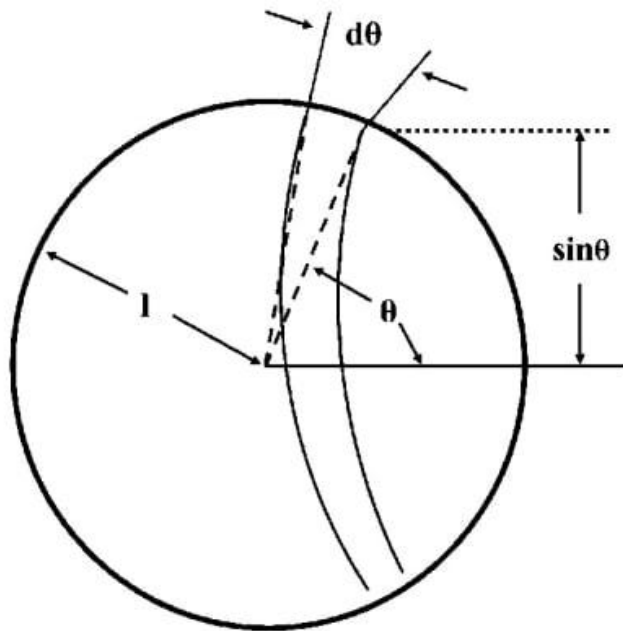
Introduction

Problem statement

Develop a theoretical model that describes the relationship between the **dielectric permittivity** of a substance and its **polarisability**.

Assemble the installation and conduct an experiment with a liquid with **non-polar** molecules and check whether the experimental data are described by the obtained law.

Theoretical model



Consider the polarization density.

$$P = N(\alpha_e + \alpha_i + \frac{p^2}{3\epsilon T})\epsilon_0 E \quad (1)$$

It is determined by three factors:

- α_e - electron polarization e
- α_i - ionic polarizability
- N - number of molecules per unit volume
- E - electric field acting on the molecule

$$dS = 2\pi \sin(\theta) d\theta$$

$$dN = A e^{pE \cos(\theta)/\epsilon T} \sin(\theta) d\theta$$

$$A = \frac{N}{\int_0^\pi e^{pE \cos(\theta)/\epsilon T} \sin(\theta) d\theta}$$

$$dP = p \cos(\theta) dN$$

$$P = \int_{\theta}^{\pi} dN p \cos(\theta) = Ap \int_{\theta}^{\pi} e^{pE \cos(\theta)/\epsilon T} \sin(\theta) \cos(\theta) d\theta$$

$$P = Np \frac{\int_{\theta}^{\pi} e^{pE \cos(\theta)/\epsilon T} \sin(\theta) \cos(\theta) d\theta}{\int_{\theta}^{\pi} e^{pE \cos(\theta)/\epsilon T} \sin(\theta) d\theta}$$

By replacing of

$$\begin{aligned} b &= pE/\epsilon T \\ x &= \cos\theta \\ dx &= -\sin(\theta) d\theta \end{aligned}$$

The previous equation simplifying to the form below:

$$P = Np \frac{\int_{-1}^1 e^{bx} x dx}{\int_{-1}^1 e^{bx} dx}$$

Integrating by parts, we get:

$$P = Np(\coth(b) - \frac{1}{b}) = NpL(b) \quad (2)$$

$L(b) = \coth(b) - \frac{1}{b}$ — the Langevin function introduced by Langevin in 1905

At room temperature, for typical electric fields in laboratories, $pE \ll \epsilon T$

So, we get:

$$L(b) \approx \frac{b}{3} = \frac{pE}{3\epsilon T}$$

Via Eq. (2) we get relation for the average dipole moment below:

$$P = \frac{Np^2 E}{3\epsilon T}$$

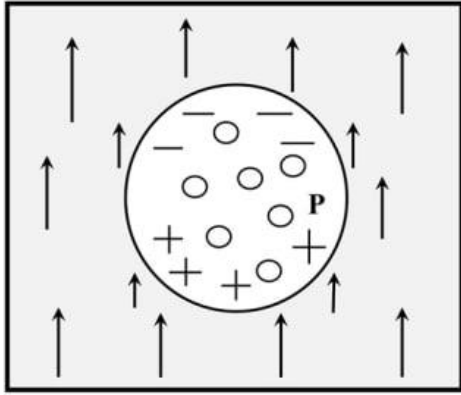


Fig. 2. Parallel plate capacitor

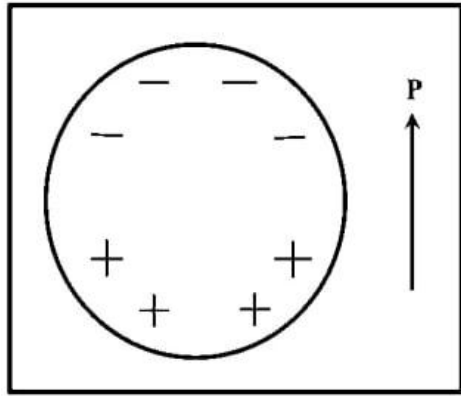


Fig. 3. Spherical cavity

The orientational polarizability per molecule:

$$\alpha_0 = \frac{P/N}{E\epsilon_0} = \frac{p^2}{3\epsilon\epsilon_0 T}$$

The electric susceptibility $\chi = \epsilon - 1$, also,

$$P = \epsilon_0 \chi E$$

We will get via substituting Eq. (1):

$$\chi = \epsilon - 1 = N\left(\alpha_e + \alpha_i + \frac{p^2}{3\epsilon T}\right)$$

Figures 2 and 3 will help to understand the total electric field in a dielectric better:

$$E_{total} = E_1 + E_2 + E_3$$

A uniformly polarized dielectric sphere creates an electric field:

$$E_p = -\nabla\phi = -\frac{P}{3\epsilon_0}$$

The charge distribution is the same, so:

$$E_2 = -E_p = \frac{P}{3\epsilon_0}$$

It has been shown that

$$E_3 = 0$$

for materials whose atoms have a simple cubic lattice structure.

Then we will have:

$$E_{total} = E_1 + \frac{P}{3\epsilon_0}$$

We defined the polarizability of a molecule of the dielectric:

$$p_{total} = \epsilon_0 \alpha E_{total}$$

$$P = N p_{total} = \epsilon_0 N \alpha E_{total} = \epsilon_0 N \alpha \left(E + \frac{P}{3\epsilon_0} \right) \qquad P = \frac{N\alpha}{1 - N\alpha/3} \epsilon_0 E$$

Thus, we obtain a relation linking the macroscopic dielectric constant with the microscopic polarizability:

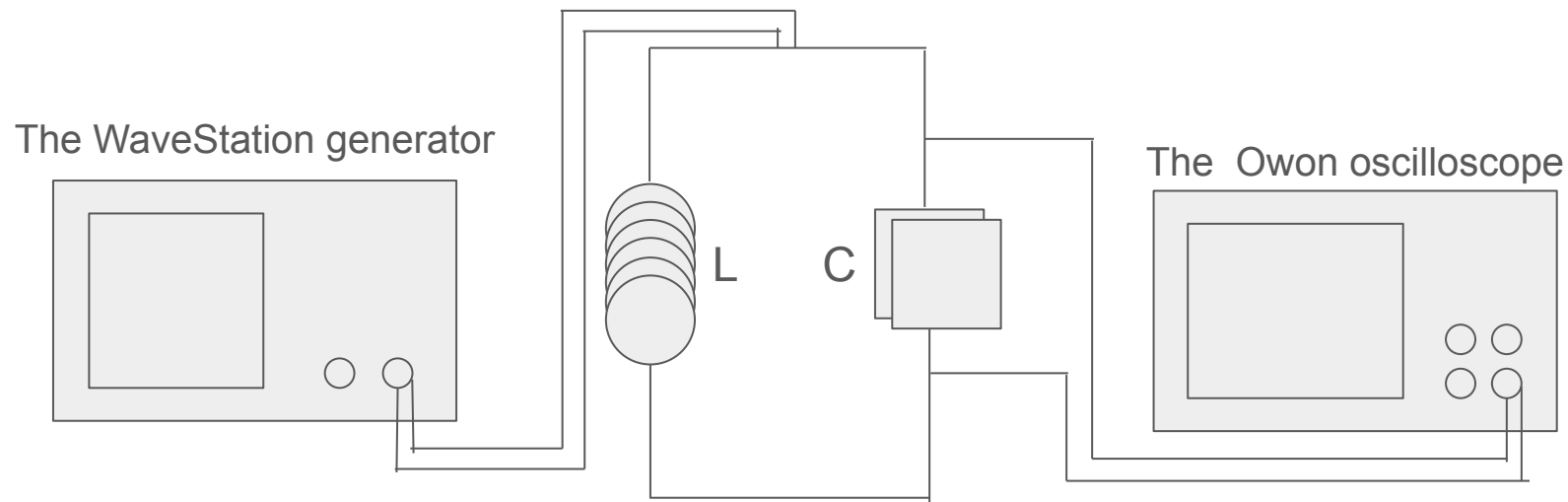
$$\chi = \epsilon - 1 = \frac{N\alpha}{1 - N\alpha/3} \longrightarrow \alpha = \frac{3}{N} \left(\frac{\epsilon - 1}{\epsilon + 2} \right)$$

We will have in the GHS system for a binary mixture of substances:

$$\frac{\epsilon - 1}{\epsilon + 2} = \frac{4\pi}{3} (N_1 \alpha_1 + N_2 \alpha_2)$$

The Clausius-Mossotti equation

Experimental setup



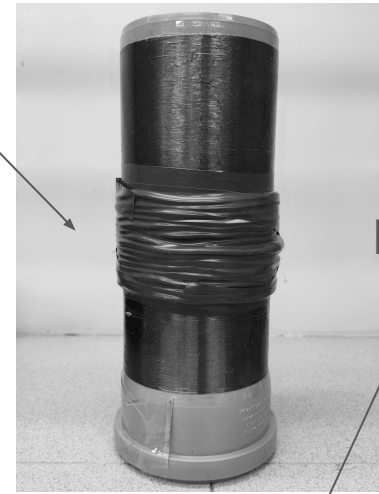
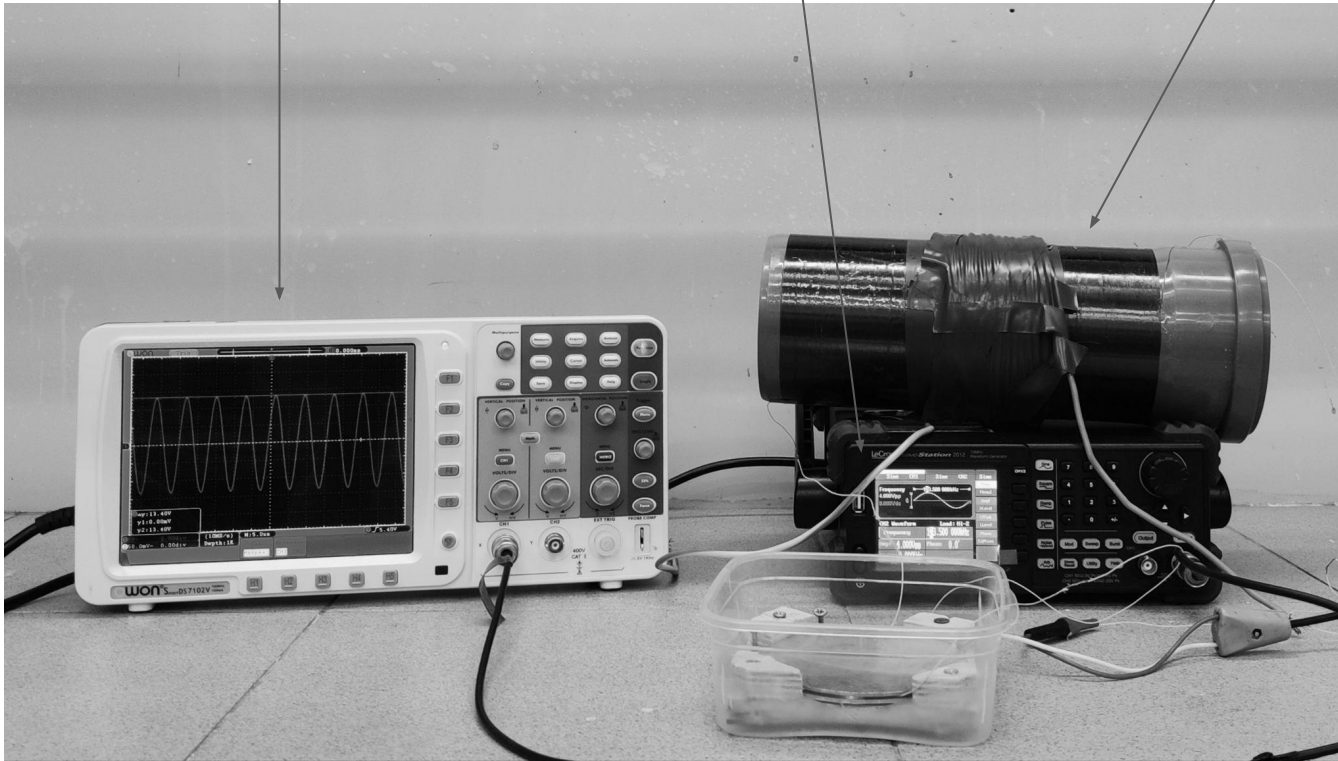
The oscilloscope

The generator

The coil

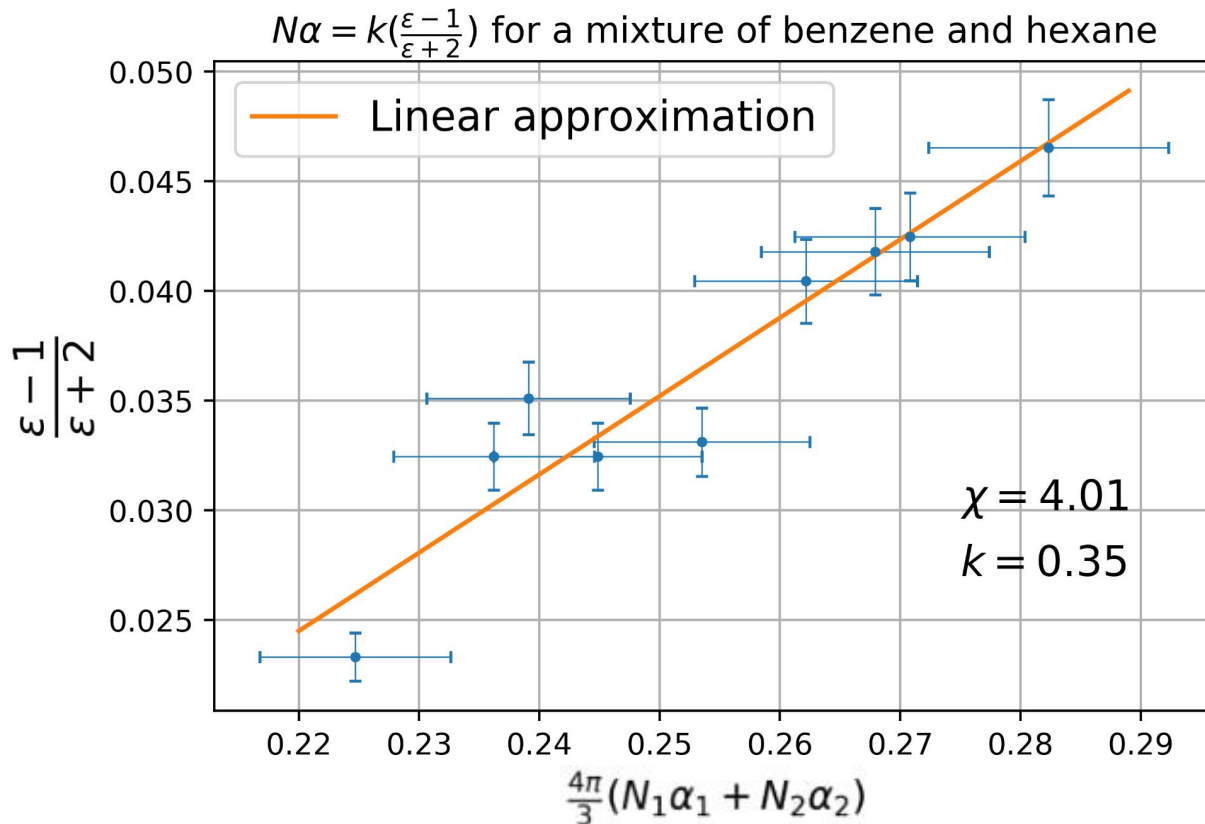
Fixing
sides

Capacitor plates

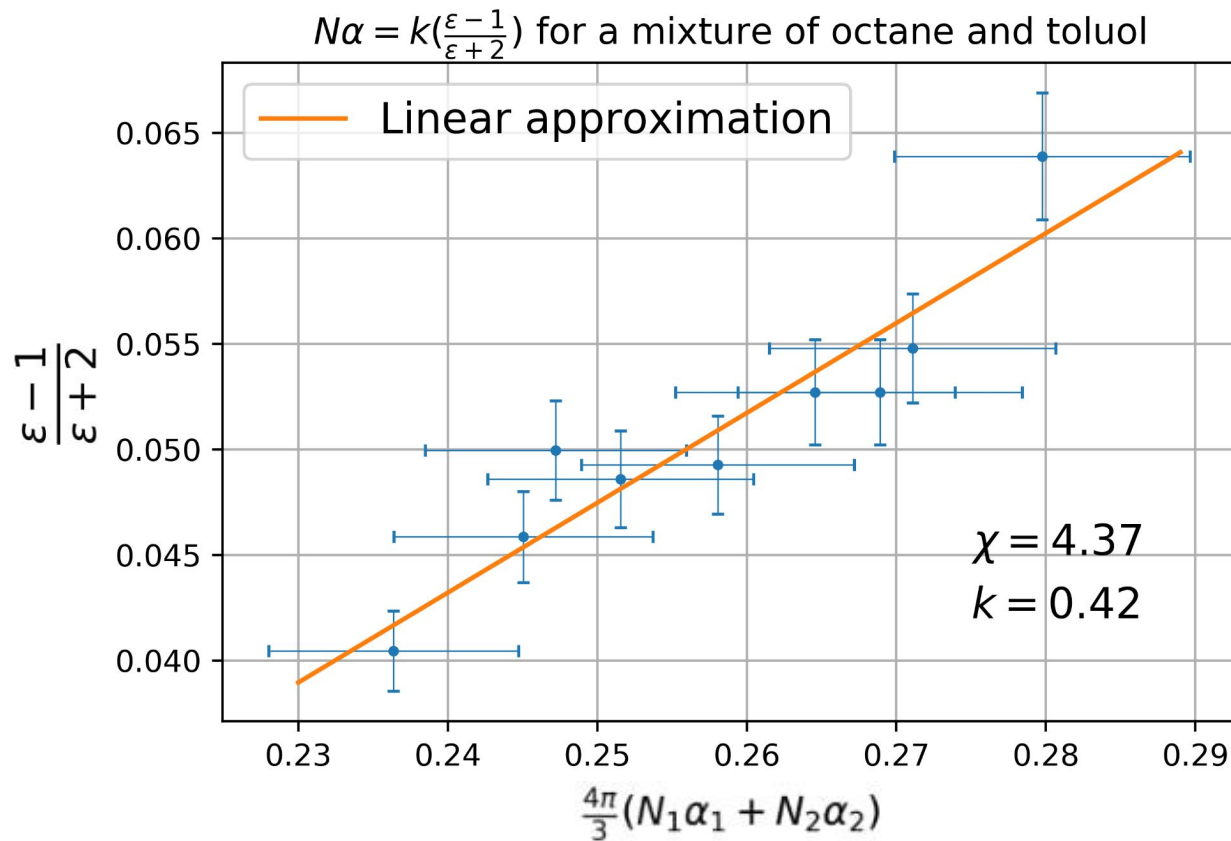


Experiment

A benzene and hexane mixture



An octane and toluol mixture



Results and analysis

Results:

- The dependence is **linear**, as expected from the theory
- **Similar results** were obtained for **two different** mixtures of liquids
- Experimental and theoretical data **converges in order**

$$k_1 = 0.35 \pm 0.08, k_2 = 0.42 \pm 0.08, k = 1$$

Analysis:

- Atoms have a **cubic lattice structure**
- The material under investigation is a **crystalline** substance

It is natural to assume that taking into account these corrections, we could see a **more** accurate **similarity** with the theory.

References

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**Thank you for your
attention!**