

# Quantum Chemistry by Levine

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Sep 2023

## 1 Chapter 4 The Harmonic Oscillator

### 4.1

It is a Taylor series.

$$f = \sum_0^{\infty} \frac{f^{(n)}(x=a)}{n!} (x-a)^n$$

### 4.2

given the equation in 4.1

$$(a) \sin x = x - \frac{x^3}{3!} + \frac{x^5}{5!}$$

$$\sin x = \sum_{k=0}^{\infty} (-1)^k \frac{x^{2k+1}}{(2k+1)!}$$

$$(b) \cos x = 1 - \frac{x^2}{2!} + \frac{x^4}{4!} - \dots$$

$$\cos x = \sum_{k=0}^{\infty} (-1)^k \frac{x^{2k}}{(2k)!}$$

### 4.3

given the equation in 4.1

$$(a) e^x = 1 + x + \frac{x^2}{2!} + \dots$$

$$(b) e^{ix} = 1 + ix - \frac{x^2}{2} + \dots = \cos x + i \sin x$$

$$e^{-ix} = 1 - ix - \frac{x^2}{2} + \dots = \cos x - i \sin x$$

summing up:

$$\cos x = \frac{e^{ix} + e^{-ix}}{2} = 1 - \frac{x^2}{2} + \dots$$

### 4.4

$$E = T + V$$

$$V = \frac{1}{2} k x^2$$

$$T = \frac{1}{2} m \left( \frac{dx}{dt} \right)^2$$

$$x = A \sin(2\pi \nu t + b)$$

Combining equations above:

$$E = \frac{1}{2} k A^2 = 2\pi^2 \nu^2 m A^2$$

#### 4.5

(a) The original equation is:

$$(1 - x^2)y'' - 2xy' + 3y = 0$$

Suppose:

$$y = \sum_{k=0}^{\infty} c_n x^n$$

$$c_{n+2} = \frac{n^2 + n - 3}{(n+1)(n+2)} c_n$$

$$(b) c_4 = \frac{3}{12} \frac{-3}{2} c_0 = \frac{-3}{8} c_0$$

$$c_5 = \frac{9}{20} \frac{-1}{6} c_1 = \frac{-3}{40} c_1$$

#### 4.6

(a) odd

(b) even

(c) odd

(d) neither

(e) even

(f) odd

(g) neither

(h) even

#### 4.7

Let  $f(x), g(x)$  are odd functions,  $k(x), l(x)$  are even functions.

$$f(x)g(x) = -f(-x) * [-g(-x)] = f(-x)g(-x)$$

$$k(x)l(x) = k(-x)l(-x)$$

$$f(x)k(x) = -f(-x)k(-x)$$

#### 4.8

(a) Given  $f(x) = f(-x)$

$$\frac{df(x)}{dx} = \frac{df(-x)}{dx} = \frac{df(-x)}{d(-x)}$$

$$\frac{df(x)}{dx} = -\frac{df(-x)}{d(-x)}$$

(b) Given  $f(x) = -f(-x)$

$$\frac{df(x)}{dx} = -\frac{df(-x)}{dx} = -\frac{f(x)}{dx} \frac{df(-x)}{dx} = \frac{df(x)}{dx}$$

(c) Given  $f(x) = f(-x)$

$$\frac{df(x)}{dx} = -\frac{df(x)}{dx}$$

$$f'(0) = 0$$

#### 4.9

(a) in the lowest energy state:

$$\Psi_0 = \frac{\alpha^{1/4}}{\pi} e^{-\alpha x^2/2}$$

$$<T> = \int_{-\infty}^{\infty} \Psi^* T \Psi d\tau$$

=

$$\frac{1}{2} \hbar \nu = \int_{-\infty}^{\infty} \Psi^* T \Psi dx = \int_{-\infty}^{\infty} \Psi^* \left( -\frac{\hbar^2}{2m} \frac{d^2}{dx^2} \right) \Psi dx$$

$$= \hbar \nu / 4$$

Given in this energy state  $E = \frac{1}{2} \hbar \nu$ :

$$\langle V \rangle = \frac{\hbar \nu}{4}$$

Elsewise we can also use the integration for  $\langle V \rangle$ :

$$\langle V \rangle = \int_{-\infty}^{\infty} \Psi^* T \Psi dx$$

where:  $T = 2\pi^2 \nu^2 m x^2$

$$\langle V \rangle = \frac{\hbar \nu}{4}$$

So:

$$\langle V \rangle = \langle T \rangle$$

#### 4.10

(a) for  $v = 1$

$$\Psi = c_1 x e^{-\alpha x^2/2}$$

Normalize it:

$$1 = \int_{-\infty}^{\infty} |\Psi|^2 dx$$

$$c_1 = \sqrt{2} \alpha^{3/4} \pi^{-1/4}$$

(b) for  $v = 2$

$$\Psi = (c_0 + c_2 x^2) e^{-\alpha x^2/2}$$

Given the recursive equation:

$$c_2 = -2\alpha c_0$$

For normalization:

$$c_0 = 2^{-1/2} \alpha^{1/4} \pi^{1/4}$$

#### 4.11

Given  $v = 3$

$$\Psi = (c_1 x + c_3 x^3) e^{-\alpha x^2/2}$$

From the recursion:

$$c_3 = -2\alpha c_1/3$$

For normalization:

$$1 = \int_{-\infty}^{\infty} |\Psi|^2 dx$$

$$|c_1| = \sqrt{3} \alpha^{3/4} \pi^{-1/4}$$

#### 4.12

for  $v = 4$

$$\Psi = e^{-\alpha x^2/2} (c_0 + c_2 x^2 + c_4 x^4)$$

using recursive equation:

$$c_2 = -4\alpha c_0$$

$$c_4 = \frac{4}{3} \alpha^2 c_0$$

For normalization:

$$1 = \int_{-\infty}^{\infty} |\Psi|^2 dx$$

$$\Psi_4 = c_0 e^{-\alpha x^2/2} (1 - 4\alpha x^2 + \frac{4}{3}\alpha^2 x^4)$$

#### 4.13

for  $v = 1$

$$\Psi = c_1 x e^{-\alpha x^2/2}$$

the max prob point is:

$$\frac{d|\Psi|^2}{dx} = 0$$

where  $x = 0$  is the min point

$x = \frac{1}{\sqrt{\alpha}}$  and  $x = -\frac{1}{\sqrt{\alpha}}$  are the max point

#### 4.14

for  $v = 5$

it is an odd function, and has 6 max/min points

#### 4.15

for quantum number  $v$ . the function:

$$\Psi = e^{-\alpha x^2/2} (c_0 + c_2 x^2 + c_4 x^4 + \dots) \text{ for even } v$$

$$\Psi = e^{-\alpha x^2/2} (c_1 x + c_3 x^3 + c_5 x^5 + \dots) \text{ for odd } v$$

$$\langle x \rangle = \int_{-\infty}^{\infty} \Psi^* x \Psi d\tau$$

#### 4.16

(a) T.

odd  $v$  means odd function

(b) T.

(c) F.

it can multiply by  $-1$  and the function still holds.

(d) T.

$$E = (v + \frac{1}{2})h\nu$$

(e) T.

there is only one  $v$  for each energy level.

#### 4.17

for PIB,  $n = n_0$  there are  $n_0$  max/min points

when  $n = 0$  we have  $E = 0$

for harmonic oscillator,  $v = 0, 1, 2 \dots$  has  $v + 1$  max/min points

when  $v = 0$ , we have  $E = \frac{1}{2}h\nu$

#### 4.18

(a) the classic equation:

$$x = A \sin(2\pi\nu t + b)$$

so:

$$t = \frac{1}{2\pi\nu} (\sin^{-1}(\frac{x}{A}) - b)$$

$$(b) \quad dt = f(x)dx$$

at the turning points, the speed of classic object is 0

So it means the object will stay static there

the time to stay there (prob density) is inf

#### 4.19

for  $x < 0$  it is PIB condition

$$E = \frac{n^2 h^2}{8ml^2}$$

for  $x \geq 0$  it is harmonic oscillator condition:

$$E = (\frac{1}{2} + v)h\nu$$

for continuity  $E(0) = 0$

So the harmonic oscillator has to be a odd function with  $v = 2k + 1$

So overall:

$$E = (2k + 1 + \frac{1}{2})h\nu$$

#### 4.20

(a) the Schrodinger equation becomes:

$$-\frac{\hbar^2}{2m} \nabla^2 \Psi + \frac{1}{2}(k_x^2 + k_y^2 + k_z^2) \Psi = E \Psi$$

if  $\Psi = f(x)g(y)h(z)$

then separate the variables:

$$-\frac{\hbar^2}{2m} \frac{1}{f} \frac{d^2 f}{dx^2} + \frac{1}{2} k_x^2 = E$$

formulate it to be:

$$-\frac{\hbar^2}{2m} \frac{d^2 f}{dx^2} + \frac{1}{2} k_x^2 f = E f$$

which is the same as one-dimension harmonic oscillator

$$E_x = (\frac{1}{2} + v_x)h\nu_x$$

overall:

$$E = (\frac{1}{2} + v_x)h\nu_x + (\frac{1}{2} + v_y)h\nu_y + (\frac{1}{2} + v_z)h\nu_z$$

(b) when  $k_x = k_y = k_z$

it means  $\nu_x = \nu_y = \nu_z = \nu$

$$E = (\frac{3}{2} + v_x + v_y + v_z)h\nu$$

the lowest energy is  $v_x = v_y = v_z = 0$  not degenerate.

$$E = \frac{3}{2}h\nu$$

So the lowest degeneracy is 1, 0, 0 or 0, 1, 0 or 0, 0, 1

#### 4.21

(a)  $n = 0, 1, 2, 3$  can approve

$$(b) \quad zH_n(z) = nH_{n-1}(z) + \frac{1}{2}H_{n+1}(z)$$

(c) can approve for  $v = 0$

$$\Psi_0 = \frac{\alpha^{1/4}}{\pi} e^{-\alpha x^2/2}$$

**4.23** (a) the function:

$$\Psi = e^{-\alpha x^2/2}(c_0 + c_2x^2 + c_4x^4 + \dots) \text{ for even } v$$

$$\Psi = e^{-\alpha x^2/2}(c_1x + c_3x^3 + c_5x^5 + \dots) \text{ for odd } v$$

$$(b) c_{n+2} = \frac{\alpha + 2\alpha n - 2mE\hbar^{-2}}{(n+1)(n+2)} c_n$$

**4.24**

(a) from classic equation:

$$\nu = \frac{1}{2\pi} \sqrt{\frac{k}{m}}$$

So here:

$$k = 4\pi^2\nu^2\mu \text{ where } \mu = \frac{m_1m_2}{m_1+m_2} = \frac{1}{N} \frac{1*35}{1+35}$$

$$k = 481N/m$$

(b) zero point vibration energy:

$$E = \frac{1}{2}h\nu = 2.87 * 10^{-20} J$$

$$(c) \nu = \frac{1}{2\pi} \sqrt{\frac{k}{m}}$$

$$\text{So } \frac{\nu_2}{\nu_1} = \frac{\sqrt{m_1}}{\sqrt{m_2}}$$

$$\text{where } m_2 = \frac{1}{N} \frac{2*35}{2+35}$$

$$\nu_2 = \sqrt{\frac{m_1}{m_2}} \nu_1 = 6.20 * 10^{13} s^{-1}$$

**4.25**

(a) the emission from  $v_1$  to  $v_2$ :

$$v_{light} = (v_2 - v_1)v_e - v_ex_e(v_2^2 - v_1^2 + v_2 - v_1)$$

in this problem:

from  $v_1 = 0$  to  $v_2 = 1$ :

$$v_{light} = v_e - 2v_ex_e = 2889.98 cm^{-1} * c$$

from  $v_1 = 0$  to  $v_2 = 2$ :

$$v_{light} = 2v_e - 6v_ex_e = 5667.98 cm^{-1} * c$$

$$v_e = 8.69 * 10^{13} s^{-1}$$

$$v_ex_e = 1.559 * 10^{12} s^{-1}$$

(b) from  $n = 0$  to  $n = 3$

$$v_{light} = 3v_e - 12v_ex_e$$

$$\frac{1}{\lambda} = \frac{v_{light}}{c} = 8346.00 cm^{-1}$$

**4.26**

$$(a) \Delta E = \frac{hc}{\lambda} = 6.626 * 10^{34} Js * 3 * 10^{10} cm/s * 1359 cm^{-1} = 2.7 * 10^{-20} J$$

this is nondegenerate, using Boltzmann distribution equation:

$$\frac{N_1}{N_0} = e^{-\Delta E/kT}$$

where  $k$  is the Boltzmann constant

when  $T = 298K$ :

$$\frac{N_1}{N_0} = 0.0014$$

when  $T = 473K$ :

$$\frac{N_1}{N_0} = 0.016$$

(b) similarly for ICl:

$$\Delta E = \frac{hc}{\lambda} = 7.57 * 10^{-21} J$$

where  $\lambda = 381cm^{-1}$

then  $\frac{N_1}{N_0} = 0.16$  at  $T = 298K$

$\frac{N_1}{N_0} = 0.31$  at  $T = 473K$

#### 4.27

$$E_{vib} = (\frac{1}{2} + v)h\nu - (v + \frac{1}{2})^2 hv_e x_e$$

given  $v_2 = v_1 + 1$

$$\frac{\Delta E}{h} = v_e - 2v_e x_e (v_1 + 1)$$