

Quantum Chemistry by Levine

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1 Chapter 1 The Schrodinger Equation

1.1

(a) F.

$$E = h\nu$$

(b) T.

$$c = \nu\lambda$$

(c) F.

Energy has different level. in quantumized way.

1.2

$$(a) E = h\nu = h * \frac{c}{\lambda} = 6.626 * 10^{-34} * 3 * 10^8 / 1064 * 10^{-9} = 1.867 * 10^{-19} J$$

$$(b) n = \frac{5 * 10^6 * 2 * 10^{-8}}{E} = 5 * 10^{17}$$

1.3

Energy of 1 mol photon:

$$E = \frac{hc}{\lambda} * N = \frac{6.02 * 10^{23} * 6.626 * 10^{-34} * 3 * 10^8}{300 * 10^{-9}} = 399 kJ$$

1.4

$$(a) T = \frac{hc}{\lambda} - \Phi = \frac{6.626 * 10^{-34} * 3 * 10^8}{200 * 10^{-9}} - 2.75 * 1.602 * 10^{-19} = 5.53 * 10^{-19} J$$

(b) It only needs to give energy as large as Φ :

$$\lambda = \frac{hc}{E} = \frac{6.626 * 10^{-34} * 3 * 10^8}{2.75 * 1.602 * 10^{-19}} = 451 nm$$

$$(c) T = E - \Phi$$

The left kinetic energy is larger than pure Na.

1.5

(a) Planck's blackbody equation:

$$B_\nu(\nu, T) = \frac{2h\nu^3}{c^2} * \frac{1}{\exp(\frac{h\nu}{k_B T}) - 1}$$

Get its derivative to ν :

$$\frac{dB_\nu(\nu, T)}{d\nu} = 3(e^u - 1) - ue^u$$

where:

$$u = \frac{h\nu}{k_B T}$$

At peak wavelength:

$$3(e^u - 1) - ue^u = 0$$

then:

$$\nu = \frac{2.8214k_B T}{h}$$

$$(b) e^u = 1 + u$$

1.6

$$\lambda = \frac{h}{mV} = \frac{6.626 \cdot 10^{-34}}{9.109 \cdot 10^{-28} \cdot \frac{1}{137} \cdot 3 \cdot 10^8} = 0.332 nm$$

1.7

$$x = -\frac{1}{2}gt^2 + (gt_0 + v_0)t + C$$

1.8

Using Schrodinger equation:

$$-\frac{\hbar}{i} * \frac{\partial \Phi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Phi}{\partial x^2} + V\Phi$$

Given:

$$\Phi = ae^{-ibt}e^{-bmx^2/\hbar}$$

$$V = 2b^2mx^2$$

1.9

(a) F.

Not all has a phase term. Not separable. Not stationary.

(b) F.

Not all Ψ can be separated to $F(x)G(\psi)$

1.10

Given the postulation of E : $-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + V\Psi = E\Psi$

$$E = \frac{3\hbar^2 c}{m}$$

1.11

Only time-dependent

1.12

(a) The infinitesimal is small enough:

$$Prob = |\Psi|^2 dx = 3.29 \cdot 10^{-6}$$

(b) The distance is not small enough:

$$Prob = \int_0^{2nm} |\Psi|^2 dx = 0.0753$$

(c) $x = 0$

$$(d) \int_{-\inf}^{\inf} |\Psi|^2 dx = 1$$

1.13

The infinitesimal is small enough not to use integral:

$$|\Psi|^2 dx = 0.000216$$

1.14

force to use integral:

$$\int_{15nm}^{1.5001nm} |\Psi|^2 dx = 4.978 * 10^{-6}$$

Almost the same

1.15

- (a) not real
- (b) negative
- (c) can find a plausible b

1.16

- (a) Not independent: $\frac{1}{3}$
- (b) Independent: $\frac{1}{2}$

1.17

the prob of 138 peak:

$$0.9889^2$$

the peak of 139 is $C_{12}C_{13}F_6$

$$: 0.9889 * 0.0111 * 2 / (0.9889 * 0.9889) * 100 = 2.24$$

the peak of 140 is $C_{13}^2 F_6$:

$$0.0111^2 / (0.9889 * 0.9889) * 100 = 0.0126$$

1.18

26 cards of 2 spade, 24 non-spade:

one getting 13 non-spade:

$$\frac{24}{26} * \frac{23}{25} * \dots * \frac{12}{14} = \frac{6}{25}$$

the prob of getting one spade to each person is:

$$1 - \frac{6}{25} * 2 = \frac{13}{25}$$

1.33

- (a) T.
prob density always positive.
- (b) F.
function can have negative value.
- (c) F.
function can have imaginary value, like oscillating.
- (d) T.

- (e) F.
To all normalized system.
- (f) T.