Comparison of chromatographic stationary phases using Bayesian-based multilevel modeling

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ABSTRACT: We used a Bayesian multilevel model of chromatographic retention to compare five RP-HPLC stationary phases: XBridge Shield RP18, XTerra MS C18, XBridge Phenyl, XBridge C8 and Xterra MS C8. For that purpose we used a large dataset of retention times collected using chromatographic techniques coupled with mass spectrometry detection. The experiments were conducted in a gradient mode for an initial mixture of 300 small analytes for a wide range of pH values, in MeOH and ACN, at two temperatures, and for three gradient durations. Our analysis is based on a mechanistic model derived from principles and fundamentals of liquid chromatography and utilizes the literature prior knowledge about chromatographic parameters. The data and model allowed us to characterize the between column differences in chromatographic parameters of neutral, acidic, and basic analytes. The analysis provides an interpretable summary of stationary phase properties that can be used in decision-making, i.e. finding the best chromatographic conditions given limited experimental data. The proposed approach seems to be an interesting alternative to existing approaches used to compare chromatographic stationary phases.

The selection of chromatographic stationary phases is an important aspect of analytical work. Over the years, many methods and approaches has been proposed to classify, select, and characterize chromatographic stationary phases.1–4 All the current methods use various sets of probe analytes and a set of chromatographic conditions to estimate the most important chromatographic characteristics, e.g. column efficiency, hydrophobicity, silanol activity, ion-exchange capacity and steric selectivity and/or to calculate various similarly metrics. Usually, the comparison is based on empirical and simple models. This limits the practical usefulness of the derived comparison metrics in extrapolation to other analytes, other chromatographic conditions, and other stationary phases than those used in a particular method. Also a very detailed physical (and thermodynamic) description of the chromatographic systems is possible using enthalpies and entropies of transfer or absorption isotherms, nevertheless it requires extensive sets of carefully controlled experiments and as such are possible for few selected problems.5

The chromatographic stationary phases can also be compared based on retention time data collected for a group of preferably structurally heterogenous compounds. Nowadays, it is possible to collect such large datasets using chromatographic techniques coupled with mass spectrometry detection. Since such datasets carry much information about chromatographic retention, they can be used to determine a detailed multidimensional characterization of chromatographic stationary phases, to build a mechanistic model, and to predict retention time based on various number of preliminary experiments (e.g. to predict retention time for a set of analytes given several measurements collected using a different stationary phase). In principle the model build based on such data can provide predictions that generalize to other analyte, stationary phases and chromatographic conditions. The retention time predictions in liquid chromatography that are accurate or with well calibrated uncertainties are required for rapid column screening, computer-assisted method development and method transfer.6

In this work we applied the previously developed Bayesian multilevel framework7–10 to characterize chromatographic gradient retention time datasets collected using a multicomponent mixtures of analytes, five stationary phases (XBridge Shield RP18, XTerra MS C18, XBridge Phenyl, XBridge C8 and Xterra MS C8), and a wide range of chromatographic conditions (pH, organic modifier, temperature, gradient program). The general idea was to develop a mechanistic model of chromatographic retention that jointly characterizes the retention of available data, allows for estimating the stationary phase (column) effects on chromatographic parameters for acids, bases and neutral analytes, quantifies various sources of variation, and characterizes uncertainty. The usefulness of the model and results in decision-making was illustrated by searching chromatographic conditions leading to the desired separation given access to different types of preliminary data, e.g. predicting retention on the tested columns given XBridge Shield RP18 data. This part was based on statistical decision theory.

# EXPERIMENTAL SECTION

## Data

The data were collected by performing 84 different liquid chromatography experiments using an initial mixture of 300 analytes. The experiments differed with respect to gradient duration (30, 90, and 270 min), pH of the mobile phase (from 2.5 to 10.5), type of organic modifier (methanol (MeOH) or acetonitrile (ACN)) and column temperature (25 °C and 35 °C).

Briefly the liquid chromatography experiments were carried out using an Agilent Technologies 1260 Infinity system (Agilent Technologies, Waldbronn, Germany) and a 6224 time of flight (TOF) mass spectrometer with a dual electrospray ionization source (Dual ESI) in positive polarity, using an XBridge Shield RP18, XTerra MS C18, XBridge Phenyl, XBridge C8 and Xterra MS C8 Waters Ltd., Milford, MA, USA, 3 mm × 50 mm, 2.5 μm). The extra column volume (*Ve*) and system dwell volume (*Vd*) equaled 0.020 mL and 1.05 mL, respectively. The column hold-up volume (*V0*) was 0.266 mL 0.271, 0.271, 0.276, 0.284 mL and the flow rate (*F*) was 0.5 mL/min.

Ammonium bicarbonate, ammonium acetate and ammonium formate were selected as buffers to control the pH of the mobile phase during chromatographic separation. The pH of the buffers (nominal aqueous pH) was adjusted to the desired pH (ammonium formate: 2.5, 3.3, 4.1, 8.9, and 9.7; ammonium acetate: 4.9 and 5.8; and ammonium bicarbonate: 6.8 and 10.5) by an appropriate addition of formic acid, acetic acid and ammonia, respectively. The pH was measured at 25 °C and 35 °C using an S220 pH meter (Mettler Toledo, Greifensee, Switzerland) with an InLab® Routine Pro ISM electrode after mixing an organic modifier with the buffer solution.

The MassHunter Profinder B.08.00 (Agilent Technologies, Waldbronn, Germany) was selected to find all the matches per formula using “Batch Targeted Feature Extraction” (containing 300 predefined mases for each analyte included in the mixture). The data for analysis were restricted to analytes that had “IdentificationScores” higher than 95%, that were present on at least half of chromatograms (total and per column), and that had less than 2 dissociation steps in a pH range from 2 to 11.

The functional groups and structural elements were determined using Checkmol (version 0.5b N. Haider, University of Vienna, 2003-2018).11 The lipophilicity (*log P*), dissociation constant (*pKalit*) were calculated using the ACD/Labs program12 based on the structures of analytes generated from SMILES strings.

More details about data extraction can be found in our previous work.7 Raw data for selected analytes is shown in Figure S1.

## Structural Model

The details of the model are provided in supplementary material. Briefly, a standard chromatographic model was employed in this work.13,14 For each analyte the effect of pH was accounted for by the function describing the relationship between the isocratic retention factor and *pH* for an analyte with *R* dissociation steps and *R+1* forms15:

(1)

where *r* represents thedissociation step, denotes the *rth* dissociation constant, andrepresents the retention factor of a particular form of the analyte in a given chromatographic condition. Furthermore, it was assumed that depends on the organic modifier content, pH and temperature according to the following equation:

(2)

where represents the logarithm with the base of 10 of retention factors extrapolated to 0% of organic modifier content at 25 °C for mobile phase pH 7 for the neutral and dissociated forms of the analyte; and denote slopes in the Neue equation; denotes the change in due to the increase in temperature by 10 °C, *apHr* denotes pH effects for cations and anions. In this parametrization of the Neue equation, the *S1* parameter reflects the difference between the logarithm of retention factors between the 0% organic modifier content and 100% organic modifier content.

Furthermore, a linear relationship between *pKa* values and organic modifier content was assumed:

(3)

where denotes dissociation constants of an analyte in given chromatographic conditions, denotes aqueous *pKa*, and denotes the slope due to changes in the organic modifier.

## Measurement-Error Model

The observed retention factors (*tRobs,z*) were modeled using the following model:

(4)

where *z* denotes the *zth* measurement and *student\_t* denotes the student’s t-distribution with the mean given by the predicted retention time *tR,z*, scale *σ* (analyte and column specific) and normality parameter . The retention time *tR,z* under an organic modifier gradient was calculated utilizing the following integral equation:

(5)

where *t0*denotes column hold-up (dead) time, *te* extra column time, and the instantaneous isocratic retention factor corresponding to the mobile phase composition at time *t* at the column inlet for a particular observation. The numerical solution of this integral equation was carried out using the method of steps with 4 and 10 steps for methanol and acetonitrile gradients using the method proposed by Nikitas et al.13

## Analyte-Level Model

Parameter were assumed to be different for each analyte to account for between analyte variability (BAV). The relationship was of the form:

(6)

where *Ri* denotes an individual value of a parameter *R*, is the mean value that correspond to an analyte with *logP=*2.2, is a regression coefficient, anddenotes standard deviation. For correlated parameter a multivariate normal distribution was used instead.The BAV was assigned for *logkwN* and *S1mN* (*S1* in MeOH) for neural (N) form of analytes (these parameters were correlated (ρ) and depended on *logP* (*β*)). Also the effects of ACN on *S1* (*dS1N*), the effect of temperature on logkw (*dlogkT*), the effect of dissociation on *logkw*, *S1m* and *dS1* (*dlogkw*, *dS1m*, *ddS1)* separately for acids (A) and bases (B)), and the effect of column on these parameters (*clogkw*, *cS1m*, *cβ*, *cdS1*, *cdlogkT*, *cdlogkw*, *cdS1m*, *cddS1*) were used to characterize the impact of chromatographic conditions on model parameters. Also the *clogkw* parameters for neutral forms of analytes was found to be correlated (cρ) during the model building process. The pH effects for cations and anions (*apHA* and *apHB*) were assumed to be the same across analytes but different across the columns (*capHA* and *cpHB*). S2 parameter was assumed to be the same across analytes and columns but different in MeOH and ACN (*logS2m*, and *dlogS2* for ACN effect). Also the BAV was assigned for dissociation related parameters: *pKaw* (dependent on literature *pKa* values), the parameters for acids and bases in MeOH (, and the effect of ACN on (. The *pKa* and parameters were assumed to be independent of the column. The standard deviation of BAV was denoted *ω* for the parameters of the neutral form of analyte on XBridge Shield RP18 and *cω* for the column effects, *ωT*and *cωT* for temperature effects, *κ* and *cκ* for the parameter related to dissociated forms and *τ* for the parameters related to dissociation constant.

## Priors

The Bayesian model requires specification of priors that allow to incorporate domain expertise into inferences. In this work priors were selected based on literature knowledge as discussed previously.7 The means for the priors describing column effects were assumed 0. The standard deviation were set to about half of the standard deviations/scales used for XBridge Shield RP18 parameters. It assumes a very weak similarity of the compared stationary phases across various conditions. The exact values are given in Supplementary Materials.

## Bayesian Inference

**Technical.** Multilevel modeling was performed in Stan/cmdstanr16 software linked with Rstudio17. For the inference we used 8 Markov chains with 500 iterations after 1000 warm up iterations. The *reduce\_sum* function was used to accelerate the calculations by parallelizing the execution of a single Stan chain across multiple cores. Convergence diagnostics were checked using Gelman-Rubin statistics and trace plots. No divergence was reported in the model. The R code, data and Stan code used to analyze the data are publicly available on GitHub repository (https://github.com/wiczling/columncomparison). The raw data are also available through a osf.io repository18–22. The calculations were run on the Tryton computing cluster in Centre of Informatics Tricity Academic Supercomputer and networK.

**Predictions Using a Limited Set of Experiments.** The model applicability was illustrated using 6 analytes with different acidic/basic properties: acridine (monoprotic acid), baclofen (zwitterion: acidic and basic group), hydrocortisone (neutral), pioglitazone (zwitterion: basic and acidic group), quinine (diprotic: 2 basic groups), and tolbutamide (monoprotic base). In this work we present the limited data predictions that correspond to a future predictions given all the experimental data collected for XBridge Shield RP18 column. For this problem all the population parameters were fixed to the a posteriori estimates. The individual parameters for six selected analytes were re-estimated using limited data only. It allowed to predict retention for the other columns in a situation of having a very good understating of retention in one reference column (specifically XBridge Shield RP18 column).

**Uncertainty chromatogram**. The predictions were summarized as uncertainty chromatograms (posterior distribution of retention times expected for a given set of chromatographed analytes under given conditions).12 Here we used typical model predictions (without residual variability) for a particular analyte. The uncertainty chromatogram visualizes the uncertainty for the locations of the maximum of each peak on a given chromatogram.

**Decision making.** The Bayesian optimal decision was sought based on the maximum of expected utility calculated across a wide range of chromatographic conditions. Simulations were performed for *fio* ranging from 0.05 to 0.15 by 0.02, *pHo* ranging from 2.5 to 10.5 by 0.2, *tg* ranging from 20 to 200 by 20, using MeOH and ACN as organic modifiers and for five tested columns at 25oC. The utility function (*U*) was used to calculate the value of each possible chromatogram (posterior predicted retention times). It was defined based on the lowest retention time among analytes (*mintr*), the highest retention time among analytes (*maxtr*), and the difference in retention times between the critical pair of analytes (*res*). This utility was assumed zero if at least one of the analytes has retention higher than 40 or less than 2 min, and if the *res* is less than 2. Otherwise, the utility was linearly related to *maxtr* favoring shorter runs.

U (7)

The expected utility was plotted for a dense grid of chromatographic conditions yielding the expected utility map. The graph was used to identify chromatographic conditions with high/the highest utilities.

# RESULTS AND DISCUSSION

In this work, we applied a mechanistic model to describe the retention data of small molecules obtained for a wide range of chromatographic conditions and for five chromatographic columns. The model was built based on known fundamentals of gradient chromatography and prior knowledge available in the literature. The methodology allowed us to characterize the stationary phases properties using the commonly used chromatographic parameters (specifically *logkw*, *S1* and *S2* of the Neue model). It allows for an easy interpretation of the estimated effects and application of results in decision making. The model parameters are summarized in Table S1 and Figure S2. The key parameters are also provided in Figures 1-3.

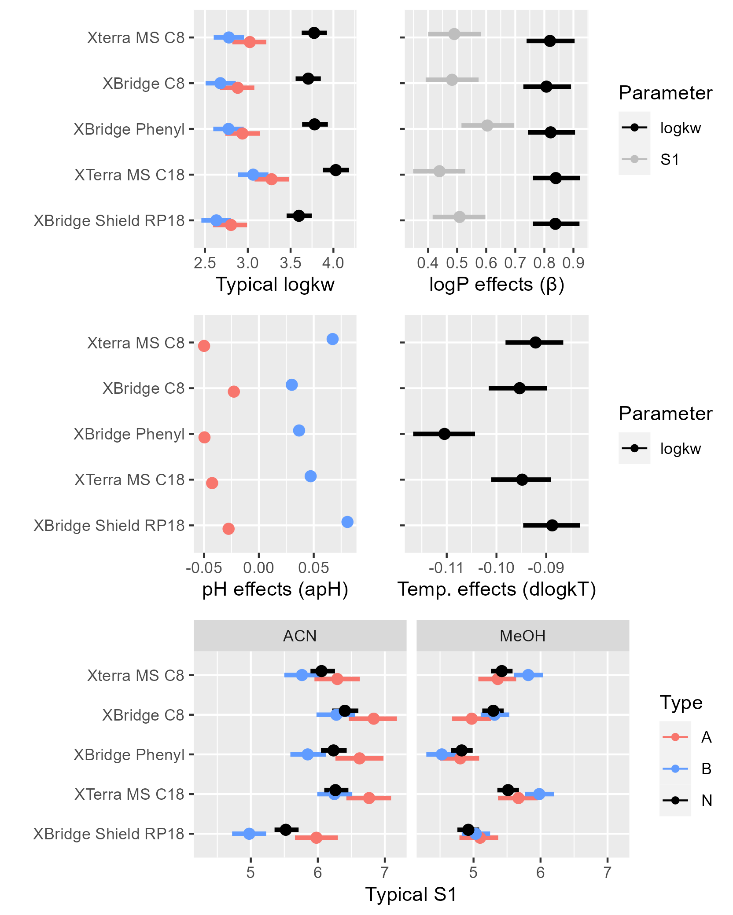


Figure 1. Summary of marginal posterior distributions of the key population-level parameters characterizing retention of analytes for all the tested columns.

These parameters provide a concise summary of influence of various factors affecting retention of analytes on the tested columns. Basically, each parameter isolates and quantifies an effect of a chromatographic condition/analyte properties (pH, organic modifier type and content, temperature, column type, log P) on the retention of neutral, acidic and basic form of an analyte. For simplicity we will discuss the mean values of posterior distribution summarizing the expected value of a particular parameters. The whole posterior distribution is presented graphically in the attached figures and tables. The typical *logkw* of a neutral form of an analyte with logP of 2.2 (a measure of hydrophobicity) is 3.6 for XBridge Shield RP18 at 25oC, and is 0.42, 0.17, 0.10, and 0.17 higher for XTerra MS C18, XBridge Phenyl, XBridge C8 and Xterra MS C8 columns, respectively. It corresponds to a typical 2.64, 1.48, 1.26 1.48-fold higher retention factors for a neutral form of an analyte in neat water eluents in comparison to XBridge Shield RP18. The typical slope in MeOH is 4.96 and the difference between XBridge Shield RP18 and the other columns is 0.59, -0.12, 0.36, 0.48. The BAV for *logkw* and *S1* (*ωlogkw* and *ωS1*) for XBridge Shield RP18 is about 0.9 with a high correlation of 0.87 implying large mutual information of the analyte-specific values of these two parameters. The BAV for column effects on *logkw* (*cωlogkw*) are small (0.10-0.13). They are also correlated with correlations (*cρ*) ranging from 0.55-0.92 depending on the compared columns. The BAV for column effects on S1 in MeOH (*cωS1m*) are also small (0.02-0.15). Since the between analyte variabilities are on a logarithmic scale with base 10, all standard deviations have to be interpreted accordingly to understand the BAV of retention factor. For example the between analyte variability of about 0.1 for *clogkw* implies that the predicted effects of the column are on the order of ± 0.1, which corresponds to a multiplicative factors for retention factor in water-reach eluents ranging from 10-0.1=0.79 to 100.1=1.26. High values of BAV for column effects indicate column orthogonality with respect to the retention behavior controlled by a particular parameter.

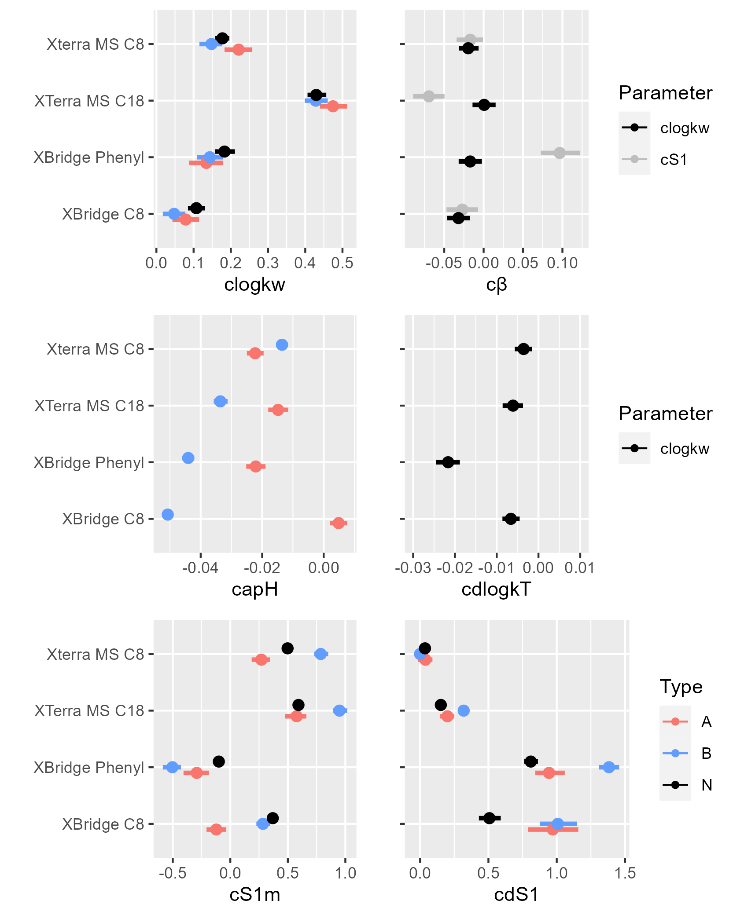


Figure 2. Summary of marginal posterior distributions of the key population-level parameters characterizing the difference in retention of analytes between the indicated column and XBridge Shield RP18 column (column effects).

The log P effect for *logkw* and *S1m* has a slope of 0.83 and 0.48 for XBridge Shield RP 18. The column effects on this parameter are close to zero implying similar effects of *logP* on analyte retention. The largest effect of 0.1 was observed for XBridge Phenyl *cS1m* parameter.

The *S1* is higher in ACN than in MeOH by 0.61 for XBridge Shield RP18 column. In addition it is higher by 0.15, 0.81, 0.51, 0.04 for XTerra MS C18, XBridge Phenyl, XBridge C8 and Xterra MS C8, respectively. The BAV for dS1 (ωdS1) is about (0.55) and for column effects it ranges from (0.14-0.47).The S2 parameter of the Neue equation is 0.49 in MeOH and 1.3 in ACN. The temperature effect for XBridge Shield RP18 are small and consistent with prior knowledge (-0.09 per increase in 10oC) with BAV of 0.03. The column effects on temperature and corresponding BAV are small. The largest effect of -0.02 was observed for XBridge Phenyl. This suggest the quantification of temperature effect on one of the columns allows fairly precisely predicted this effects on other columns.

The logkw for acids and bases is lower by -0.79 and -0.97, respectivly on XBridge Shield RP18 with BAV of 0.59. The column effects for that parameter are very small (from -0.04 to 0.04) with BAV ranging from 0.07-0.11. The pH effects on logkw for acids and bases (*apH*) are small and negative for acids (-0.03 per unit pH) and positive for bases (0.08 per unit pH) for XBridge Shield RP18. The column effects for *apH* vary across columns and range for acids from -0.02 to 0 and for bases from 0.05 to -0.01. The *apH* parameter quantifies the silanol activity.

The S1m for acids and bases is higher by 0.17 and 0.12 on XBridge Shield RP18 with BAV of 0.69. The column effects for that parameter vary from -0.49 to 0.36 depending on column and analyte form with BAV ranging from 0.04-0.20. The slope for acids and bases is different in ACN (relative to *S1m*) by 0.28 for acids and -0.67 for bases on XBridge Shield RP18 with BAV of 0.55. The column effects for that parameter vary from -0.04 to 0.57 depending on column and analyte form with BAV ranging from 0.07-0.72 (the highest value was observed for XBridge C8).

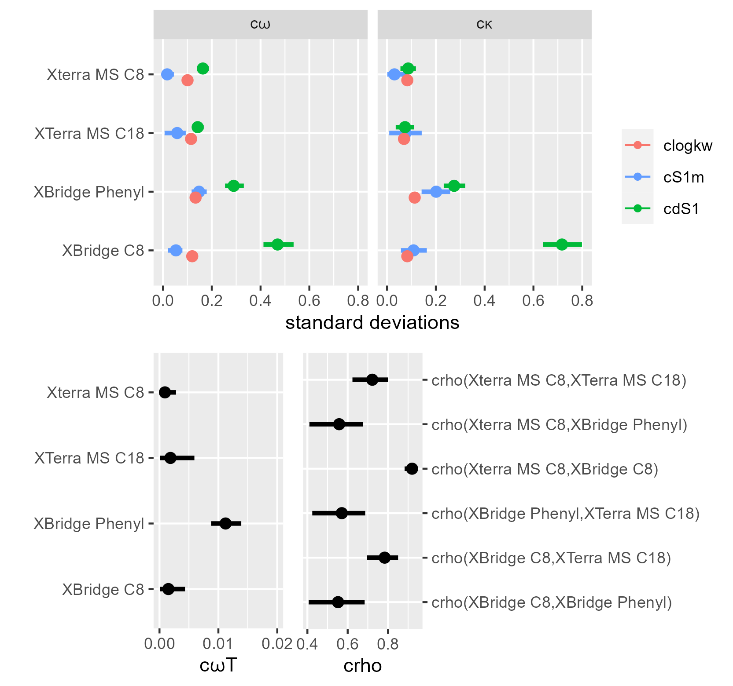


Figure 3. Summary of marginal posterior distributions of the BAV parameters characterizing the difference in retention of analytes between the indicated and reference XBridge Shield RP18 column.

The model predictions are well calibrated with the data for all the columns as shown in Figure S3. The individual and population predictions versus observed retention times are relatively symmetrically distributed around the line of identity indicating model applicability for predictions. The individual and population predictions for several analytes are also shown in Figure S4 and S5. The individual prediction are very precise and close to observed data. The population predictions are also well calibrated but (as expected) are less sharp/precise. The limited data predictions are shown in Figure S6. By comparing them to the population predictions one is able to assess the added predictive value of XBridge Shield RP18 data. Uncertainty chromatograms for population, individual and limited data predictions under the same chromatographic conditions are shown in Figure S7. They illustrate retention uncertainties given access to different data. Clearly the added predictive value of XBridge Shield RP18 data is large. As an example the standard deviation of retention time predictions is reduced from about 10 min for population predictions to about 2 min for limited data predictions and further to about 0.1 for individual predictions. It confirms that the population predictions are very uncertain and have limited practical usefulness. The reason is that they are mostly driven by unexplained BAV *ω*, which is large. However, by adding experimental information uncertainty can be reduced. For example, the predictions based on all the experimental data are very precise. For the limited data predictions the uncertainty is almost entirely reduced for XBridge Shield RP18 column (as there is a lot of data collected using that particular column). For all the other columns there is still some proportion of uncertainty left. In this case, it mostly depends on c*ω*. Since cωis small (about 0.1 for neutral forms of analyte)one can expect a fairly precise predictions, i.e. one is able to predict isocratic retention factor with an uncertainty of about 20-25%. It might be of practical usefulness for simple problems involving few analytes.

All model parameters jointly affect analyte retention. To better illustrate this joined effect of parameters, we simulated the retention factors for the typical acidic, basic and neutral analyte with log P of 0, 3 and 6. The results are present in Figure 4. Based on that graph column characteristics can be directly compared across a wide range of chromatographic conditions. We also provide various isocratic prediction (with uncertainties) for 6 selected analytes to illustrate the impact of parameters on isocratic retention (Figure S8 and S9).

The model can be used in decision making to help identify chromatographic conditions leading to the desired separation. In this work it was illustrated using statistical decision theory. For that purpose we proposed a very simply utility function that favors shorter runs within a separation window between 2 and 40 min and ensure at least 2 min difference in retention of critical pair of analytes. The utility maps are presented on Figures S10 and S11 and the uncertainty chromatograms for the highest expected utility are shown on Figures 5 and 6 for individual and limited data predictions. Based on individual predictions one can identify chromatographic conditions leading to large expected utilities for all of the analyzed columns. The highest expected utility among the simulated chromatographic conditions was identified for Xterra Phenyl. Nevertheless, the differences in expected utilities are small suggesting that similar separations (with respect to the proposed utility) can be identified also for other columns. Based on utility maps one can also identify the regions of chromatographic conditions leading to similar separations.

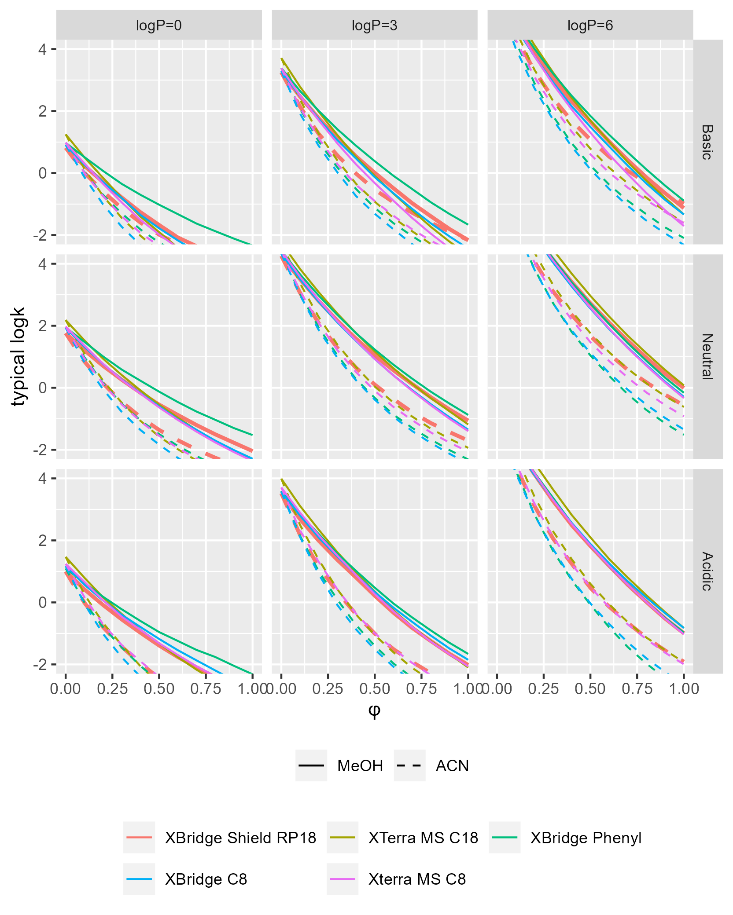


Figure 4. Graphical display of the typical retention profiles (log k vs φ) for a typical basic, neutral and acidic form of an analyte with log P of 0, 3, and 6 at 25oC. The uncertainty is not provided to improve readability.

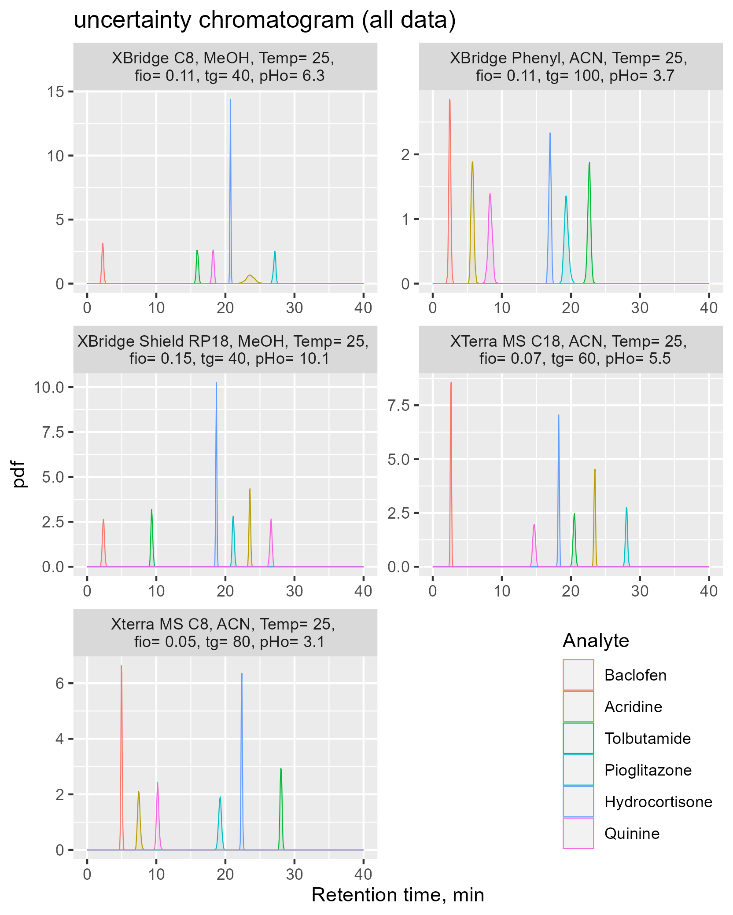


Figure 5. The optimal Bayesian-based chromatographic conditions found based on individual data predictions.

It is also possible to identify optimal conditions based on the limited data (e.g. conditional on XBridge Shield RP18 data). This is a situation of predicting retention on a different column having a fairly good understanding of retention of analytes for one reference column. In our scenario, the uncertainties of the resulting chromatograms are fairly small (standard deviation is in a range of few minutes for gradient conditions). However, they are not sufficient to identify the best conditions with confidence. Clearly some additional experiments are required to improve precision. There are several options of how to proceed. One can try to find the best experimental design ensuring the highest informativeness of a set of additional experiments. Since the proposed model provides a priori information for subsequent analysis this prior information can be utilized to decrease the number of future experiments. One can also perform experiments sequentially and refine predictions after every experiment.23 The clear benefit of having the model that quantifies uncertainty is that it allows to identify chromatographic conditions for which the desired separation is plausible at any state of knowledge about the problem. It is equivalent of having access to some preliminary data.

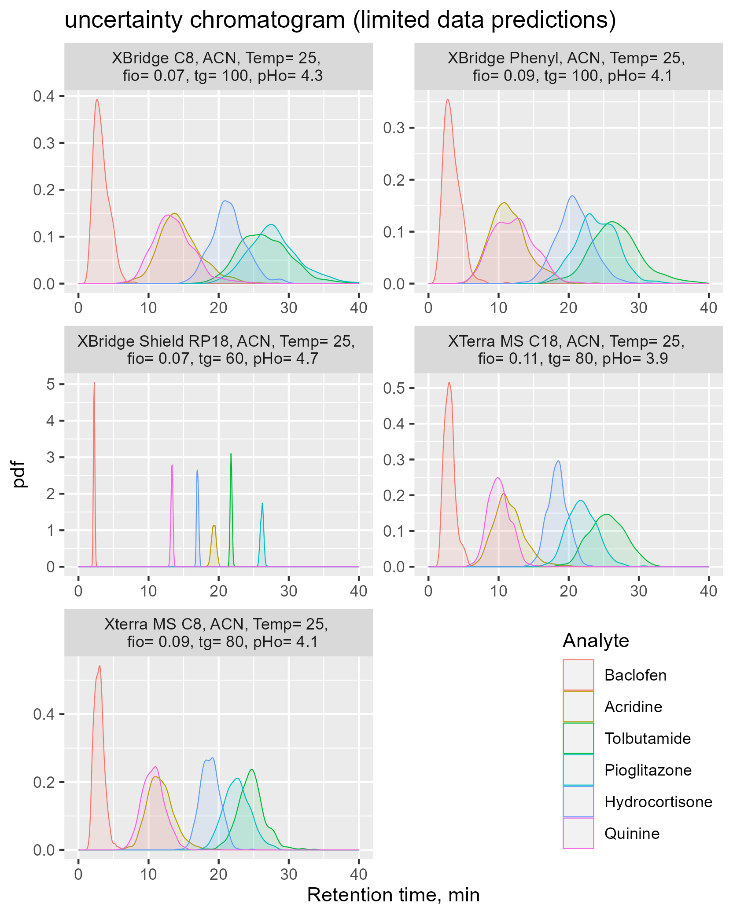


Figure 6. The optimal Bayesian-based chromatographic conditions found based on limited data predictions.

The proposed model is complex, but there are still several improvements that can be considered (e.g. temperature effects for the effects of dissociation and *pKa*, BAV for *S2* and *apH*, etc.). Specifically, the S2 parameters were assumed similar across columns. This is a strong assumption, but justified by the fact that any small change of S2 can be equally well accounted for by changes in S1. We also encountered several technical difficulties during the model building process that required simplifications to run the MCMC without any convergence issues, e.g. excluding analytes with more than 2 dissociations steps and analytes which were incorrectly identified (with a mismatch between observed and ACD-predicted pH-profile). In the present model the column effects were incorporated as fixed effects. It is also possible to include them as a random effects and add column-level predictors. It would allow to build a model that generalized to other columns in a similar way the proposed model generalized to other analytes. This idea is worth pursuing, however requires data collected for a wider range of columns. In this work we decided to use log P as the only predictor relating molecular structure to chromatographic parameters. In principle various functional groups can be added that would allow to assess the impact of molecular structure on retention in a more nuanced fashion9. It would lead to a more detailed characterization of column properties. It is possible with current dataset but would be more accurate with data comprising larger number of structurally diverse analytes.

Our results confirm that it is difficult to predict analyte retention precisely without a set of preliminary chromatographic measurements. Thus, any ranking of columns in terms of their usefulness in obtaining a desired separation in a situation of knowing only analyte structure is very uncertain (one has to use population predictions). To have practically useful results one needs to perform few measurements using at least one stationary phase. But even, having access to the extensive data collected for one column, the isocratic retention factor can be predicted with ~25% precisions for the most optimistic scenario of a neutral compound. It might be sufficient for some simple problems encountered in practice, but is not sufficient for more complex problems involving several analytes of different acidic/basic characteristics. We show this using a fairly difficult problem of finding the optimal separation for 6 analytes of different acidic/basic properties. The proposed utility function can be more complex. i.e. can be based on the cost of performing particular chromatographic experiment, or can favor the most similar/the most orthogonal separations with respect to the any reference separation. It would allow to answer more tailored questions.

It is hard to propose a single number that characterize various interactions occurring in the chromatographic system and uniquely characterizes stationary phase properties. In our opinion it is more valuable to provide a model quantitating the variability and uncertainty of relevant chromatographic parameters, that is sufficiently complex to generalize to other chromatographic conditions, analytes and stationary phases. Such a model might be built once for a particular stationary phase and used by others to solve their specific problem. Since the proposed approach is very general, it can also serve to determine all the currently used metrics (if the same analytes and chromatographic conditions are considered).

# CONCLUSIONS

This pilot work demonstrates the application of a Bayesian multilevel model to compare five stationary phases using large datasets of retention times collected for a wide range of gradient chromatographic conditions. This analysis characterizes the chromatographic retention of neutral, acidic, and basic analytes and assesses the effects of pH, temperature, log P, *pKa* organic modifier type and content on analyte retention across a range of columns. The model along with the estimated parameters can be used to solve various decision problems. The proposed method provides a tool allowing for a very comprehensive assessment of stationary phase properties that seems to be an interesting alternative to existing methods.

# ASSOCIATED CONTENT

## Supporting Information

The following Supporting Information is available free of charge at the ACS website:

# 1. Model description; 2. Table S1. Summary of the MCMC simulations of the marginal posterior distributions of population-level model parameters; 3. Figure S1. Raw data for 6 selected analytes; 4. Figure S2. Summary of the MCMC simulations of the marginal posterior distributions of population-level model parameters; 5. Figure S3. Goodness of fit plots; 6. Figure S4. Individual gradient predictions; 7. Figure S5. Population gradient predictions; 8. Figure S6. Limited data gradient predictions; 9 Figure S7. Uncertainty chromatograms. 10. Figure S8. Individual isocratic predictions; 11. Figure S9. Population isocratic predictions; 12. Figure S10. Utility maps based on individual predictions; 13. Figure S11. Utility maps based on limited data predictions.

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## Author Contributions

PW and AK analyzed the data; PW wrote the paper; PW conceived of the presented idea, designed the study and supervised the project.

Notes  
The authors declare no competing financial interest.

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