

# PHYSICS

## SOLUTION OF NUMERICALS

SOLVED BY:

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Q.2(i) DATA:

$$m = 1.67 \times 10^{-27} \text{ Kg}$$

$$e = 1.6 \times 10^{-19} \text{ coulomb}$$

$$\Delta V = 14.32 \times 10^{-9} \text{ Volt}$$

Distance between two plates =  
 $d = ?$

SOLUTION:

According to equation of P.D. between two oppositely charged plates,

$$\Delta V = E \times d$$

$$\Rightarrow \frac{\Delta V}{E} = d$$

$$\Rightarrow d = \frac{\Delta V}{E} \text{ --- (i)}$$

To find the value of "E":

In case of balanced charged particle between two oppositely charged plates,

$$F_{(\text{Elec})} = W_{(\text{Particle})}$$

$$\Rightarrow Ee = mg$$

$$\Rightarrow E = \frac{mg}{e}$$

$$\Rightarrow E = \frac{(1.67 \times 10^{-27}) \times (9.8)}{(1.6 \times 10^{-19})}$$

$$\Rightarrow E = 1.023 \times 10^{-7} \text{ Volt/meter}$$

Now, Substituting the value of "E" in eqn(i)

$$\therefore \text{eqn (i)} \Rightarrow d = \frac{\Delta V}{E}$$

$$\Rightarrow d = \frac{14.32 \times 10^{-9}}{1.023 \times 10^{-7}}$$

$$\Rightarrow d = 0.13998 \text{ meter}$$

$$\text{OR } d = 13.998 \text{ cm}$$

Q.2(ii) See in Theory

Q.2(iii) DATA:

$$T(^{\circ}\text{C}) = ?$$

$$v_{(\text{r.m.s.})} = 3300 \text{ m/s}$$

Mass of Hydrogen molecule =

$$m = 3.32 \times 10^{-27} \text{ Kg}$$

Given that:

Boltzman's constant =  $K =$

$$1.38 \times 10^{-23} \text{ J/K}$$

SOLUTION:

As,

$$v_{(\text{r.m.s.})} = \sqrt{\frac{3KT}{m}}$$

By squaring on both sides,

$$v^2_{(\text{r.m.s.})} = \frac{3KT}{m}$$

$$\Rightarrow (3300)^2 = \frac{3 \times (1.38 \times 10^{-23}) \times T}{(3.32 \times 10^{-27})}$$