

STACK EMISSION MONITORING DATA RECORDING SHEET

1.	Sample ID:	SA/2105/0125	2.	Instrument ID:	DFD12						
3.	Date & Time of Sampling:24/06/21										
4.	Stack Identity:32										
5.	Stack attached to:32										
6.	Stack height (m):32										
7.	Stack diameter/Duct Dimensions at sampling point (m):32										
8.	Height of port hole from ground level (m ²):55										
9.	Are 8D and 2D Criteria met: Yes / No:True										
10.	Material of Construction:323										
11.	Stack shape at top:66										
12.	Fuel used: i) Type:65		ii) Consumption: 66 kg/d or Ud								
13.	Stack Area (A) (m ²):77										
14.	Whether Sampling port and platform exists? Yes / No:True										
15.	Whether Air pollution control equipment exists? Yes /No (Specify)True										
16.	Barometric Pressure (Ba): 323.000 mm Hg										
17.	Fuel Gas composition:	CO ₂ :%:22.000	O ₂ :%:34.000	CO:%:45.000	N ₂ :%:56.000						
18.	Moisture Content (M) = 323.000 %		ii) Bwo: (M /100) =23.0000								
19.	Duration (h) / day of run of boiler/ process:77.00										
20.	Ambient Temperature :		22.00	°C + 273 =	56.00 °K						
21.	Stack Temperature(Ts) :		12.00	°C + 273 =	23.00 °K						
22.	Velocity of Stack Gas (V):		32.00	Pitot Tube factor (K):56.0000							
	m/s			C = K x 0.2295 =45.00							
Details		1	2	3	4	5	6	7	8	9	10
a) Traverse Point distance from the port, cm											
b) Differential Pressure (ΔP) mmH ₂ O											
c) Static Pressure (Ps)mmHg		Ps= 234.0000 mmH ₂ O/13.6 = mmHg									
d) Absolute Stack Pressure (PA)		PA (Ba ± Ps) = 345.0000 mmHg									
e) Velocity V in m/s V=C√AP x Ts. °K		23									
f) Flue Gas Quantity (Dry) Nm ³ /h Q = A x V x 3600X(298/Ts) X(PA/760) X(1-Bwo)		456.0000									

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Note =* Physical Details of Sack Provided by Customer

19. Sampling Rate for Isokinetic conditions:

a. Nozzle Constant: Area of Nozzle(mz) X 60 X 1000 = 32.0000											
b. Sampling Flow Rate (LPM) =Velocity x Nozzle constant x Ta/Ts x PA/ Ba-Pv	12.0000										
c. Sampling duration in Minutes	23.0000										
d. Vacuum PV in mm Hg	Initial	23.0000									
	Final	34.0000									
	Average	45.0000									
e. Sampling gas Temp(Tg) °K	Initial	34.0000			Final	56.0000			Average	67.0000	0
Thimble No:-323											

20. Total gas passed:(Flow Rate) x Duration= 323.0000 Liters

OR

Dry Gas Meter Reading: **Final** **Initial** **Diff .** m3

21. Gaseous Sampling Data:

Parameter	Flow Rate (LPM) (R)	Sampling time in minutes (T)	Gas Temp (°K)(Tg)	Barometric Pressure mm Hg (Ba)	Dry gas meter Reading (m³)			Bottle No.	Absorbing solution used			Preservation Done Parameter (If Any)
					Initial	Final	Total		Solution	Conc.	Vol. (ml)	
Carbon Dioxide (CO ₂)	5.000	4	5.000	4.000	5.0000	4.0000	5.0000	1	4	5.00	4.00	5

$$V_s \text{ (in liters at STP)} = R \times T \times (Ba/760) \times (298/ Tg) = y$$

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22. Sampling Data for NO_x

$$V_{sc} = (T_{std}/P_{std})(V_f - V_a) (P_f/T_f - P_i/T_i)$$

Where,V_s = 1 (Volume of flask & vdlve in ml)V_a = 2 (Volume of absorbing solution, 25 ml)P_f = 3 (Final absolute pressure of flask, mm Hg)P_i = 4 (Initial absolute pressure of flask, mm Hg)P_{std} = 6 (Standard absolute pressure, 760 mm Hg (29.92 inch of Hg))T_f = 8 (Final absolute temperature of flask, °K)T_i = 6 (Initial absolute temperature of flask, °K)T_{std} = 9 (Standard absolute temperature, 298.15 °K)V_{sc} = Sample Volume at standard condition (dry basis), ml

$$V_{sc} = (\text{----}) (\text{ - }) (\text{ - - - }) = 32.0000 \text{ ml}$$

$$= \frac{\text{---}}{1000} \text{ ml}$$

$$= \text{---} \text{ L}$$

Preservation Done (If Any) :

Sample received in lab by: **Sign:** _____ **Date:** 26/06/21**Probable date of report:** 30/06/21**Sampling done by:** Ram Lakshman **Sign:** _____**Name and Address of Customer:** Monica
Nagpur**Sample ID:** SA/2105/0125

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