

Name: \_\_\_\_\_

The table below is the portion of the periodic table of elements for semiconductor materials:

II	III	IV	V	VI
	B	C	N	
	Al	Si	P	S
Zn	Ga	Ge	As	Se
Cd	In		Sb	Te

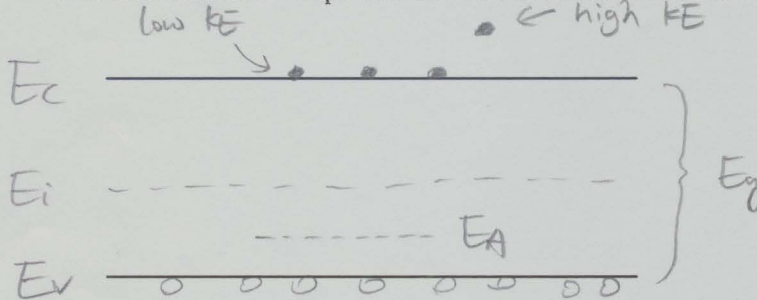
Consider a silicon sample that is doped with gallium (Ga):

1) Is the material n or p-type? (1 point) *p-type*

2) Justify your answer (1 points)

*Ga atom has one less electron than Si*

The two lines below represent the conduction band and the valence band of a Si semiconductor:

3) Label the conduction band ( $E_C$ ) and the valence band ( $E_V$ ) (1 point).Indicate schematically the following:4) The position of the intrinsic level ( $E_i$ ), the position of the gallium level ( $E_D$ , if it is a donor;  $E_A$ , if it is an acceptor). (2 points)

5) The energy gap. (1 point)

6) The electrons by solid dots ●, and the holes by open dots ○. (2 points)

7) An electron with high kinetic energy and an electron with low kinetic energy. (1 point)

8) Are there more electrons or holes? Why? (1 points)

*more holes. p-doped with Ga.*