

Chapter 10: Amines

10.1 Amines

Classification

1°

2°

3°

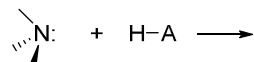
4°

Structure and Properties of Amines

sp³-hybridized (aliphatic amines)

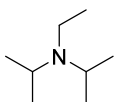
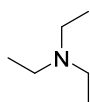
highly polar

Amines are basic compounds and react with acids:

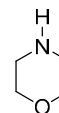
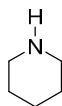


10.1

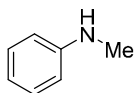
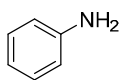
Aliphatic Amines



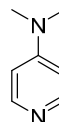
Heterocyclic Aliphatic Amines



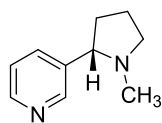
Aryl Amines



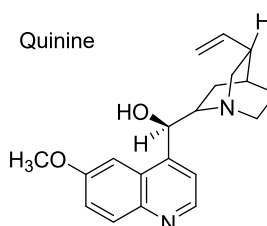
Heterocyclic Aromatic Amines



Natural Products



Nicotine



Quinine

10.2

10.2 Nomenclature of Amines**Rules:**

If there are no higher-priority functional groups:

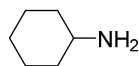
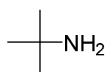
- Name the three substituents in alphabetical order and add “amine”.
- Use the prefixes “di” and “tri” if the groups are the same.
- If two or three -NH₂ groups are present use di and tri before the parent name

With unsymmetrically N-substituted secondary and tertiary amines:

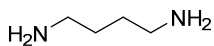
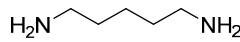
- The largest organic group is the parent chain
- The location of the smaller groups are indicated by the prefix “N”

With higher-priority functional groups:

- Name -NH₂ as a substituent- “amino”

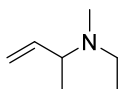
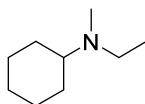
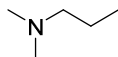
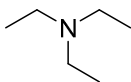
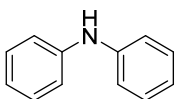
Examples:

Common names

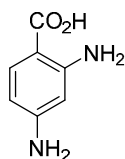
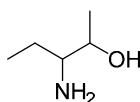
**Putrescine****Cadaverine**

IUPAC

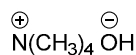
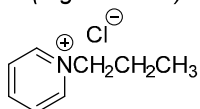
10.3



10.4

**4° ammonium salts**

- Replace amine with ammonium (or pyridinium)
- Add the name of the anion (e.g. chloride)

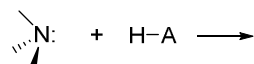


10.5

10.3 Structure and Properties of Amines

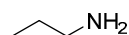
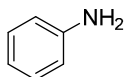
sp³-hybridized (aliphatic amines)
highly polar

Amines are basic compounds and react with acids:



Much stronger bases than alcohols, ethers, or water

Aryl amines are weaker bases than alkyl amines due to resonance moving electrons away from the N



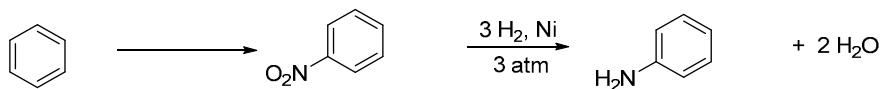
Amides (RCONH₂) are non-basic due to resonance

10.6

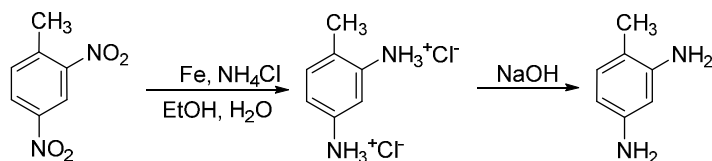
10.4 Synthesis of Arylamines

Nitration followed by reduction.

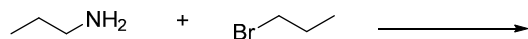
Hydrogenation:



Reduction by a metal (Fe, Zn, Sn) in presence of a proton source:



10.7

10.5 Amines as Nucleophiles

10.8