Exam 2

Started: Nov 8 at 1:01am

Quiz Instructions

Question 1	3 pts
What is the principle of operation of electrochemical sensing portable instruments?	
Chemical oxidation of the test gas	
 Creates organic ions by passing a hydrogen gas through a flame and then measu conductivity change in the flame as a concentration of the gas 	ures the
ouses an ultraviolet lamp to ionize organic compounds to detect and quantify them	
Chemical reagent changes color	

Question 2 3 pts

If you have a 1-L balloon of nitrogen gas at an initial temperature and pressure of 20 C and 760 mmHg and you take that balloon to the top of a mountain where the temperature and pressure are only 5 C and 650 mmHg, what is the new volume of nitrogen in the balloon in liters (assume no resistance to expansion from the balloon membrane)?

1.11

0.29

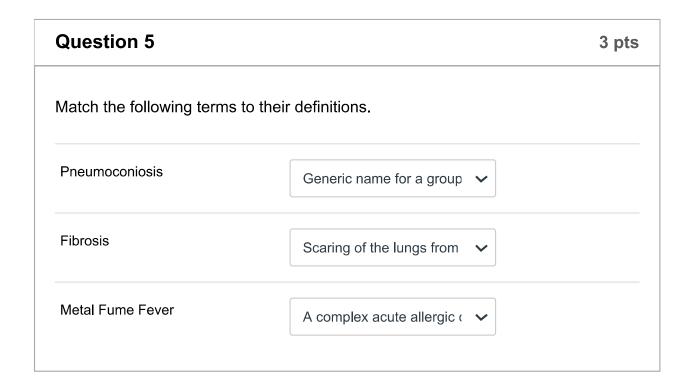
O 1.23

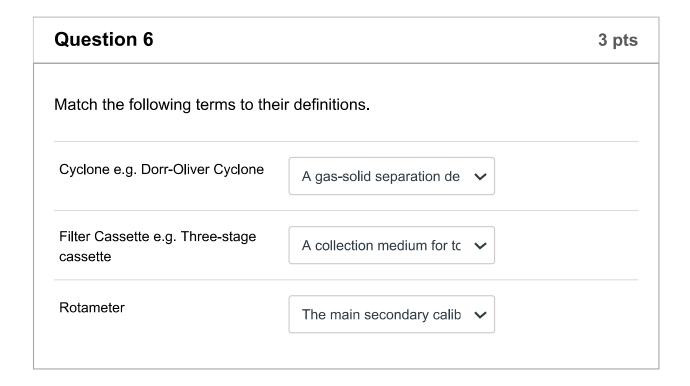
0.81

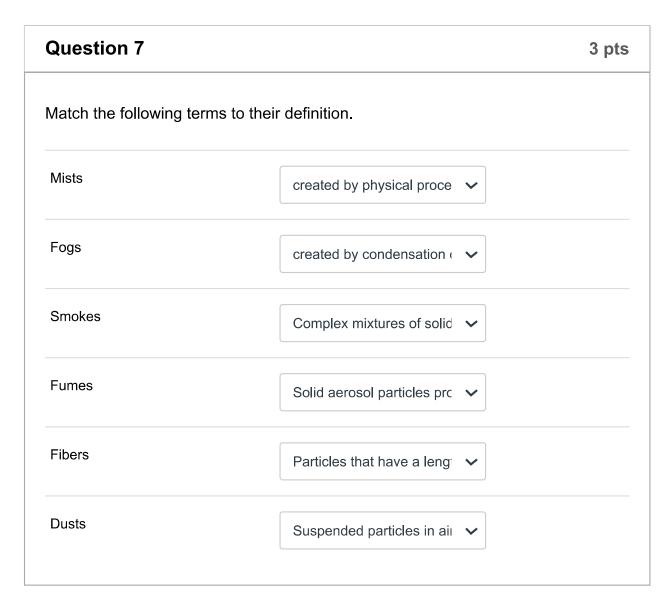
Question 3 3 pts

acetone has	an SG of 0.79, how many mL does 2.0 g of acetone occupy?
2.53	
1.58	
Not enough i	formation to answer
No answer te	kt provided.

Question 4	3 pts
According to a NIOSH analytical method, the limit of detection (LOD) chemical is 50 µg, the recommended exposure limit for this chemical mg/m^3. If the detection criteria is an exposure of 10% of the exposurany liters of air must be collected to obtain the minimum sample ma	is 1.0 re limit, how
<u> </u>	
○ 10○ 50	







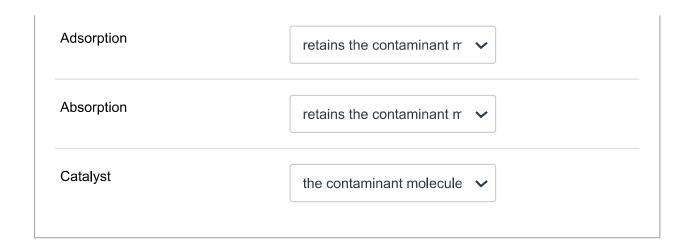
Question 8 3 pts

	s to their definitions.
Sedimentation	The processes by which ε 🗸
mpaction	Occurs when an aerosol t 🗸
nterception	Takes place when an aerc ➤
Brownian Diffusion	The deposition of an aero 🗸

Question 9	3 pts
What are the three types of centrifugal fans?	
Forward-curved, Backward-curved, Radial Blade Fan	
Squirrel cage, forward curve, backward curve	
Radial, Impeller, backward curve	
○ Tubeaxial, propeller, radial	

Question 10	3 pts
What do we call the air movement at a given distance from the front of a house is necessary to overcome opposing air currents and cause a contaminant to into the hood?	
Capture Velocity	
○ Face Velocity	
O Duct Velocity	

Question 11	3 pts
The velocity of air in a duct is 3,582 ft/min. What is	the velocity pressure?
0.7999	
○ 1.250	
O 239,700	
○ 54.1	
O 1,800	
O.2	
Question 12	3 pts
	nave a minimum of 12 ACH
If it is recommended that a patient's hospital room h	
If it is recommended that a patient's hospital room is and the room is 15x25x12 ft., what quantity of air no	
	eeds to be supplied to (and
and the room is 15x25x12 ft., what quantity of air no	eeds to be supplied to (and
and the room is 15x25x12 ft., what quantity of air no exhausted from) the room in cfm? Assume perfect	eeds to be supplied to (and
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and the room is 15x25x12 ft., what quantity of air no exhausted from) the room in cfm? Assume perfect 900 54000	eeds to be supplied to (and
and the room is 15x25x12 ft., what quantity of air no exhausted from) the room in cfm? Assume perfect 900 54000 375	eeds to be supplied to (and



Question 14	3 pts
Of the following respirator types, which is the only one acceptable for use i IDLH atmosphere?	n an
SCBA	
○ Filtering Facepiece	
○ PAPR	
Full-Face Cartridge Respirator	
○ Half-Face Cartridge Respirator	

Question 15	3 pts
All of the following are requirements of the OSHA respiratory protection pro except:	gram
Psychological Evaluation of employees using respirators	
○ Fit- testing of respirators	
Training and training procedures	
O Procedures for when respirators are not required	

Question 16	3 pts
All of the following are steps in the respirator fit-testing procedure except:	
Shouting	
○ Normal Breathing	
○ Talking	
○ Moving head up and down	
Question 17	3 pts
All of the following are primary functions of the skin except:	
Provide a barrier to the environment	
O Protect muscles and tissues	
○ Let necessary materials in	
○ Temperature control	
Question 18	3 pts
All of the following are means of permeation of agents through the skin exc	<u>ept</u> :

Via telekenetic action

○ Transcellular permeation

O Through appendages in the skin

 $\bigcirc\,$ Travel on the intracellular lipid pathway

Question 19	3 pts
All of the following are <u>chemical factors</u> that may ir and permeation through the skin <u>except</u> :	ncrease chemical absorption
Broken skin	
Chemical volatility and vapor pressure	
Molecular weight	
Hydrophilic or lipophilic properties	
Specific chemical structure	
Question 20	3 pts
oermeation through the skin <u>except</u> : Hydrophilic or lipophilic properties 	
Anatomical location	
○ Skin thickness	
Metabolic rates	
○ Individual difference	
Question 21	3 pts
Given the background oxygen concentration is air of a chemical in air is 6,500 ppm, what would you concentration to read on an instrument?	

○ 3.25%			
12.25%			
○ Insufficient data to ca	lculate		
O 20.10%			

Question 22	3 pts
All of the following are advantages of colorimetric tubes except:	
High precision readings	
O Do not require calibration	
O Do not require batteries	
○ They are inexpensive	

Question 23	3 pts
When measuring the airborne concentration of Ethylbenzene with a PID with a 10.6 eV lamp, the meter displays a value of 162 ppm. Using the c factor of 0.47, what is the actual concentration of Ethylbenzene in ppm?	
76	
○ 83	
○ 318	
○ 59	
○ 344	
○ 59	

Question 24

If a combustible gas meter reads less than zero, what should you assume about the concentration of flammable vapors in the environment?

● The concentration is >UEL

○ The concentration is <UEL

○ The instrument needs calibration

○ The instrument needs to be replaced

When collecting personal air samples, the inlet of the sampling device should be placed in the ____.

Breathing Zone

Work Zone

Anywhere on the worker

Anywhere in the work space

Question 26 3 pts

Using the table of the sampling results below, calculate the 8-hour TWA for the sample period in ppm assuming that the un-sampled time has the same concentration as the highest recorded concentration.

Sample	Time	Concentration
No.	(min)	(ppm)
1	101	123
2	91	154
3	94	118

4	99	149		
O 136				
O 109				
139				
<u> </u>				
O 141				

Question 27 3 pts

Using the table of the sampling results below, calculate the TWA for the sample period in ppm.

Sample	Time	Concentration
No.	(hrs)	(ppm)
1	1.2	56
2	2.3	51
3	1.5	44
4	2.0	64

54.	1

47.3

55.3

○ 23

O 29

Question 28 3 pts

Using the Brief & Scala Formula provided below, calculate the adjustment factor for converting an 8-hr PEL into a 10-hr PEL and then determine the 10-hr PEL for

Acetonitrile that has an 8-hr PEL of 40 ppm.

$$Q=\frac{8}{h}x\,\frac{(24-h)}{16}$$

- 28
- O 2
- 33
- O 100
- O 57
- O 10

Question 29 3 pts

When taking air samples, you collect the pre and post calibration measurements from the sampling pump as given in the table below. What do you record for the flow rate of the sample pump?

Measurement	Pre-Cal	Post Cal
1	1.735	1.668
2	1.741	1.653
3	1.743	1.657
4	1.733	1.649
5	1.738	1.629

- 1.651
- O 1.695
- O 1.743
- O 1.629

Question 30	3 pts
If the 95UCL of a distribution of personal air s TLV is 200 ppm, what AIHA exposure categorexposure?	• • • • • • • • • • • • • • • • • • • •
○ 2	
O 1	
3	
0 4	
Question 31	3 pts
8hr TWA. If the PEL is 250 ppm and the SAE method is +/-9.5%, how would you classify thi compliance? O Possible Violation	
Violation	
○ No Violation	
No answer text provided.	
Question 32	3 pts
Question 32 A rectangular flammable liquids storage room Determine the volumetric flow rate of air in cfr in compliance with the OSHA 1910.106(d)(4)(has a floor that is 12 ft by 12 ft. m necessary to ventilation the room

○ 168	
○ 150	
O 120	
Question 33	3 pts
You are performing Total Dust air monitoring for exposure to Parti Otherwise Regulated (PNOR) per NIOSH Method 0500 that has a analytical error of 11.04%. Given the PEL for total dust is 15 mg/r result was 15.57 mg/m ³ , how would you classify this exposure?	a sampling and
○ Violation	
Possible Violation	
○ No Violation	
Question 34	3 pts
Question 34 For an atmosphere containing Acetonitrile that has an 8hr TWA P and an IDLH value of 137 ppm, what is the maximum use concentace powered air purifying respirator?	EL of 40 ppm
For an atmosphere containing Acetonitrile that has an 8hr TWA P and an IDLH value of 137 ppm, what is the maximum use concen	EL of 40 ppm
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For an atmosphere containing vegetable oil mist that has an 8hr TWA PEL of 15 mg/m^3 and no established IDLH value, what is the maximum use concentration for a half face air purifying respirator?

To mg/m^3

150 mg/m^3

15,000 mg/m^3

375 mg/m^3

Question 36 3 pts

When performing a respiratory protection review at your workplace, you observe an employee in a paint booth spraying a paint that is organic solvent based. The worker is using a full face cartridge respirator with cartridges pictured below. Assuming that there is sufficient oxygen and that previous sampling indicates that a full face APR is appropriate, what is your course of action?



- Stop work and issue the worker cartridges with a green band and magenta cap
- Stop work and issue the worker cartridges with an olive band
- Ocontinue, the worker has selected the correct cartridge
- Allow the work to continue, but inform the supervisor to have the worker change cartridges after the next break

Question 37

OSHA's silica standard for construction 1926.1153 - Respirable crystalline silica, Appendix A calls for the use of NIOSH Method 7602 for the analysis of silica samples. The LOD for NIOSH 7602 is 5 μg/sample. If the flow rate of the cyclone is 2.2 lpm, what is the minimum detectable TWA concentration in air for an 8-hour shift as a percentage of the PEL of 50 μg/m³?

9.5%

0.0095%

1056%

36.2%

12.3%

Question 38

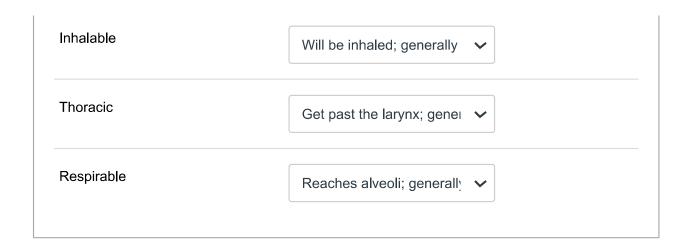
3 pts

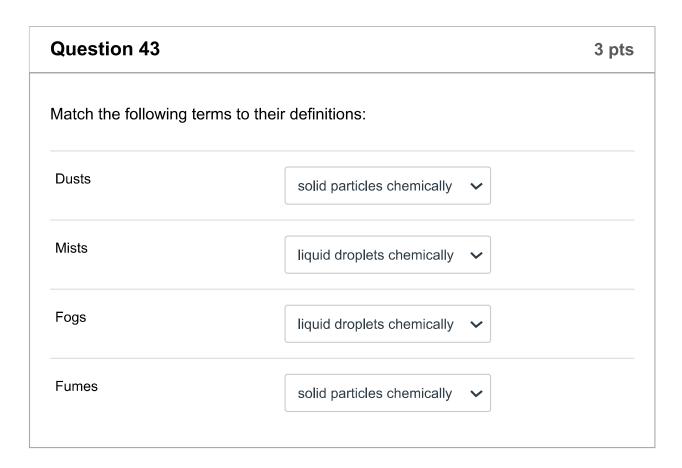
Question 38	3 pts
Upon reporting your results to your client for lead dust sampling that gar of 28 ug/m^3, the client asks you to convert your answer to ppm. What answer? Note: the molecular weight of lead is 207.2 g/mol.	
This conversion cannot be calculated	
0.0033 ppm	
○ 3.3 ppm	
○ 1.5E6 ppm	

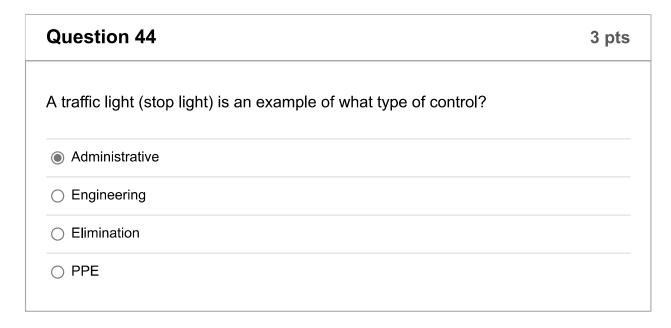
Question 39 3 pts

One mole of substance that is a liquid at NTP (25 C and 760 mmHg) that evaporates will occupy how many liters?

24.45	
O 22.04	
○ 25.6	
○ 24.04	
Question 40	3 pts
In industrial hygiene, when using the term ppm to describe chemical eair, ppm is a ration of what basis?	exposures in
volume/volume	
○ mass/mass	
○ mass/volume	
○ volume/mass	
Question 41	3 pts
The acronym PNOR stands for:	
✓ Particulates Not Otherwise Regulated	
☐ Protection for No Other Regulation	
☐ Particulates Not Often Recognized	
☐ Particulates in-Need of Regulation	
Question 42	3 pts
Match the following aerosol size fractions with their descriptions:	







Question 45	3 pts
Process Automation is an example of what type of control?	
Engineering	
○ Administrative	
○ PPE	
○ Substitution	

Random variations in sampling equipment e.g. fluctuations in pump flow, effect:

Precision
Accuracy
No answer text provided.

No answer text provided.

Question 47 3 pts

What type of respirator is pictured below?



Half-Face Cartridge Respirator	
Full-Face Cartridge Respirator	
Self-Contained Breathing Apparatus	
Question 48	3 pts
After an employee has received proper BBP training, the Hepatitis B Vaccination (unless declined by the employee previously received the vaccination) within how many day	, or the employee has
10 working days	
○ 10 calendar days	
○ 7 working days	
○ 7 calendar days	
Question 49	3 pts
Quantitative fit testing is required when the necessary AP	F exceeds:
100	
○ 50	
○ 1000	
○ 25	

○ Immersion		
○ Splashes		
Deposition		
Contact with contaminated su	faces	

Quiz saved at 1:59am

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