Exam1

Due No due date **Points** 100 **Questions** 21

Available Oct 1 at 9:18pm - Oct 2 at 12:08am 2 hours and 50 minutes

Time Limit 170 Minutes

Instructions

Please write your answer based on the given instruction in each question. See below.

If the calculated result value is 894.6548, the rounding rule is as below.

- the right answer rounded to the nearest ones is 895
- the right answer rounded to the nearest tens is 890
- the right answer rounded to the first decimal place is 894.7
- the right answer rounded to the second decimal place is 894.65

Unless specified, assume density factor = 1.

Best luck!

Attempt History

	Attempt	Time	Score
LATEST	Attempt 1	99 minutes	88 out of 100

Score for this quiz: 88 out of 100

Submitted Oct 1 at 10:58pm This attempt took 99 minutes.

Question 1 3 / 3 pts

What is the value of "X" for the following stoichiometric chemical reaction?

Correct!

5

orrect Answers

5 (with margin: 0)

Question 2

3 / 3 pts

What is the number of x for the mixture to represent air? Assume that air is composed of 21 vol% of Oxygen and 79 Vol% of Nitrogen. Round your answer to the second decimal place and do not include any units.

air: xO2+2.75N2

Correct!

0.73

orrect Answer

0.73 margin of error +/- 0.01

Question 3

3 / 3 pts

Calculate the heat of combustion of C3H8 burning in air in kJ/mole using the following values. Write down your answer to the first decimal place without any units.

Heat of formation: C3H8 = -103.8 kJ/mole,
 CO2 = -393.5 kJ/mole, H2O = -241.8 kJ/mole

Correct!

2.043.9

orrect Answers

2,043.9 (with margin: 1)

Question 4

0 / 3 pts

Calculate the heat of combustion of C2H4 burning in air in kJ/g using the following values. Write down your answer to the first decimal place without any units.

Heat of formation: C2H4 = 52.5 kJ/mole,
 CO2 = -393.5 kJ/mole, H2O = -241.8
 kJ/mole

ou Answered

1,323.1

orrect Answers

47.3 (with margin: 1)

Question 5

3 / 3 pts

What is the representative heat of combustion value that we agreed for most hydrocarbon fuels in kJ/g?

Write down your answer rounded to the nearest ones without any units.

Correct!

45

orrect Answers

45 (with margin: 1)

Question 6

3 / 3 pts

Calculate the enthalpy of propane in kJ/mole at 1,018K assuming a representative specific heat value of 128.7 J/mole-K from 298 K to 1,018K. The heat of formation of propane is -103.8 kJ/mole. Write down your answer rounded to the first decimal place without units.

Correct!

-11.1

orrect Answer

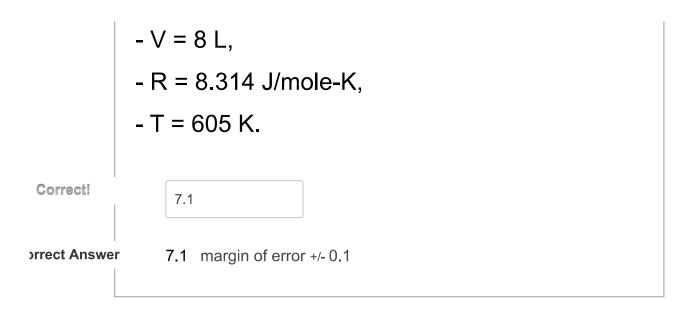
-11.1 margin of error +/- 0.1

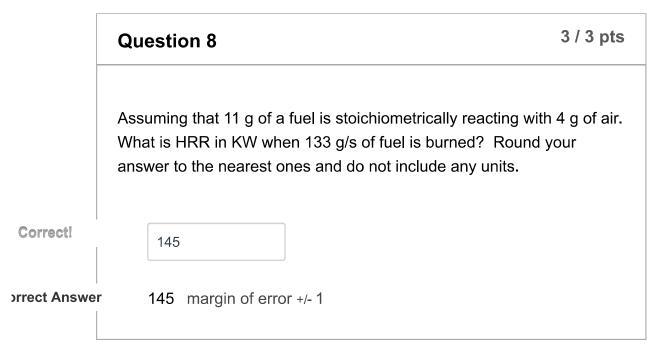
Question 7

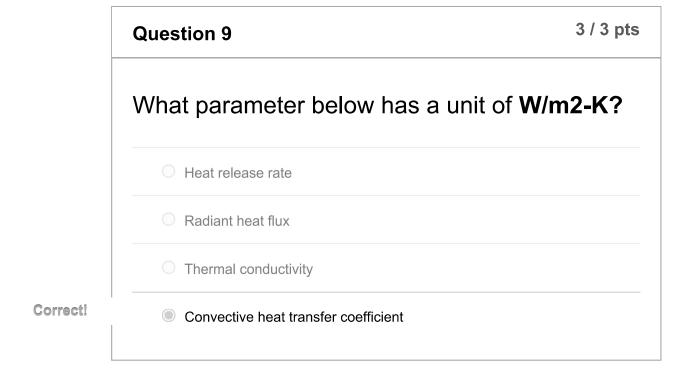
3 / 3 pts

Calculate the mass of propane (C3H8) in gram for the following conditions. Round your answer to the first decimal place and do not include any units.

- P = 1 atm = 101325 Pa = 1.01325 bar = 14.7 psi,







Question 10 3 / 3 pts

Calculate the mass flow rate [kg/s] of air in a 11 inch round duct (in diameter).

Air flow velocity is 9,893 feet per min and the temperature is 56 °C.

Use air density = 1.2 kg/m³ at 20°C. Round your answer to the first decimal place and do not include any units.

Correct!

3.3

orrect Answer

3.3 margin of error +/- 0.05

Question 11 0 / 3 pts

Calculate the average velocity of the outflow from the compartment with the following conditions in m/s. Round your answer to the first decimal place and do not include any units.

- Single room with one door opening
- Temp: 25°C, Amb. Density: 1.2 kg/m³
- Opening height: 2.1 m from the floor
- Neutral plane height: 41% of the opening height

• Compartment upper layer temperature is 280 °C

ou Answered

orrect Answer

3 margin of error +/- 0.1

Question 12 0 / 3 pts

What would be the room upper temperature in Celsius to generate 2.31 m/s average velocity of incoming flow in a room fire? Write your answer rounded to the nearest ones without units.

Assumption:

- Single room with one door opening
- Ambient air Temp. = 20 °C,
- Ambient air density = 1.2 kg/m3
- gravity = 9.81 m/s2
- Opening height = 2.1 m
- Opening width = 0.9 m
- Neutral plane height from the bottom of opening = 0.91 m

ou Answered

622

Question 13

3 / 3 pts

Calculate adiabatic flame temperature (K) of Butane(C4H10) burning in the air with the initial temperature of 298K and 1 atm. Write down your answer to the nearest ones without units.

- Specific heat: CO2 = 54.3 J/mole-K, H2O = 41.3 J/mole-K, N2 = 32.7 J/mole-K.
- Heat of formation: fuel = -124.7 kJ/mole,
 CO2 = -393.5 kJ/mole, H2O = -241.8
 kJ/mole

Correct!

2,472

orrect Answers

2,472 (with margin: 3)

Question 14

3 / 3 pts

A fire in a room increases the interior wall surface temperature and maintains it at 1,111 °C. The outside ambient temperature is 20 °C. If the wall is made of 200 mm brick, calculate

the steady-state exterior wall surface temperature in Celsius? Write down your answer rounded to the nearest ones without any units.

- Thermal conductivity of brick = 0.7 W/m-K
- Convective heat transfer coefficient = 13
 W/m2-K

Correct!

251

orrect Answer

251 margin of error +/- 1

Question 15 3 / 3 pts

Assuming thermally thick behavior, 39 kW/m² heat flux was applied to a 3 mm thick wood veneer for 22 sec. The initial ambient temperature was 299 K and convective heat transfer coefficient was 20 W/m²-K. calculate the surface temperature of the wood veneer in Celsius. Write down your answer rounded to the nearest ones without units.

Wood veneer properties:

Thermal conductivity of wood veneer = 0.15 W/m-K, density = 580 kg/m3, specific heat = 1750 J/kg-K, surface absorptivity = 0.85.

Below is the complementary error function table.

					Hundredth	s digit of x				
x	0	1	2	3	4	5	6	7	8	9
0.0	1.00000	0.98872	0.97744	0.96616	0.95489	0.94363	0.93238	0.92114	0.90992	0.89872
0.1	0.88754	0.87638	0.86524	0.85413	0.84305	0.83200	0.82099	0.81001	0.79906	0.78816
0.2	0.77730	0.76648	0.75570	0.74498	0.73430	0.72367	0.71310	0.70258	0.69212	0.68172
0.3	0.67137	0.66109	0.65087	0.64072	0.63064	0.62062	0.61067	0.60079	0.59099	0.58126
0.4	0.57161	0.56203	0.55253	0.54311	0.53377	0.52452	0.51534	0.50625	0.49725	0.48833
0.5	0.47950	0.47076	0.46210	0.45354	0.44506	0.43668	0.42838	0.42018	0.41208	0.40406
0.6	0.39614	0.38832	0.38059	0.37295	0.36541	0.35797	0.35062	0.34337	0.33622	0.32916
0.7	0.32220	0.31533	0.30857	0.30190	0.29532	0.28884	0.28246	0.27618	0.26999	0.26390
0.8	0.25790	0.25200	0.24619	0.24048	0.23486	0.22933	0.22390	0.21856	0.21331	0.20816
0.9	0.20309	0.19812	0.19323	0.18844	0.18373	0.17911	0.17458	0.17013	0.16577	0.16149
1.0	0.15730	0.15319	0.14916	0.14522	0.14135	0.13756	0.13386	0.13023	0.12667	0.12320
1.1	0.11979	0.11647	0.11321	0.11003	0.10692	0.10388	0.10090	0.09800	0.09516	0.09239
1.2	0.08969	0.08704	0.08447	0.08195	0.07949	0.07710	0.07476	0.07249	0.07027	0.06810
1.3	0.06599	0.06394	0.06193	0.05998	0.05809	0.05624	0.05444	0.05269	0.05098	0.04933
1.4	0.04771	0.04615	0.04462	0.04314	0.04170	0.04030	0.03895	0.03763	0.03635	0.03510
1.5	0.03389	0.03272	0.03159	0.03048	0.02941	0.02838	0.02737	0.02640	0.02545	0.02454
1.6	0.02365	0.02279	0.02196	0.02116	0.02038	0.01962	0.01890	0.01819	0.01751	0.01685
1.7	0.01621	0.01559	0.01500	0.01442	0.01387	0.01333	0.01281	0.01231	0.01183	0.01136
1.8	0.01091	0.01048	0.01006	0.00965	0.00926	0.00889	0.00853	0.00818	0.00784	0.00752
1.9	0.00721	0.00691	0.00662	0.00634	0.00608	0.00582	0.00557	0.00534	0.00511	0.00489
2.0	0.00468	0.00448	0.00428	0.00409	0.00391	0.00374	0.00358	0.00342	0.00327	0.00312
2.1	0.00298	0.00285	0.00272	0.00259	0.00247	0.00236	0.00225	0.00215	0.00205	0.00195
2.2	0.00186	0.00178	0.00169	0.00161	0.00154	0.00146	0.00139	0.00133	0.00126	0.00120
2.3	0.00114	0.00109	0.00103	0.00098	0.00094	0.00089	0.00085	0.00080	0.00076	0.00072
2.4	0.00069	0.00065	0.00062	0.00059	0.00056	0.00053	0.00050	0.00048	0.00045	0.00043
2.5	0.00041	0.00039	0.00037	0.00035	0.00033	0.00031	0.00029	0.00028	0.00026	0.00025
2.6	0.00024	0.00022	0.00021	0.00020	0.00019	0.00018	0.00017	0.00016	0.00015	0.00014
2.7	0.00013	0.00013	0.00012	0.00011	0.00011	0.00010	0.00009	0.00009	0.00008	0.00008
2.8	0.00008	0.00007	0.00007	0.00006	0.00006	0.00006	0.00005	0.00005	0.00005	0.00004
2.9	0.00004	0.00004	0.00004	0.00003	0.00003	0.00003	0.00003	0.00003	0.00003	0.00002
3.0	0.00002	0.00002	0.00002	0.00002	0.00002	0.00002	0.00002	0.00001	0.00001	0.00001
3.1	0.00001	0.00001	0.00001	0.00001	0.00001	0.00001	0.00001	0.00001	0.00001	0.00001
3.2	0.00001	0.00001	0.00001	0.00000	0.00000	0.00000	0.00000	0.00000	0.00000	0.00000

Correct!

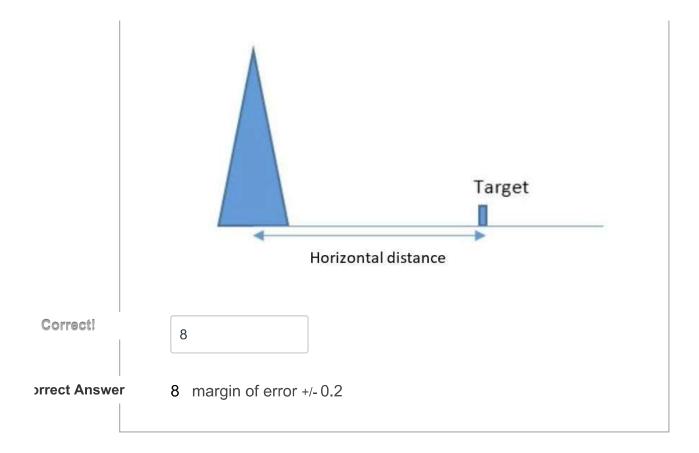
394

orrect Answer

394 margin of error +/- 2

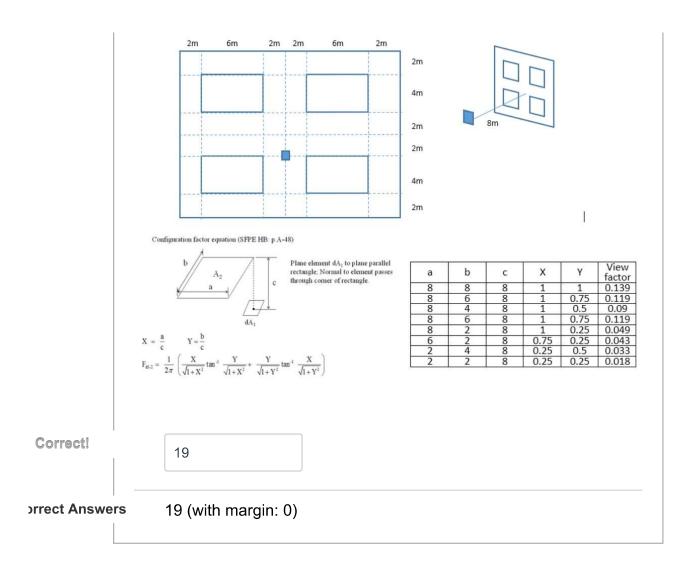
Question 16 3 / 3 pts

Heptane spills and forms a pool fire in a dike area of 8 m diameter. Calculate heat flux in kW/m2 on the exposed surface of the target horizontally 22 m away from the base of fire. Assume the followings: point source height = 13 m, heat release rate = 116 MW, radiative fraction = 0.66. Write down your answer rounded to the first decimal place without any units.



Question 17 3 / 3 pts

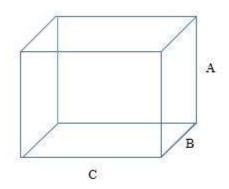
Calculate the radiative heat flux in kW/m2 on the target which is 8 m away from the surface of the window openings. Assume the emissivity of the window opening is 0.9 and the surface temperature 1200K. Write down your answer rounded to the nearest ones without any units.



Question 18 0 / 3 pts

Calculate the fraction of the radiation leaving the walls and ceiling to the floor for the following configuration. Write down your answer rounded to the second decimal point without units.

A=4 m, B= 8 m, C = 9 m



ou Answered

0.21

orrect Answer

0.35 margin of error +/- 0.01

Question 19

3 / 3 pts

Only one side of a 2 mm thick steel wall was suddenly exposed to 500 °C air stream.

Calculate the surface temperature of the steel in Celsius at 12 sec with a time step of 4 sec.

Write down your answer rounded to the nearest ones without units.

- Initial temperature = 20 °C
- Convective heat transfer coefficient on both sides = 10 W/m2-K
- Steel specific heat = 460 J/kg-K
- Steel density = 7500 kg/m3

Correct!

28

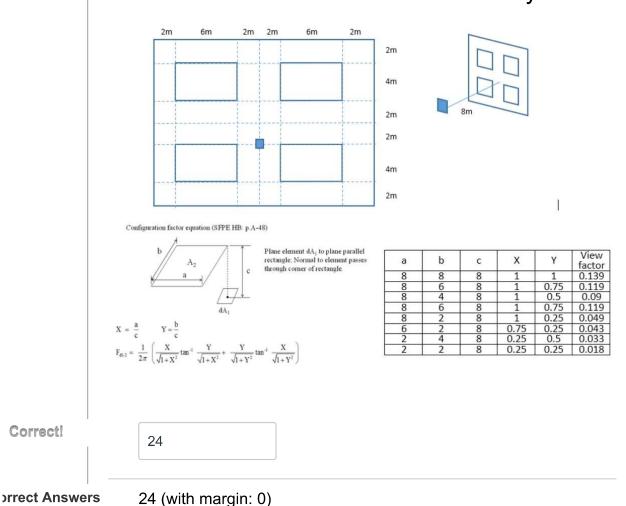
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Question 20 3 / 3 pts

Calculate the radiative heat flux in kW/m2 on the target which is 8 m away from the surface of the window openings. Assume the emissivity of the window opening is 0.9 and the surface temperature 1000 C. Write down your answer rounded to the nearest ones without any units.



Correct!



Correct!	2
	O 119

Quiz Score: 88 out of 100