

Written test „Structure of Matter“ 2018, December 4th

-Mid term test-

solution proposals

Name:

Matriculation number:

1. Multiple-choice test: Please tick the **box(es)** with the correct answer(s)!
(correctly ticked box: +1/2 point; wrongly ticked box: -1/2 point)

A CO ₂ laser generates light at a wavelength of 10.6μm. This wavelength belongs to the	Ultraviolet spectral range	
	Visible spectral range	
	Infrared spectral range	x
Vibrations of nuclei in molecules and solids cause light absorption in the	Ultraviolet spectral range	
	Visible spectral range	
	Infrared spectral range	x
The Paschen series in the emission spectrum of hydrogen atoms is observed in the	Ultraviolet spectral range	
	Visible spectral range	
	Infrared spectral range	x
Doppler broadening of spectral lines in gases results in a linshape that is	Gaussian	x
	Lorentzian	
	triangular	
The Einstein coefficient A ₂₁ responsible for spontaneous light emission is given in	s ⁻¹	x
	W/m ²	
	m/s	
With increasing temperature, the light intensity emitted by a heated black body will	increase	x
	decrease	
	remain constant	
Electrons belong to the class of	fermions	x
	bosons	
	Neither of them	
Slater determinants describe the wavefunction of systems of non-interacting	fermions	x
	bosons	
	Neither of them	

$$\epsilon_0 = 8.86 \cdot 10^{-12} \text{F/m}$$

$$\mu_0 = 1.256 \cdot 10^{-6} \text{Vs/(Am)}$$

Plancks constant: $h = 6.625 \cdot 10^{-34} \text{Ws}^2$

elementary charge: $e = 1.602 \cdot 10^{-19} \text{As}$

electron rest mass: $m_e = 9.108 \cdot 10^{-31} \text{kg}$

proton rest mass: $m_p = 1.672 \cdot 10^{-27} \text{kg}$

Boltzmanns constant: $k_B = 1.38 \cdot 10^{-23} \text{Ws/K}$

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2. Quantum mechanical treatment of the one - dimensional harmonic oscillator (decide whether the statement is true or wrong and tick the appropriate box):

Statement	true	wrong
The harmonic oscillator has equidistant energy levels.	<input checked="" type="checkbox"/>	<input type="checkbox"/>
All wavefunctions of the harmonic oscillator are even functions of x.	<input type="checkbox"/>	<input checked="" type="checkbox"/>
In the ground state, $\langle x^2 \rangle = 0$	<input type="checkbox"/>	<input checked="" type="checkbox"/>
In the ground state, $\langle x^3 \rangle = 0$	<input checked="" type="checkbox"/>	<input type="checkbox"/>
Electric dipole-allowed quantum transitions occur only between adjacent energy levels	<input checked="" type="checkbox"/>	<input type="checkbox"/>

(5 points): 1 point per correct decision, 0 points per wrong or no decision

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3. Assume an electron confined in a rectangular potential box of length $L = 1\text{nm}$ with impermeable walls (one-dimensional case). What is the transition matrix element x_{jk} and the corresponding oscillator strength for the transition $k = 2 \rightarrow j = 4$?

(3 points)

matrix element = 0 because both wavefunctions are odd with respect to symmetry centre, so their product is even, while x is odd. Once $x_{kj}=0$, the oscillator strength must be zero, too.

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4. Calculate the relation between the ground state energies of an electron in the hydrogen atom and in the Li^{2+} ($Z = 3$, mass number = 7) ion!

(4 points)

$$\frac{E_{\text{Li}^{2+}}}{E_H} = Z_{\text{Li}^{2+}}^2 \frac{\mu_{\text{Li}^{2+}}}{\mu_H} = 9 \frac{\mu_{\text{Li}^{2+}}}{\mu_H} = 63 \frac{m_e + m_n}{m_e + 7m_n} = 63 \frac{1 + \frac{m_n}{m_e}}{1 + 7 \frac{m_n}{m_e}} \approx 63 \frac{1 + 1836}{1 + 7 * 1836} \approx 9.004$$

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5. Calculate the commutation relation $[\hat{x}, \hat{p}_x^2]$ (x - Cartesian coordinate; p-momentum)!

(6 points)

$$[\hat{x}, \hat{p}_x^2] = 2i\hbar\hat{p}_x = 2\hbar^2 \frac{\partial}{\partial x}$$

a) direct calculation using $\hat{p}_x = -i\hbar \frac{\partial}{\partial x}$

$$\begin{aligned} b) [\hat{x}, \hat{p}_x^2] &= \hat{x}\hat{p}_x^2 - \hat{p}_x^2\hat{x} = \hat{x}\hat{p}_x^2 - \hat{p}_x^2\hat{x} + \hat{p}_x\hat{x}\hat{p}_x - \hat{p}_x\hat{x}\hat{p}_x = \hat{p}_x\hat{x}\hat{p}_x - \hat{p}_x^2\hat{x} + \hat{x}\hat{p}_x^2 - \hat{p}_x\hat{x}\hat{p}_x = \\ &= \hat{p}_x(\hat{x}\hat{p}_x - \hat{p}_x\hat{x}) + (\hat{x}\hat{p}_x - \hat{p}_x\hat{x})\hat{p}_x = \hat{p}_x(i\hbar) + (i\hbar)\hat{p}_x = 2i\hbar\hat{p}_x \end{aligned}$$

$$c) [\hat{x}, \hat{p}_x^2] = \hat{x}\hat{p}_x^2 - \hat{p}_x^2\hat{x} = (i\hbar + \hat{p}_x\hat{x})\hat{p}_x - \hat{p}_x(\hat{x}\hat{p}_x - i\hbar) = 2i\hbar\hat{p}_x$$

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6. Consider the hydrogen state as given by $n=2, l=1, m=0$. Its wavefunction is:

$$|210\rangle = \frac{1}{4\sqrt{2\pi}a_0^{\frac{3}{2}}} \frac{r}{a_0} e^{-\frac{r}{2a_0}} \cos \theta \quad \text{with } a_0 - \text{Bohr's radius}$$

- What is the total energy of the electron in this state (in Ry)? (2 points)
- Calculate the expectation value $\left\langle \frac{1}{r} \right\rangle$ in this state by direct integration in spherical coordinates! (8 points)
- Indicate the expectation value of the potential energy (in Ry) in this state! (2 points)

$$E_{\text{ground state}} = -\frac{Ry}{n^2} = -\frac{Ry}{4}$$

$$|210\rangle = \frac{1}{4\sqrt{2\pi}a_0^{\frac{3}{2}}} \frac{r}{a_0} e^{-\frac{r}{2a_0}} \cos \theta$$

$$\Rightarrow \left\langle \frac{1}{r} \right\rangle = \frac{1}{32\pi a_0^5} \underbrace{\int_0^\infty r^3 e^{-\frac{r}{a_0}} dr}_{6a_0^4} \underbrace{\int_0^{2\pi} d\varphi}_{2\pi} \underbrace{\int_0^\pi \cos^2 \theta \sin \theta d\theta}_{\frac{2}{3}} = \frac{1}{4a_0} \Rightarrow$$

$$\Rightarrow \langle U(r) \rangle = -\frac{e^2}{4\pi\epsilon_0} \left\langle \frac{1}{r} \right\rangle = -\frac{e^2}{16\pi\epsilon_0 a_0} = -\frac{Ry}{2}$$

