Homework for Chapter 6

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1. 已知钒原子的基态是 ${}^4F_{3.}$ (1)问钒原子束在不均匀横向磁场中将分裂为几束? (2)求基态钒原子的有效磁矩 μ_J .

$$l=3 \qquad j=\frac{3}{2} \qquad s=\frac{3}{2}$$

 \Rightarrow

$$g = 1 + \frac{j(j+1) - l(l+1) + s(s+1)}{2j(j+1)} = 0.4$$

 \Rightarrow

$$m = j, j - 1, \dots, -j = \frac{3}{2}, \frac{1}{2}, -\frac{1}{2}, -\frac{3}{2}$$

Since m has 4 different values, the beam of atoms in magnet field will split into 4 sub-beams.

$$\mu_l = g \sqrt{j \, (j+1)} \mu_B = 7.18361 \times 10^{22} \, \, \mathrm{A \cdot m^2}$$

3. Li 漫线系的一条 $(3^2D_{\frac{3}{2}} \rightarrow 2^2P_{\frac{1}{2}})$ 在磁场中将分裂成多少条光谱线? 试作出相应的能级跃迁图.

For $3^2D_{\frac{3}{2}}$

$$l=2 \qquad j=\frac{3}{2} \qquad s=\frac{1}{2}$$

 \Rightarrow

$$m = j, j - 1, \dots, -j = \frac{3}{2}, \frac{1}{2}, -\frac{1}{2}, -\frac{3}{2}$$

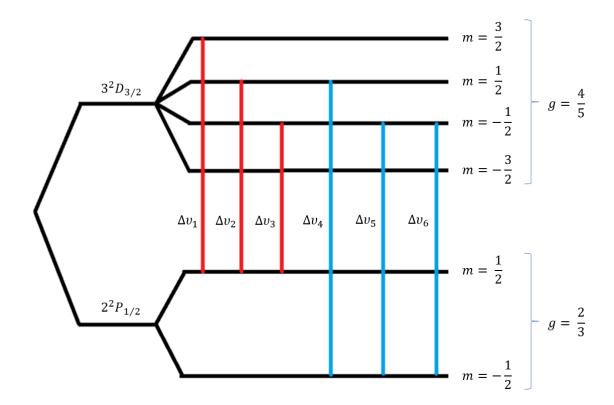
$$g=1+\frac{j\left(j+1\right)-l\left(l+1\right)+s\left(s+1\right)}{2j\left(j+1\right)}=\frac{4}{5}$$

For $2^2 P_{\frac{1}{2}}$

$$l = 1$$
 $j = \frac{1}{2}$ $s = \frac{1}{2}$

 \Rightarrow

$$m = j, j - 1, \dots, -j = \frac{1}{2}, -\frac{1}{2}$$
$$g = 1 + \frac{j(j+1) - l(l+1) + s(s+1)}{2j(j+1)} = \frac{2}{3}$$



$$\nu_{1} = \left(\frac{3}{2} \cdot \frac{4}{5} - \frac{1}{2} \cdot \frac{2}{3}\right) L = \frac{13}{15} L$$

$$\nu_{2} = \left(\frac{1}{2} \cdot \frac{4}{5} - \frac{1}{2} \cdot \frac{2}{3}\right) L = \frac{1}{15} L$$

$$\nu_{3} = \left(-\frac{1}{2} \cdot \frac{4}{5} - \frac{1}{2} \cdot \frac{2}{3}\right) L = -\frac{11}{15} L$$

$$\nu_{4} = \left(\frac{1}{2} \cdot \frac{4}{5} + \frac{1}{2} \cdot \frac{2}{3}\right) L = \frac{11}{15} L$$

$$\nu_{5} = \left(-\frac{1}{2} \cdot \frac{4}{5} + \frac{1}{2} \cdot \frac{2}{3}\right) L = -\frac{1}{15} L$$

$$\nu_{6} = \left(-\frac{3}{2} \cdot \frac{4}{5} + \frac{1}{2} \cdot \frac{2}{3}\right) L = -\frac{13}{15} L$$

5. 氦原子光谱中波长为 6678.1Å $(1s3d^1D_2\rightarrow 1s2p^1P_1)$ 及 7065.1Å $(1s3s^3S_1\rightarrow 1s2p^3P_0)$ 的两条谱线,在磁场中发生塞曼效应时各分裂成几条? 分别作出能级跃迁图.

 $1s3d^{1}D_{2} \rightarrow 1s2p^{1}P_{1}$:

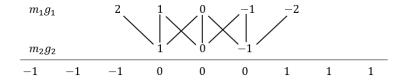
$$l_{1} = 2 l_{2} = 1$$

$$s_{1} = 0 s_{2} = 0$$

$$j_{1} = 2 j_{2} = 1$$

$$g = 1 + \frac{j(j+1) - l(l+1) + s(s+1)}{2j(j+1)} \Rightarrow \begin{cases} g_{1} = 1 \\ g_{2} = 1 \end{cases}$$

$$m = j, j - 1, \dots, -j \Rightarrow \begin{cases} m_{1} = 2, 1, 0, -1, -2 \\ m_{2} = 1, 0, -1 \end{cases}$$



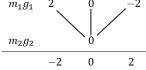
The spectral line will split into 3 components in Zeeman effect. $1s3s^3S_1 \to 1s2p^3P_0 \colon$

$$l_1 = 0$$
 $l_2 = 1$
 $s_1 = 1$ $s_2 = 1$
 $j_1 = 1$ $j_2 = 0$

 \Rightarrow

$$\begin{cases} g_1 = 2 \\ g_2 = 0 \end{cases}$$

$$m = j, j - 1, \dots, -j \Rightarrow \begin{cases} m_1 = 1, 0, -1 \\ m_2 = 0 \end{cases}$$



The spectral line will split into 3 components in Zeeman effect.

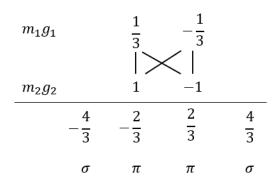
6. Na 原子从 $3^2P_1 \rightarrow 3^2S_1$ 跃迁的光谱线波长为 5896 Å, 在 B=2.5 韦伯/米²的磁场中发生塞曼分裂。问从垂直于磁场方向观察, 其分裂为多少条光谱线。其中波长最长和最短的两条光谱线的波长各多少 Å?

$$l_1 = 1$$
 $l_2 = 0$
 $s_1 = \frac{1}{2}$ $s_2 = \frac{1}{2}$
 $j_1 = \frac{1}{2}$ $j_2 = \frac{1}{2}$

 \Rightarrow

$$\begin{cases} g_1 = \frac{2}{3} \\ g_2 = 2 \end{cases}$$

$$m = j, j - 1, \dots, -j \Rightarrow \begin{cases} m_1 = \frac{1}{2}, -\frac{1}{2} \\ m_2 = \frac{1}{2}, -\frac{1}{2} \end{cases}$$



Observing in perpendicular orientation, the spectral line will split into 4 components.

$$\Delta\left(\frac{1}{\lambda}\right) = -\frac{\Delta\lambda}{\lambda^2} = \frac{4}{3}L \Rightarrow \Delta\lambda_{max} = \frac{8}{3}L\lambda^2 = 1.08~\text{Å}$$