## Curium (Cm)

Atomic number: 96 S Mass number: 247

**Description:** Curium is named in honor of Pierre and Marie Curie

Discoverer by: Glenn Seaborg and colleagues in 1944.

## **Properties:**

Melting point: 1618K Boiling Point: NA

Density: 13.51gm/cm<sup>3</sup> State: Solid

Group: Actinides Period: 7 Block: f

Electronic configuration: [Rn]5f<sup>7</sup>6d<sup>1</sup>7s<sup>2</sup>

Isotopes: <sup>243</sup>Cm, <sup>248</sup>Cm

Appearance: Radioactive metal silver in color.

Tarnish rapidly in air

## Uses:

1) Used to provide power to electrical equipment on space mission.

Source: www.rsc.org

