

YTTERBIUM

Ytterbium compounds:

- * Ytterbium (iii) chloride
- Ytterbium (iii) oxide
- Ytterbium (ii) chloride

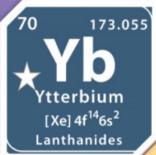
Ytterbinm was isolated in 1878 by
Jean Charles Galissard de
Marignac who was a Swiss
chemist working at the University
of Geneva at the time. Its
discovery can be traced back to the
oxide yttria. Origin of name:
named after the village of
"Ytterby" near Vaxholm in
Sweden.

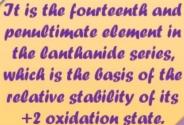


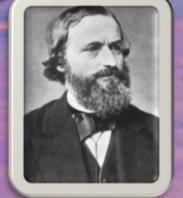
😶 🛮 🛂 tterbium metal tarnishes slowly in air.

Mixtures of powdered ytterbium with polytetrafluoroethylene or hexachloroethane burn with luminous emerald-green flame.

- tterbium dissolves slowly in water, but quickly in acids, liberating hydrogen gas.
- Vtterbium is quite electropositive and it reacts slowly with cold water and quite quickly with hot water to form ytterbium (III) hydroxide.









Boiling point: 1469 K (824°C, 2185°F)

Melting point: 1097 K (824°C, 1515°F)



It is the 44th most abundant element in Earth's crust.

It is monazite.

It is one of the

most common rare

earths, present at

about 2.7 to 8 ppm in the crust.

common in the mineral

Uses: ytterbium metal has possible use in improving the grain refinement, strength, and other mechanical properties of stainless steel. One isotope is reported to have been used as radiation source substitute for a portable X-ray machine where electricity is unavailable. Few other uses have been found.

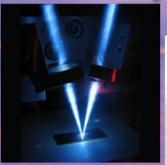




- The isotopes of ytterbium range in atomic weight 147.967 u to 180.9562 u.
- Twenty seven radioisotopes have been characterized, with the most stable being 166^Vb, 169^Vb, 175^Vb.
- This element also has 12 meta states, with the most stable being 169m^2/b.











DID YOU KNOW

The world's most stable atomic clock is made with Ytterbium metal