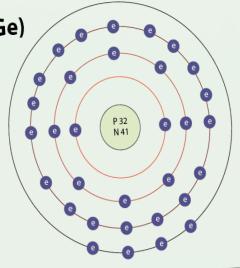
## ELEMENTO-PEDIA GERMANIUM

32 Ge 72.64

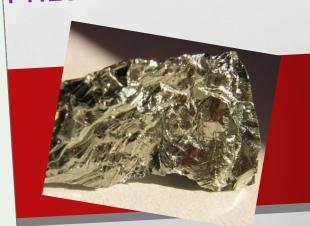
Germanium

## Germanium (Ge)

Energy levels: 4 Protons: 32 Neutrons: 41 Electrons: 32



PRESENTATION BY-T SANJANA



XI C

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## **INTERESTING FACTS**

85% of the World's Consumption of germanium is in fibre optics.

Germanium did not become a Useful element until 1945, when researchers discovered its excellent semi-conducting properties. It is a metalloid .

Germanium is one of the few elements that expands when it freezes, like water

The name,
Germanium, derives
from the latin word
'germania'(Germany
) and was given to
the element by
Winkler.

## **USES OF GERMANIUM**

- BEING A SEMICONDUCTOR, IT HAS PROPERTIES SIMILAR TO TIN AND SILICON AND SO, GERMANIUM IS USED BY THE SEMICONDUCTOR INDUSTRIES ON A REGULAR BASIS.
- IT IS USED IN MANUFACTURE OF WIDE ANGLE CAMERA LENSES.
- THE ELEMENT IS USED IN TRANSISTORS.
- GERMANIUM IS ONE OF THE ELEMENTS ON INSIDE OF FLUORESCENT BULBS (USED AS A PHOSPHOR).
- IT IS ALSO ADDED TO ALLOYS.



