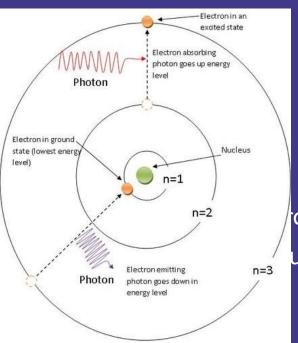
Imperial College London

Materials at Manufactur



THE CHEMISTRY OF COLOURED GLASS

Glass is coloured in 3 main ways. It can have transition or rare earth metal ions added; it can be due to colloidal particles formed in the glass; or it can be due to particles which are coloured themselves. This graphic shows some of the typical chemical elements that are used to colour glass.

SODA-LIME GLASS

COMPOSITION

SiO₂ 70-74%

SILICON DIOXIDE

CaO 10-14%

CALCIUM OXIDE

Na₂O 13-16% SODIUM OXIDE

Soda-lime glass is the most common glass type, making up an estimated 90% of all manufactured glass. Its uses include containers, windows, bottles, and drinking glasses. The above percentages are a general composition only; other compounds are also present in smaller amounts.





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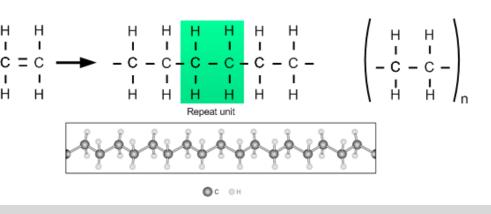
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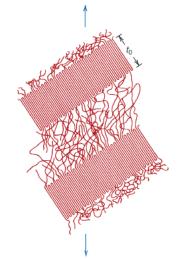
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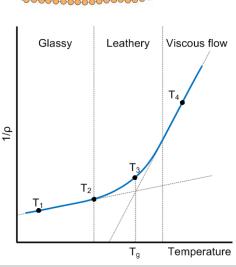
Last time on M&M...

Polymers

- Types of plastics: thermoplastics, thermosets and elastomers
- Polymerisation of monomers
- Interatomic (covalent) and intermolecular (van der Waals) forces
- Chain types: linear, branched and cross linked
- Packing: crystalline, semi-crystalline and amorphous polymers
- Tensile properties of polymers Chain entanglement
- Temperature dependent properties The glass transition temperature
- Visco-elastic properties







Learning objectives

- Define what a ceramic is
- Describe the flexural strength, 3-point bending tests and why ceramics are tested differently to metals
- Understand the 6 main categories of ceramics: glasses, clays, refractories, abrasives, cements and advanced ceramics, giving examples of their applications



Why is it important

Ceramics and glasses are a class of material with some unique properties which, as engineers, we wish to utilize. Despite having a low toughness, ceramics have high hardness and excellent high temperature and corrosion resistance and are thermal and electrical insulators. This opens up several important applications for which ceramics are the optimum material.

With advanced understanding of these materials new applications have come about which have transformed our daily lives



What are ceramics/glasses?

Ceramics are typically non-metallic, inorganic solids. The so called traditional ceramics include claybased bricks, tiles, porcelain, pipes and crockery.

As engineers we are mainly concerned with and interested in modern, high performance ceramics.

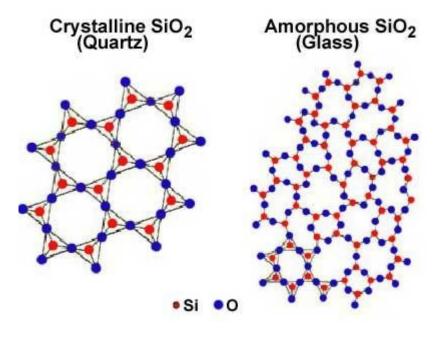
Ceramics have crystalline, semi-crystalline and amorphous structures whereas glass is typically amorphous





Quartz vs glass

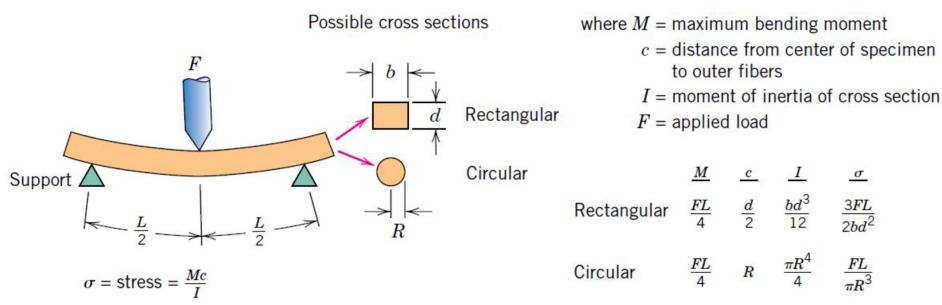




Mechanical properties of ceramics

The stress-strain behaviour of brittle ceramics is not usually ascertained by a tensile test, for three reasons:

- 1. It is difficult to prepare and test specimens having the required geometry
- 2. It is difficult to grip brittle materials without fracturing them
- Ceramics fail after only about 0.1% strain, which necessitates that tensile specimens be perfectly aligned to avoid the presence of bending stresses, which are not easily calculated
- Transverse loading is often more suitable



Mechanical properties of ceramics

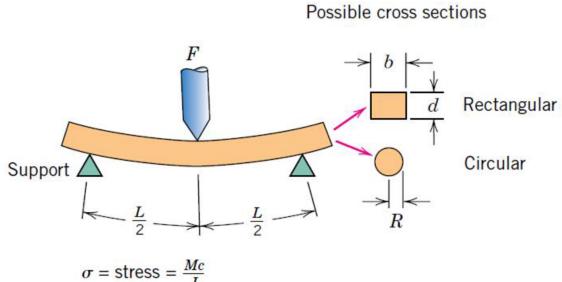
 The stress at fracture using this flexure test is known as the flexural strength, modulus of rupture, fracture strength, or the bend strength, an important mechanical parameter for brittle ceramics

$$\sigma_{fs} = \frac{3F_fL}{2bd^2}$$
 Rectangular

$$\sigma_{fs} = \frac{F_f L}{\pi R^3}$$

Cylindrical

- σ_{fs} will depend on the thickness of the sample
- With increasing specimen volume (that is, specimen volume exposed to a tensile stress) there is an increase in the probability of the existence of a crack-producing flaw and consequently, a decrease in flexural strength

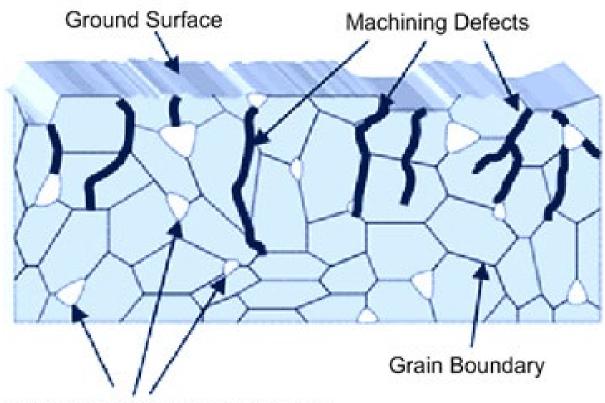


where M = maximum bending momentc = distance from center of specimen to outer fibers I = moment of inertia of cross sectionF = applied load

Rectangular $\frac{FL}{4}$ $\frac{d}{2}$ $\frac{bd^3}{12}$ Circular $\frac{FL}{4}$ R $\frac{\pi R^4}{4}$

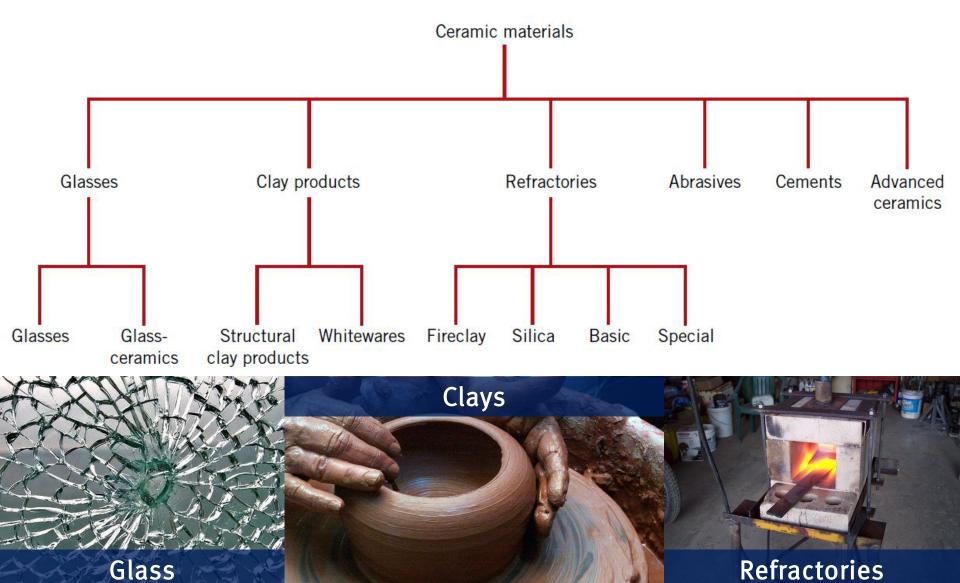
Stress concentrators

Grinding Force + Excessive Force



Intrinsic Defects (Pore, Impurity)

Types of ceramics



Glasses

- The glasses are a familiar group of ceramics; containers, lenses, and fiberglass represent typica applications
- They are non-crystalline materials containing other oxides, notably CaO, Na₂O, K₂O, and Al₂O₃
- A typical soda-lime glass consists of approximately 70 wt% SiO₂, the balance being mainly Na₂O (soda) and CaO (lime)
- Optically transparent and easy to make





Glass type	Charactierstics and applications
Fused silica	High melting temperature, very low coefficient of expansion (thermally shock
	resistant)
96% Silica (Vycor TM)	Thermally shock and chemically resistance - laboratory ware
Borosilicate (Pyrex TM)	Thermally shock and chemically resistant-ovenware
Container (soda-lime)	Low melting temperature, easily worked, also durable
Fiberglass	Easily drawn into fibers-glass resin composites
Optical flint	High density and high index of refraction-optical lenses
Glass-ceramic	Easily fabricated; strong; resists thermal shock-ovenware
(Pyrocerm TM)	

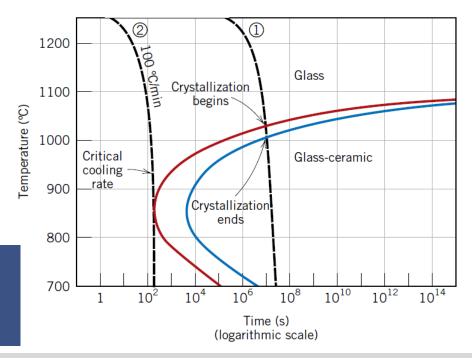
Glass-ceramics

- Most inorganic glasses can be made to transform from a non-crystalline state to one that is crystalline by the proper high-temperature heat treatment
- This process is called crystallization, and the product is a fine-grained polycrystalline material which is often called a glass-ceramic
- High mechanical strength, low thermal expansion, high temperature operation, good dielectric properties and good biological compatibility

Relatively easy to make



Continuous cooling transformation diagram for the crystallization of lunar glass



Clays

- One of the most widely used ceramic raw materials is clay
- This inexpensive ingredient, found naturally in great abundance, often is used as mined without any upgrading of quality
- Easy to form when mixed with water
- Fired at elevated temperatures to improve mechanical properties

Structural clays
Bricks, tiles sewer pipes





Refractories

- Refractory ceramics are able to withstand high temperature without melting or decomposing as well as remaining inert
- Often used to provide thermal insulation
- Applications include: furnace linings, glass manufacturing, metallurgical heat treatment and power generation
- Types include: fire-clay, silica, basic and special refractories



Abrasives

- Abrasive ceramics are used to wear, grind, or cut away other material
- Prime requisite for this group of materials is hardness or wear resistance; in addition, a high degree of toughness is essential to ensure that the abrasive particles do not easily fracture
- High temperatures may be produced from abrasive frictional forces, so some refractoriness is also desirable
- Common abrasives include: diamonds, silicon carbide, tungsten carbide, silica sand etc
- Abrasives are used in 3 main forms

Bonded to grinding wheels

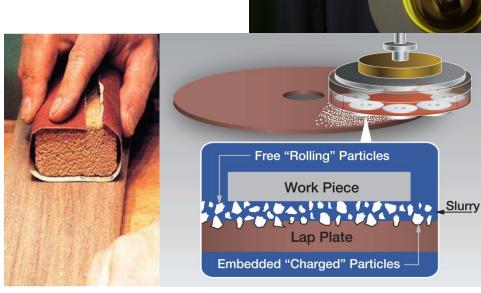
 Abrasive particles are bonded to a wheel by means of a glassy ceramic or an organic resin

Coated

 Abrasives are bonded to some sort of paper or cloth

Loose grains

- Abrasives delivered in a carrier fluid (water or oil)
- Grinding, lapping and polishing



Cements

- Several familiar ceramic materials are classified as inorganic cements: cement, plaster of paris, and lime
- When mixed with water, they form a paste that subsequently sets and hardens
- Portland cement is probably the most used of the cements
- Produced by mixing clay and lime-bearing materials and heating (calcination)
- Resulting clinker is then ground and a gypsum (CaSO₄-2H₂O) is added to set (retard)
- Hydraulic cement develops its hardness with reaction with water
- Non-hydraulic cement involve other compounds to harden the material (i.e. CO₂)

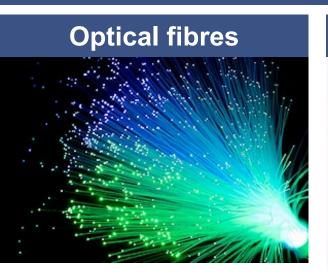


Advanced ceramics

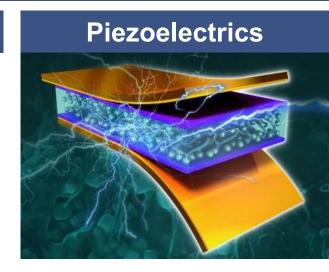
Although the traditional ceramics discussed previously account for the bulk of the production, the development of new and what are termed advanced ceramics has begun and will continue to establish a prominent niche in our advanced technologies.

In particular, electrical, magnetic, and optical properties and property combinations unique to ceramics have been exploited in a host of new products

Examples

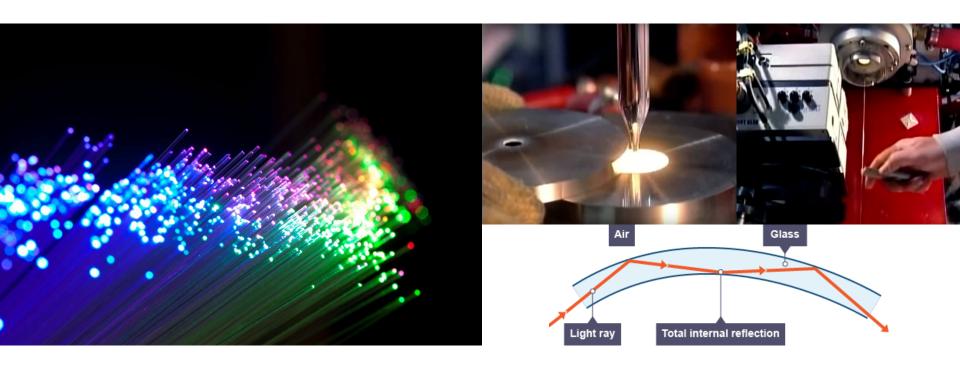






Optic fibres

- Critical component in our modern optical communications systems is the optical fibre
- The optical fiber is made of extremely high-purity silica, which must be free of even minute levels of contaminants and other defects that absorb, scatter, and attenuate a light beam



Ceramic bearings

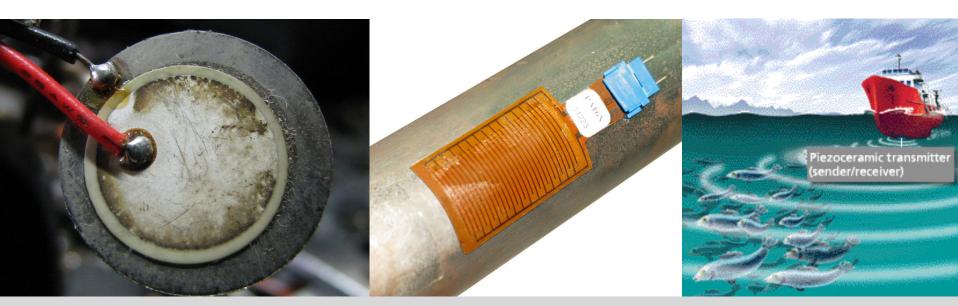
- A bearing consists of balls and races that are in contact with and rub against one another when in use
- In the past, both ball and race components traditionally have been made of bearing steels that are very hard
- Recently silicon nitride (Si₃N₄) balls have begun replacing steel balls
- Lighter weight
 - Density of Si₃N₄ is much less than steel (3.2 versus 7.8 g/cm³)
- Higher speed
 - Lower centrifugal loads (20-40% faster)
- Reduced noise and vibrations
 - Higher Young's modulus (320 GPa vs ~200 GPa) so less deformation and vibrations
- Higher lifetime
 - Better wear resistance (~3-5x better)
- Used in bikes, skates, electric motors, chemical equipment, dental drills





Piezoelectrics

- A few ceramic materials (as well as some polymers) exhibit the unusual phenomenon of piezoelectricity - electric polarisation (i.e. an electric field or voltage) is induced in the ceramic crystal when a mechanical strain (dimensional change) is imposed on it
- Can be used as transducers between mechanical and electrical energies
- Early uses include sonar
- Modern applications include: microactuators for hard disks, ink-jet printing heads, strain gauges, ultrasonic welders, energy harvesting etc



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Next time on M&M...

