PS3

a)

Metabolites:

1	ATP	10	citrulline
2	AMP	11	argininosuccinate
3	Diphosphate	12	arginine
4	phosphate	13	ornithine
5	H ₂ O	14	NADPH
6	Carbamoyl Phosphate	15	H ⁺
7	Aspartate	16	O ₂
8	Fumarate	17	NO
9	urea	18	NADP ⁺

Fluxes:

v1: $ATP + Citru + Aspar \rightarrow AMP + diphos + succi$

 $v2:succi \rightarrow fumar + argin$

 $v3:argin + H2O \rightarrow ornith + urea$

 $v4:carbamoyl + ornith \rightarrow phosph + citru$

 $v51:argin + 1.5NADPH + 1.5H^+ + 202 \rightarrow citru + NO + 1.5NADP + 2H_{2}O$

v52: $citru + NO + 1.5NADP + 2H₂O → <math>argin + 1.5NADPH + 1.5H^{+} + 2O2$

 $b1:source \rightarrow carbamoyl$

b2: $source \rightarrow aspar$

b3: $fumar \rightarrow sink$

b4: $urea \rightarrow sink$

b5: $source \rightarrow ATP$

b6: $source \rightarrow water$

b7: diphos → sink

b8: $phos \rightarrow sink$

b9: AMP → sink

b10: $source \rightarrow NADPH$

b11: source →H⁺

b12: $source \rightarrow O_2$

b13: $NO \rightarrow sink$

b14: source →NADP+

Stoichiometric matrix: Rows are metabolites and columns are fluxes.

b)

Atom matrix: The rows are C, H, N, O, P respectively, and the columns are metabolites 1~18.

10	10	0	0	0	1	4	4	1	6	10	6	5	21	0	0	0	21
16	14	4	3	2	4	7	4	4	13	18	14	12	30	1	0	0	29
5	5	0	0	0	1	1	0	2	3	4	4	2	7	0	0	1	7
13	7	7	4	1	5	4	4	1	3	6	2	2	17	0	2	1	17
3	1	2	1	0	1	0	0	0	0	0	0	0	3	0	0	0	3

A*S=E

E =

From Matlab, we can get E which is shown below.

From E, we can see that there are zeroes on the first 6 columns(for v1~v52).

Therefore, it's elementally balanced for the fluxes.

c)

The goal is to calculate the maximum rate of urea production. To do so, we need to calculate the upper bounds of each flux.

Using the formula from lecture note:

$$0 \le v \le V max * theta * \left(\frac{a}{a+Km}\right)$$
, where Vmax=k_cat*E

and the bound from the problem set:

$$0 \le bj \le \frac{10mmol}{gDW-hr}$$

After calculating these bound values, we further use Flux.jl which is used to solve the FBA problem to get the maximum urea flux which is 657.56 miumol/gDW-hr.