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Methods for lithium extraction—hard rock (1/2)

Two major approaches for lithium extraction: **hard rock extraction** and **brine extraction**. Hard rock mining: mainly used in Australia. Three major approaches—traditional pyrometallurgy, pressure leaching, and bioleaching.

1. **Traditional pyrometallurgy:** incurring significant energy costs and expenses, particularly with heat-sensitive minerals like spodumene. The most commonly used approach due to high recovery rate. **Environmentally the worst!**
2. Pressure leaching: using sodium chloride and calcium hydroxide under high pressure to facilitate lithium dissolution. **Requires very stringent condition of lithium ore.**
3. Bioleaching: microorganisms to dissolve lithium without high energy costs. **Too slow!**

Methods for lithium extraction—brine (1/3)

- **Brine extraction:** 6 primary methods involved: precipitation, chromatography, ion exchange, traditional liquid-liquid extraction, ionic liquid extraction, and membrane processes.
 1. **Precipitation:** Involves forming lithium aluminate to obtain high-purity lithium carbonate. Cheap but highly sensitive to pH and temperature conditions.
 2. **Chromatography:** Separates lithium from magnesium ions using materials like polyacrylamide gel. Effective but costly and challenging to scale.

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 3. **Ion exchange:** Offers high lithium selectivity but requires handling and costly materials, making it less practical for industrial use.

Methods for lithium extraction—brine (2/3)

- **Brine extraction (continued):**
 4. **Traditional liquid-liquid extraction:** Employs solvents, such as tributyl phosphate ($((CH_3CH_2CH_2CH_2O)_3PO)$), which are effective in achieving high lithium recovery rates but raise environmental concerns due to solvent toxicity.

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 6. **Membrane process:** Uses reverse osmosis and nanofiltration. This method offers efficient lithium recovery but is highly sensitive to brine composition and operational factors like pH and pressure.

b

Solar evaporation ponds

Waste

Reagents

Increasing LiCl concentration

Brine

Lithium carbonate

Saline aquifer

Freshwater aquifer

Figure: Source—(Vera et al., 2023)

Environmental problem

In brine mining, two distinct aquifers are exploited, **brine** and **fresh water**. The extraction and re-injection of aquifer leads to 3 major concerns:

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2. **Freshwater usage:** direct lithium extraction (DLE) processes demand 500 m^3 of freshwater per tonne of lithium carbonate.
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It is hard to balance production costs with externality!

Modeling lithium mineral production (1/2)

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 2. a bell-shaped production curve where production rises, peaks, and then declines symmetrically;
 3. a stable extraction technology and costs throughout the resource's lifecycle (which is surely violated)
 4. a constant economic and political policy that ensures continuous resource extraction (which is also likely to be violated due to resource war (Acemoglu et al., 2012; Yared, 2010)).
- The functional form I use for prediction is:

$$q(t) = \frac{Q_{\max} \cdot m \cdot e^{-m \cdot (t - t_{\text{peak}})}}{(1 + e^{-m \cdot (t - t_{\text{peak}})})^2}, \quad (1)$$

Modeling lithium mineral production (2/2)

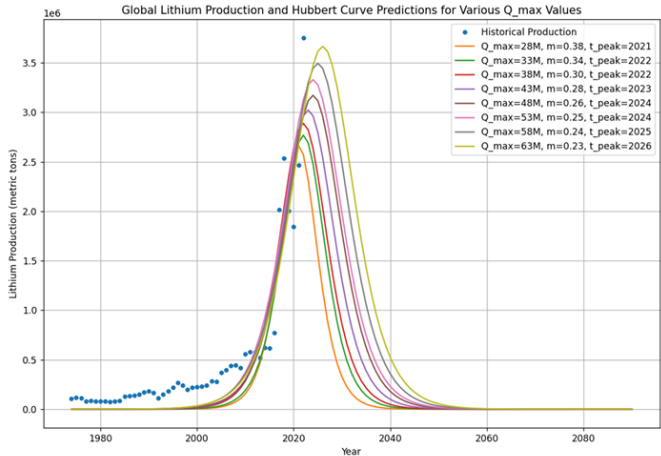


Figure: Lithium production via HCM. Data source: (British Geological Survey).

Modeling lithium mineral demand (1/3)

Issues with demand modeling:

1. **Technology growth factor:** it is hard to weight different technology bundles that may lead to one bundle of factors economically dominates another one. Cobalt manganese oxide (NCM), lithium nickel cobalt aluminum oxide (NCA), and lithium iron phosphate (LFP) each have different lithium composition. A minor factor advantage in any of the path above may lead to a significant *R&D* input that pushes the lithium production onto the corresponding path.

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4. **Recycle:** accrued battery waste plus resource shortage incentivize *R&D* input in recycling (Harner et al., 2010; Gaines et al., 2020).

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- I performed calibration using the early period to compute the conversion rate from lithium mineral to LCE. The conversion rate is $\alpha = 71198651.09897064 \times 10^{-6}$.

Modeling lithium mineral demand (3/3)

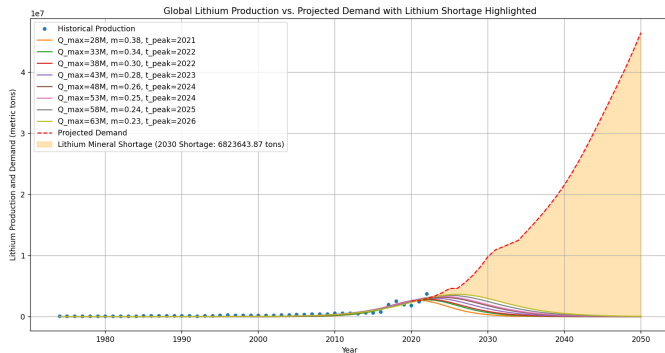


Figure: Data source—(Xu et al., 2020) and (British Geological Survey)

Compared with other modeling

Lithium carbonate global equivalent demand 2030, supply 2021 and 2030 by country, kt

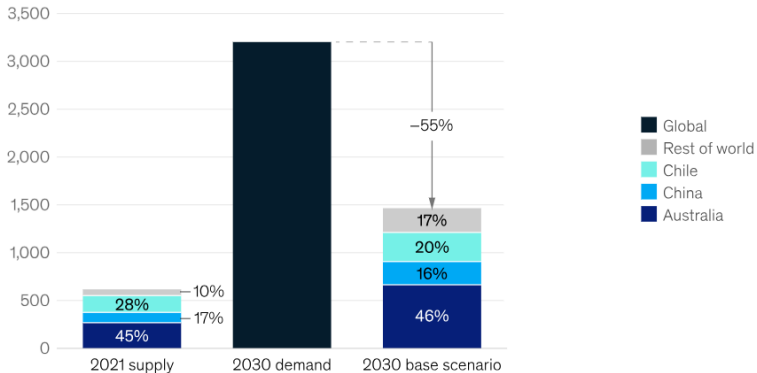


Figure: Source—(Fleischmann et al., 2023)

Lithium battery production–illustration

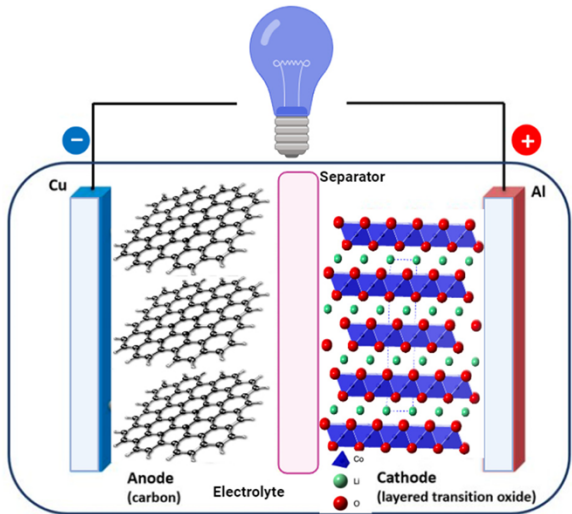


Figure: Source–(Zanoletti et al., 2024)

Lithium-Ion Battery (LIB) Manufacturing Process (1/3)

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- **Roots and Adaptations:**

- LIB manufacturing originated from protocols for consumer electronics, adapted for electric vehicles (Ahmed et al., 2016).
- Core manufacturing stages consistent across designs (cylindrical, pouch, prismatic cells).

Lithium-Ion Battery (LIB) Manufacturing Process (2/3)

Manufacturing Stages:

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 - Coat electrodes onto collectors: aluminum (cathodes) and copper (anodes), followed by calendaring.
- **Cell Assembly:**
 - Dry room assembly: layer, wind, or stack electrodes with separators.

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- Ultrasonic welding commonly used for current collectors.
- Electrolyte filling and final sealing.

Lithium battery production

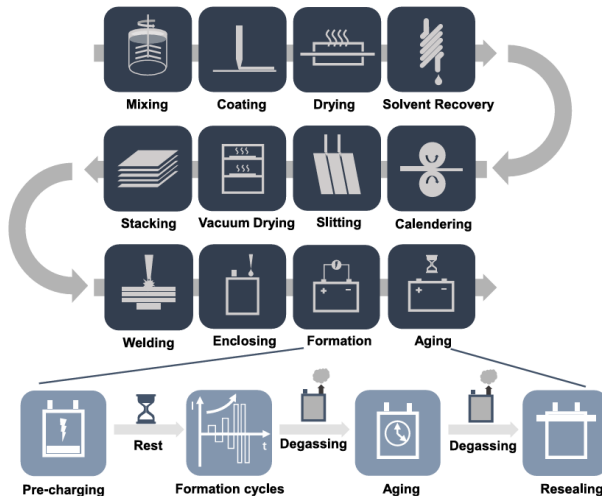


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Costs and energy consumption

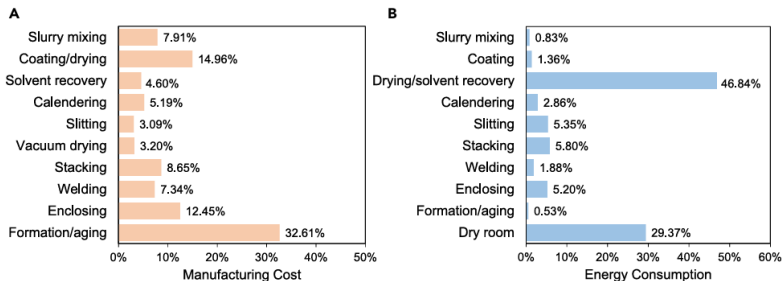


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 5. **Solvometallurgy:** uses ionic liquids and deep eutectic solvents (DESS) to dissolve metals. It has high recovery rates of metals like cobalt, nickel, and lithium, but **too expensive to be applied in industry!**

Lithium recycle (2/3)

Table: Direct comparison of main LIB recycling technologies and pollution characteristics.

	Pyrometallurgy	Hydrometallurgy	Biometallurgy	Solvometallurgy (Ionic Liquid)	Solvometallurgy (DES)
Advantages	Short process flow, low equipment requirements, strong operability	Low energy consumption, great versatility, high product purity, high recovery efficiency	Complete metal recovery, simplicity, cost-effectiveness, low energy consumption, mild conditions	Nonflammable, low volatility, tunable	Nonflammable, low recovery cost, green process, cheap and easy preparation, low toxicity
Disadvantages	High energy consumption, poor metal purity, difficulty in lithium recovery	Need to dispose of large amount of acid and toxic wastewater, long recovery process	Long processes and low kinetics, vulnerability to pollution	Expensive	Difficulty to scale-up, low cathode/DES ratio
Applied at industrial level	Yes	Yes	No	No	No
Main source of pollution	Emission of polluting gases and production of slags	Release of toxic gases (e.g. NO _x , SO _x , Cl ₂)	-	-	-

Note: source—(Zanoletti et al., 2024).

Lithium recycle (3/3)

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 1. **Anodes recycling:** involves pre-treatment, pyrolysis, hydrometallurgy, supercritical fluid techniques, and water treatment to separate the active material (e.g graphite)

(d) Ganfeng

Resource nationalism

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- In Chile, President Gabriel Boric announced plans to create a state-owned lithium company and implement policies that would give the Chilean government majority control over all lithium extraction projects. This move would increase state participation in the industry, particularly for new projects or when renewing existing contracts with private companies such as SQM and Albemarle.

Historical lesson

The War of the Pacific (1879–1884), fought between Chile, Bolivia, and Peru is widely regarded as a resource-driven conflict, primarily over control of the valuable guano (bird excrement), nitrates, and saltpeter deposits in the Atacama Desert. The causes of the war is characterized as the following:

1. Gigantic resource wealth: in the pre-war period, nitrates accounted for 20% of Peru's government revenue and later, 48% of Chile's revenues after its victory.

Research question

Lithium value chain demonstrates the following facts:

1. The Southern American countries control the majority of accessible lithium mine

Empirical method

$$Y_{pt} = \alpha + \beta PP_{pt} + X_{pt}\Lambda + \epsilon_{pt} \quad (2)$$

- $PP_{pt} = \sum_c \frac{x_{pc,2021}}{x_{p,2021}} IMP_{tc}^{country} = \sum_c \frac{x_{pc,2021}}{x_{p,2021}} \frac{IM_{tc}^{country}}{D_{2021} + IM_{2021} - EX_{2021}}$.
- IM_t^c = import from country c to China at time t; D_{2021} = the total lithium demand in China in 2021; EX_{2021} = total export to China in 2021.
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- Many ways of decomposing the full penetration to region/industry/ethnic group specific—by relative import, employment, population density

Empirical method—How could it be wrong?

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- Placebo test on pre-trends (i.e. run zero-first stage test on monthly data 2019-2020).

Panel A: Import						Panel B: Export				
Commodity	Obs	Mean	Std. Dev.	Min	Max	Obs	Mean	Std. Dev.	Min	Max
Lithium	224	794,409.6	2,612,620	176	24,300,000	391	5,255,659	7,188,171	44	63,000,000
Lithium Iron Phosphate	196	809,847.1	1,296,698	170	9,879,130	520	526,394.5	1,312,929	6	9,026,640
Lithium Carbonate	1,371	81,300,000	446,000,000	15	11,400,000,000	827	7,156,669	28,200,000	6	376,000,000
Lithium Chloride	607	741,900.3	2,271,543	7	20,800,000	382	377,905.9	1,089,251	7	15,400,000
Lithium Enriched in Lithium-6	5	17,459.6	3,954.5	12,615	22,755	–	–	–	–	–
Lithium Hexafluorophosphate	161	2,684,438	4,866,950	20	31,000,000	552	7,095,791	9,427,186	11	59,900,000
Lithium Hydroxide	452	5,911,896	18,500,000	36	176,000,000	847	54,600,000	184,000,000	7	1,980,000,000
Lithium Manganate	101	915,832.7	1,325,719	126	10,100,000	233	543,795.1	1,800,999	70	13,700,000
Lithium Nickel Cobalt Aluminum Oxides	345	53,600,000	121,000,000	758	651,000,000	142	8,275,086	20,800,000	7	107,000,000
Lithium Nickel Cobalt Manganese Oxide	1,094	76,000,000	220,000,000	20	2,230,000,000	1,412	29,500,000	97,600,000	7	1,170,000,000
Lithium Oxide	85	72,807.1	334,249.6	99	2,935,759	59	1,701,989	3,797,940	7	18,100,000
Lithium Primary Cells & Batteries	12,631	696,479.9	1,923,192	2	27,200,000	12,798	610,870.2	1,957,702	0	56,500,000
Lithium-ion Electric Accumulators	18,773	6,624,487	26,000,000	6	624,000,000	40,830	13,900,000	88,100,000	2	3,710,000,000

Interpretation on FS

- Instruments are not strong enough neither individually nor jointly. The first-stage F-statistics ($F = 0.81$ for Chile and $F = 2.23$ for Mexico) indicate weak instrument inference ?.
- This leads to:
 1. IV estimator is not asymptotically normal and incorrect SE ?—need bootstrapped SE.
 2. Severe bias— Weak IV could potentially be more biased than OLS estimator ?!
- This means that the renegotiation does not drop the total supply in the resource-rich country.

