Lithium value chain-past, present, and future

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November 18, 2024

Introduction

The electric vehicle stock has increased strongly from a few thousands in 2010 to 11.3 million in 2020, and 142 million electric vehicles are forecast to be on the road by 2030.

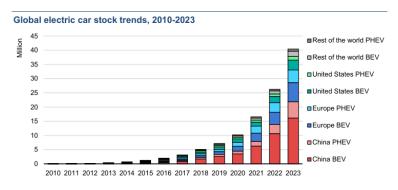


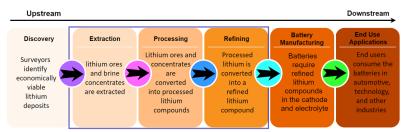
Figure: Electric vehicle growth trend. Source: IEA analysis

Lithium value chain



Figure: Source-(Fleischmann et al., 2023)

Lithium value chain



Source: Willis et al, "Australia's Opportunity in the Lithium Battery Boom," January 30, 2018; Graphic developed by USITC staff.

Figure: Source-(?)

What this presentation is about

- Questions I will answer:
 - How the following processes happen: lithium mining, refinery, LIB manufacturing, and LIB recycling?
 - 2. How to model lithium supply and demand (not to very depth)?
 - 3. Which countries and firms are the main players?
 - 4. What is resource nationalism?
 - 5. Why resource rich country could make lithium a powerful weapon?
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- Example question I will not be able to answer:
 - How to characterize off-path strategy under Markov Perfect Equilibrium?
 - 2. Should I go purchase Tianqi Lithium stock?

Contents

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Evolution of Lithium chain

Mining Lithium battery production Recycling

Major players in the value chain

Resource nationalism

Resource war theory

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Data

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Conclusion



Methods for lithium extraction-hard rock (1/2)

Two major approaches for lithium extraction: **hard rock extraction** and **brine extraction**. Hard rock mining: mainly used in Australia. Three major approaches—traditional pyrometallurgy, pressure leaching, and bioleaching.

1. Traditional pyrometallurgy: incurring significant energy costs and expenses, particularly with heat-sensitive minerals like spodumene. The most commonly used approach due to high recovery rate. Environmentally the worst!

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- 3. Bioleaching: microorganisms to dissolve lithium without high energy costs. Too slow!

Methods for lithium extraction-hard rock (2/2)

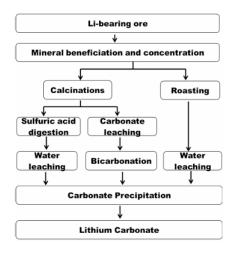


Figure: Source-(Swain, 2017)

Methods for lithium extraction–brine (1/3)

- Brine extraction: 6 primary methods involved: precipitation, chromatography, ion exchange, traditional liquid-liquid extraction, ionic liquid extraction, and membrane processes.
 - Precipitation: Involves forming lithium aluminate to obtain high-purity lithium carbonate. Cheap but highly sensitive to pH and temperature conditions.

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 - Ion exchange: Offers high lithium selectivity but requires handling and costly materials, making it less practical for industrial use.

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 - Membrane process: Uses reverse osmosis and nanofiltration. This
 method offers efficient lithium recovery but is highly sensitive to
 brine composition and operational factors like pH and pressure.

Methods for lithium extraction-brine (3/3)

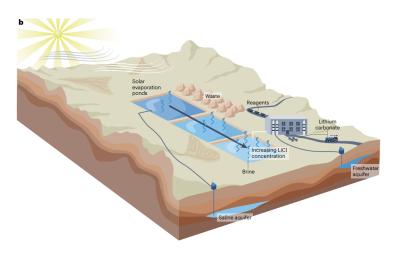


Figure: Source-(Vera et al., 2023)

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It is hard to balance production costs with externality!

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- The functional form I use for prediction is:

$$q(t) = \frac{Q_{\text{max}} \cdot m \cdot e^{-m \cdot (t - t_{\text{peak}})}}{\left(1 + e^{-m \cdot (t - t_{\text{peak}})}\right)^2},\tag{1}$$

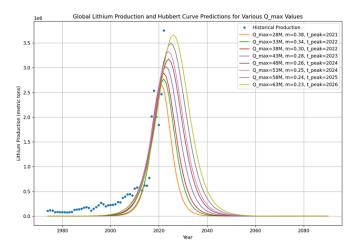


Figure: Lithium production via HCM. Data source: (British Geological Survey).

Issues with demand modeling:

1. **Technology growth factor:** it is hard to weight different technology bundles that may lead to one bundle of factors economically dominates another one. Cobalt manganese oxide (NCM), lithium nickel cobalt aluminum oxide (NCA), and lithium iron phosphate (LFP) each have different lithium composition. A minor factor advantage in any of the path above may lead to a significant R&D input that pushes the lithium production onto the corresponding path.

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- 4. **Recycle:** accrued battery waste plus resource shortage incentivize *R&D* input in recycling (Harper et al., 2019; Gaines et al., 2020).



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- I performed calibration using the early period to compute the conversion rate from lithium mineral to LCE. The conversion rate is $\alpha=71198651.09897064\times10^{-6}$.

Modeling lithium mineral demand (3/3)

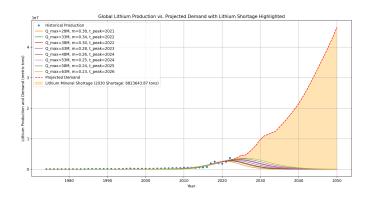


Figure: Data source-(Xu et al., 2020) and (British Geological Survey)

Compared with other modeling

Lithium carbonate global equivalent demand 2030, supply 2021 and 2030 by country, kt

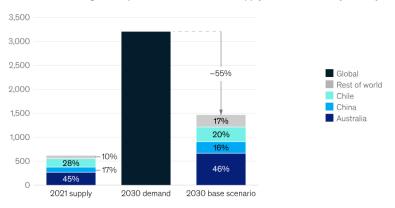


Figure: Source-(Fleischmann et al., 2023)

Lithium battery production-illustration

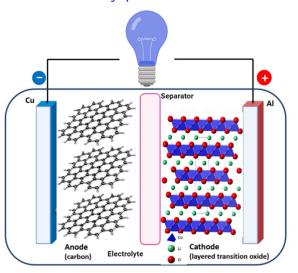


Figure: Source–(Zanoletti et al., 2024)



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- Gas venting and cell aging to stabilize SEI and electrolyte.

Costs and Energy Allocation:

- Cost Allocation (top 3): Formation/aging (32.61%), coating/drying (14.96%), enclosing (12.45%).
- Energy Usage (top 2): Drying/solvent recovery (46.84%), dry room (29.37%).

Lithium battery production

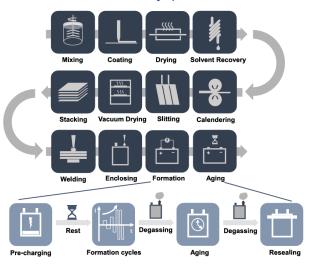


Figure: Source–(Liu et al., 2021)



Costs and energy consumption

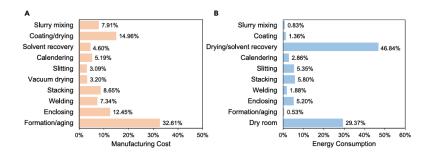


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 - 5. **Solvometallurgy:** uses ionic liquids and deep eutectic solvents (DESs) to dissolve metals. It has high recovery rates of metals like cobalt, nickel, and lithium, but too expensive to be applied in industry!

Table: Direct comparison of main LIB recycling technologies and pollution characteristics.

	Pyrometallurgy	Hydrometallurgy	Biometallurgy	Solvometallurgy (Ionic Liquid)	Solvometallurgy (DES)
Advantages	Short process flow, low equipment re- quirements, strong operability	Low energy con- sumption, great versatility, high product purity, high recovery efficiency	Complete metal recovery, simplicity, cost-effectiveness, low energy consumption, mild conditions	Nonflammable, low volatility, tunable	Nonflammable, low recovery cost, green process, cheap and easy preparation, low toxicity
Disadvantages	High energy con- sumption, poor metal purity, dif- ficulty in lithium recovery	Need to dispose of large amount of acid and toxic wastew- ater, long recovery process	Long processes and low kinetics, vulner- ability to pollution	Expensive	Difficulty to scale- up, low cath- ode/DES ratio
Applied at in- dustrial level	Yes	Yes	No	No	No
Main source of pollution	Emission of pollut- ing gases and pro- duction of slags	Release of toxic gases (e.g. NO_x , SO_x , CI_2)	-	-	-

Note: source-(Zanoletti et al., 2024).

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 - 1. **Anodes recycling:** involves pre-treatment, pyrolysis, hydrometallurgy, supercritical fluid techniques, and water treatment to separate the active material (e.g graphite)
 - 2. **Electrolyte recycling:** still at early stage. Methods using organic solution could retract high share of lithium carbonate at high purity, but the condition is hard to mimic in industry (Xu et al., 2023)

National Level (1/3)

- Major Lithium Producers:
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- United States: Contains substantial lithium reserves of 1100000 metric tons, yet remains largely unexploited due to environmental regulations, geopolitical strategies, and risk management (Sanchez-Lopez, 2023).

National level (2/3)

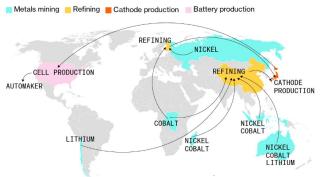


Figure: Source-(British Geological Survey)

National level (3/3)

A New Industry for the U.S.: Battery Materials

Redwood wants to eliminate this 50,000 mile supply chain for battery components



Note: 50,000 miles describes the route, by land and sea, that some materials travel before reaching the car manufacturer as finished battery cells.

Bloomberg

Figure: Source-Bloomberg



Disparities in the LIB Value Chain

- Upstream activities like mining and refining contribute only 2% of the value (\$3.8 billion), while downstream processes generate 46%.
- The final stage of EV manufacturing accounts for 54% of the total value, highlighting the concentration of profits in later stagesTrade et al. (2018).
- Controlling downstream processes is strategically importantHeredia et al. (2020).

Gigafactories:

- Large-scale battery manufacturing plants, pioneered by Tesla, are crucial for meeting global demand and attracting foreign direct investment (FDI)Cooke (2020).
- In 2021, Asia (led by China) accounted for 71% of global Gigafactory capacity, with China alone contributing 69%.
- CATL, a leading Chinese company, held 22% of the world's 500 GWh capacity in 2021 and is expanding operations globally Moores (2021).
- Gigafactories are both economic assets and "geopolitical hot potatoes" Moores (2021).



Innovation, Geopolitics, and Challenges

Innovation in Battery Technology:

- From 2014 to 2018, Japan led battery patenting (41%), followed by South Korea, Germany, the U.S., and ChinaEPO (2020).
- Since 2018, China has rapidly caught up, aligning technological advancements with industrial policiesKalantzakos (2020).
- Japan's leadership in patents has not translated into market dominance (2% global EV share in 2019)Global (2021).

Geopolitical Implications:

- Concentrated manufacturing capacity in Asia creates trade imbalances and supply risks for Europe and the U.S.Månberger and Johansson (2019); Overland (2019).
- Lithium-rich countries like Bolivia and Argentina remain in lower-value supply chain stages, struggling to industrialize their reservesKalantzakos (2020).
- China controls significant portions of the LIB value chain, including:
 - 66% of global graphite
 - 68% of silicon
 - 7% of lithiumLebedeva et al. (2016)
- Integration of raw material control with manufacturing has made



Lithium price

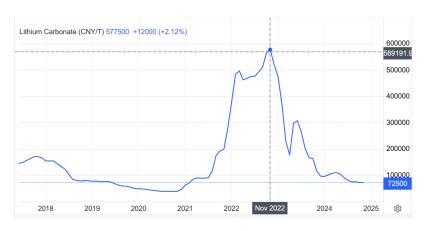
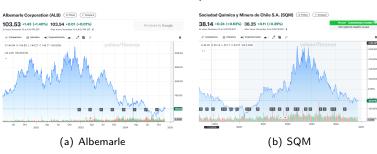


Figure: Source-Trading Economics

Stock market performance





(c) Tiangi







Resource nationalism

 Resource nationalism refers to the efforts by governments to assert control over natural resources located within their borders, often by restricting foreign involvement, renegotiating profit split, increasing taxes, or nationalizing industries. The goal is usually to maximize the country's benefit from the extraction and sale of these resources, which are considered national assets.

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- In 2022, Mexican President Andrés Manuel López Obrador declared lithium a strategic mineral and established a state-run company, LitioMx, to control lithium extraction in the Bacanora Lithium Project, aiming to limit foreign involvement in the burgeoning industry.

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- In Chile, President Gabriel Boric announced plans to create a state-owned lithium company and implement policies that would give the Chilean government majority control over all lithium extraction projects. This move would increase state participation in the industry, particularly for new projects or when renewing existing contracts with private companies such as SQM and Albemarle.

The War of the Pacific (1879–1884), fought between Chile, Bolivia, and Peruis widely regarded as a resource-driven conflict, primarily over control of the valuable guano (bird excrement), nitrates, and saltpeter deposits in the Atacama Desert. The causes of the war is characterized as the following:

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- Competitive market equilibrium: firms are as price takers and induces externality when each extra unit is produced. Attempts for monopolization failed due the conflict among major European banking groups.



Figure: Source: Janitoalevic

Model setup

Basic analysis

Simple implication

Theory-fact isomorphism

Lithium value chain demonstrates the following facts:

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Research question: How will the upstream resource nationalization affect downstream production and price?

Data description

- Chinese customs data: I accessed the monthly Chinese customs data from 2019 January until 2024 September. The data set decomposes lithium related product by country, product type, trade code, and by time. The dataset I constructed consists of import (a total number of 36,557 records) and export (a total number of 61,007 records).
- Firm stock market performance: I use Alpha Vantage API and FMP API to construct a dataset consisting of the Chinese firms that play an important role in the whole manufacturing process.
- Lithium trade penetration to other major manufacturing countries: I consult the Nations (n.d.) to build a dataset on lithium related product import.
- Province level covariates: The data is extracted from China Census Bureau.
- Demand in 2021: Non-EV demand would represent approximately 150 billion Yuan.

Empirical method

$$Y_{pt} = \alpha + \beta P P_{pt} + X_{pt} \Lambda + \epsilon_{pt}$$
 (2)

- $PP_{pt} = \sum_{c} \frac{x_{pc,2021}}{x_{p,2021}} IMP_{tc}^{country} = \sum_{c} \frac{x_{pc,2021}}{x_{p,2021}} \frac{IM_{tc}^{country}}{D_{2021} + IM_{2021} EX_{2021}}.$
- IM_t^c = import from country c to China at time t; D_{2021} = the total lithium demand in China in 2021; EX_{2021} = total export to China in 2021.
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- Many ways of decomposing the full penetration to region/industry/ethnic group specific-by relative import, employment, population density



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• For instance, policy interventions (EV subsidy) may affect PP_{pt} and Y_{pt} , as high subsidy gives enterprise incentive to invest in the subsidy-rich province and pushes up the product price (although payment made by consumer may be lower).

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- $\frac{x_{pc,2021}}{x_{p,2021}}$ could be a bad decomposition approach as the first entrance to China is likely along national boarder!
- Placebo test on pre-trends (i.e. run zero-first stage test on monthly data 2019-2020).

Data description

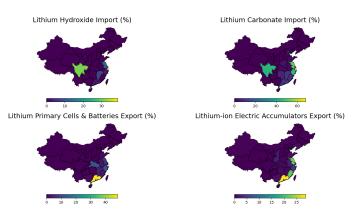


Figure: 01/2021 - 09/2024 Lithium related product share (import and export) by province)

Summary stats (in RMB)

Panel A: Import Panel B: Export										
Commodity	Obs	Mean	Std. Dev.	Min	Max	Obs	Mean	Std. Dev.	Min	Max
Lithium	224	794,409.6	2,612,620	176	24,300,000	391	5,255,659	7,188,171	44	63,000,000
Lithium Iron Phosphate	196	809,847.1	1,296,698	170	9,879,130	520	526,394.5	1,312,929	6	9,026,640
Lithium Carbonate	1,371	81,300,000	446,000,000	15	11,400,000,000	827	7,156,669	28,200,000	6	376,000,000
Lithium Chloride	607	741,900.3	2,271,543	7	20,800,000	382	377,905.9	1,089,251	7	15,400,000
Lithium Enriched in Lithium-6	5	17,459.6	3,954.5	12,615	22,755	-	-	-	-	-
Lithium Hexafluorophosphate	161	2,684,438	4,866,950	20	31,000,000	552	7,095,791	9,427,186	11	59,900,000
Lithium Hydroxide	452	5,911,896	18,500,000	36	176,000,000	847	54,600,000	184,000,000	7	1,980,000,000
Lithium Manganate	101	915,832.7	1,325,719	126	10,100,000	233	543,795.1	1,800,999	70	13,700,000
Lithium Nickel Cobalt Aluminum Oxides	345	53,600,000	121,000,000	758	651,000,000	142	8,275,086	20,800,000	7	107,000,000
Lithium Nickel Cobalt Manganese Oxide	1,094	76,000,000	220,000,000	20	2,230,000,000	1,412	29,500,000	97,600,000	7	1,170,000,000
Lithium Oxide	85	72,807.1	334,249.6	99	2,935,759	59	1,701,989	3,797,940	7	18,100,000
Lithium Primary Cells & Batteries	12,631	696,479.9	1,923,192	2	27,200,000	12,798	610,870.2	1,957,702	0	56,500,000
Lithium-ion Electric Accumulators	18,773	6,624,487	26,000,000	6	624,000,000	40,830	13,900,000	88,100,000	2	3,710,000,000

Summary stats-trade penetration

Table: Lithium Import Penetration Summary

Variable	Obs	Mean	Std. Dev.	Min	Max		
Panel A: Monthly Lithium Trade Penetration in China							
mex_pen	388	1.08×10^{-6}	2.44×10^{-6}	1.15×10^{-10}	0.0000264		
$\mathtt{chi}_\mathtt{pen}$	486	0.0016991	0.0060186	5.26×10^{-10}	0.0935313		
Panel B: Lithium Import Penetration by Country (from Mexico)							
USA	249	0.0001605	0.0000834	0.0000602	0.0003873		
Japan	98	0.0000488	0.0000311	1.15×10^{-6}	0.0001448		
Korea	103	0.0000388	0.0000259	8.41×10^{-6}	0.000105		
Sweden	84	0.0003622	0.0004598	1.70×10^{-7}	0.0024563		
Panel C: Lithium Import Penetration by Country (from Chile)							
USA	134	0.0001637	0.0001244	6.53×10^{-6}	0.0005268		
Japan	106	0.0075517	0.0063659	0.0004869	0.0302904		
Korea	101	0.0546896	0.0516539	0.0112835	0.1972923		
Sweden	3	1.39×10^{-6}	2.15×10^{-6}	9.37×10^{-8}	3.87×10^{-6}		

First Stage Results

Variable	Coefficient	Std. Err.	\mathbf{t}	P> t	$[95\% \ CI]$		
Panel A: Chile (Year $= 2022, 2023, 2024; Obs = 96$)							
Penetration USA	4.322	10.872	0.40	0.692	[-17.271, 25.916]		
Penetration Japan	0.266	0.198	1.34	0.183	[-0.128, 0.660]		
Penetration Korea	-0.008	0.028	-0.30	0.768	[-0.063, 0.047]		
_cons	0.001	0.003	0.23	0.822	[-0.006, 0.008]		
Panel B: Mexico (Year = 2022 , 2023 , 2024 ; Obs = 64)							
Penetration USA	0.010	0.005	2.14	0.036	[0.001, 0.019]		
Penetration Japan	-0.008	0.020	-0.43	0.669	[-0.048, 0.031]		
Penetration Korea	-0.021	0.014	-1.52	0.134	[-0.048, 0.007]		
_cons	-0.000	0.002	-0.09	0.932	[-0.003, 0.003]		

Note: Trade penetration of Sweden is omitted due to insufficient observations. The dependent variable is trade penetration in China.

Main results

Takeaway

References I

- Daron Acemoglu, Mikhail Golosov, Aleh Tsyvinski, and Pierre Yared. A dynamic theory of resource wars. The Quarterly Journal of Economics, 127(1):283–331, 2012.
- Shabbir Ahmed, Paul A Nelson, Kevin G Gallagher, and Dennis W Dees. Energy impact of cathode drying and solvent recovery during lithium-ion battery manufacturing. *Journal of Power Sources*, 322:169–178, 2016.
- British Geological Survey. British geological survey. https://www.bgs.ac.uk. Accessed: 2024-11-01.
- Philip Cooke. Gigafactory logistics in space and time: Tesla's fourth gigafactory and its rivals. Sustainability, 12(5):2044, 2020.
- B. Dunn, H. Kamath, and J. M. Tarascon. Beyond lithium: Alternative batteries for grid and transportation applications. Science, 334 (6058):928–935, 2011. doi: 10.1126/science.1212741.
- IEA EPO. Innovation in batteries and electricity storage—a global analysis based on patent data, 2020.
- Jakob Fleischmann, Mikael Hanicke, Evan Horetsky, Dina Ibrahim, Sören Jautelat, Martin Linder, Patrick Schaufuss, Lukas Torscht, and Alexandre van de Rijt. Battery 2030: Resilient, sustainable, and circular. McKinsey & Company, pages 2–18, 2023.
- T. Gaines, M. Reuter, and S. Andersson. Lithium recycling and sustainability. Market impacts and policy incentives. Waste Management, 102:18–26, 2020. doi: 10.1016/j.wasman.2019.12.028.
- EV Global. Global ev outlook 2021. Paris: IEA, 2021.
- G. Harper, R. Sommerville, and P. Kendrick. Recycling and second life of lithium-ion batteries: The rise of sustainable solutions. *Nature Reviews Materials*, 4(4):234–240, 2019. doi: 10.1038/s41578-019-0128-0.
- Florencia Heredia, Agostina L Martinez, and Valentina Surraco Urtubey. The importance of lithium for achieving a low-carbon future: overview of the lithium extraction in the 'lithium triangle'. Journal of Energy & Natural Resources Law, 38(3):213–236, 2020.
- M. King Hubbert. Nuclear energy and the fossil fuels. Technical Report 95, Shell Development Company, 1956. Publication No. 95, Shell Development Company.
- M. King Hubbert. Techniques of prediction with application to the petroleum industry. Technical report, Shell Development Company, 1959. Shell Development Company.
- Sophia Kalantzakos. The race for critical minerals in an era of geopolitical realignments. The International Spectator, 55(3):1–16, 2020.
- A. Kushnir and B. Sandén. Economics of battery materials: Price trends, material demand, and substitution options. *Journal of Industrial Ecology*, 16(2):194–201, 2012. doi: 10.1111/j.1530-9290.2012.00486.x.
- Natalia Lebedeva, Franco Di Persio, and Lois Boon-Brett. Lithium ion battery value chain and related opportunities for europe. European Commission, Petten, 2016.



References II

Tao Li, Xue-Qiang Zhang, Peng Shi, and Qiang Zhang. Fluorinated solid-electrolyte interphase in high-voltage lithium metal batteries. Joule, 3(11):2647–2661, 2019.

Yangtao Liu, Ruihan Zhang, Jun Wang, and Yan Wang. Current and future lithium-ion battery manufacturing. IScience, 24(4), 2021.

André M\u00e4nberger and Bengt Johansson. The geopolitics of metals and metalloids used for the renewable energy transition. Energy Strategy Reviews, 26:100394, 2019.

William Mitchell Mathew. Peru and the british guano market, 1840-1870. The Economic History Review, 23(1):112-128, 1970.

R. Moores, E. Yang, and D. Belton. Lithium market dynamics and the supply-demand balance: How price is shaped. Resources Policy, 63: 101417, 2019. doi: 10.1016/j.resourpol.2018.101417.

Simon Moores. The global battery arms race: lithium-ion battery gigafactories and their supply chain. In Oxford Energy Forum, number 126, pages 26–30, 2021.

United Nations. UN Comtrade Database. https://comtrade.un.org/, n.d. Accessed: 2024-11-13.

Indra Overland. The geopolitics of renewable energy: Debunking four emerging myths. Energy Research & Social Science, 49:36–40, 2019.

Maria Daniela Sanchez-Lopez. Geopolitics of the li-ion battery value chain and the lithium triangle in south america. Latin American Policy, 14(1):22–45, 2023.

B. Sovacool, M. Ali, and M. Bazilian. The geopolitics of mineral resources for renewable energy technologies. Nature Sustainability, 4: 231-238, 2021. doi: 10.1038/s41893-020-00656-4.

Basudev Swain. Recovery and recycling of lithium: A review. Separation and Purification Technology, 172:388–403, 2017.

Australian Trade, Investment Commission, et al. The lithium-ion battery value chain: new economy opportunities for australia. 2018.

U.S. Geological Survey. Mineral commodity summaries: Lithium, January 2024. URL https://pubs.usgs.gov/periodicals/mcs/. U.S. Department of the Interior. Accessed: November 12, 2024.

María L Vera, Walter R Torres, Claudia I Galli, Alexandre Chagnes, and Victoria Flexer. Environmental impact of direct lithium extraction from brines. Nature Reviews Earth & Environment, 4(3):149–165, 2023.

David L Wood, Jianlin Li, and Seong Jin An. Formation challenges of lithium-ion battery manufacturing. Joule, 3(12):2884–2888, 2019.

Chengjian Xu, Qiang Dai, Linda Gaines, Mingming Hu, Arnold Tukker, and Bernhard Steubing. Future material demand for automotive lithium-based batteries. Communications Materials. 1(1):99, 2020.

Rui Xu, Shuya Lei, Tianyu Wang, Chenxing Yi, Wei Sun, and Yue Yang. Lithium recovery and solvent reuse from electrolyte of spent lithium-ion battery. Waste Management, 167:135–140, 2023.

Pierre Yared. A dynamic theory of war and peace. Journal of Economic Theory, 145(5):1921-1950, 2010.

Alessandra Zanoletti, Eleonora Carena, Chiara Ferrara, and Elza Bontempi. A review of lithium-ion battery recycling: Technologies, sustainability, and open issues. Batteries, 10(1):38, 2024.

