### Chapter 3: Angular Momentum

Yuquan Chen

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# 1 Coupled spin- $\frac{1}{2}$ system

Suppose we have an electron orbiting around an In the electron's frame, the positive charge orbits around  $e^-$  and create a magnetic field

$$\vec{L} \Rightarrow \vec{B} \propto \vec{L}$$
 (1)

similar to  $\vec{\mu} \cdot \vec{B}$  spin under of a magnetic field. Here we have

$$H \propto \vec{S} \cdot \vec{L}$$
, where  $\begin{cases} \vec{S} \text{ is spin of } e \\ \vec{L} \text{ is} \end{cases}$  (2)

generalize to quantum,

$$H = \Omega \vec{S} \cdot \vec{L} = \Omega \left( \hat{S}_x \otimes \hat{L}_x + \hat{S}_y \otimes \hat{L}_y + \hat{S}_z \otimes \hat{L}_z \right)$$
 (3)

### 1.1 Recap on angular momentum

With total spin  $\vec{S}$ , projection along z to be  $S_z$ , we can define eigenstate  $|s, m_s\rangle$ 

$$\begin{cases} \vec{S}^2 | s, m_s \rangle = \hbar^2 s(s+1) | s, m_s \rangle \\ S_z | s, m_s \rangle = \hbar m_s | s, m_s \rangle \end{cases}$$
(4)

with atomic physics, we know

$$|l - s| \le j \le s + l \tag{5}$$

where j is the quantum number of  $\vec{J}$ . The possible choice of j is

$$|l-s|, |l-s|+1, |l-s|+2, ..., s+l-1, s+l$$
 (6)

#### Box 1.1: Examples

1. The ground orbital of *H*-atom,  $l=0, s=\frac{1}{2},$  ignore nuclei for now. Then

$$j = \frac{1}{2}$$
,  $j_z = -\frac{1}{2}$  or  $\frac{1}{2}$ 

2. *H*-atom, first excited orbital  $l=1,\ s=\frac{1}{2}.$  Then

$$j = \frac{1}{2} \text{ or } \frac{3}{2}$$

## 1.2 Termsymbol