Chapter 3

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1 Coupled spin- $\frac{1}{2}$ system

Suppose we have an electron orbiting around an In the electron's frame, the positive charge orbits around e^- and create a magnetic field

$$\vec{L} \Rightarrow \vec{B} \propto \vec{L}$$
 (1)

similar to $\vec{\mu} \cdot \vec{B}$ spin under of a magnetic field. Here we have

$$H \propto \vec{S} \cdot \vec{L}$$
, where $\begin{cases} \vec{S} \text{ is spin of } e \\ \vec{L} \text{ is} \end{cases}$ (2)

generalize to quantum,

$$H = \Omega \vec{S} \cdot \vec{L} = \Omega \left(\hat{S}_x \otimes \hat{L}_x + \hat{S}_y \otimes \hat{L}_y + \hat{S}_z \otimes \hat{L}_z \right)$$
 (3)

1.1 Recap on angular momentum

With total spin \vec{S} , projection along z to be S_z , we can define eigenstate $|s, m_s\rangle$

$$\begin{cases} \vec{S}^2 | s, m_s \rangle = \hbar^2 s(s+1) | s, m_s \rangle \\ S_z | s, m_s \rangle = \hbar m_s | s, m_s \rangle \end{cases}$$
(4)

with atomic physics, we know

$$|l - s| \le j \le s + l \tag{5}$$

where j is the quantum number of \vec{J} . The possible choice of j is

$$|l-s|, |l-s|+1, |l-s|+2, ..., s+l-1, s+l$$
 (6)

Box 1.1: Examples

1. The ground orbital of *H*-atom, $l=0, s=\frac{1}{2},$ ignore nuclei for now. Then

$$j = \frac{1}{2}$$
, $j_z = -\frac{1}{2}$ or $\frac{1}{2}$

2. *H*-atom, first excited orbital $l=1,\ s=\frac{1}{2}.$ Then

$$j = \frac{1}{2} \text{ or } \frac{3}{2}$$

1.2 Termsymbol