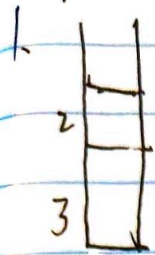


PSX 1/5/2005



$$P = \rho g h_1 + \rho g h_2 = \rho g (h_1 + h_2) = 455 \text{ Pa}$$
~~$$1.82 \rho g (h_1 + h_2) = 5 \times 9.8 \times 1.82 = 89.2 \text{ Pa}$$~~

2. $\Delta P = \rho g h = 1000 \times 9.8 \times 1.2 = 1.18 \times 10^4 \text{ Pa}$

3. $P_0 + \frac{1}{2} \rho v^2 + \rho g (h - y) = P_0 + \rho g (h - y)$

$$v = \sqrt{2g(h-y)}$$

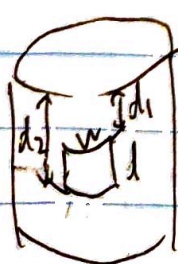
$$y = \frac{1}{2} g t^2 \Rightarrow t = \sqrt{\frac{2y}{g}}$$

$$x = vt = \sqrt{\frac{4g(h-y)y}{g}} = 2\sqrt{(h-y)y}$$

i) $\frac{dx}{dy} = \sqrt{(h-y)y} (h-2y) = 0$
 $y = \frac{h}{2}$

ii) $x = 2\sqrt{(h-y)y} = y$
 $4(h-y)y = y^2$
 $4h^2 - 4hy = 0$
 $y(4y - 4h) = 0$
 $y = \frac{4h}{5}$

iii) $x_{\text{max}} = x(\frac{h}{2}) = h$



i) $v = \sqrt{2gd}$
 flow rate = $\int_0^{d_2} \int_{d_1}^{d_2} \sqrt{2gd} \, dd = W \left[\sqrt{2g} \cdot d^{\frac{3}{2}} \cdot \frac{3}{2} \right]_{d_1}^{d_2}$
 $= \frac{3W\sqrt{2g}}{2} (d_2^{\frac{3}{2}} - d_1^{\frac{3}{2}}) \text{ m}^3/\text{s}$

$$ii) \frac{2w}{3} \sqrt{2g} \left[(d_1 + \delta)^{\frac{3}{2}} - d_1^{\frac{3}{2}} \right]$$

$$= \frac{2w}{3} \sqrt{2g} \left[d_1 \left(1 + \frac{\delta}{d_1} \right)^{\frac{3}{2}} - d_1^{\frac{3}{2}} \right]$$

$$= \frac{2w}{3} \sqrt{2g} \left[d_1 \left(1 + \frac{\delta}{d_1} \right)^{\frac{3}{2}} - d_1^{\frac{3}{2}} \right]$$

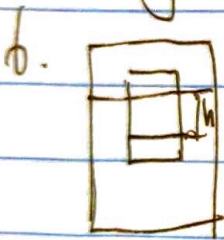
$$= \frac{2w}{3} \sqrt{2g} d_1 \left(\frac{3}{2} \frac{\delta}{d_1} \right)$$

$$= w \delta \sqrt{2g} d_1$$

$$5. V = \frac{m}{\rho}$$

$$T = F_b - mg = \rho_w V g - mg = \frac{\rho_w}{\rho} mg - mg$$

$$\frac{T}{mg} = \frac{\rho_w}{\rho} - 1 = \frac{1}{0.22} - 1 = 3.5$$



$$ma = -\rho V g + mg$$

$$= -\rho A g h + mg$$

$$w = \sqrt{\frac{12}{m}} = \sqrt{\frac{\rho A g}{m}}$$

$$T = 2\pi \sqrt{\frac{m}{\rho A g}}$$

$$7. P_1 + \frac{1}{2} \rho v_1^2 = P_2 + \frac{1}{2} \rho v_2^2$$

$$v_1 A_1 = v_2 A_2$$

$$v_2 = \frac{v_1 A_1}{A_2}$$

$$\rho v_1^2 = (P_2 - P_1) + \rho \left(\frac{v_1 A_1}{A_2} \right)^2$$

$$\rho v_1^2 = 2(P_2 - P_1) + \rho v_1^2 \left(\frac{A_1}{A_2} \right)^2$$

$$\left(\left(\frac{A_1}{A_2} \right)^2 - 1 \right) \rho v_1^2 = 2(P_2 - P_1)$$

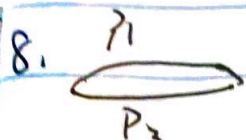
$$v_1 = \sqrt{\frac{2(P_2 - P_1)}{\rho \left(\left(\frac{A_1}{A_2} \right)^2 - 1 \right)}}$$

$$v_1 = \sqrt{\frac{1}{\rho} (2(P_2 - P_1)) + \left(\frac{v_1 A_1}{A_2} \right)^2}$$

$$= \sqrt{\frac{2}{1000} (-2 \times 10^4)}$$

$$= \sqrt{\frac{4 \times 10^4}{10^3 \left(\left(\frac{10^2}{10^2} \right)^2 - 1 \right)}} = \sqrt{\frac{40}{24}} = 8.1 \times 10^{-3}$$

$$v_1 A_1 = 25 \times 10^{-4} \times v_1 = 2.5 \times 10^{-3} \times 8.1 \times 10^{-3}$$



$$P_2 - P_1 = \frac{1}{2} \rho (v_1^2 - v_2^2) + \rho g (h_1 - h_2)$$

$$F = (P_2 - P_1) A = \rho A \left(\frac{1}{2} (v_1^2 - v_2^2) + g (h_1 - h_2) \right)$$

$$= 1.16 \times 50 \left(\frac{1}{2} (10^2 - 70^2) \right) = 9.28 \times 10^4 \text{ N}$$



$$p(h) = \rho g h$$

$$dF = p(h) dA = \rho g h dA$$

$$F = \int_0^L \int_0^h \rho g x dx dy$$

$$= L \int_0^h \rho g x dx = \frac{L \rho g}{2} x^2 \Big|_0^h = \frac{L \rho g h^2}{2}$$

$$= \frac{10 \times 10^3 \times 9.8 \times 2^2}{2} = 1.96 \times 10^5 \text{ N}$$

PS XI 1/5/2025

$$C_w = 4.2 \times 10^3 \text{ J/(kg} \cdot \text{K)}$$

$$C_{ice} = 2060 \text{ J/(kg} \cdot \text{K)}$$

$$L_{vapor} = 1850 \text{ J/(kg} \cdot \text{K)}$$

$$L_{fusion} = 334 \text{ kJ/kg}$$

$$L_{evaporation} = 2260 \text{ kJ/kg}$$

$$1. Q_1 = m C_{ice} \Delta T = 10 \times 2.06 \times 10^3 \text{ J} = 2.06 \times 10^4 \text{ J}$$

$$Q_2 = m L_f = 3.34 \times 10^5 \text{ J}$$

$$Q_3 = m C_w \Delta T = 4.2 \times 10^5 \text{ J}$$

$$Q_4 = m L_v = 2260 \times 10^3 \text{ J} = 2.26 \times 10^6 \text{ J}$$

$$Q = \sum Q = 3.03 \times 10^6 \text{ J}$$

$$2. \quad C_m(T_1) + m_i L_f = C_m(T_2 - T)$$

$$m_i(C_i(T_1) + L_f) = C_m(T_2 - T)$$

$$0.4(206 \times 10^3(7+2) + 334 \times 10^3) = 4200(21-T)$$

$$T(C_m m_i + C_m m_2) = C_m T_2 + C_i m_i T_1 - m_i L_f - 2 C_m$$

$$T = \frac{C_m T_2 + C_i m_i T_1 - m_i L_f - 2 C_m}{C_m(m_i + m_2)}$$

$$= \frac{-2060 \times 0.4 \times 2 + 4200 \times 21 - 0.4 \times 334000}{1.4 \times 4200} = 21$$

$$Q_{ice} = C_{ice} m_{ice} \times 2 = 2.1 \times 0.4 \times 2 = 1.7 \text{ kJ}$$

$$Q_{water} = C_w m_w \times 21 = 4.2 \times 21 = 88.2 \text{ kJ}$$

$$Q_f = Q_w - Q_i = 86.5 \text{ kJ} = m_f L_f$$

$$m_f = \frac{86.5}{L_f} = \frac{86.5}{334} = 0.26 \text{ kg}$$

$$m_{ice} = m_{ice} - m_f = 0.14 \text{ kg}$$

\therefore produce is 0.14 kg ice and 1.26 kg water.

$$3. \quad C_{cu} = 0.385 \text{ kJ/(kg} \cdot \text{K)}$$

$$C_{cu} m_{cu}(T - T_1) + C_w m_w(T - T_1) + C_x m_x(T - T_1) = C_w m_w(T_2 - T)$$

$$(T - T_1) C_x m_x = C_w m_w(T_2 - T_1) + C_{cu} m_{cu}(T_1 - T)$$

$$C_x = \frac{C_w m_w(T_2 - T_1) + C_{cu} m_{cu}(T_1 - T)}{m_x(T - T_1)}$$

$$C_x = \frac{4.2 \times 100 \times 31}{29} + 0.39 \times 30 - 4.2 \times 65}{75} = 2.5 \text{ kJ}$$

$$4. \quad PV = nRT$$

$$n = \frac{PV}{RT} = \frac{100 \times 10^3}{300 \times 8.31} = 2000 \text{ mol}$$

5.

$$P_1 V_1 = n R T_1$$

$$n = \frac{P_1 V_1}{R T_1} = \frac{P_2 V_2}{R T_2}$$

$$P_2 V_2 = n R T_2$$

$$V_2 = \frac{n R T_2}{P_2} = \frac{P_1 V_1 T_2}{P_2 T_1}$$

$$= \frac{(100 + 10^3 \times 9.8 \times 30) \pi (0.02)^2 \times 300}{10^5 \times 280}$$

$$= 1.4 \times 10^{-4} \text{ m}^3$$

6. $V = \frac{n R T}{P}$

$$= \frac{8.31 \times 273}{1.01 \times 10^5} = 0.0225 \text{ m}^3 = 22.5 \text{ L}$$

7. $T = \frac{P V}{n R} = \frac{2.02 \times 10^5 \times 10}{8.31} = 2.43 \times 10^5 \text{ K}$

$$U = \frac{3}{2} k T = \frac{3}{2} \times 1.38 \times 10^{-23} \times 2.43 \times 10^5 = 5.03 \times 10^{-18} \text{ J}$$

$$= \frac{1}{2} m v^2 \quad v = \sqrt{\frac{2U}{m}} = 1.47 \times 10^5 \text{ m/s}$$

8.

$$\frac{dQ}{dt} = -K A \frac{\Delta T}{L}$$

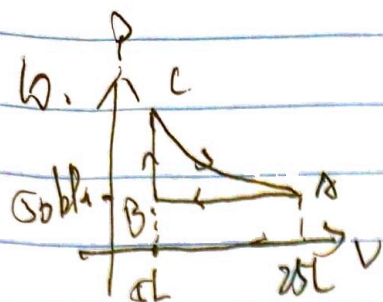
$$= -421 \times (\pi \cdot (0.02)^2) \frac{100}{0.5}$$

$$= -101 \text{ W}$$

9.

$$\frac{dQ}{dt} = -K A \frac{\Delta T}{L}$$

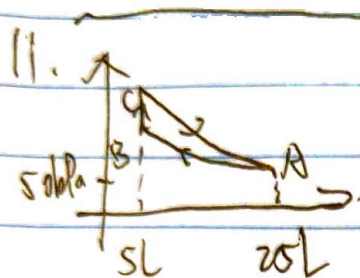
$$= -1 \times 100 \times \frac{17}{0.2} = -8500 \text{ W}$$



$$P_A V_A = P_C V_C$$

$$P_C = \frac{P_A V_A}{V_C} = \frac{5 \times 10^4 \times 25 \times 10^{-3}}{5 \times 10^{-3}} = 2.5 \times 10^5 \text{ Pa}$$

$$\therefore C = (2.5 \times 10^5 \text{ Pa}, 5 \text{ L})$$



C → A adiabatic
A → B isothermal

$$P_A V_A = P_B V_B$$

$$P_B = \frac{P_A V_A}{V_B}$$

$$= \frac{5 \times 10^4 \times 25 \times 10^{-3}}{5 \times 10^{-3}}$$

$$= 2.5 \times 10^5 \text{ Pa}$$

$$T_{AB} = \frac{P_A V_A}{nR} = \frac{5 \times 10^4 \times 25 \times 10^{-3}}{8.31} = 150 \text{ K}$$

$$P_A V_A = P_C V_C$$

$$P_C = \frac{P_A V_A}{V_C} = \frac{5 \times 10^4 \times (25 \times 10^{-3})^{\frac{4}{3}}}{(5 \times 10^{-3})^{\frac{4}{3}}}$$

$$= 5.7 \times 10^4 \text{ Pa} = 427 \text{ kPa}$$

$$T_C = \frac{P_C V_C}{nR} = 257 \text{ K}$$

$$W_{C \rightarrow A} = P_A V_A \ln\left(\frac{V_A}{V_C}\right)$$

$$= 5 \times 10^4 \times 25 \times 10^{-3} \times \ln\left(\frac{25}{5}\right)$$

$$= 2012 \text{ J}$$

$$W_{A \rightarrow B} = P \Delta V = 5 \times 10^4 \times (25 - 5) \times 10^{-3} = -10^3 \text{ J}$$

$$W_{B \rightarrow C} = 0$$

$$W = 1012 \text{ J}$$

$$U = Q - W$$

$$Q = W = 1012 \text{ J absorbed}$$

$$W_{C \rightarrow A} = P_C V_C - P_A V_A$$

$$= 2 \left(\frac{427}{8.31} \times 5 \times 10^{-3} \right) - 5 \times 10^4 \times 25 \times 10^{-3}$$

$$= 2655 \text{ J}$$

$$W_{A \rightarrow B} = P_A V_A \ln\left(\frac{V_B}{V_A}\right)$$

$$= 50 \times 25 \times \ln\left(\frac{5}{25}\right) = -2012$$

$$W_{B \rightarrow C} = 0$$

$$Q = W = 643 \text{ J absorbed}$$