## GOVT. DEGREE COLLEGE HINGORJA

Test Paper - Subject: Combined (undefined)

A) Magnesium nitrate

C) Magnesium nitro

Date:		Time:		
Instructor:	undefined	Duration:	undefined	
Max Marks:	199	Min Marks:	299	
Class:	undefined			
STUDENT		SECT	ION:	
		f bikes sold is increased by $10\%$ . Wha		
C) remains same		B) Decreases D) may increase or may	decrease	
2. The radius of circle is in	ncreased by $20\%$ its area is inc	reased by:		
A) $44\%$		B) $40\%$		
C) $20\%$		D) no change		
3 A 30-ounce solution i	s 25% alcohol. If 60-ounces	of water are added to it what percent	of the new solution is alcohol?	
A) 8%	5 2070 diconoi. Il 00 odiloce	B) 8.33%	of the new solution is deconor.	
C) 8.35%		D) 8.50%		
		,		
that are not black?	orking in a company wear jeans		ack what percent of the workers wear jeans	
A) 30%		B) 60.00%		
C) $60.55\%$		D) $64\%$		
5. Two trains travel in the $15\ seconds.$ The length	,	nd $32~\mathrm{km/h}$ . A man in slower train o	bserves that the faster train passes him in	
A) 75m		B) 100m		
C) 150m		D) $341.66$ m		
6. A train 110 meters lo	ong passes a telegraph pole in $3$	seconds. How long will it take to cr	oss a platform 165 meters long?	
A) 4 seconds	01 011	B) 4.5 seconds	1	
C) 7 seconds		D) 10 seconds	•	
Rawalpindi will the two tr		,	ad $70~\mathrm{km/h}$ . How many kilometers from	
A) 100 km		B) $190~\mathrm{km}$		
C) 210 km		D) $150~\mathrm{km}$		
8. A train traveling at $36\mathrm{l}$ A) $45\mathrm{m}$	m km/h took $10~ m seconds$ to pa	ass a stationary man. What was the lenses $80\mathrm{m}$	gth of the train?	
C) 100m		D) 110m		
9. A train 100 meters lo	ong traveling at $48~ m km/h$ com	pletely crosses a bridge in $30~{ m second}$	ds. How long is the bridge:	
A) 100m	·	B) 150m		
C) 200m		D) 300m		
10. If part of tank filled in	one hour = $\frac{1}{10}$ then time requir	end to fill the tank		
A) 8 hours	one nour = 10 then time requir	B) 12 hours		
C) 10 hours		D) 15 hours		
11. An inlet pipe can fill a both pipes are opened.	tank in $10~\mathrm{hrs}$ and an outlet p	ipe empties the same full tank in $20~\mathrm{h}$	${f rs}$ . The time required to fill the tank when	
A) 10 hrs		B) $20~\mathrm{hrs}$		
12. The number of signific	cant figures in $0.01010$ , $2.00$	0 and $50.00$		
A) $4, 4, 4$		B) $6, 4, 4$		
C) $5, 1, 4$		D) $5, 4, 2$		
13. Good Report answer 2	2 SF			
A) 10		В) 100		
C) 100.		D) NOT		
	shal of tungeters also the	2,1.01		
14. which is chemical sym	iboi of tungsten elements	די (ם		
A) T C) W		B) Tl D) Tu		
•	NO )	<i>D)</i> 10		
15. correct name of $Mg(1)$	$NO_3)_2$			

B) Magnesium nitrite

D) Magnesium nitrile

16. free Radical contain	
A) ve charge	C) even number electron
D) odd number of electron	
17. combining capacity of nitrogen	
A) $+2$	$_{ m B)}-2$

C) + 3	D) $-3$
18. Chemistry is the branch of	

A) physical	B) chemical
C) natural	D) biological

19. correct formula of magnesium phosphate				
A) $MgSO_4$	B) CaSO <sub>4</sub>			
C) ${ m Mg}_3({ m PO}_4)_2$	D) $\mathrm{MgPO}_4$			
20. 2 mole of nitrogen gas contain				
A) 28g	B) $4N_A$ atoms of nitrogen			
C) 14g	D) 42g			
21. meaning of mole	D) newigle			
A) unit C) pile	B) particle D) nano			
22. which is example of mono atomic molecule	•			
A) nitrogen	B) bromine			
C) neon	D) iodine			
23. 4 moles of hydrogen gives moles of water	D.			
A) one C) three	B) two D) four			
24. Concept of relative atomic mas given by				
A) Dalton	B) J. Thomson			
C) Avogadro's	D) NOT			
25. 8kg of water contain grams of hydrogen				
A) 222g	B) 444g			
C) 1000g	D) 888g			
26. Find out mass of $\mathrm{CO}_2$ ? $\mathrm{CaCO}_3$ which is $50\%$ pure decompose $\mathrm{CO}_2$ ?				
A) 11g	B) 22g			
C) 5.5g	D) NOT			
27. plus on reactant side called A) react with	B) and			
C) what	D) AOT			
28. number of atoms of oxygen in $45\mathrm{g}$ of glucose				
A) $7.4 imes10^{23}$	B) $3.01 imes10^{23}$			
C) $9.03 imes10^{22}$	D) $2.8 imes10^{22}$			
29. How many moles of oxygen in 45 liters				
A) 1	B) 2			
C) <b>3</b>	D) 5			
30.20g of oxygen react with $30g$ nitrogen to ammonia which is limiting	g and excess respectively			
A) $\mathrm{O}_2$ and $\mathrm{N}_2$	B) $N_2$ and $O_2$			
C) both consume at same time				
31. if atom loss electron carry charge	D)			
A) +ve C) no charge	B) —ve D) NOT			
32. $40 \mathrm{g SO}_3$ Produce $\mathrm{dm}^3$ of oxygen	<i>b</i> )1101			
A) 5.6 dm <sup>3</sup>	B) $11.2~\mathrm{dm}^3$			
C) $22.4 \text{ dm}^3$	D) $33.6 \text{ dm}^3$			
33. actual yield always than practical yield	2,00.0 411			
A) less	B) greater			
C) equal	D) AOT			
34. $4g$ of nitrogen and $4g$ of hydrogen produce $NH_3$ grams				
A) 8g	B) 5g			
C) 22.6g	D) 11g			
35. find out percentage yield of ${ m CO}_2$ when ${ m 50g~CaCO}_3$ gives only ${ m 6g}$				
A) 27%	B) 50%			
C) 17%	D) $25\%$			
36. $20\mathrm{ml}$ of oxygen react with $20\mathrm{ml}$ hydrogen to form ammonia which				
A) Hydrogen C) water	B) oxygen D) NOT			
S) Huite	2,1.01			