

LAB #: F140529-2605-1 **PATIENT: Matthew Webber**

ID: WEBBER-M-00088

SEX: Male AGE: 23

CLIENT #: 39238 DOCTOR:, MD

Neurological Research Institute, Llc

279 Walkers Mills Rd Bethel, ME 04217 U.S.A.

Toxic Metals; Feces

TOXIC METALS										
		RESULT mg/kg Dry Wt	REFERENCE INTERVAL	PERCENTILE 68 [™] 95 th						
Mercury	(Hg)	0.016	<.05 w/o amalgams*							
Mercury	(Hg)	0.016	<0.5 with amalgams*	•						
Antimony	(Sb)	0.097	< 0.080							
Arsenic	(As)	0.16	< 0.30							
Beryllium	(Be)	0.004	< 0.009							
Bismuth	(Bi)	0.251	< 0.050							
Cadmium	(Cd)	0.32	< 0.50							
Copper	(Cu)	34	< 60							
Lead	(Pb)	0.20	< 0.50							
Nickel	(Ni)	3.1	< 8.0							
Platinum	(Pt)	< dl	< 0.003							
Thallium	(TI)	0.009	< 0.020							
Tungsten	(W)	0.176	< 0.090							
Uranium	(U)	0.498	< 0.120							

WATER CONTENT											
	RESULT	REFERENCE	MEAN								
	% H₂O	INTERVAL	-2SD	-1SD	72.5%	+1SD	+2SD				
% Water Content	81	60 - 85%									

INFORMATION

Analysis of elements in feces provides a comprehensive evaluation of environmental exposure, accumulation and endogenous detoxification of potentially toxic metals. For several toxic elements such as mercury, cadmium, lead, antimony and uranium, biliary excretion of metals into feces is the primary natural route of elimination from the body. Studies performed at DDI demonstrate that the fecal mercury content and number of amalgam surfaces are highly correlated, as is the case for post-DMPS urine mercury levels and amalgam surface area.

Results are reported as mg/kg dry weight of feces to eliminate the influence of variability in water content of fecal specimens. The reference values that appear in this report have been derived from both published data and in-house studies at DDI. *Due to exposure to mercury in the oral cavity, people with dental amalgams typically have a considerably higher level of mercury in the feces than individuals without dental amalgams; therefore, two reference ranges have been established for mercury.

To provide guidance in interpretation of results, patient values are plotted graphically with respect to percentile distribution of the population base. Since this test reflects both biliary excretion and exposure (metals to which the patient is exposed may not be absorbed), it may not correlate with overt clinical effects. Further testing can assist in determining whether the metals are from endogenous (biliary excretion) or exogenous (oral exposure) sources.

- Bjorkman, L, Sandborgh-Englund, G, and Ekstand, J,. Mercury in Saliva and Feces after Removal of Amalgam Fillings. Toxicology & Applied Pharmacology 144: 156-162 (1997)
- Zalups, R, Progressive Losses of Renal Mass and the Renal and Hepatic Disposition of Administered Inorganic Mercury. Toxicology & Applied Pharmacology 130: 121-131 (1995)
 Adamsson, E., Piscator, M., and Nogawa, K., Pulmonary and Gastrointestinal Exposure to Cadmium Oxide Dust in a Battery Factory. Environmental Health Perspectives, 28: 219-222 (1979) 3.
- Smith, J., et al., The Kinetics of Intravenously Administered Methyl Mercury in Man. Toxicology & Applied Pharmacology 128:251-256 (1994) Bass, D., et al., "Measurement of Mercury in Feces", Poster presentation 1999 AACC

SPECIMEN DATA

Comments:

Date Collected: 05/25/2014 Provocation: Dental Amalgams: No

Detoxification Agent: Quantity: Date Received: 05/29/2014

Date Completed: 06/05/2014 Methodology: Dosage: ICP-MS V08.10 Lab number: F140529-2605-1 Fecal Toxics
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URANIUM HIGH

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FecalU

The levels of Uranium (U) in feces have been used to arrive at the total daily intake of U while urinary U represents a smaller fraction of the total daily elimination of U. Most U passes through the intestine unabsorbed. Some excretion occurs from blood into the intestine, via bile (endogenous excretion).

U is a nonessential element that is very abundant in rock, particularly granite. U is present at widely varying levels in ground (drinking) water, root vegetables, and present in high phosphate fertilizers. Other sources of U include: ceramics, some colored glass, many household products (uranyl acetate) and tailings from U mines.

Uranyl cations bind tenaciously to protein, nucleotides, and bone, where it substitutes for Ca. Published data are sparse, but there appears to be a correlation between U exposure, nephrotoxicity and all forms of cancer. Kidney and bone are the primary sites of U accumulation.

All isotopes of U are radioactive; U-238 is the most abundant and lowest energy emitter. It is important to note that the measured result, which is U-238, does NOT indicate or imply exposure to highly enriched U-235, which is used in nuclear power and weaponry.

Hair elements analysis can be performed to assess chronic exposure to U. Uranium is infrequently found to be elevated in unprovoked urine and only reflects ongoing exposure. To date, there are no available chelating agents that can be definitively used to assess U accumulation in the body.

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