

Exercise IV: Free energies

Background: In the lecture (Sec. 2.4), we discussed the free energy landscape $\Delta F(r) = -k_B T \ln g(r)$ with the radial distribution function (RDF) $g(r)$. The RDF is an important observable to evaluate the internal ordering of particles within gaseous, liquid, solid, or amorphous phases, respectively. It can be calculated from MD simulations, and experimentally measured by e.g. X-ray and neutron scattering.

-Task I: simulation

- Create a 2D box with the dimensions 10 nm x 10 nm with periodic boundary conditions.
- Create 100 spherical particles with radius $r = 0.2$ nm (either at random positions or on a regular grid) and mass $m = 2$ g / mol.
- Assign a velocity \mathbf{v} with $|\mathbf{v}| = 0.5$ nm/ns and random direction to each particle.
- For collisions, use the “hard spheres” collision method from the first exercise.
- Carry out a simulation of 5000 steps with a time step length $\Delta t = 0.1$ ns per step.

-Task II: radial distribution function

The RDF displays the average number of particles dn that can be found for each particle at a distance r within the sphere shell (in 3D) or the ring area (in 2D) between r and $r+dr$. Use a dr of ca. 0.01 nm.

- What is the maximal distance r for which you can evaluate the RDF?
- Calculate and display the time- and particle-averaged 2D RDF $g(r) = \frac{1}{2\pi r \rho} \frac{dn(r)}{dr}$ with the distance r and the number density ρ of the simulated box.
- *Hint I:* as for the case with the time step: in data evaluation, infinitesimally small intervals need to be approximated by suitably small finite intervals.
- *Hint II:* remember that you deal here with periodic boundary conditions, i.e., you need to calculate distances over the borders, as well.

-Task III: potential of mean force

- You first need to determine the system temperature from the mean kinetic energy per particle. Remember that we simulate in 2D, so the particles exhibit 2 degrees of freedom.
- What is the temperature in the simulated box?
- Calculate and plot the two-particle $\Delta F(r)$.
- The value of $\Delta F(r)$ may appear odd at first glance. Why is that so?