

Apple Observations: Conservation of Mass

Objective:	In this activity, we will use simple materials to demonstrate the Law of Conservation of Mass . We will put one apple in a closed jar, and one in an open jar. We will then observe how their masses change over time.
	→ Answer This: Define the Law of Conservation of Mass:
	→ Answer This: Generate a focus question. On the lines below, ask a question that relates to the objective of this activity.
• 2 ja	nilar apples rs (mason jars, spaghetti sauce jars, jelly jars, etc.) e beam balance or digital scale
2. Place or3. Place the plastic value	two similar apples. The of the apples in an open jar and leave it open. The other apple in a jar that you can seal very tightly shut. (Use a piece of wax paper or wrap across the opening before you screw on the lid if this will help it seal better.) The mass of each jar with apple today and every other day for several weeks.
	→ Answer This: Which container represents a closed system?
	An open system?
Hypothesis	: Make a hypothesis about what you think will happen to the mass of each jar over time.

Data Tables:

Data – Open Jar				
Date	Mass (g)	Observations Describe what the apple looks like and how it is changing (or not).		

Name			

Data – Closed Jar				
Date	Mass (g)	Observations Describe what the apple looks like and how it is changing (or not).		

	Namealysis: Write a summary of your observations in this activity. Be sure to include how the appearance of the apples changed in each jar, as well as how the mass changed (or did not) in each jar.
2.	Did your results support or refute your original hypothesis? Explain your answer using data from the activity.
3.	Which of the jars best demonstrated the Law of Conservation of Mass? Explain your answer using data from the activity.

4. Which of these jars would be a good analogy to logs burning in a campfire? Explain your answer using data from the activity.

5. Which of these jars would be a good analogy to combining baking soda and vinegar in a flask that is capped by a balloon? Explain your answer using data from the activity.

6. Complete the Claims/Evidence/Reasoning grid below to answer the following question: During a chemical reaction in a closed system, is matter lost?

My Claim	My Evidence	My Reasoning

7. Restate the Law of Conservation of Mass in your own words.