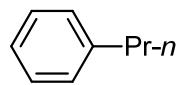


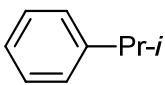
第七章 芳 烃

1. 写出单环芳烃 C₉H₁₂ 的同分异构体的构造式并命名之。

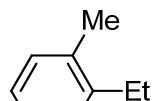
解答：分子式 C₉H₁₂ 的单环芳烃的同分异构体构造式和命名如下：



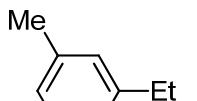
正丙苯



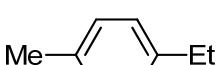
异丙苯



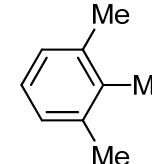
邻乙基甲苯



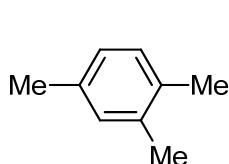
间乙基甲苯



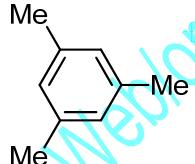
对乙基甲苯



连三甲苯



偏三甲苯



均三甲苯

2. 写出下列化合物的构造式。

(1). 2-硝基-3,5-二溴甲苯

(2). 2, 6-二硝基-3-甲氧基甲苯

(3). 2-硝基对甲苯磺酸

(4). 三苯甲烷

(5). 反二苯基乙烯

(6). 环己基苯

(7). 3-苯基戊烷

(8). 间溴苯乙烯

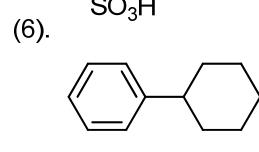
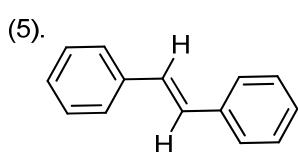
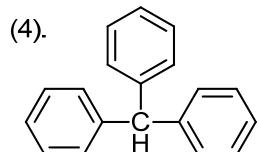
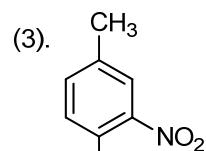
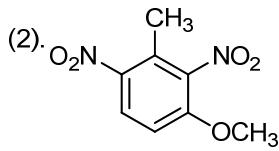
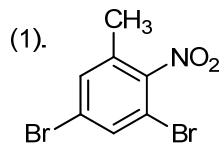
(9). 对溴苯胺

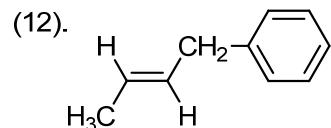
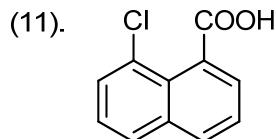
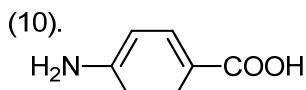
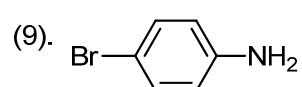
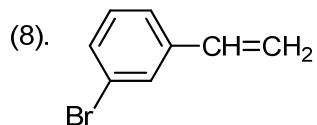
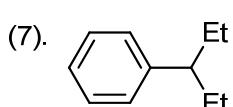
(10). 对氨基苯甲酸

(11). 8-氯-1-萘甲酸

(12) (E)-1-苯基-2-丁烯

解答：





3. 写出下列化合物的结构式。

(1). 2-nitrobenzoic acid

(2). *p*-bromotoluene

(3). *o*-dibromobenzene

(4). *m*-dinitrobenzene

(5). 3,5-dinitrophenol

(6). 3-chloro-1-ethoxybenzene

(7). 2-methyl-3-phenyl-1-butanol

(8). *p*-chlorobenzenesulfonic acid

(9). benzyl bromide

(10). *p*-nitroaniline

(11). *o*-xylene

(12). tert-butylbenzene

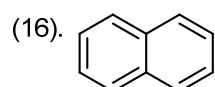
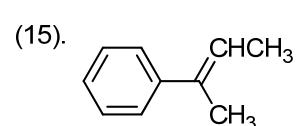
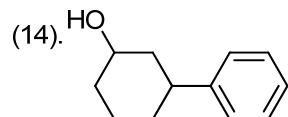
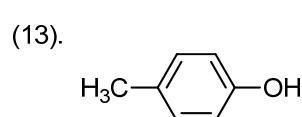
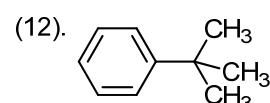
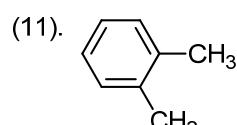
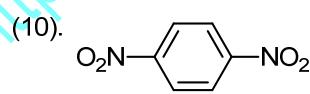
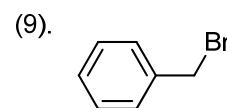
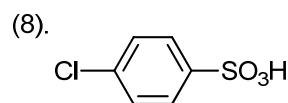
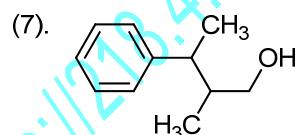
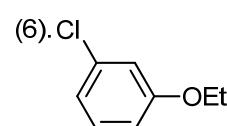
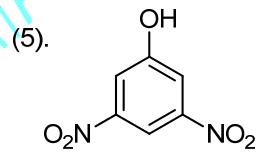
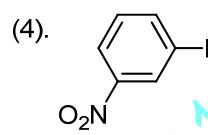
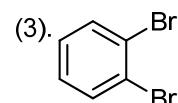
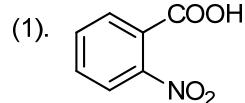
(13). *p*-cresol

(14). 3-phenylcyclohexanol

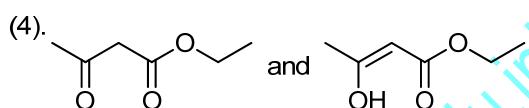
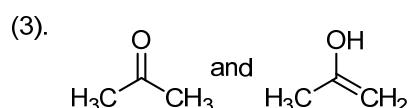
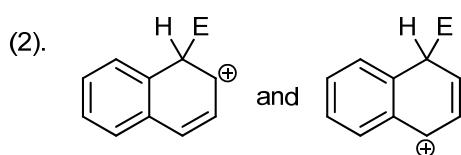
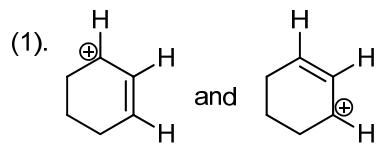
(15). 2-phenyl-2-butene

(16). naphthalene

解答:



4. 在下列各组结构中应使用“ \longleftrightarrow ”或“ \rightleftharpoons ”才能把它们正确地联系起来，为什么？



解答：

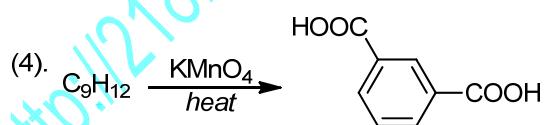
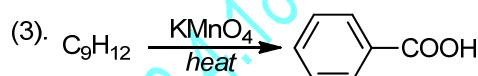
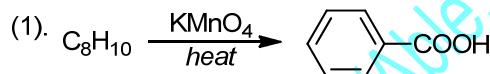
(1). \longleftrightarrow (共振结构);

(2). \longleftrightarrow (共振结构);

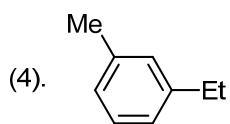
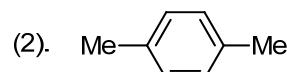
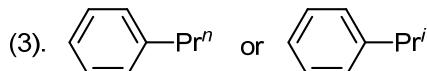
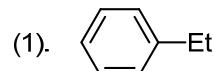
(3). \rightleftharpoons (互变异构);

(4). \rightleftharpoons (互变异构).

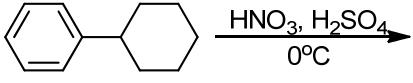
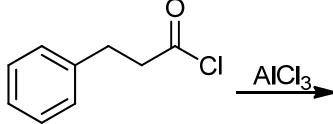
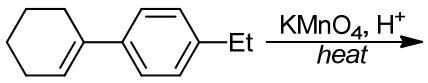
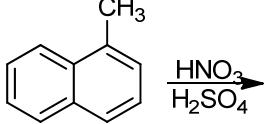
5. 写出下列反应的反应物构造式。



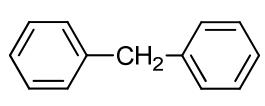
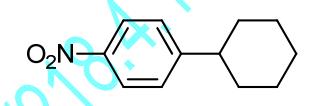
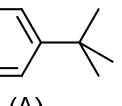
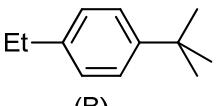
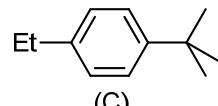
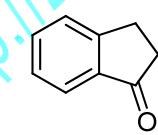
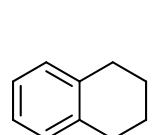
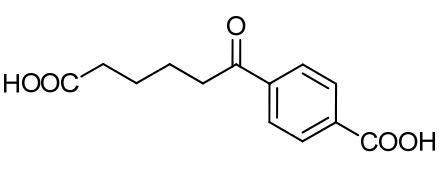
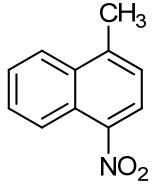
解答：



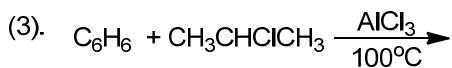
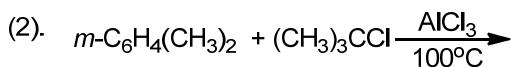
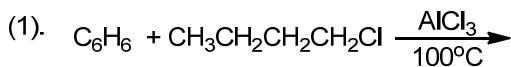
6. 完成下列反应。

- (1).  + Cl-CH2-CH2-CHCl → AlCl3
- (2).  (excessive) + CH2Cl2 → AlCl3
- (3).  HNO3, H2SO4, 0°C
- (4).  $\xrightarrow[\text{HF}]{\text{H}_3\text{C}-\overset{\text{CH}_2}{\underset{\parallel}{\text{C}}}-\text{CH}_3}$ (A) $\xrightarrow[\text{AlCl}_3]{\text{C}_2\text{H}_5\text{Br}}$ (B) $\xrightarrow[\text{H}_2\text{SO}_4]{\text{K}_2\text{CrO}_7}$ (C)
- (5).  AlCl3
- (6).  2H2, Pt → (A) $\xrightarrow[\text{AlCl}_3]{\text{CH}_3\text{COCl}}$ (B)
- (7).  KMnO4, H+, heat
- (8). 

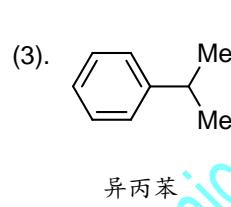
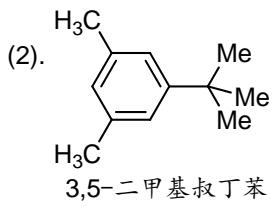
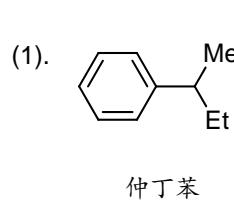
解答：

- (1). 
- (2). 
- (3). 
- (4). (A)  (B)  (C) 
- (5). 
- (6). (A) 
- (7). 
- (8). 

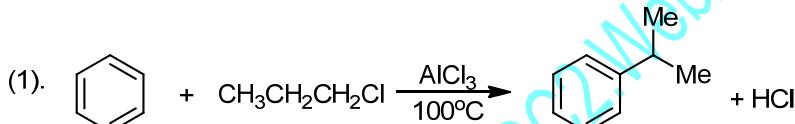
7. 写出下列反应的主要产物的构造式和名称。



解答：



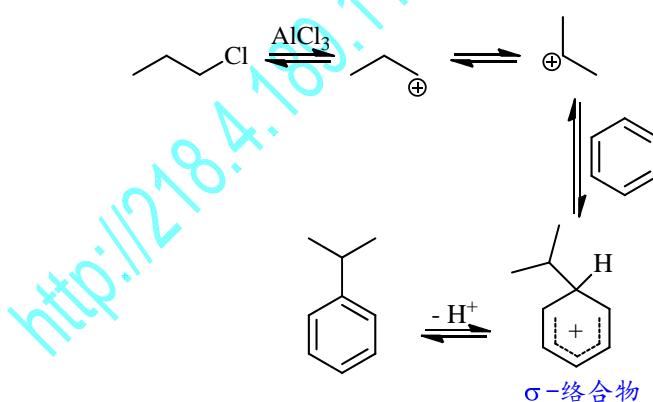
8. 试解释下列傅-克反应的实验事实：



(2). 苯与 RX 在 AlCl₃ 存在下进行单烷基化需要使用过量的苯。

解答：

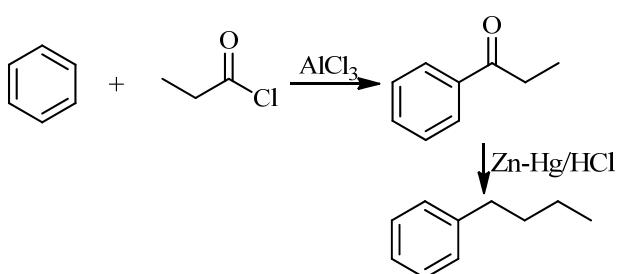
(1).



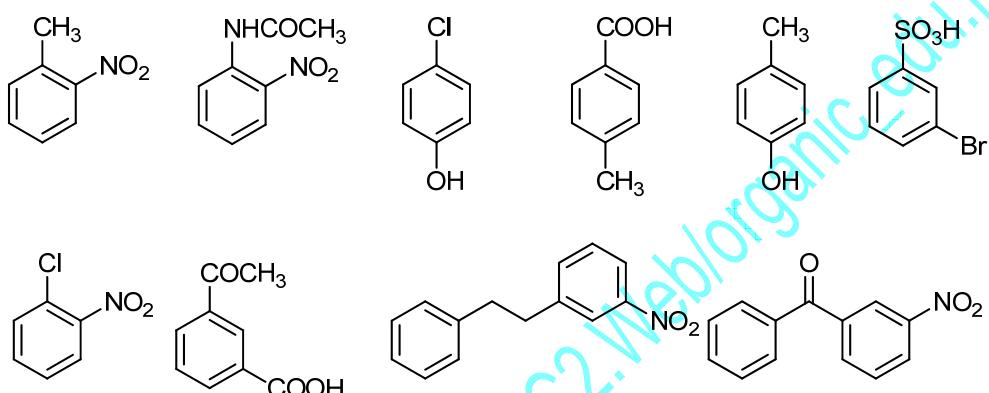
(2). 因为单烷基苯比本更容易进行烷基化反应，因此需要大大过量的苯；同时，次反应中苯还作为反映的溶剂，因而用量较大。

9. 怎样从苯和脂肪族化合物制取丙苯？用反应方程式表示。

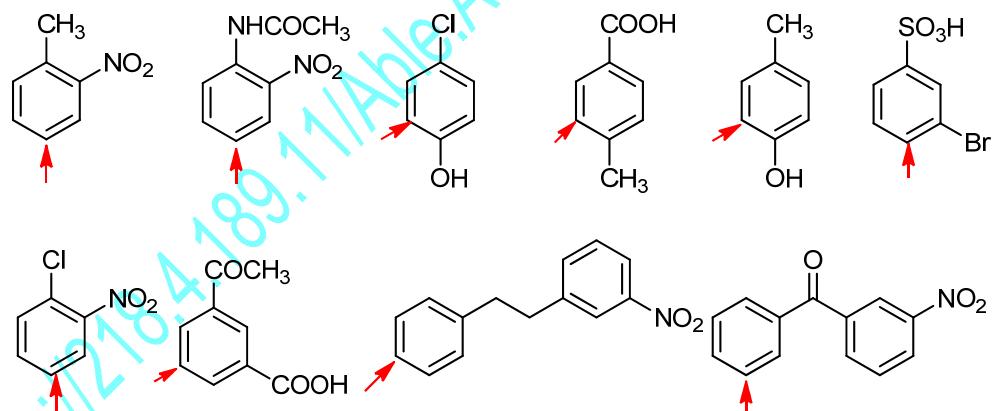
解答：



10. 将下列化合物进行一次硝化，试用箭头表示硝基进入的位置(指主要位置)。



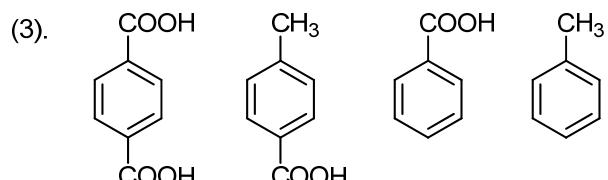
解答：

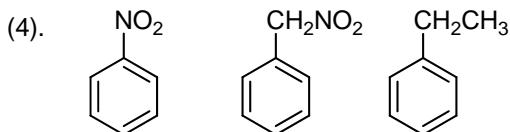


11. 比较下列各组化合物进行硝化反应时的难易。

(1). 苯; 1,2,3-三甲苯; 甲苯; 间二甲苯;

(2). 苯; 硝基苯; 甲苯;

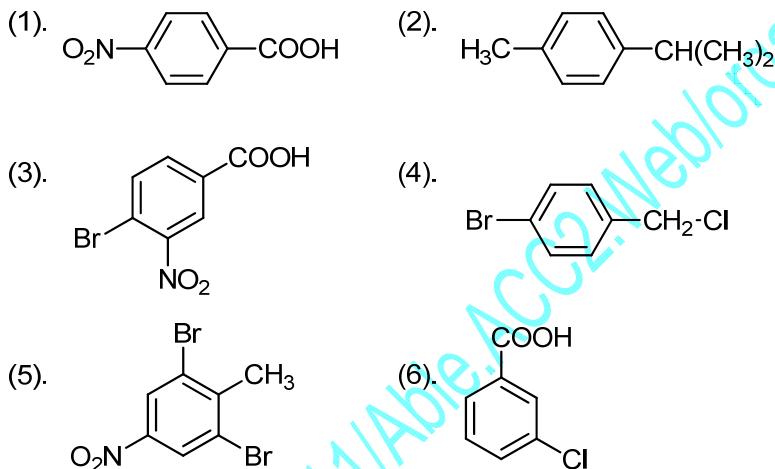




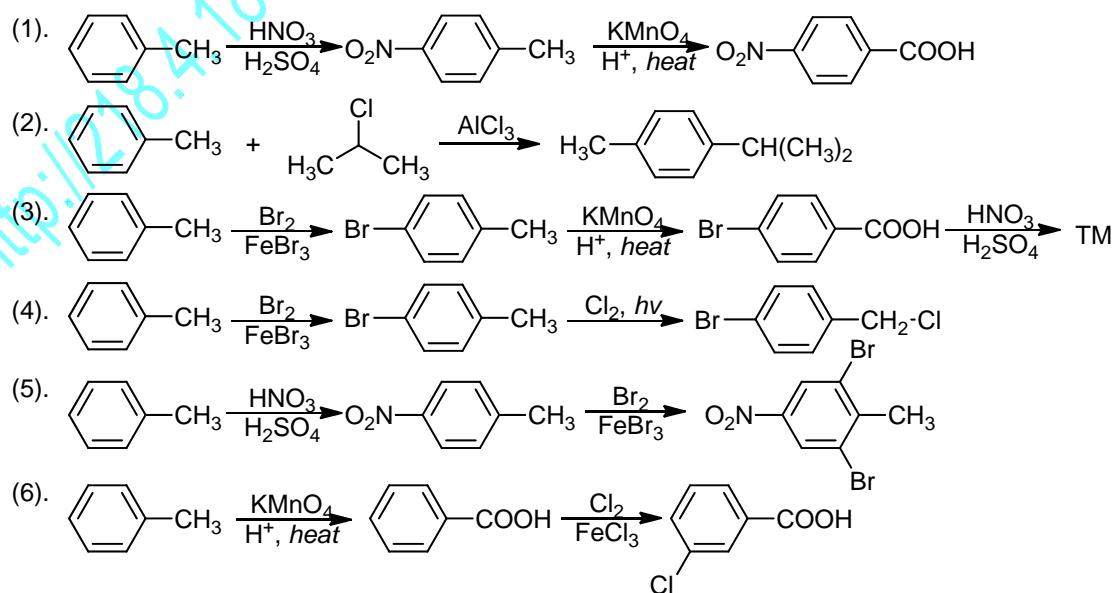
解答:

- (1). 1,2,3-三甲苯>间二甲苯>甲苯>苯
- (2). 甲苯>苯>硝基苯
- (3). 甲苯>对甲苯甲酸>苯甲酸>对苯二甲酸
- (4). 乙基苯> α -硝基甲苯>硝基苯

12. 以甲苯为原料合成下列各化合物。请你提供合理的合成路线。



解答:

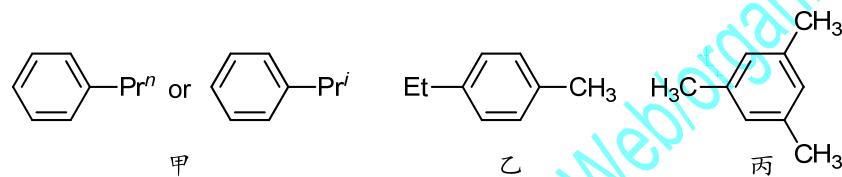


13. 某芳烃其分子式为 C_9H_{12} , 用重铬酸钾的硫酸溶液氧化后得一种二元酸, 将原来的芳烃进行硝化所得的一元硝基化合物主要有两种, 问该芳烃的可能构造式如何? 并写出各步反应式。



14. 甲、乙、丙三种芳烃分子同为 C_9H_{12} , 氧化时甲得一元羧酸, 乙得二元羧酸, 丙得三元羧酸。但经硝化时, 甲和乙分别得到两种一硝基化合物, 而丙只得一种一硝基化合物, 求甲、乙、丙三者的结构。

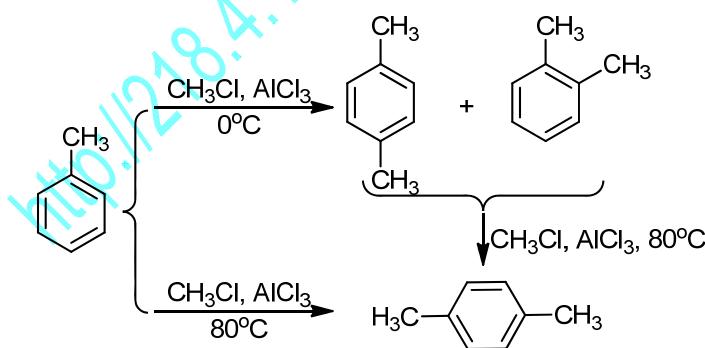
解答: 化合物甲、乙、丙的结构分别为:



15. 比较下列碳正离子的稳定性。 R_3C^+ $ArCH_2^+$ Ar_3C^+ Ar_2CH^+ CH_3^+

解答: 稳定性顺序分别为: $Ar_3C^+ > Ar_2CH^+ > ArCH_2^+ > R_3C^+ > CH_3^+$

16. 下列傅-克反应过程中, 哪一个产物是速率控制产物? 哪一个是平衡控制产物?



解答: 在低温 $0^\circ C$ 下进行的反应为速率控制产物, 在 $80^\circ C$ 下进行的反应为平衡控制产物。

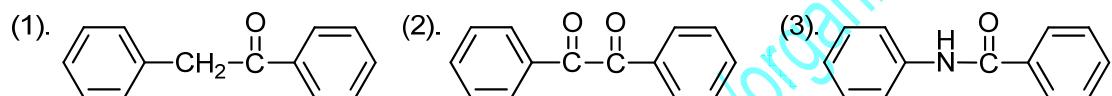
17. 解释下列事实:

- (1). 甲苯硝化可得到 50% 邻位产物，而叔丁基苯硝化只得 16% 的邻位产物。
- (2). 用重铬酸钾酸性溶液做氧化剂，使甲苯氧化成苯甲酸，反应产率差，而将对硝基甲苯氧化成对硝基苯甲醛，反应产率好。

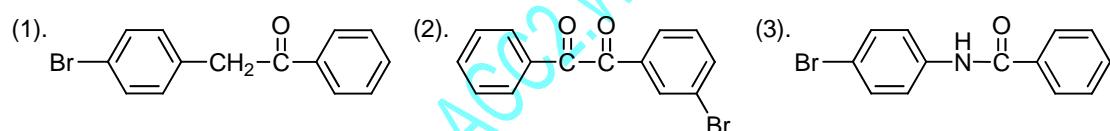
解答：

- (1). 因为叔丁基的位阻大于甲基的，所以邻位产物少；
- (2). 这是由于硝基的吸电子诱导效应将甲基上的电荷吸引到苯环上，从而是甲基碳原子上的电子云密度降低，这进一步降低了 C—H 键之间的电子云密度，使 C—H 键容易被氧化剂进攻而断裂，所以对硝基甲苯的氧化活性高；

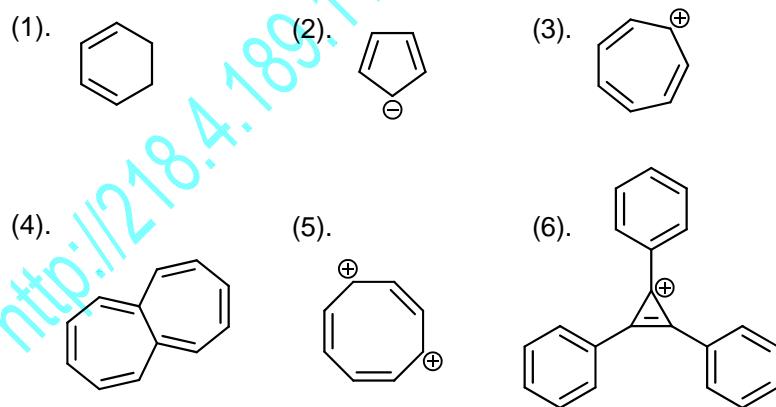
18. 下列各化合物在 Br_2 和 FeBr_3 存在下发生溴代反应，将得到什么产物？



解答：



19. 下列化合物或离子哪些具有芳香性，为什么？



解答：(2), (3), (5), (6) 具有芳香性，因为它们的 p 电子数分别为 6, 6, 6, 2，且都为平面分子，符合休克尔规则。而(1)和(4)的 p 电子数分别为 4 和 12，不符合休克尔规则。