

Heat can travel from one place to another place with 3 ways

- ① Conduction for liquid type objects
- ② Conduction for solid type object
- ③ radiation send matter for the type of the material it could be solid, liquids, ...

Different between energy and power?  
Power is the amount of energy that consumes per time unit

Energy balance for wall:

$$\underbrace{Q_{in} - Q_{out}}_{\text{heating the wall}} = \frac{dE_{wall}}{dt}$$

$Q$ : energy

$Q = \frac{dQ}{dt} \text{ (J/s)} \rightarrow W$  other definition of power

Conductive material how the material will to transfer heating and it indicates with  $k$   
water is high conductive and wood or plastic is low conductive

$$\frac{\Delta T}{\Delta x} \Rightarrow \text{Homo Assumption} \rightarrow \frac{dT}{dx}$$

$$Q_{\text{cond, wall}} = -kA \frac{dT}{dx} \rightarrow -kA \frac{T_i - T_r}{L}$$

$$Q_{\text{cond, wall}} = \frac{T_i - T_r}{R_{\text{wall}}}$$

$$\rightarrow R_{\text{wall}} = \frac{L}{kA} \text{ (C/W)}$$

the thicker wall, the less heat goes through it.

■  $\frac{d(\text{anything})}{dt}$  : How much everything change by time.

■  $\frac{dT}{dx}$  → how much temperature with the change of  $x$

$\frac{dT}{dx}$  → homogenous Assumption →  $\frac{\Delta T}{\Delta x}$

$$\dot{Q}_{\text{cond. wall}} = -kA \frac{dT}{dx}$$

$$\int_{x=0}^L \dot{Q}_{\text{cond. wall}} dx = - \int_{T_1}^{T_2} kA dT$$

$$\dot{Q}_{\text{Cond. wall}} = kA \frac{T_2 - T_1}{L} \quad (\text{W})$$

$L$  → thickness

Simplified conclusion of  
Fourier's law of heat conduction

$$\dot{Q} = KA \times \frac{\Delta T}{L}$$

• heat transfer through a wall is proportional to its Area

• It is proportional to the difference of temperature and the conductivity

• Conductivity, willingness of material to transfer heat. The unit of conductivity is  $\left(\frac{\text{W}}{\text{m} \cdot \text{K}}\right)$

$$\frac{J}{s} = W$$

$$k \text{ (W/m.K)} = \frac{W}{m.K}$$



IT inversely proportional to the thickness!  
 why? the thicker wall, the less heat goes through it!

Temperature unit,

$$K = ^\circ C + 273.15$$

if we have a  $\Delta T = 20^\circ C \rightarrow$

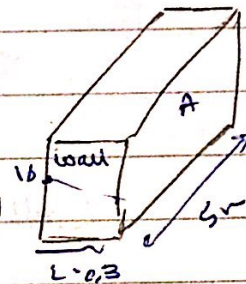
Ex:  $T_1 = 25^\circ C \rightarrow 298 K$   
 $T_2 = 5^\circ C \rightarrow 278 K$

Example :  
 $k = 0.9$

Find the rate

$$\dot{Q} = kA \frac{\Delta T}{L} \Rightarrow$$

$$\dot{Q} = 0.9 \times 15 \times \frac{16 - 2}{0.3} = 630 W$$



$$\dot{Q}_{\text{Cond. wall}} = kA \frac{T_1 - T_2}{L}$$

$$\dot{Q}_{\text{Cond. wall}} = \frac{T_1 - T_2}{R_{\text{wall}}} \quad (W)$$

$$R_{\text{wall}} = \frac{L}{kA} \quad (^\circ C/W)$$

$$T_1 \xrightarrow{R} T_2$$

$$\dot{Q} = \frac{T_1 - T_2}{R}$$

(a) heat flow

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/ / تاريخ:

موضوع:

سؤال

$$L = 0.4 \text{ m}$$

$$A = 20 \text{ m}^2$$

$$\Delta T = 25$$

$$k = 0.78 \frac{\text{W}}{\text{m}}$$

$$Q = k A \frac{\Delta T}{L} = 0.78 \cdot 20 \cdot \frac{25}{0.4}$$

$$= 975 \text{ W}$$

$$R_{\text{wall}} = \frac{L}{Ak} = \frac{0.4}{20 \cdot 0.78} = 0.0256 \frac{^\circ\text{C}}{\text{W}}$$