

Time: 01 Hour

Maximum Marks: 25

Note:

(i) Attempt all questions. Assume any suitable data, if needed.

(ii) Symbols have their usual meanings.

(iii) Marks allotted to each question and course outcome (CO) covered are indicated against each question.

Q. No.

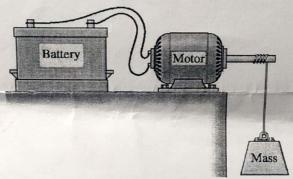
Question

CO M.M.

CO<sub>1</sub>

1(a)

As illustrated in figure below, electric current from a storage battery runs an electric motor. The shaft of the motor is connected to a pulley-mass assembly that raises a mass. Considering the motor as a system, identify locations on the system boundary where the system interacts with its surroundings and describe changes that occur within the system with time. Repeat for an enlarged system that also *includes* the battery and pulley-mass assembly.



1			
1(6)	The work done by a system is a function.	CO2	1
1(c)	The mean effective pressure on an engine piston is to length of the indicator diagram.	CO1	1
1(d)	The thermometric property of a constant-volume gas thermometer is	CO1	1
1/el	In a steady state steady flow process taking place in a device with a single inlet and a single outlet, the work done per unit mass flow rate is given by,	CO2	1

$$\omega = \int_{\text{inlet}}^{\text{outlet}} \forall dp$$

Where  $\forall$  is the specific volume and p is the pressure. The expression for  $\omega$  given above:

(a) Is valid only if the process is both reversible and adiabatic

(b) Is valid only if the process is both reversible and isothermal

(c) Is valid for any reversible process

(d) Is incorrect.

# B. Tech. (Autumn Semester)

## Mid Semester Examination

EEA1110 (Principles of Electrical Engineering) Time: 01 Hour Maximum Marks: 25 Note: Attempt all questions. Assume any suitable data, if needed. (i) (ii) Symbols have their usual meanings. Marks allotted to each question and course outcome (CO) covered are indicated against (iii) each question. Q. No. M.M. Question CO 1. An inductive coil having resistance of 8  $\Omega$  and inductance of 80 mH, is 06 CO1 connected in series with a capacitance of 100  $\mu F$  across 150 V, 50 Hz supply as shown in Fig. 1. Calculate, (a) the current, (b) the power factor, and (c) the 378.30L112.40 voltages drops across the coil. 80 mH 80 100 µF: Fig. 1 State Superposition theorem and find the current through 6 Ω resistance of CO1 06 Fig. 2 using Superposition theorem. (†)3 A Fig. 2 If the 12 Ω resistor of Fig. 3 draws a current of 1 A, find the value of 'R' by 05 3. applying Norton's theorem. Fig. 3 With the help of phasor diagram, derive the relation between line and phase 05 CO<sub>1</sub> 4. currents in delta connected three phase circuit. The initial capacitor voltage of Fig. 4 is zero. Find the steady state voltage 03 CO1 across the capacitor if the switch is closed at t=0 Fig. 4 No. of Questions: 5 No. of Pages: 1

## 2018-19

# B. Tech. (Autumn Semester)

Mid Semester Examination
ACS 1110/AC 111 (Applied Chemistry)

	ACS 1110/AC 111 (Applied Chemistry)		
Time: 01 H Note:	our Max	imum /	Marks: 25
(i) (ii) (iii)	Attempt <u>all</u> questions. Assume any suitable data, if needed.  Symbols have their usual meanings.  Marks allotted to each question and course outcome (CO) covered are indicated against each question.		
Q. No.	Question	со	M. M.
1/at	What are the requirements for a good disinfectant? Write chemical equations for disinfection of water by bleaching powder and chloramine.	CO1	04
A(p)	Calculate the temporary and permanent hardness of a sample of water containing the following impurities: $ \text{CaCl}_2 = 50.5 \text{ mg/L}; \text{ MgSO}_4 = 55 \text{ mg/L}; \text{ KCl} = 2.4 \text{ mg/L}; \text{ SiO}_2 = 9 \text{ mg/L}; \text{ Ca(HCO}_3)_2 = 11.2 \text{ mg/L}; \text{Mg(HCO}_3)_2 = 8.6 \text{ mg/L} \text{ and Na}_2 \text{SO}_4 = 9.2 \text{ mg/L}. $ (Atomic weights: H=1; C= 12; Ca= 40; Mg =24; Na= 23; K= 39; Cl= 35.5; O= 16; Si= 28 and S= 32)	CO1	04
40	Explain the terms: Break-point chlorination and Caustic embrittlement.	CO1	06
2(a)	Define corrosion? Give its significance.  OR	CO2	04
2(a)	What is wet corrosion? Discuss the mechanism of electrochemical corrosion by evolution of hydrogen.	CO2	
2(10)	Mention the requirements of good paint. Discuss the characteristics and functions of pigment.	CO2	04
2(c)	Define Pilling Bedworth rule. Give its significance.	CO2	03
	Mgsou - 120 Si05 60 Ca(HCO3)2 = 162 Mg(HCO3)2 = 146		

# **Mid Semester Examination 2018-19**

## B.Tech. First Year

# Environmental Studies (CEA1110/CE111)

## **Attempt All Questions**

Maximum Marks 25

### Time: One Hour

Q	Section A	Marks		
A.	Describe the consumptive use value as biodiversity as medicines, fuels and fibers with suitable examples.	4		
Z.	What is biodiversity and why must we conserve it?	4		
	OR			
2'	What are the different types of biodiversity? Explain any one.	4		
Section B				
3.	Write a short note on water pollution (sources, effects and control) and also make the list (only name) of various physical, chemical and biological water quality parameters?	4		
of.	Explain briefly various effects and control of soil and thermal pollution?	4		
	OR			
4'	Explain briefly the causes and control of marine and nuclear pollution?	4		
Section C				
5.	Give an account of classification of forest, what measures should be taken for conservation of forest resources?	5		
6.	What is green revolution? What are the remedial measures for solving food problems in India?	4		
	OR			
6'.	What is salinity? Give a brief account of factors responsible for salinity and its effect on agriculture.	4		

#### 2018-2019

#### B. Tech. I Year (Autumn Semester) Mid-Semester Examination AMS-1110 (Applied Mathematics)

Time: 1 hour

Note: (i) Attempt all questions.

Max. Marks: 25

(ii) Symbols have their usual meanings.

(iii) Marks allotted to each question and course outcome (CO) covered are indicated against each question.

Q. No.	Question
A. 1100	Question

CO M.M.

Define the rank of a matrix. For what values of k the system of linear equations 1(a) x + y + z = 1, 2x + y + 4z = k,  $4x + y + 10z = k^2$ 

CO1 [06] 5

have a solution and solve them in each case.

CO<sub>1</sub> [07]

Examine whether the matrix A is diagonalizable. If so obtain the matrix P such that  $P^{-1}AP$  is a diagonal matrix.

 $A = \begin{bmatrix} -2 & 2 & -3 \\ 2 & 1 & -6 \\ -1 & -2 & 0 \end{bmatrix}.$ 

OR

1(b)

Find the characteristic equation of the matrix

CO<sub>1</sub> [07]

$$A = \begin{bmatrix} 2 & 1 & 1 \\ 0 & 1 & 0 \\ 1 & 1 & 2 \end{bmatrix}$$

and hence compute  $A^{-1}$ . Also find the matrix represented by  $A^8 - 5A^7 + 7A^6 - 3A^5 +$ 

 $A^4 - 5A^3 + 8A^2 - 2A + I$  and evaluate it.

If  $y = e^{m \cos^{-1} x}$ , show that

CO<sub>2</sub> [06]

$$(1-x^2)y_{n+2} - (2n+1)xy_{n+1} - (n^2+m^2)y_n = 0.$$

Also find  $(y_n)_0$ .

Expand  $\sin x$  in powers of  $\left(x - \frac{\pi}{2}\right)$  by Taylor's theorem and hence find the value of [06] 2(b)

sin 91° correct to 4 decimal places.

OR

Find the asymptotes of the curve  $x^3 + 3x^2y - 4y^3 - x + y + 3 = 0$ .

CO<sub>2</sub> [06]

2 3 x 2 x 2 2 1 4 kg

No. of Questions: 02 No. of pages: 01