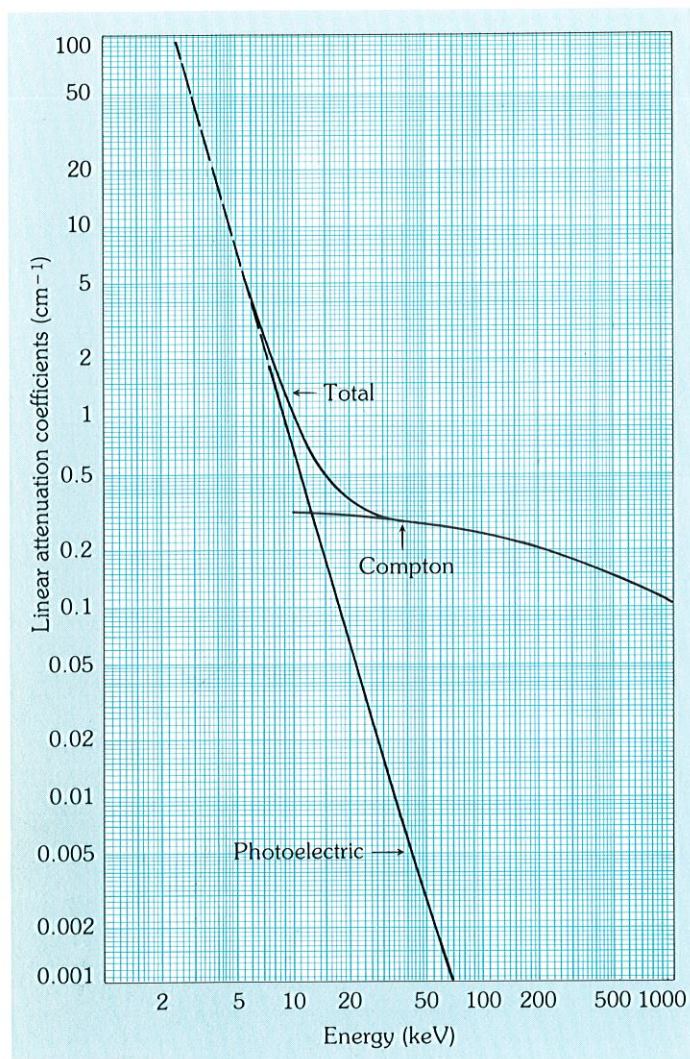
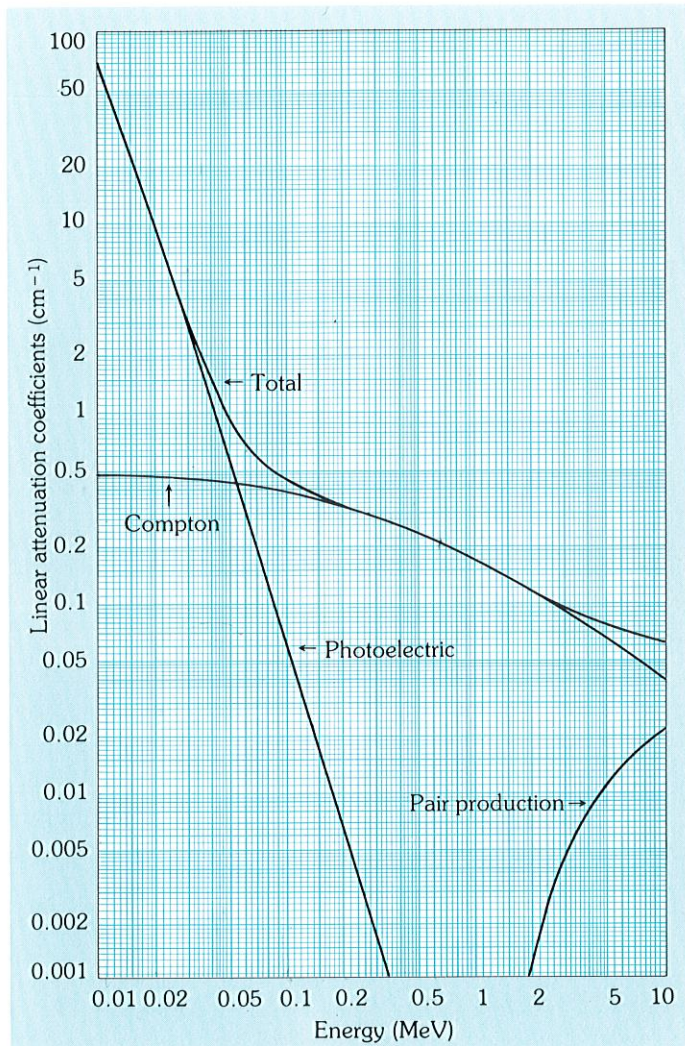


Be



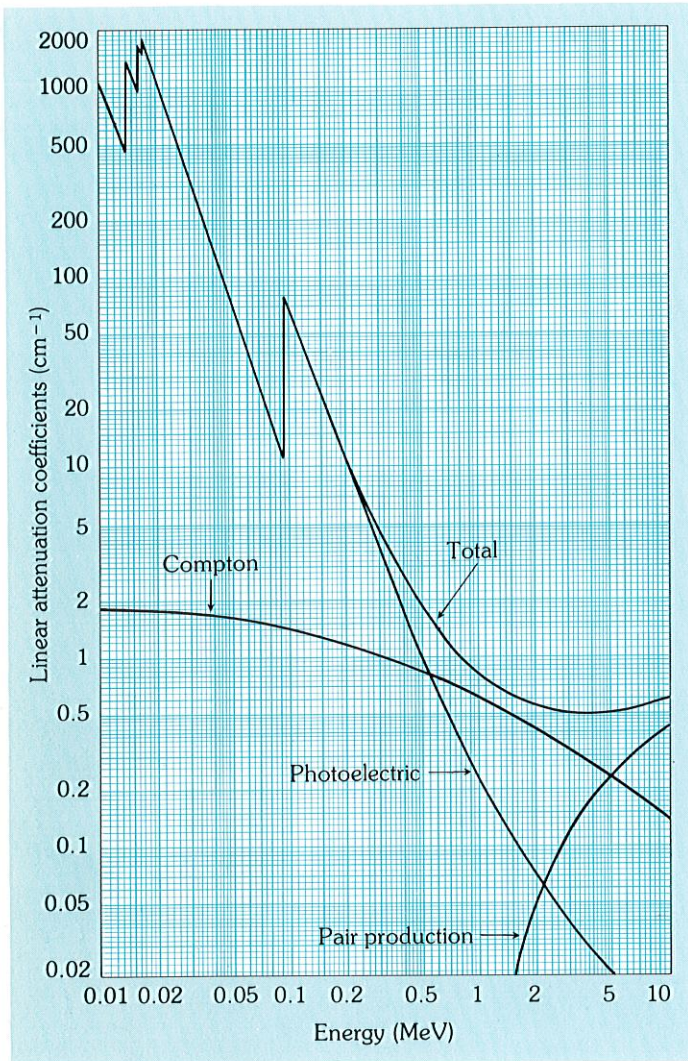
Specific mass = 1850 kg/m^3
 Atomic number: $Z = 4$
 K-Binding Energy = 0.116 keV

Al



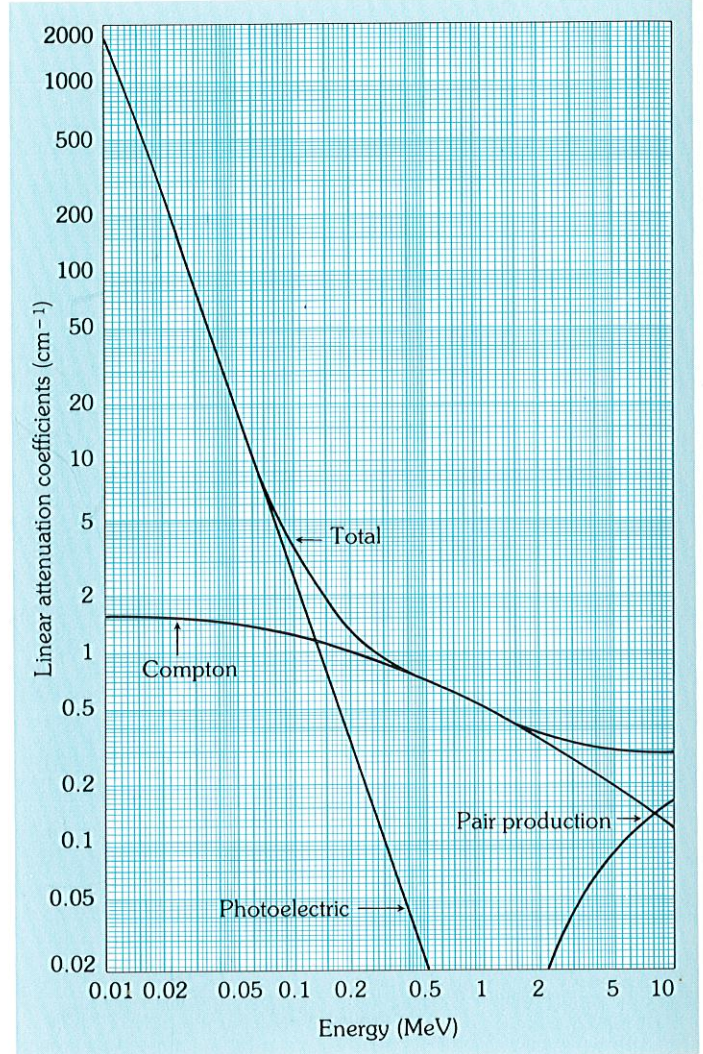
Specific mass = 2700 kg/m^3
 Atomic number: $Z = 13$
 Electron Binding Energies:
 K-edge = 1.56 keV
 Average K X-Ray Energy = 1.45 keV

Pb



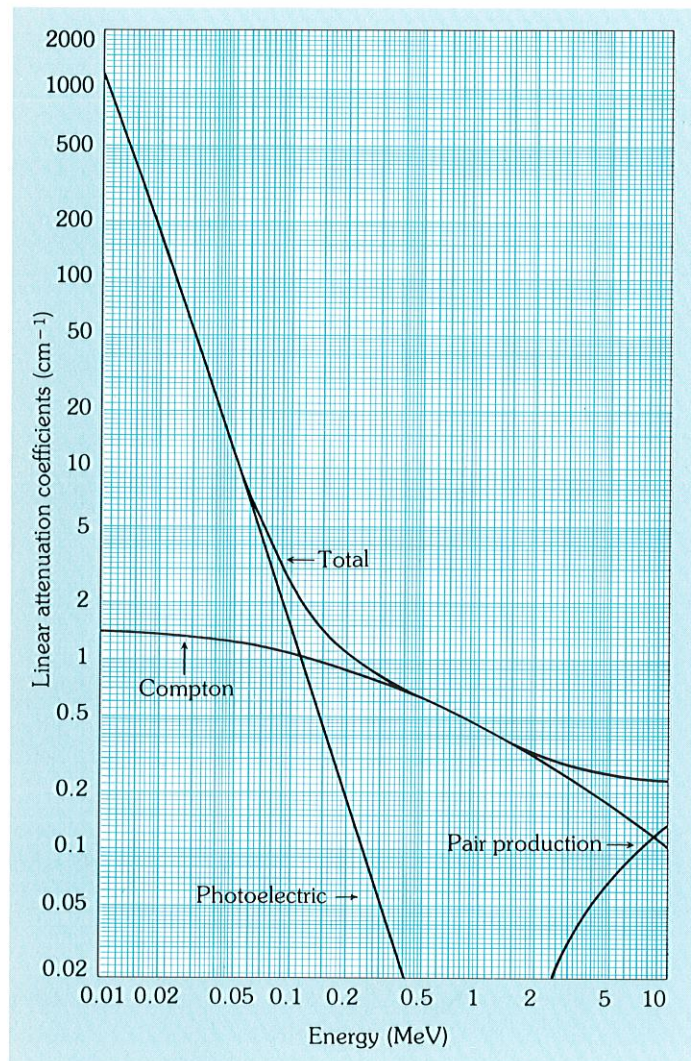
Specific mass = 11350 kg/m^3
 Atomic number: $Z = 82$
 Electron Binding Energies:
 K-edge = 88.02 keV
 L_1 -edge = 15.87 keV
 L_{11} -edge = 15.21 keV
 L_{111} -edge = 13.05 keV
 Average K X-Ray Energy = 76.74 keV

Cu



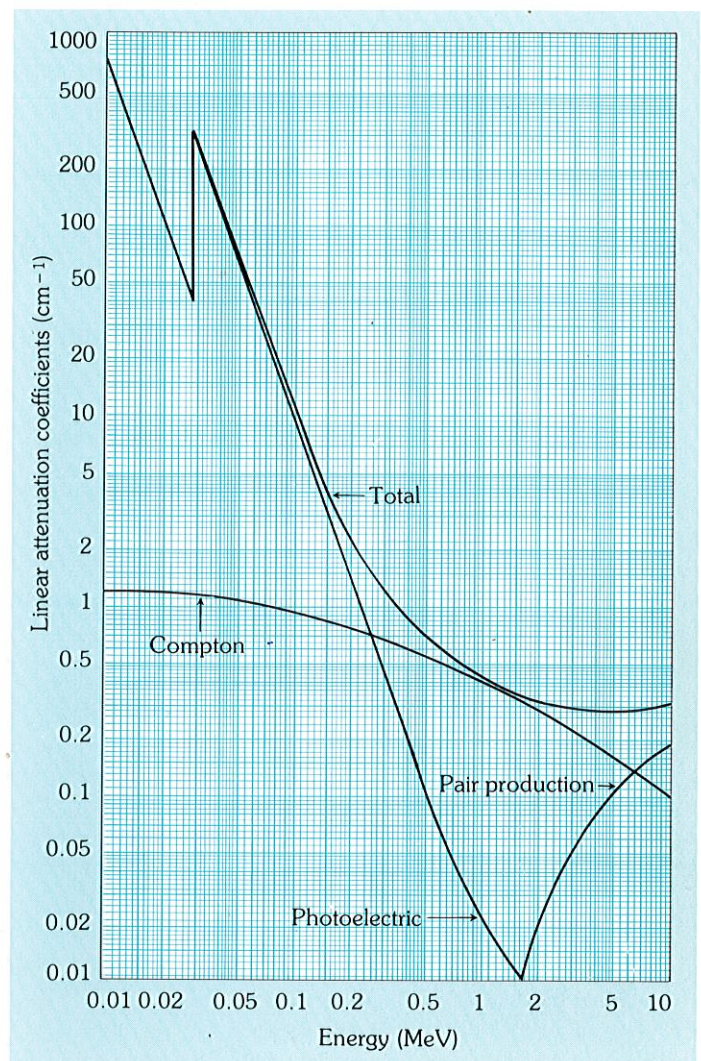
Specific mass = 8920 kg/m^3
 Atomic number: $Z = 29$
 Electron Binding Energies:
 K-edge = 8.98 keV
 L_1 -edge = 1.10 keV
 L_{11} -edge = 0.95 keV
 L_{111} -edge = 0.93 keV
 Average K X-Ray Energy = 8.11 keV

Fe



Specific mass = 7860 kg/cm^3
 Atomic Number: $Z = 26$
 Electron Binding Energies:
 K-edge = 7.11 keV
 L_1 -edge = 0.84 keV
 L_{11} -edge = 0.72 keV
 L_{111} -edge = 0.71 keV
 Average K X-Ray Energy = 6.47 keV

Sn



Specific mass = 7280 kg/m^3
 Atomic number: $Z = 50$
 Electron Binding Energies:
 K-edge = 29.20 keV
 L_1 -edge = 4.47 keV
 L_{11} -edge = 4.16 keV
 L_{111} -edge = 3.93 keV
 Average K X-Ray Energy = 25.8 keV