

Score	

Physical Experiment II

Prelab Report 14

Experiment Title:	The Photoelectric Effect	
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Date:	2018.12.04	

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Answers to Questions (20 points)

- (1) No electrons are emitted if the incident light frequency falls below some cut-off frequency, also called the threshold frequency.
- (2) The work function represents the minimum energy with which an electron is bound in the metal and if the electron is given the energy more than the work function, it could become the free electrons. And it is usually on the order of a few electron volts.

(3)

a.
$$E = h \frac{c}{\lambda}$$

= $6.626 \times 10^{-34} \times \frac{3 \times 10^8}{300 \times 10^{-6}}$
= $6.626 \times 10^{-19} I = 4.14125 eV$

b.
$$E_{kmax} = hv - w = 6.626 \times 10^{-34} \times 3 \times 10 \times 10^{-4} J - 2.46 eV = 1.68125 eV$$

c.
$$0 = E_{kmax} = hv-w$$

$$hv_c = w$$

$$w=h\frac{c}{\lambda_c}$$

$$\lambda = \frac{hc}{w} = \frac{6.626 \times 10^{-34} \times 3 \times 10^{8}}{2.46 \times 1.6 \times 10^{-19}} \approx 0.505 \mu m$$