# Crusher



- Decided to go back to 1D for now.
- Motivation was because of what appears to be a clear separation of the Tcells in the 10mm tumor.
- Added the 2D data to the 1D simulation repository.
- Created a crusher file which project all points onto some line I defined.
- Plots all the points as well as the line to get an idea of what the projection will look like.
- Also plots a pdf of all the projected points.

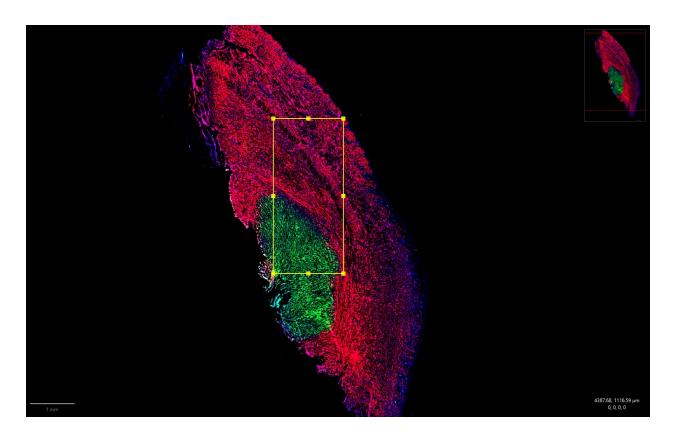


### Info

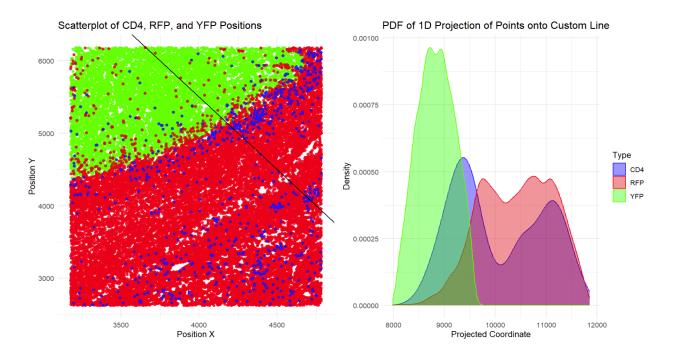
- There's a weird warning, saying that a reached elapsed time limit.
- Takes a weird amount of time to load.
- Jiggling the plot window seems to make it load better, idk why.

## 10mm\_tumor:

Crusher 1



 $\bullet \ \ {\rm Chose} \ y=-2m+13500$ 



• Seems pretty legit for now.

# 1D simulation pt2

■ Date	@March 7, 2025
	<pre>chemoking_gradient_calc.pdf</pre>
i tags	1D

- Going back because modeling the T-cell movement in 2D seems to be a little weird right now, hopefully will go back eventually.
- Will be making a change where T-cells will have a velocity that will be influenced by the gradient.

# The Setup:

- $b_{min}, b_{max}$  : smallest and largest positions observed across the entire dataset.
- Let  $X_i(0)$  denote the position of t-cell i at time step 0:

$$X_i(0) \sim Uniform(b_{min}, b_{max})$$

• Let  $V_i(0)$  denote the velocity of t-cell i at time step 0:

$$V_i(0) \sim Uniform(-1,1)$$

- Let  $X_i(t), V_i(t)$  denote the position and velocity of T-cell i at time step t

$$X_i(t) = X_i(t-1) + V_i(t-1)$$

## The Behaviors:

1D simulation pt2



#### Note

· These are subject to change

#### **Chemokine Gradient**

- Let c(x) denote the concentration of chemokine at position x:
  - $\circ$  For n cancer cells,  $x_i$  denotes the position and  $M_i$  denotes the heat of cancer cell i
  - $\circ \ D$  denotes the diffusion constant of the chemokines.
  - $\circ$   $\delta$  denotes the decay constant of the chemokines

$$c(x) = \sum_{i=1}^n rac{M_i}{2\sqrt{D\delta}} e^{-\sqrt{\delta/D}(|x-x_i|)}$$

- Can see the derivation for it in the pdf `chemokine\_gradient\_calc.pdf`.
- Can pull out some constants for faster computation

$$c(x) = rac{1}{2\sqrt{D\delta}} \sum_{i=1}^n M_i e^{-\sqrt{\delta/D}(|x-x_i|)}$$

- Let abla c(x) denote the concentration gradient at position x
  - Will point in the direction of greater concentration.

$$abla c(x) = -rac{1}{2D} \sum_{i=1}^n M_i e^{-\sqrt{\delta/D}(|x-x_i|)} \cdot sgn(x-x_i)$$

- Additionally, let  $|\nabla c(x)|$  denote the magnitude of said gradient.
- Not sure what to call this, will refer to it as the logistic function
- Will be used to determine how the gradient will influence the velocity.

$$\sigma_lpha(y) = rac{e^{lpha y}}{1+e^{lpha y}}$$

- The larger that  $\alpha$  is, the stronger that the gradient will influence the velocity.
- Will also have a speed term v

#### Chemokine Visualization 10mm Tumor

### **Behavior 0: Logistic Always Update**

- The Cancer Cells will always listen to the gradient.
- So
- If  $\nabla c(X_i(t)) > 0$

$$V_i(t) = \sigma_{lpha}(|
abla c(X_i(t))|) \cdot v$$

• If  $\nabla c(X_i(t)) < 0$ 

$$V_i(t) = -\sigma_lpha(|
abla c(X_i(t))|) \cdot v$$

## Behavior 1: Probability becomes the velocity

- The logistic function will be used as a probability that the T-cells respond to the gradient.
- let  $p \sim uniform(0,1)$
- · The gradient will affect the velocity if

$$p < \sigma_lpha(|
abla c(X_i(t))|)$$

• The velocity update:

$$\circ$$
 If  $abla c(X_i(t)) > 0$ 

$$V_i(t) = \sigma_lpha(|
abla c(X_i(t))|) \cdot v$$

$$\circ$$
 If  $abla c(X_i(t)) < 0$ 

$$V_i(t) = -\sigma_lpha(|
abla c(X_i(t))|) \cdot v$$

• When the gradient isn't affected:

$$V_i(t)=V_i(t-1)$$

### <u>results</u>

# Chemokine Visualization 10mm Tumor

- Green Represents YFP Cancer Cells Distribution
- Red Represents RFP Cancer Cells Distribution
- Black Represents the Concentration/Gradient scale is on the right



#### **Takeaways**

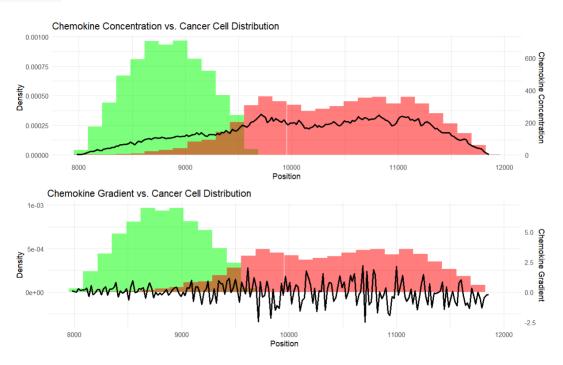
- For the RFP, YFP heat ratio, the smaller the ratio the longer it will take for a T-cell to exit it's region.
  - I think anything above a 1-2 ratio and below like 10 is reasonable.
- For Diffusion/Decay:
  - Higher diffusion/Smaller Decay should be used if we want the T-cells to reach the two noticed clusters in <u>Crusher</u> for the 10mm Tumor
  - Smaller Diffusion/Greater Decay should be used to have the T-cells be more spread out.
  - I think Diffusion 10/ Decay 0.1, is good if we want T-cells to be more spread out
  - I think Diffusion 10/ Decay 0.01, is good if we want T-cells to focus more at those peaks.

# **Modifying YFP RFP Heat Ratios**

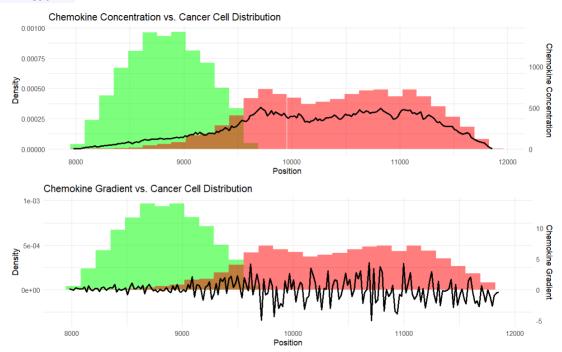
- Diffusion = 10
- Delta = 0.1

# **Modifying YFP RFP Heat Ratios**

- Diffusion = 10
- Delta = 0.1
- YFP Heat = 1
- RFP Heat = 2

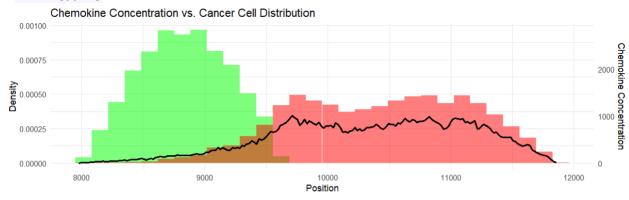


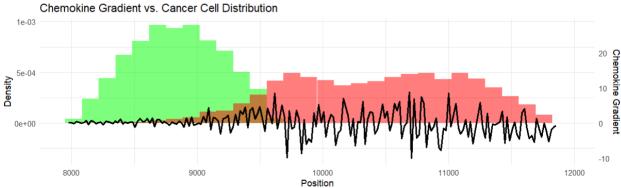
- Diffusion = 10
- Delta = 0.1
- YFP Heat = 1
- RFP Heat = 4



- Diffusion = 10
- Delta = 0.1
- YFP Heat = 1

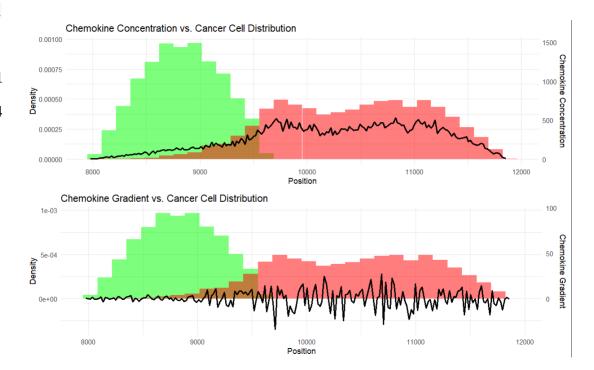
#### • RFP Heat = 8





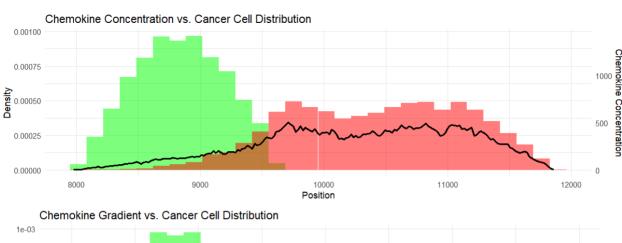
# **Modifying Diffusion**

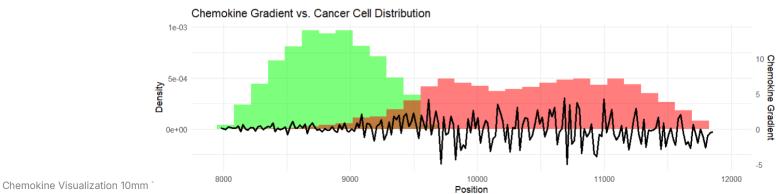
- Diffusion = 1
- Delta = 0.1
- YFP Heat = 1
- RFP Heat = 4



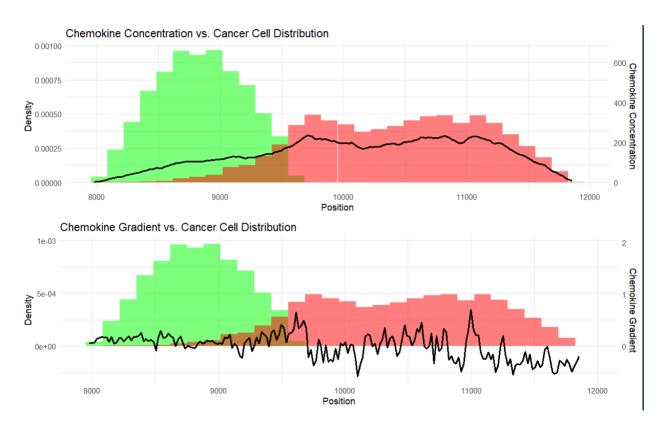


- Delta = 0.1
- YFP Heat = 1
- RFP Heat = 4



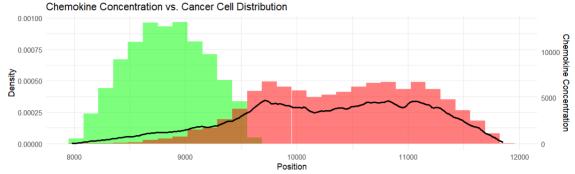


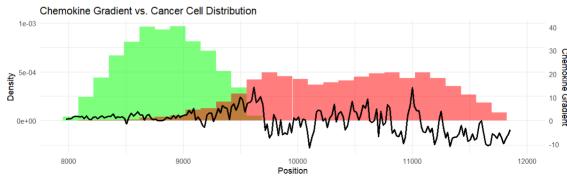
- Diffusion = 100
- Delta = 0.1
- YFP Heat = 1
- RFP Heat = 4



# **Modifying Decay**

- Diffusion = 10
- Delta = 0.01
- YFP Heat = 1
- RFP Heat = 4

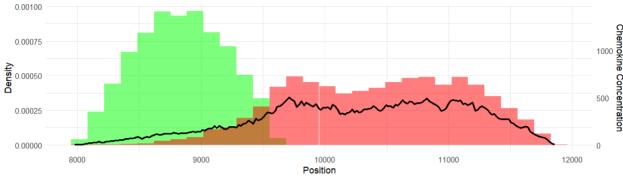


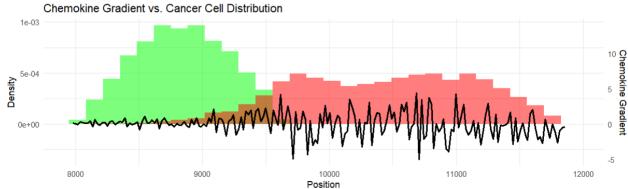


- Diffusion = 10
- Diffusion = 10 • Delta = 1
- Delta = 0.1YFP Heat = 1
- YFP Heat = 1
   RFP Heat = 4

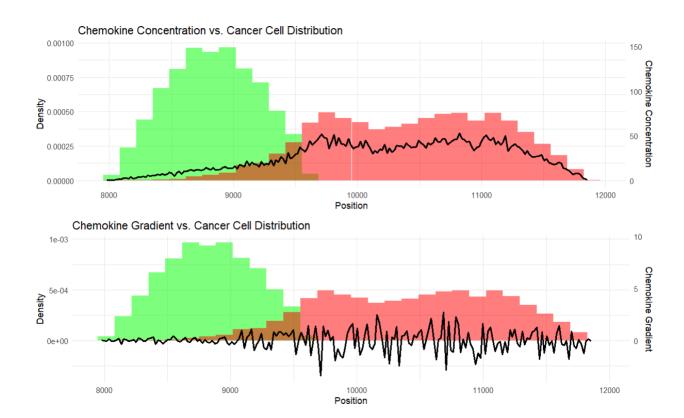
RFP Heat = 4

Chemokine Concentration vs. Cancer Cell Distribution





- Diffusion = 10=
- Delta = 1=
- YFP Heat = 1=
- RFP Heat = 4

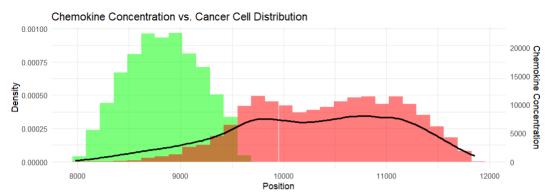


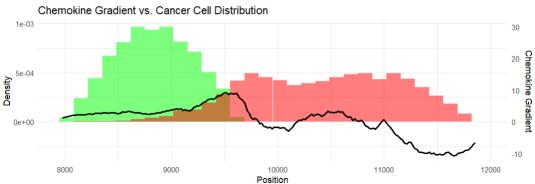
# **Paper Constants**

#### Simulation Constant Searching

#### CCL21

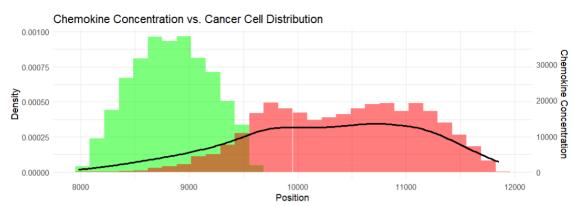
- $d_{1/2}=128.0$
- K = 0.00542
- YFP Heat = 1
- RFP Heat = 4

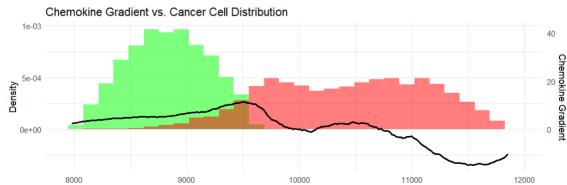




#### CCL19

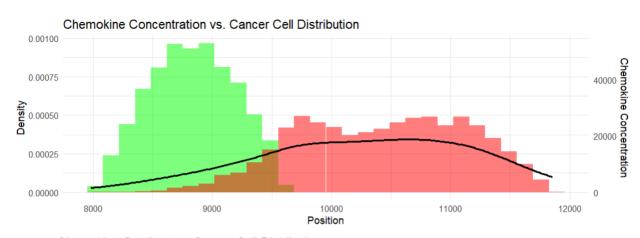
- $d_{1/2}=226.4$
- K = 0.00306
- YFP Heat = 1
- RFP Heat = 4

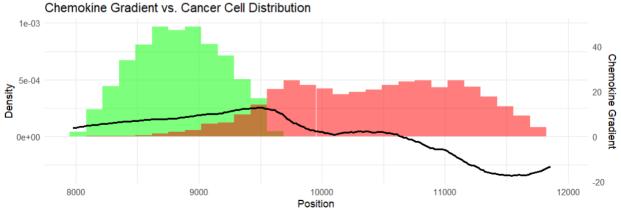




### CCL21trunc

- $d_{1/2}=338.6$
- K = 0.00205
- YFP Heat = 1
- RFP Heat = 4





# results

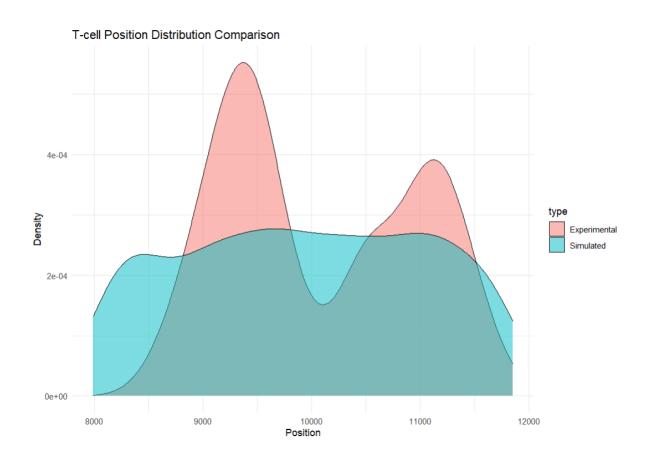
### Info

• Anything greater than alpha = 0 does not appear to have change.

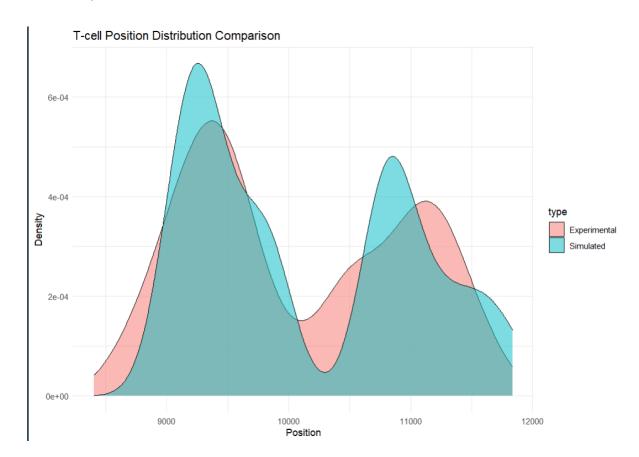
# CCL21

### **Behavior 0**

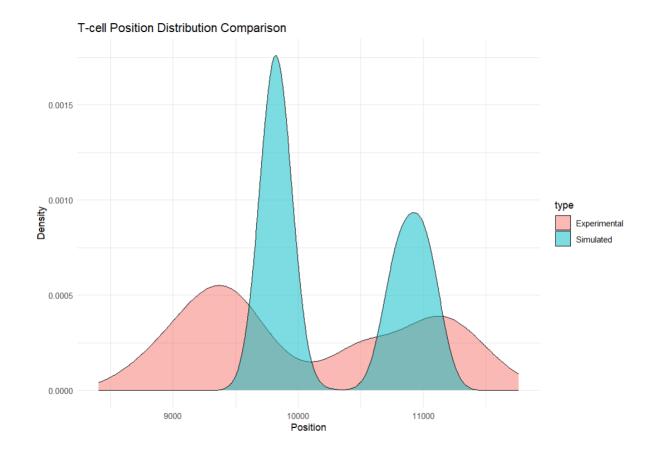
- alpha = -10
- speed = 2
- num-steps = 10000



- alpha = -1
- speed = 2
- num-steps = 10000

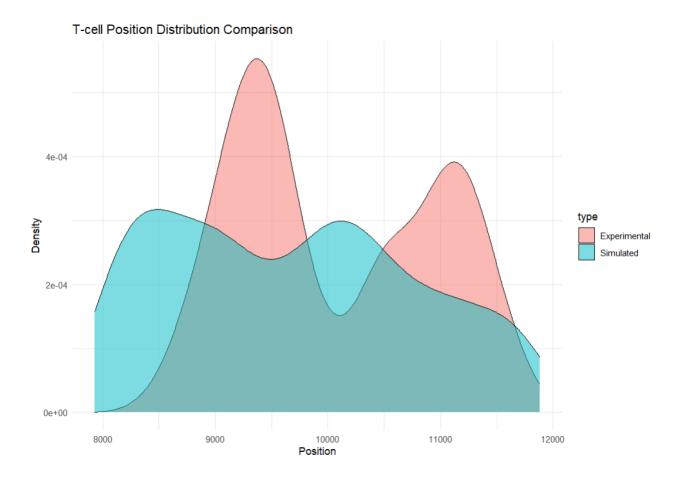


- alpha = 1e-6
- speed = 2
- num-steps = 10000

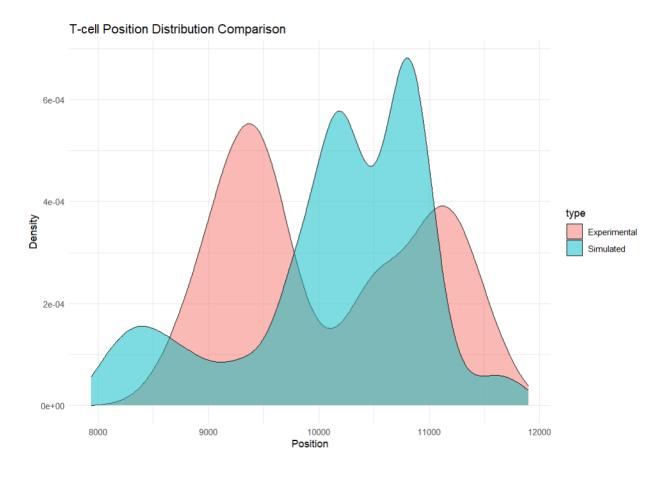


# **Behavior 1**

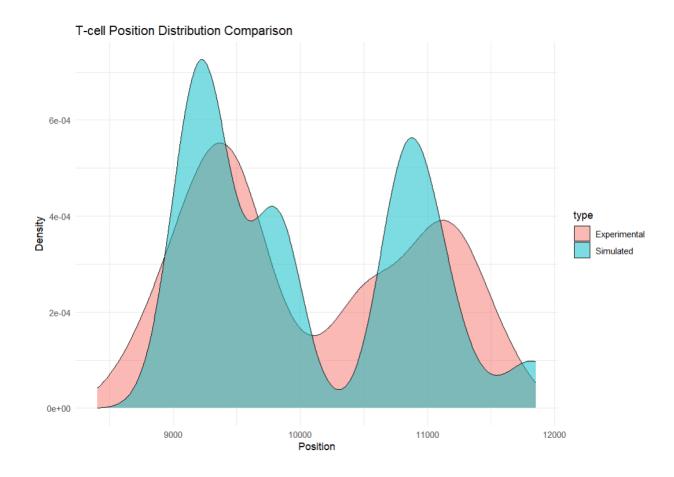
- alpha = -100
- speed = 2
- num-steps = 10000



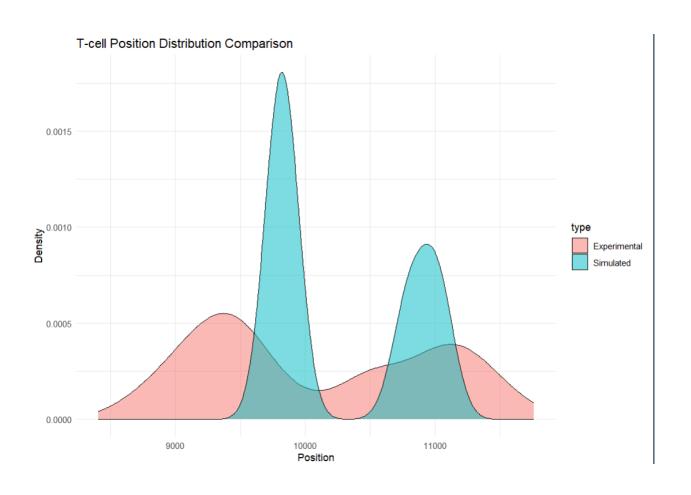
- alpha = -10
- speed = 2
- num-steps = 10000



- alpha = -1
- speed = 2
- num-steps = 10000



• alpha = 1e-6



results 7

### Steady-State Diffusion with Degradation in 2D

#### 1 Introduction

We aim to derive and solve the steady-state diffusion equation for chemokine concentration in a two-dimensional space with degradation. We assume a system where chemokines diffuse outward from cancer cells while simultaneously degrading at a constant rate.

### 2 Deriving the 2D Diffusion Equation

#### 2.1 Fick's First Law

Fick's First Law states that the diffusion flux J is proportional to the negative gradient of the concentration:

$$\mathbf{J} = -D\nabla c,\tag{1}$$

where:

- $\mathbf{J} = (J_x, J_y)$  is the diffusion flux (amount of substance per unit area per unit time),
- D is the diffusion coefficient,
- c(x, y, t) is the chemokine concentration,
- $\nabla c$  is the concentration gradient.

In Cartesian coordinates, this expands to

$$J_x = -D\frac{\partial c}{\partial x}, \quad J_y = -D\frac{\partial c}{\partial y}.$$
 (2)

#### 2.2 Fick's Second Law

Applying the principle of mass conservation, we obtain Fick's Second Law:

$$\frac{\partial c}{\partial t} = -\nabla \cdot \mathbf{J}.\tag{3}$$

Expanding the divergence in two dimensions:

$$\frac{\partial c}{\partial t} = D \left( \frac{\partial^2 c}{\partial x^2} + \frac{\partial^2 c}{\partial y^2} \right). \tag{4}$$

If chemokines degrade at a constant rate  $\delta$ , we introduce a decay term:

$$\frac{\partial c}{\partial t} = D \left( \frac{\partial^2 c}{\partial x^2} + \frac{\partial^2 c}{\partial y^2} \right) - \delta c. \tag{5}$$

#### 2.3 Steady-State Approximation

For a steady-state solution, we set  $\partial c/\partial t = 0$ :

$$D\left(\frac{\partial^2 c}{\partial x^2} + \frac{\partial^2 c}{\partial y^2}\right) = \delta c. \tag{6}$$

Dividing by D:

$$\frac{\partial^2 c}{\partial x^2} + \frac{\partial^2 c}{\partial y^2} = \frac{\delta}{D}c. \tag{7}$$

#### 3 Conversion to Polar Coordinates

Since the problem is radially symmetric, we use polar coordinates  $(r, \theta)$ , where

$$x = r\cos\theta, \quad y = r\sin\theta.$$

The Laplacian in polar coordinates (assuming no  $\theta$  dependence) simplifies to

$$\frac{\partial^2 c}{\partial r^2} + \frac{1}{r} \frac{\partial c}{\partial r} = \frac{\delta}{D} c.$$

Define

$$\lambda^2 = \frac{\delta}{D} \implies \lambda = \sqrt{\frac{\delta}{D}}.$$

The steady-state radial equation then becomes:

$$\frac{d^2c}{dr^2} + \frac{1}{r}\frac{dc}{dr} - \lambda^2 c = 0.$$

### 4 Substitution $\rho = \lambda r$

To transform the above ODE into the standard modified Bessel equation of order 0, we set

$$\rho = \lambda r \implies r = \frac{\rho}{\lambda}.$$

Using the chain rule:

1. First derivative:

$$\frac{d}{dr} = \frac{d\rho}{dr} \frac{d}{d\rho} = \lambda \frac{d}{d\rho}.$$

2. Second derivative:

$$\frac{d^2}{dr^2} = \frac{d}{dr} \left( \lambda \frac{d}{d\rho} \right) = \lambda^2 \, \frac{d^2}{d\rho^2}.$$

Substitute these into

$$\frac{d^2c}{dr^2} + \frac{1}{r}\frac{dc}{dr} - \lambda^2 c = 0.$$

We get:

$$\lambda^2 \frac{d^2 c}{d\rho^2} + \frac{1}{r} \left( \lambda \frac{dc}{d\rho} \right) - \lambda^2 c = 0.$$

Since  $r = \rho/\lambda$ , we have  $\frac{1}{r} = \frac{\lambda}{\rho}$ , so

$$\lambda \frac{1}{r} \frac{dc}{d\rho} = \lambda^2 \frac{1}{\rho} \frac{dc}{d\rho}.$$

Hence,

$$\lambda^2 \left[ \frac{d^2 c}{d\rho^2} + \frac{1}{\rho} \frac{dc}{d\rho} - c \right] = 0.$$

Dividing by  $\lambda^2$  (which is nonzero), we obtain the **modified Bessel equation** of order 0:

$$\frac{d^2c}{d\rho^2} + \frac{1}{\rho} \frac{dc}{d\rho} - c = 0.$$

#### 4.1 Solution via Modified Bessel's Equation

This is recognized as a modified Bessel's equation of order zero. By letting  $\rho = \lambda r$ , we rewrite the ODE in the standard form of the modified Bessel equation:

$$\frac{d^2c}{d\rho^2} + \frac{1}{\rho}\frac{dc}{d\rho} - c = 0. \tag{8}$$

The general solution is a linear combination of  $I_0(\rho)$  and  $K_0(\rho)$ , the modified Bessel functions of the first and second kind of order zero:

$$c(\rho) = A I_0(\rho) + B K_0(\rho). \tag{9}$$

Reverting to r, we have

$$c(r) = A I_0(\lambda r) + B K_0(\lambda r). \tag{10}$$

#### 4.2 Applying Boundary Conditions

**Decay at Infinity.** As  $r \to \infty$ ,  $I_0(\lambda r)$  grows exponentially, while  $K_0(\lambda r)$  decays exponentially . To ensure  $c(r) \to 0$  at large r, we must set A = 0. Hence, the solution simplifies to

$$c(r) = B K_0(\lambda r). \tag{11}$$