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ID# \_\_\_\_\_

**MSE 333**  
**Principles of Materials Characterization and Analysis**  
 Prof. Kannan M. Krishnan

Mid-Term

**60 Minutes + 20 mins (for online submission issues)**

This is a **closed book** exam

If something is not clear, make your best assumption and state it.  
 The time for each problem is proportional to its points. Use it wisely.

Honor Code

I have not provided or taken help from anyone or any others sources  
 (books, course pack, web sites, class or other written notes etc.)

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 Signature

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 Date

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Some formulas you may need --

$$\mathbf{a}^* = 2\pi(\mathbf{b} \times \mathbf{c}) / \mathbf{a} \cdot (\mathbf{b} \times \mathbf{c})$$

$$\mathbf{b}^* = 2\pi(\mathbf{c} \times \mathbf{a}) / \mathbf{b} \cdot (\mathbf{c} \times \mathbf{a})$$

$$\mathbf{c}^* = 2\pi(\mathbf{a} \times \mathbf{b}) / \mathbf{c} \cdot (\mathbf{a} \times \mathbf{b})$$

$$E_{\text{kin}} = hf - |E_{\text{B}}|$$

$$E_{\text{kin}} = hf - |E_{\text{B}}| - \Phi$$

$$E_n = -\frac{m_e Z^2 e^4}{8\epsilon_0^2 h^2} \frac{1}{n^2} = -13.606 \frac{Z^2}{n^2} \text{ (eV)}$$

$$\lambda = \frac{h}{p}$$

$$I_x = I_0 \exp(-\mu x)$$

$$N(E) = 4\pi \left( \frac{2m_e}{h^2} \right)^{3/2} E^{1/2}$$

$$E_{\text{vib}} = \frac{1}{2} A(R - R_0)^2$$

$$E_{\text{vib},n} = (n + 1/2) hf_{\text{vib}}$$

$$\mu_{\text{ind}} = \alpha_0 E_0 \cos(2\pi f_{\text{I}} t) + \frac{\alpha_1 E_0}{2} [\cos 2\pi(f_{\text{I}} - f_{\text{int}})t + \cos 2\pi(f_{\text{I}} + f_{\text{int}})t]$$

$$E = hf = \frac{hc}{\lambda} \quad n\lambda = 2d \sin \theta$$

Name \_\_\_\_\_

ID# \_\_\_\_\_

**For problems 1-7, pick all that is correct** (similar to your Test-your-knowledge exercises in the HWs; Note that there will be **negative grading** for wrong answers.

1. In materials characterization, the probe and signal radiations (2 points)
  - (a) Are the same
  - (b) Are different
  - (c) May be the same or may be different
2. Vibrational modes of any molecule (2 points)
  - (a) are continuous in energy
  - (b) are quantized in energy
  - (c) can be used to probe the spring constant or the strength of the bond
3. When we rotate a crystal for a diffraction experiment (2 points)
  - (i) The reciprocal lattice remains fixed/unchanged in the laboratory frame of reference
  - (ii) The reciprocal lattice attached to the crystal rotates in the laboratory frame of reference
  - (iii) The angle between the incident beam and any reciprocal lattice vector does not vary
4. H and He cannot be detected by Auger Electron Spectroscopy because (2 points)
  - (i) No one has tried to do so?
  - (ii) It is too expensive to make suitable samples
  - (iii) Their electronic structure does not allow such non-radiative decay.
5.  $[1/3 \ 1/2 \ 1]$  and  $[2 \ 3 \ 6]$  refer to (2 points)
  - (i) two different planes
  - (ii) the same line
  - (iii) a family of parallel planes passing through the origin.
6. It is best to use (2 points)
  - (i) PES to probe filled levels in the valance bands
  - (ii) inverse PES to probe unfilled levels in the conduction band
  - (iii) XAS to probe unoccupied levels
7. The **maximum** number of electrons in the  $L$ -shell ( $n=2$ ) is (2 points)
  - (i) 4
  - (ii) 6
  - (iii) 20
  - (iv) 8
  - (v) 14
8. The three broad categories of techniques for materials characterization are (6 points)
  - (a) \_\_\_\_\_
  - (b) \_\_\_\_\_

Name \_\_\_\_\_ ID# \_\_\_\_\_

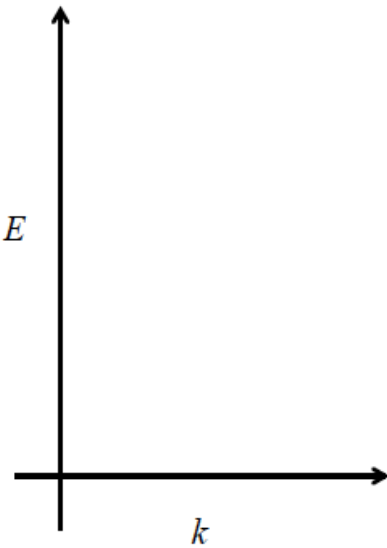
(c) \_\_\_\_\_

9. Plot the **Energy ( $E$ )** vs. **Wave-vector ( $k$ )** for (10 points)

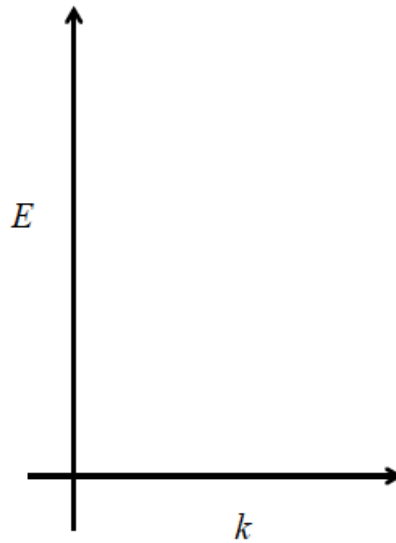
(a) **A free particle** of mass,  $m$ , whose energy is entirely **kinetic**.

(b) An electromagnetic wave **traveling in vacuum**

(a)



(b)



Name \_\_\_\_\_

ID# \_\_\_\_\_

10. For the Na atom, the binding energies of the various electrons are  $-1.071 \text{ keV}$  ( $1s$ ),  $-63.4 \text{ eV}$  ( $2s$ ),  $-30.7 \text{ eV}$  ( $2p$ ) and  $-5.14 \text{ eV}$  ( $3s$ ).

(a) Calculate the energy of the  **$K_\alpha$  X-ray** line for Na. (5 points)

(b) If  $1 \text{ eV} = 1.6 \times 10^{-19} \text{ J}$ ,  $h = 6.62 \times 10^{-34} \text{ Js}$  and  $c = 3 \times 10^8 \text{ m/s}$  calculate the wavelength of this x-ray photon. (9 points)

(c) Can we use Na  $K_\alpha$  for x-ray crystallography of real materials? Explain in a brief sentence or two. (6 points)

Name \_\_\_\_\_

ID# \_\_\_\_\_

11. What is the **key difference** between (10 points)

(a) Infra-red and Raman Spectroscopies

(b) Rayleigh scattering and Raman spectroscopy

12. For **X-ray photoelectron spectroscopy**

The binding energy of a core-level of an **elemental metal**, work function  $\sim 5$  eV, is  $-711$  eV.

- (i) If Al-K ( $\sim 1486$  eV) is used to probe this element at what energy of the photoelectron would the peak for this core-level appear in an XPS experiment, assuming  $E_B = 0$  at the Fermi level? State your assumption clearly
- (ii) Now, where would the peak appear if we assume  $E_B = 0$  at the Vacuum level ?
- (iii) If we use Mg K ( $\sim 1251$  eV) instead, where would the peak appear in (i).

(10 points)

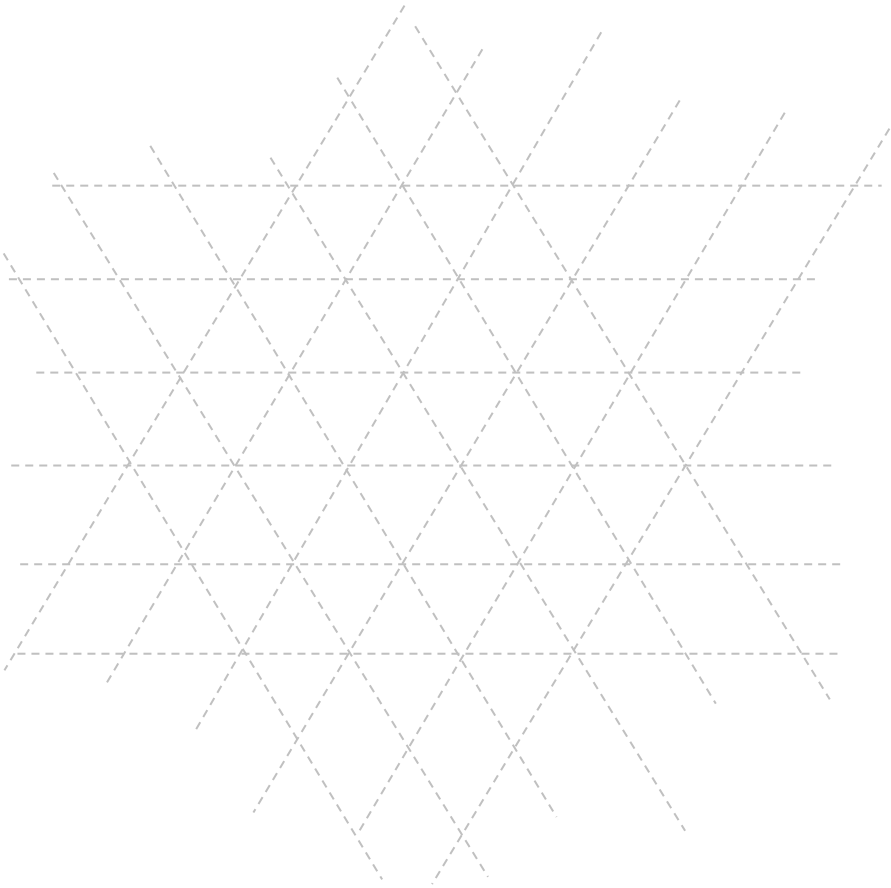
Name \_\_\_\_\_

ID# \_\_\_\_\_

13. For a two-dimensional **hexagonal** unit cell, calculate its reciprocal lattice vectors. (hint: assume a hypothetical *c-axis* normal to the page). Draw the real lattice and the reciprocal lattice (label them clearly) on the same figure below. (10 points)

Name \_\_\_\_\_

ID# \_\_\_\_\_



Name \_\_\_\_\_

ID# \_\_\_\_\_

**14. BONUS Question**

(10 points)

- (a) What is Law of diffraction in **reciprocal space** ? Show it in a figure and define/describe all the variables.
- (b) How is this related to the Bragg Law in **real space** ? Illustrate this also with a figure.