

Glycogen Metabolism

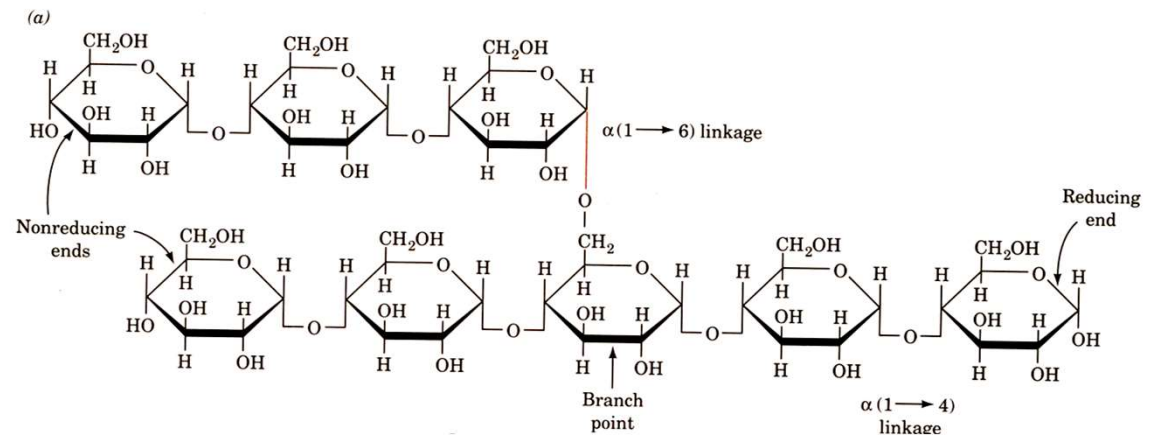
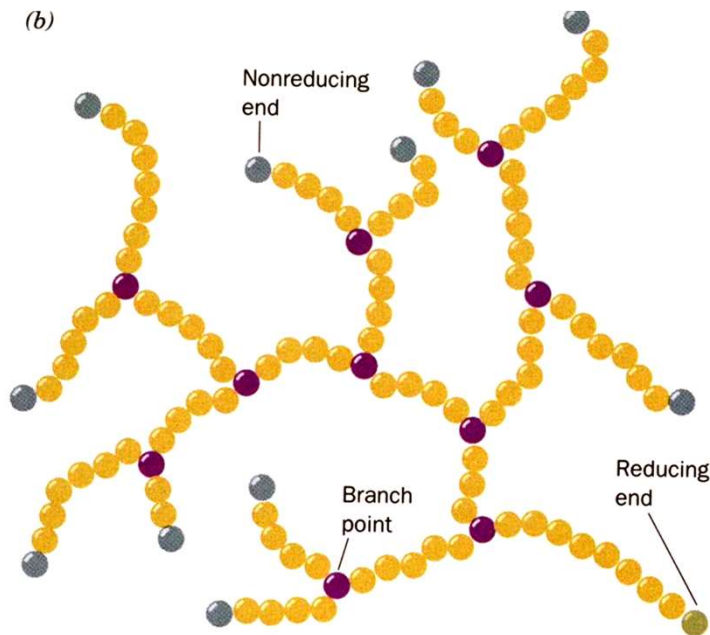
Voet & Voet, Chapter 18

Albert Einstein: “Everything should be made as simple as possible but not simpler”

- Glucose = important fuel (glycolysis, citric acid cycle)
⇒ needs to be stored to be “ready” for metabolic need

better than fat: rapidly mobilized; can replenish blood glucose; anaerobically metabolized

→ single (BIG) molecules!



➤ **branches every 8-12 residues**

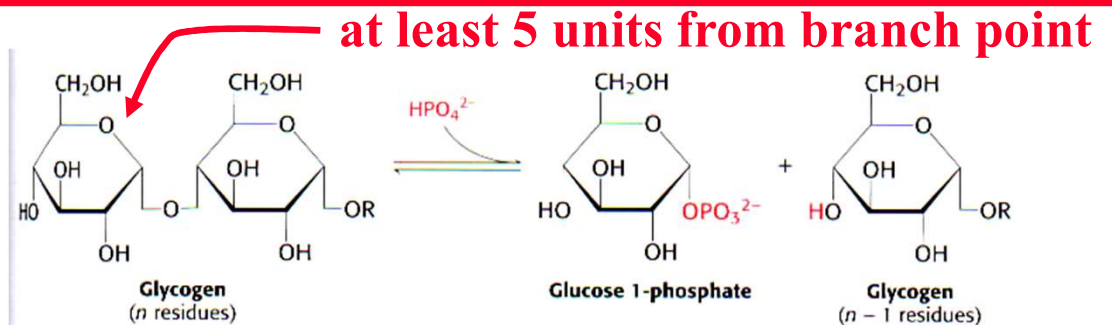
⇒ fast release at every end!

How Does the Cell Utilize Glycogen?

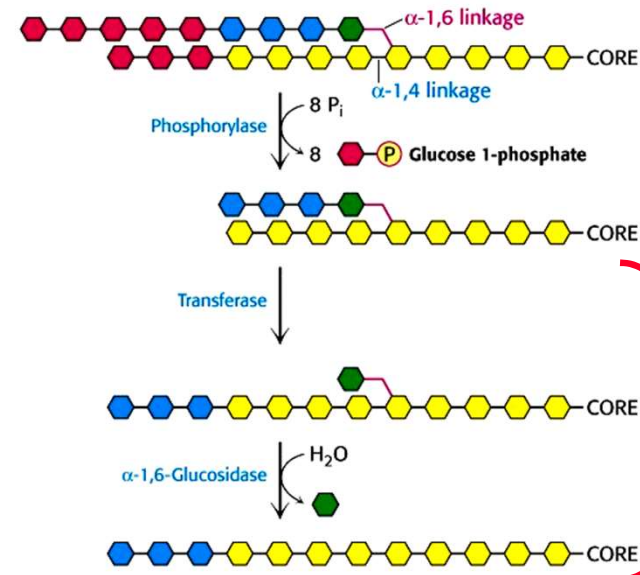
- Glycogen has to be synthesized and degraded
in a tightly regulated fashion

Glycogen Breakdown involves:

1.) (Glycogen) Phosphorylase



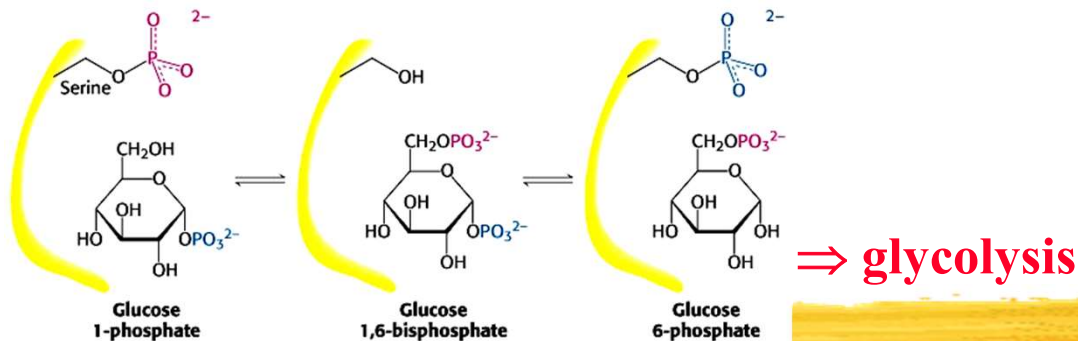
2.) Glycogen debranching enzyme



bifunctional
debranching
enzyme



3.) Phosphoglucomutase



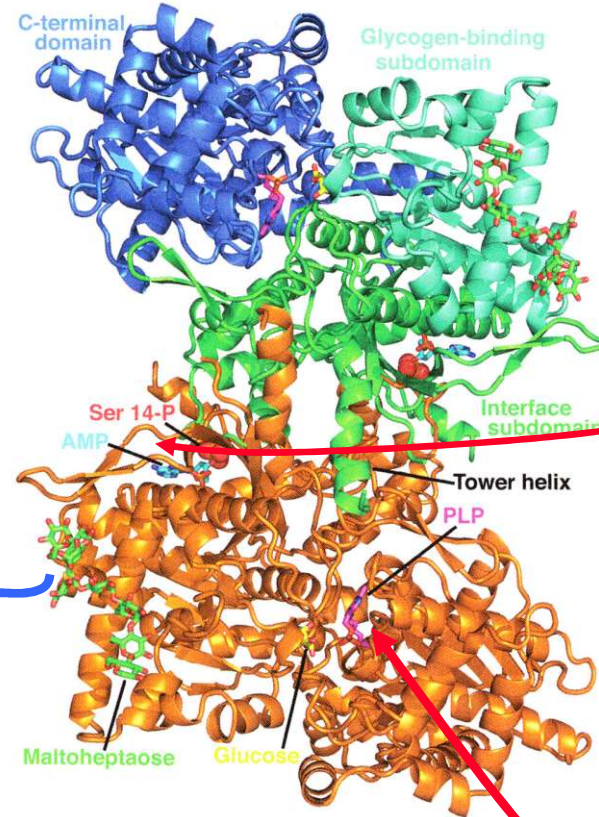
Nils Walter: Chem 451

1.) Glycogen Phosphorylase: The Structure

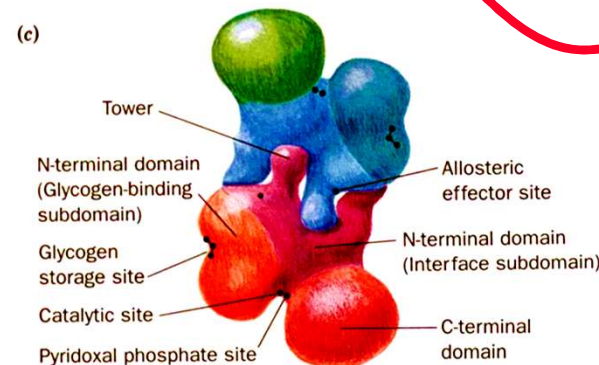
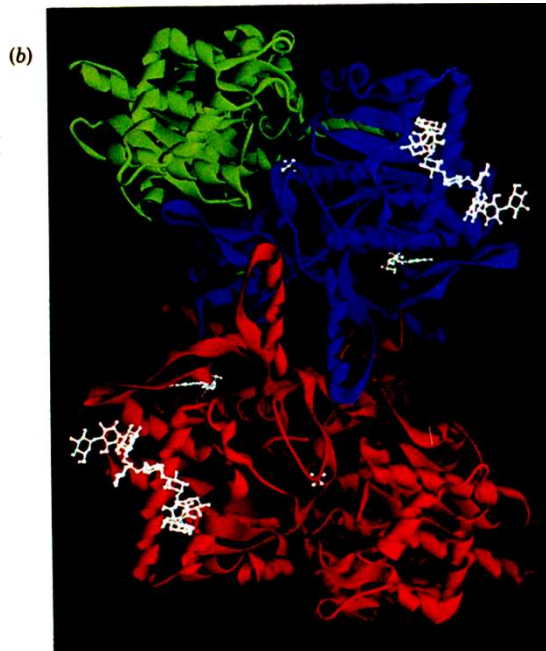
- glycogen storage site
⇒ phosphorylase holds on tightly to glycogen
- fits up to 5 glucoses but no branch!

➤ 842 aa (97 kD)

➤ two forms:
**phosphorylase a w/
Ser14-phosphoester**
and
**phosphorylase b
w/o phosphoester**



➤ dimer:

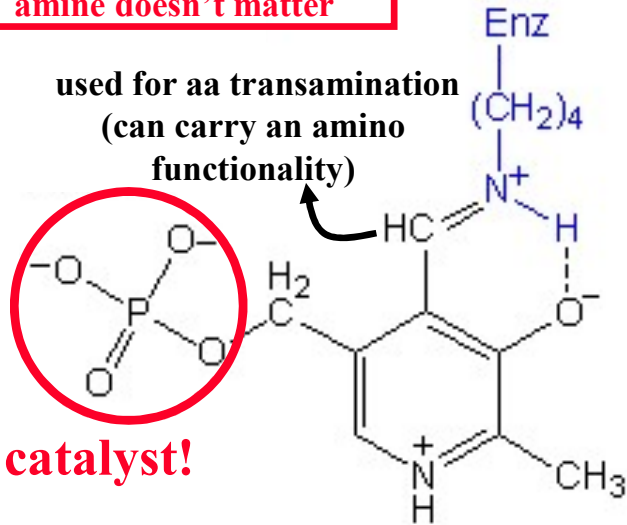


catalytic site

Glycogen Phosphorylase: The Mechanism

BUT: NaBH_4 reduction to amine doesn't matter

used for aa transamination
(can carry an amino functionality)

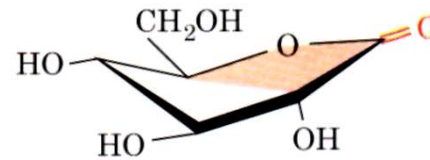


acid catalyst!

**pyridoxal-5-phosphate
(Vitamin B₆ derivative)**

Evidence:

➤ Potent inhibitor:

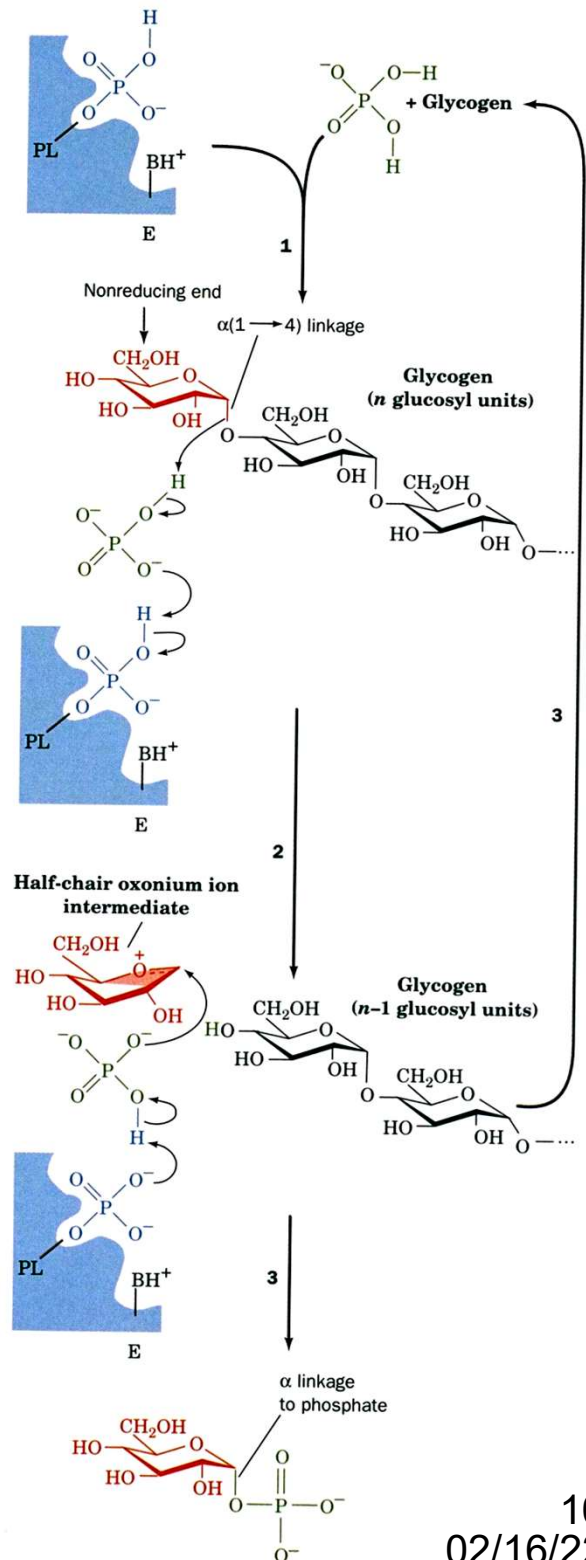
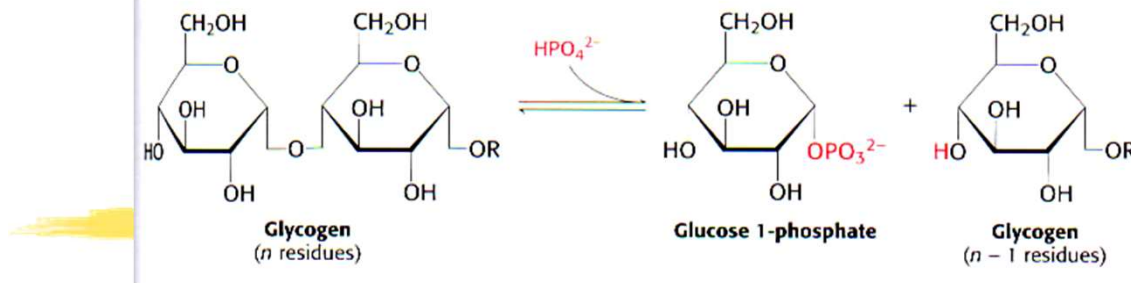


1,5-Gluconolactone

➤ Random Bi Bi kinetics
(no Ping Pong, no intermediates)

➤ Retention of configuration

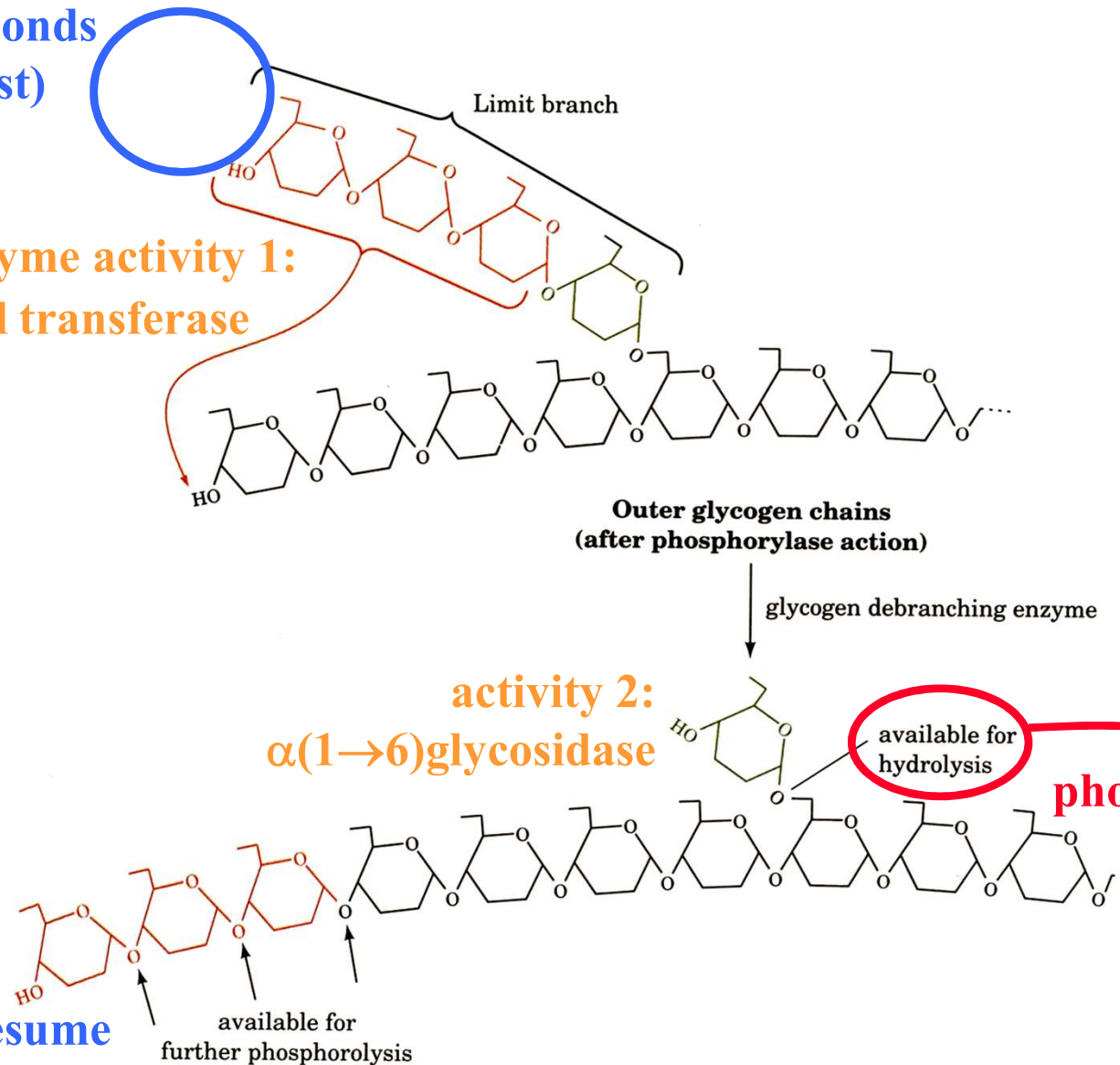
➤ Overall:



2.) Glycogen Debranching Enzyme

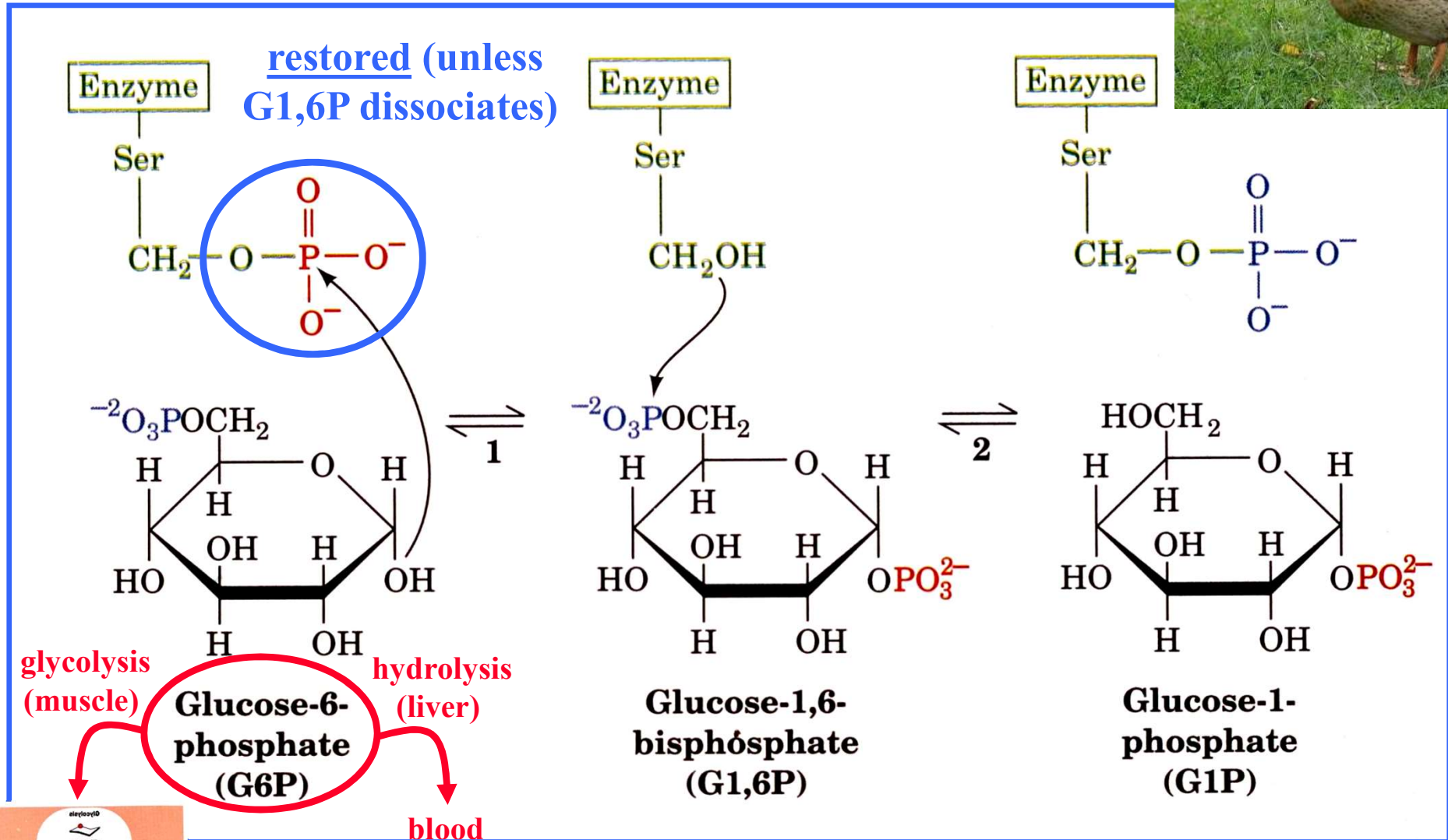
➤ chewed off by
phosphorylase in seconds
(muscle is strongest)

debranching enzyme activity 1:
 $\alpha(1 \rightarrow 4)$ glycosyl transferase

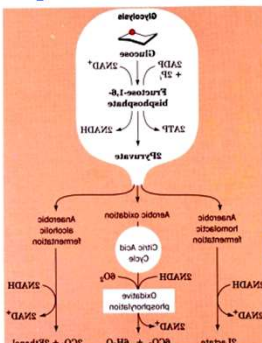


➤ phosphorylase can resume
(after lag)

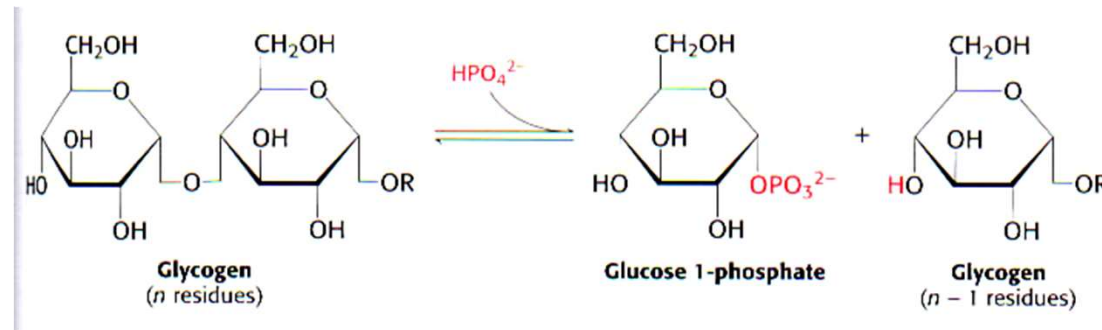
3.) Phosphoglucomutase



➤ **Similar to phosphoglycerate mutase**



Energetics of Glycogen Phosphorylation



$$\Delta G' = \Delta G^{0'} + RT \ln \frac{[\text{glycogen}_{n-1}][\text{G1P}]}{[\text{glycogen}_n][\text{P}_i]} \quad \text{and } \Delta G^{0'} = + 3.1 \text{ kJ/mol}$$

$$\Rightarrow \Delta G' = 0 \text{ (with } [\text{glycogen}_{n-1}] = [\text{glycogen}_n]) \text{ for } \frac{[\text{G1P}]}{[\text{P}_i]} = \frac{1}{3.5}$$

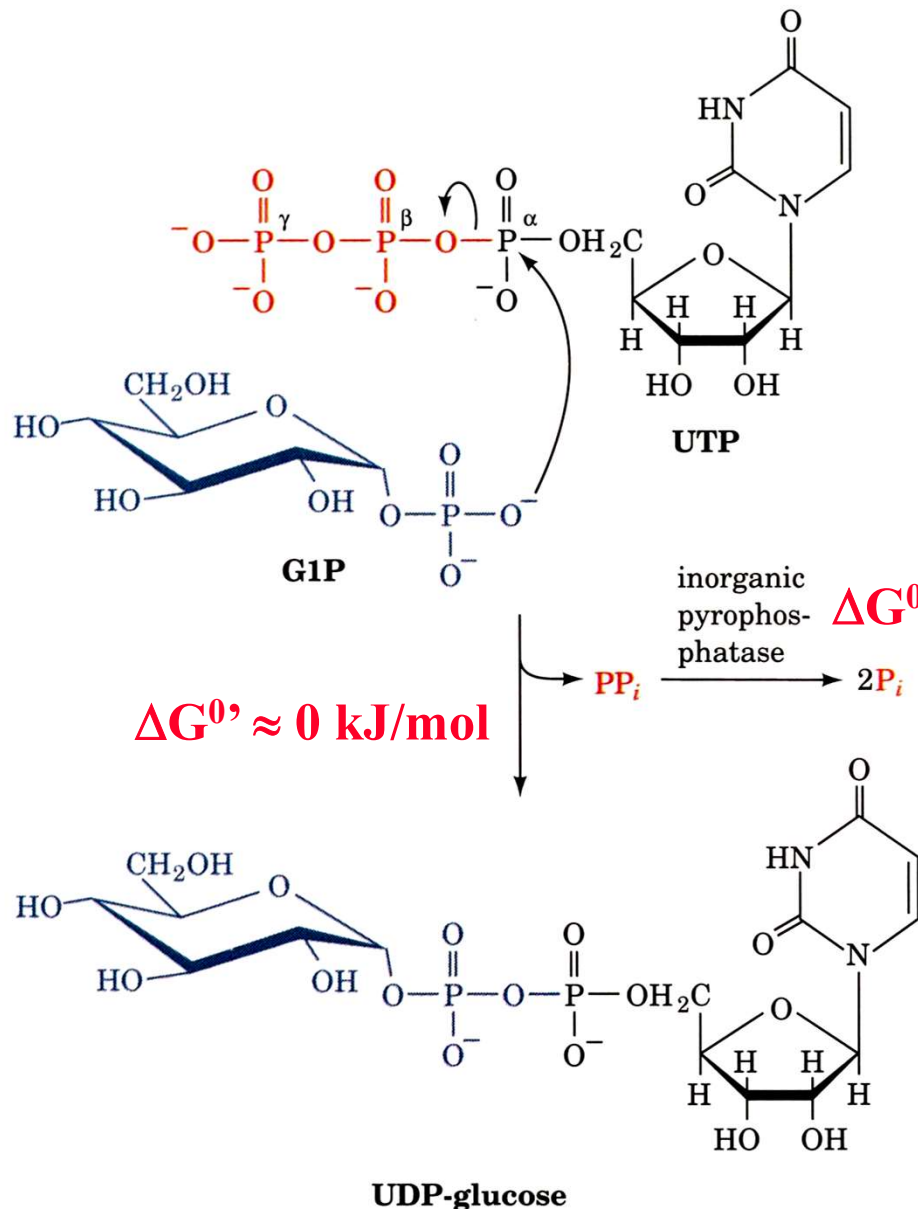
in cell: $\frac{[\text{G1P}]}{[\text{P}_i]} = \frac{1}{30} \text{ to } \frac{1}{100} \Rightarrow \Delta G' = -5 \text{ to } -8 \text{ kJ/mol}$ **exergonic**

Biosynthetic and degradative pathways are different!

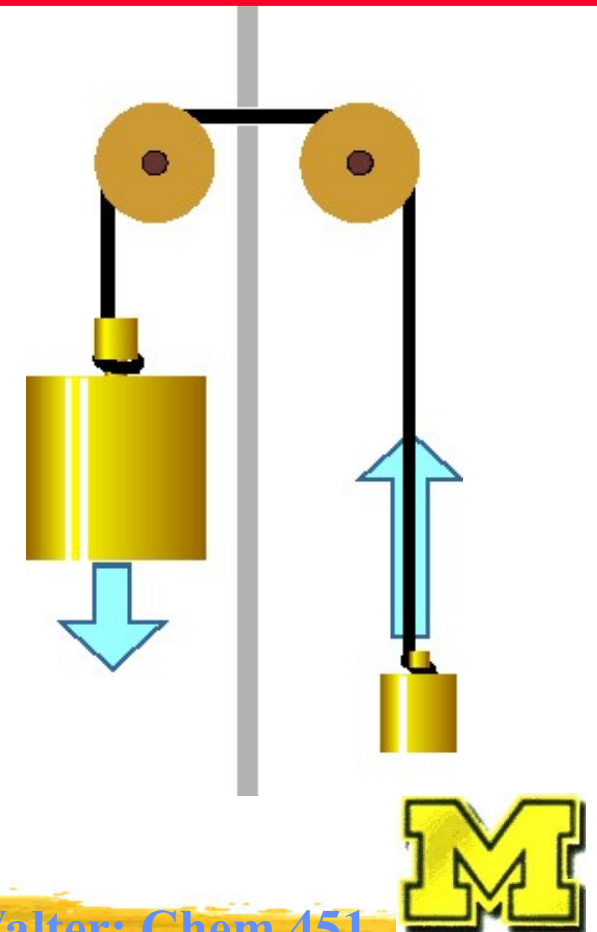
- Both pathways can be independently regulated
- Both pathways can be active at similar concs.

Glycogen Synthesis Must Take another Route

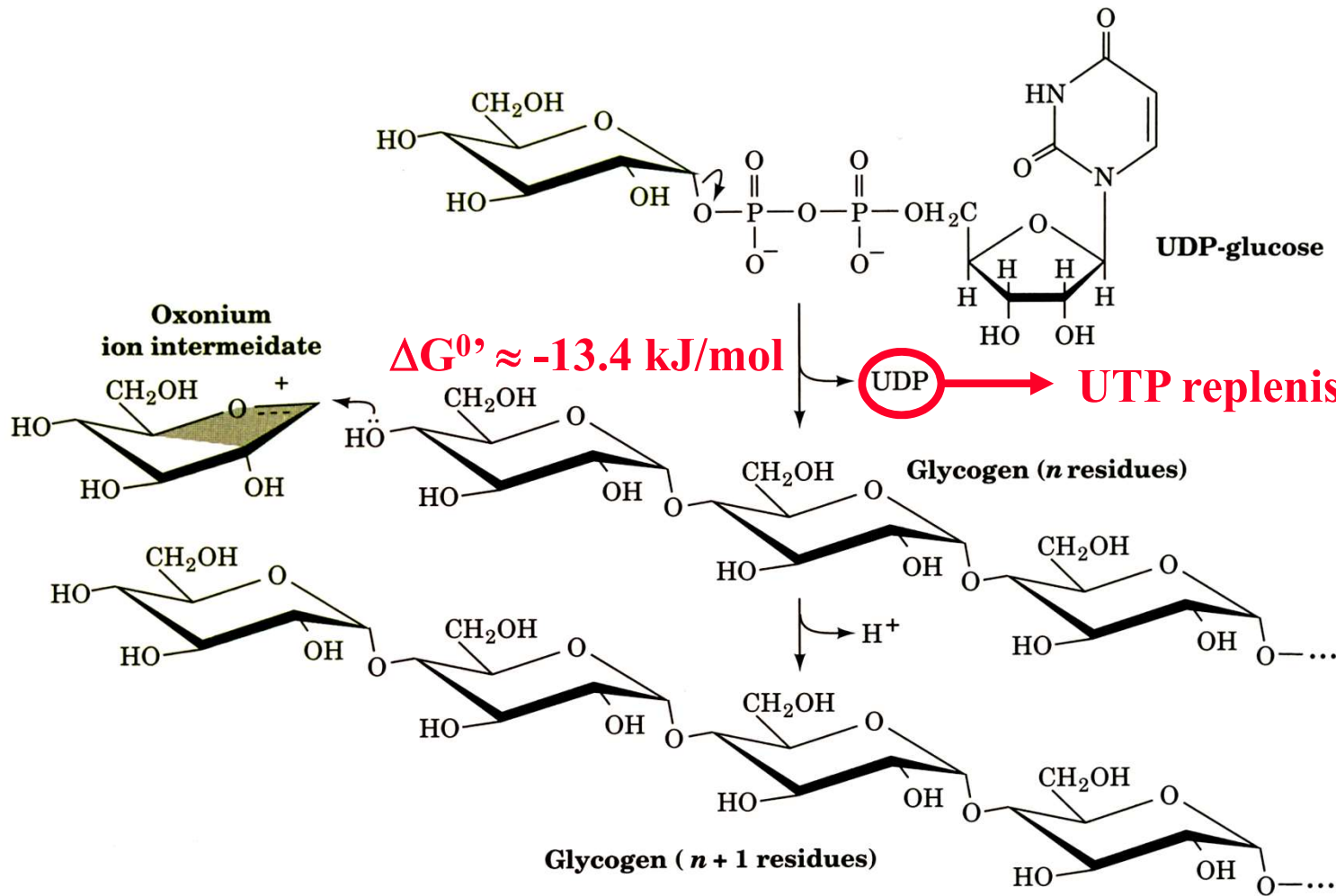
1.) UDP-Glucose Pyrophosphorylase



⇒ Overall exergonic



2.) Glycogen Synthase

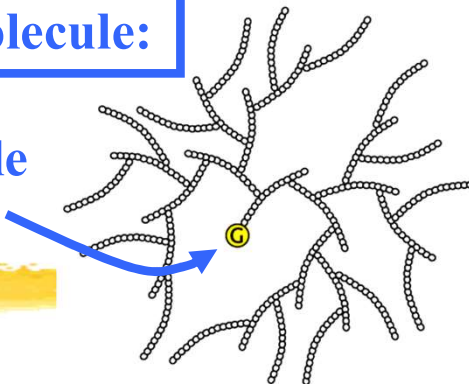


**Overall loss of
1 UTP \rightarrow UDP + P_i**

**BUT this is ONLY a
loss of 3% of the
overall energy stored
in 1 glucose
molecule!**

Priming a glycogen molecule:

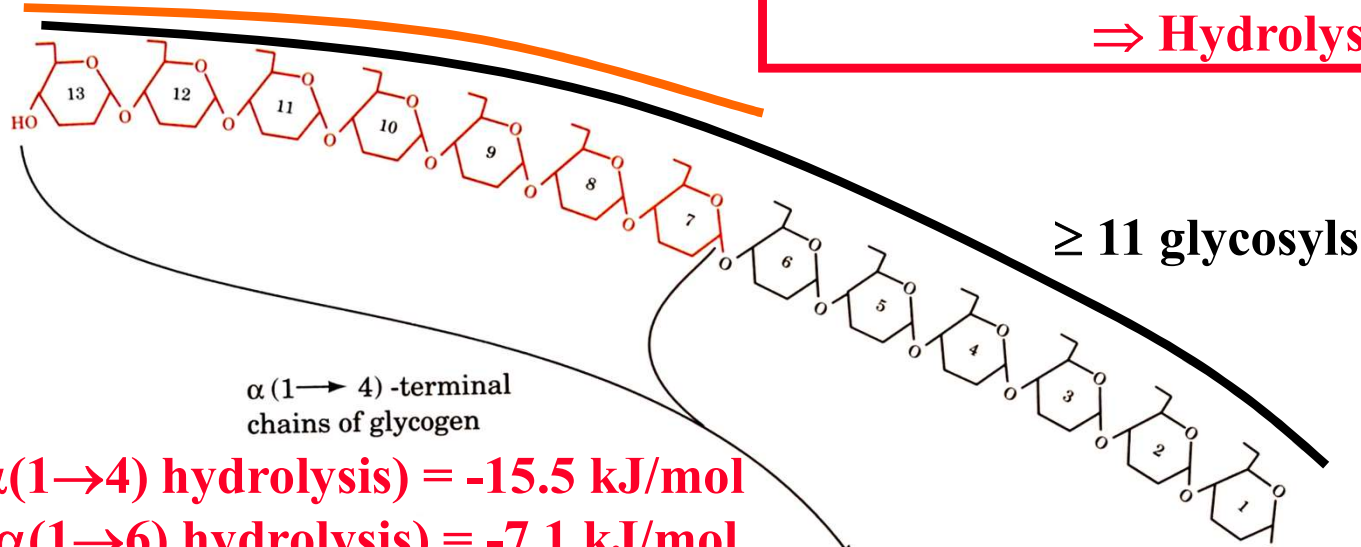
**1 Glycogenin/granule
for first 7 glucoses**



3.) Glycogen Branching Enzyme

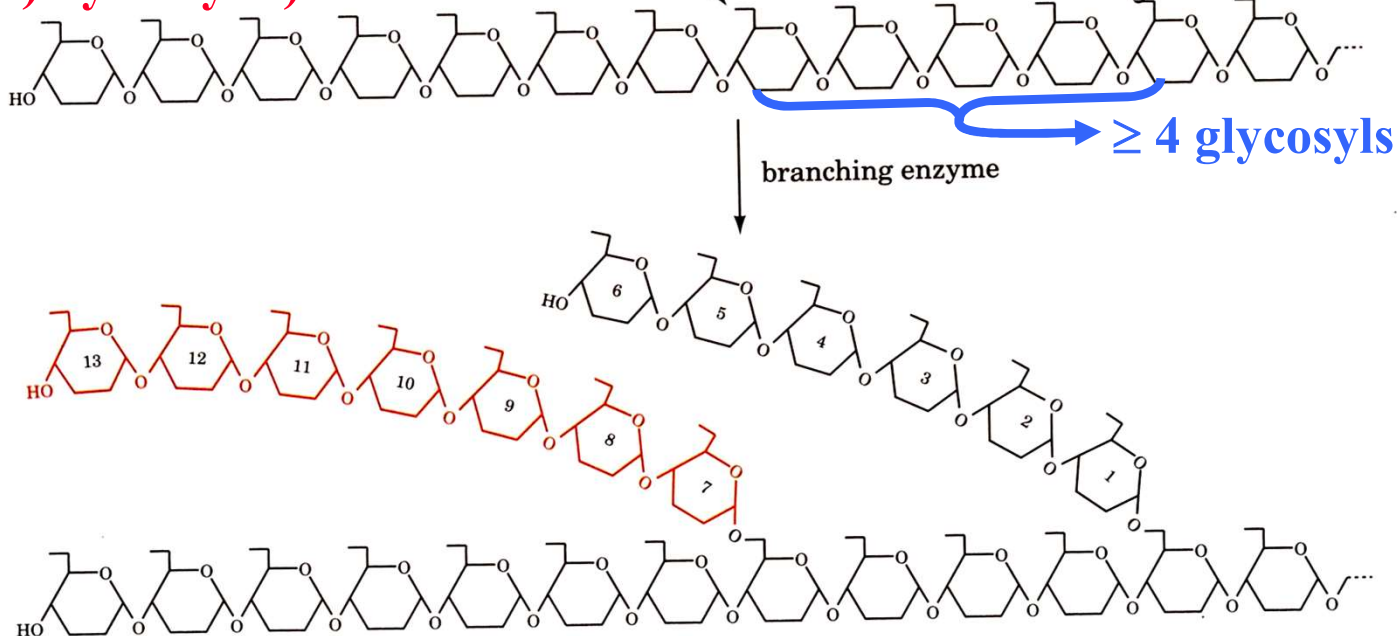
- Transfer for branching is exergonic
- Transfer for debranching would be endergonic
⇒ Hydrolysis instead!

~7 glycosyls



$\Delta G(\alpha(1 \rightarrow 4) \text{ hydrolysis}) = -15.5 \text{ kJ/mol}$

$\Delta G(\alpha(1 \rightarrow 6) \text{ hydrolysis}) = -7.1 \text{ kJ/mol}$



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