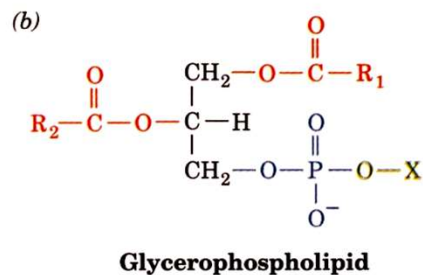
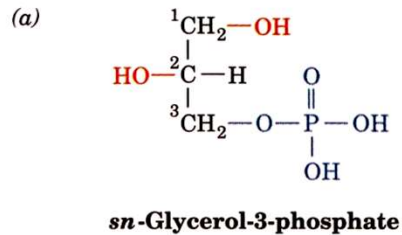
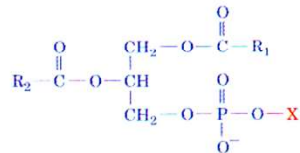
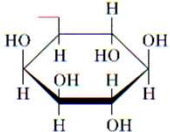

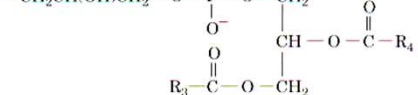


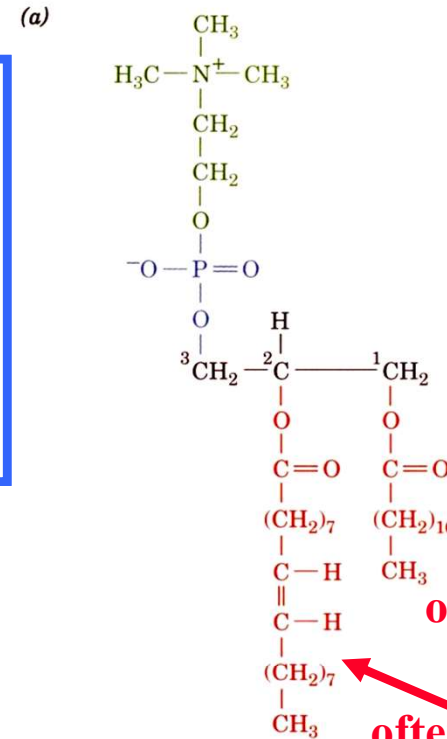
Glycerophospholipids, the Major Players



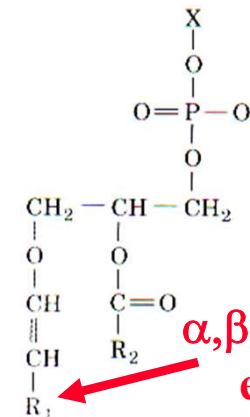
➤ **Glycerophospholipids = phosphoglycerides are amphiphilic molecules with nonpolar aliphatic “tails” and polar phosphoryl-X “heads”**



Name of X—OH	Formula of —X	Name of Phospholipid
Water	— H	Phosphatidic acid
Ethanolamine	— CH ₂ CH ₂ NH ₃ ⁺	Phosphatidylethanolamine
Choline	— CH ₂ CH ₂ N(CH ₃) ₃ ⁺	Phosphatidylcholine (lecithin)
Serine	— CH ₂ CH(NH ₃ ⁺)COO [−]	Phosphatidylserine
myo-Inositol		Phosphatidylinositol
Glycerol		Phosphatidylglycerol
Phosphatidylglycerol		Diphosphatidylglycerol (cardiolipin)



1-Stearoyl-2-oleoyl-3-phosphatidylcholine

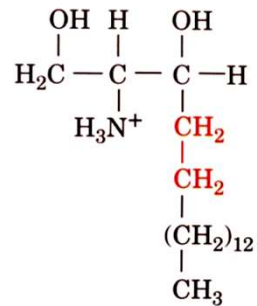
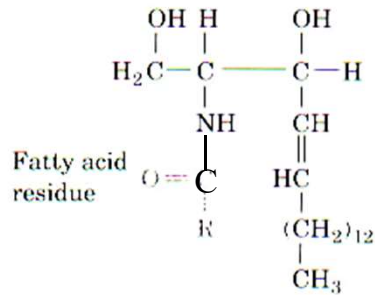
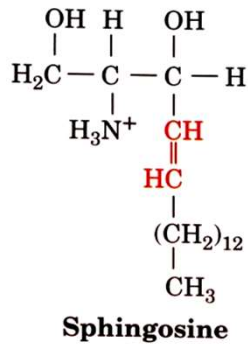


A plasmalogen



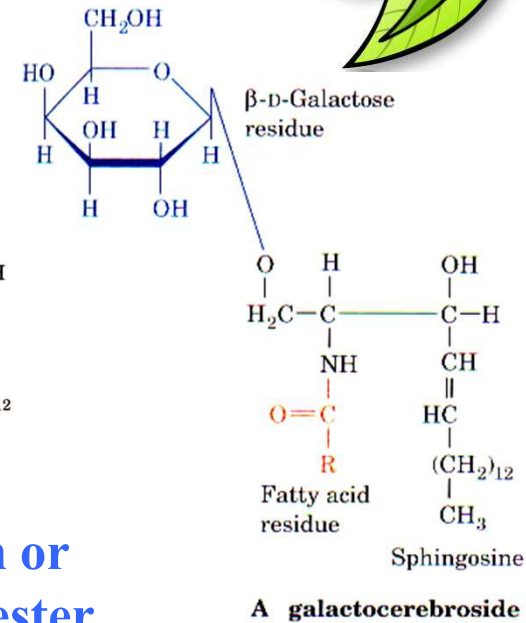
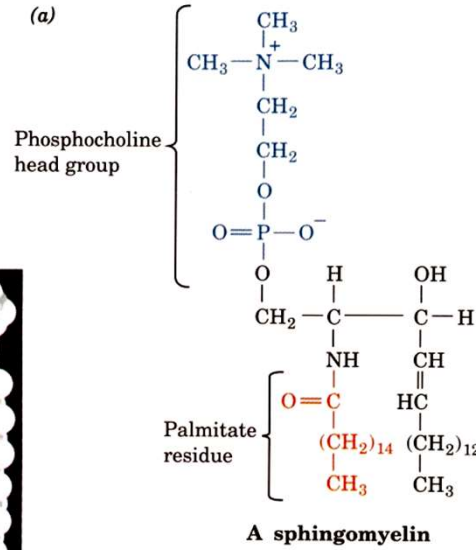
Chem 451

Sphingolipids, the Brainy Ones



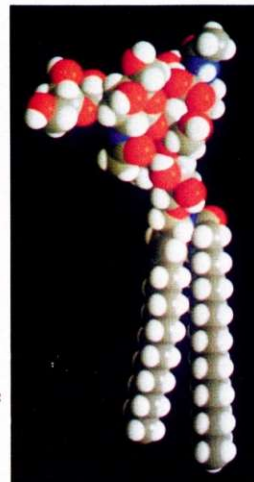
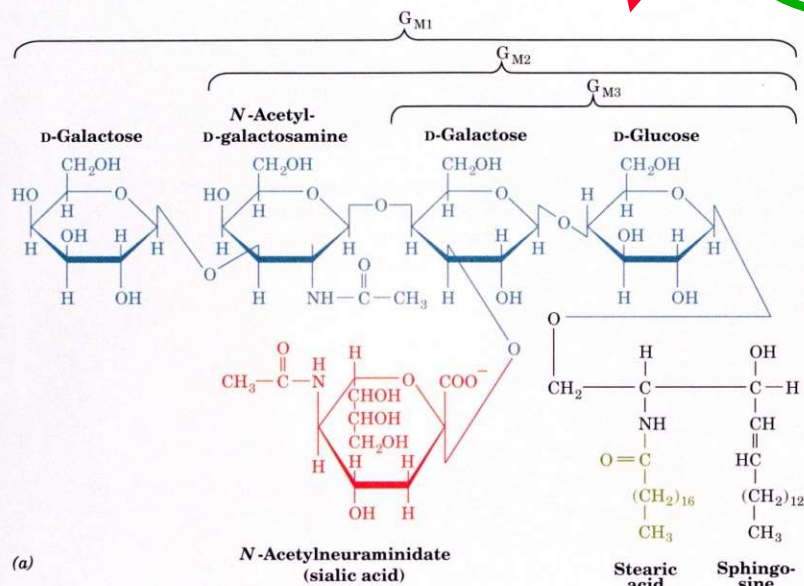
C₁₈ amino alcohols

Fatty acid amide



Phosphocholin or -ethanolamine ester (structurally similar to phospholipids)

Head group = one sugar



Most complex sphingoglycolipids

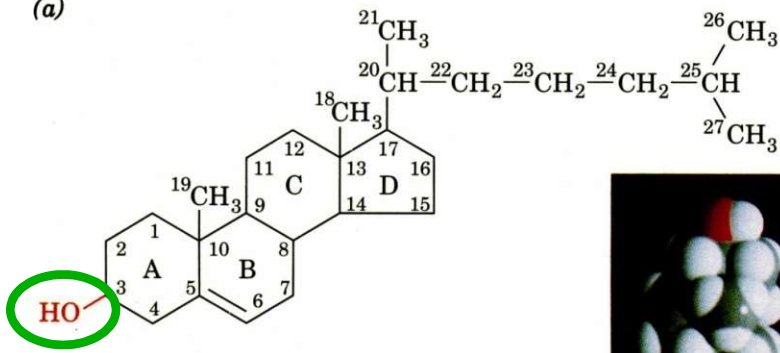
6% of brain lipids; cell-cell recognition

Nus Water: Chem 451



Cholesterol, the Unruly One

(a)

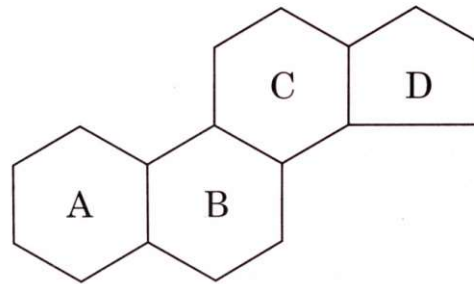
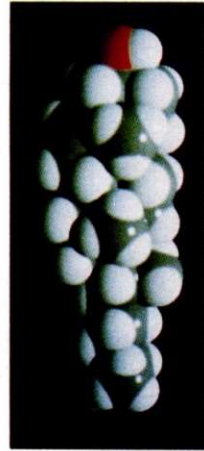


Cholesterol

➤ major component of animal plasma membranes

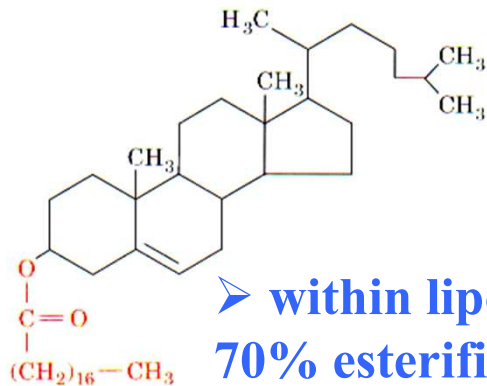
➤ weakly amphiphilic

(b)



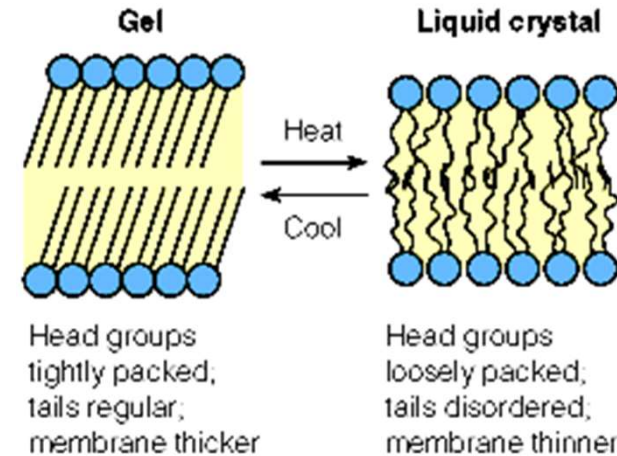
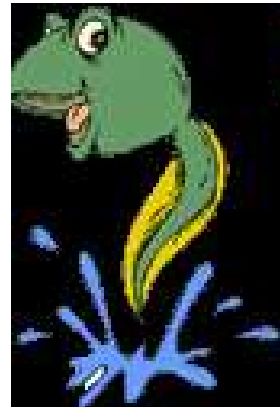
Cyclopentanoperhydrophenanthrene

➤ rigid fused ring system

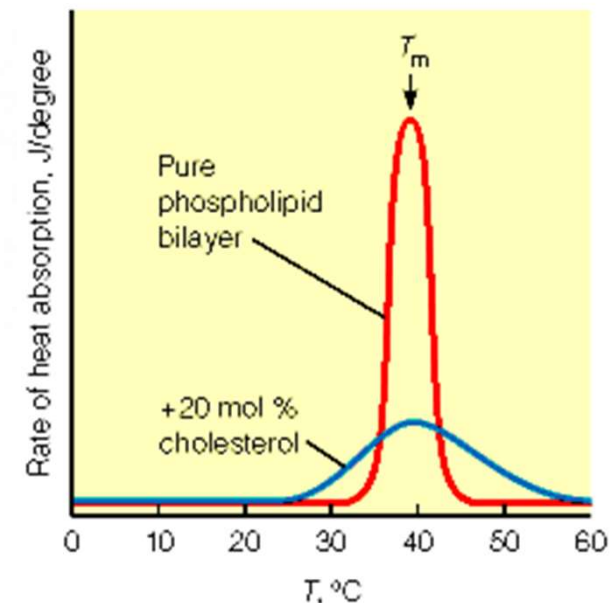


Cholesteryl stearate

➤ within lipoproteins in blood plasma:
70% esterified to fatty acids for transport



(a) Transition

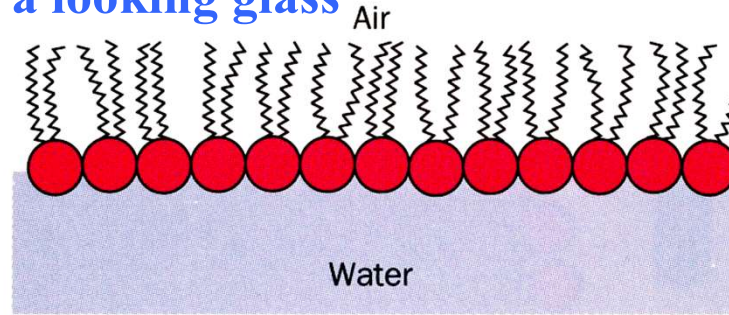


(b) Transition with and without cholesterol



The “Magical” Properties of Lipids

➤ Benjamin Franklin (1774): A drop of olive oil calms waves on a pond to become “smooth as a looking glass”



Reduction in water's
high surface tension

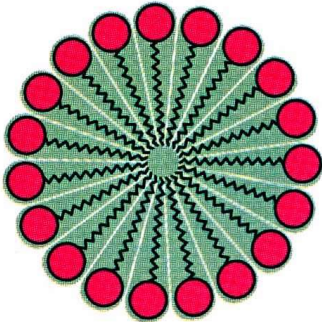
Single-tailed lipids/detergents form small (several 100 molecules containing) micelles:

(a)

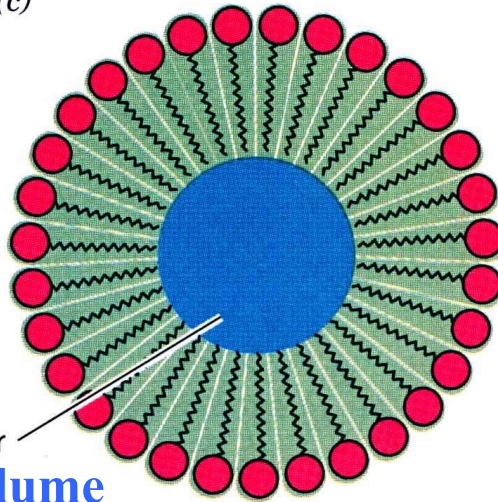


van der Waals
envelope

(b)



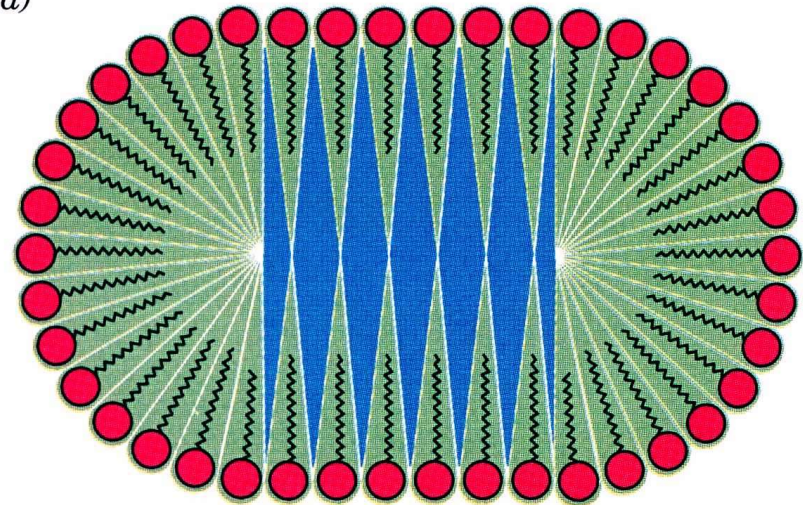
(c)



Void volume

⇒ unfavorable!

(d)



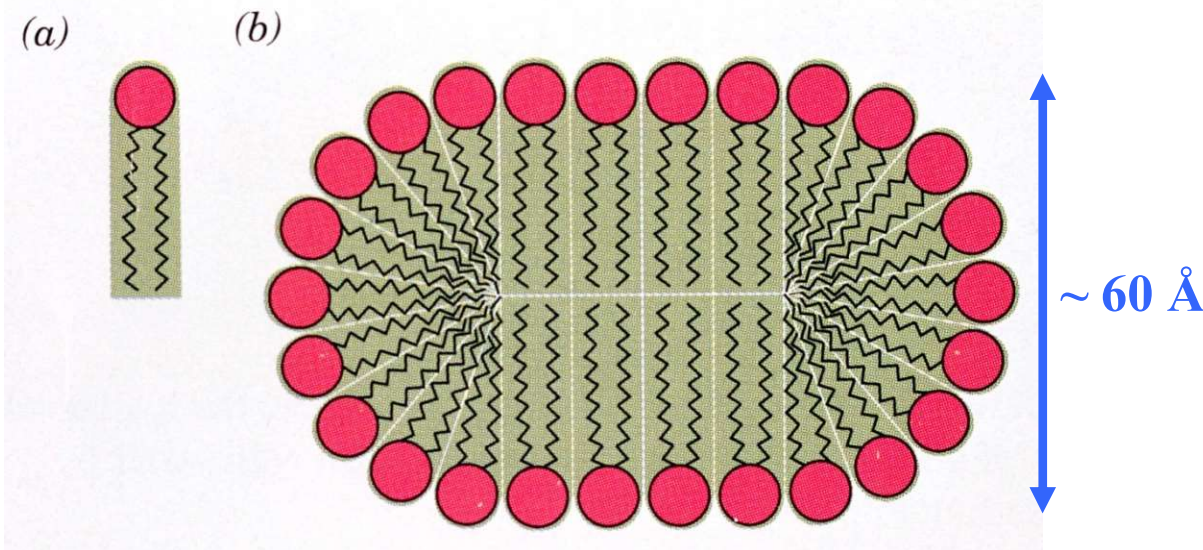
Good at emulsifying!

Nils Walter: Chem 451

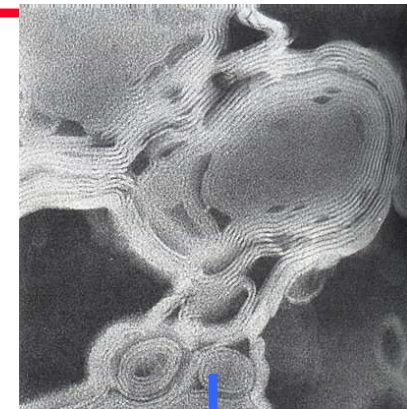
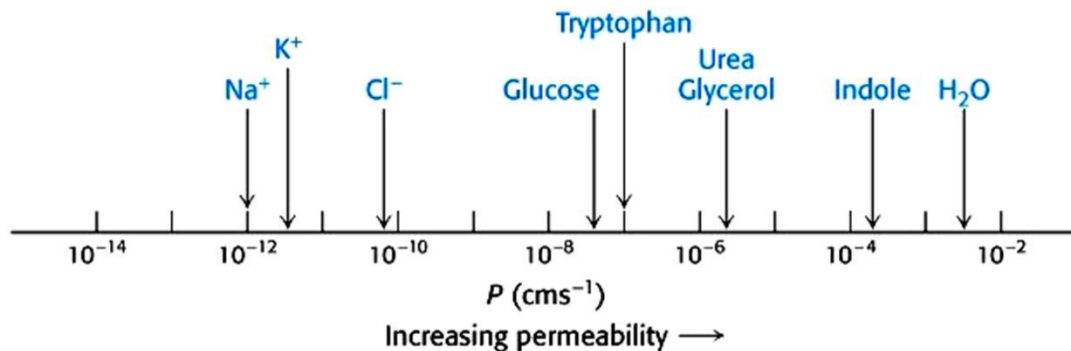


Lipid Bilayers Work Their Magic in and Around Cells

**Double-tailed lipids = phospholipids and sphingolipids form disklike micelles
⇒ lipid bilayers of membranes!**



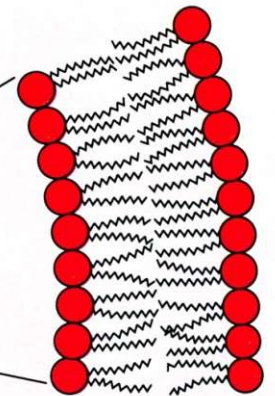
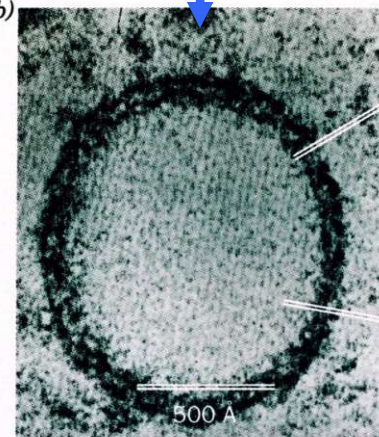
Lipid bilayers are highly impermeable to ions and polar molecules



Multilamellar phospholipid vesicle

sonication

(b)



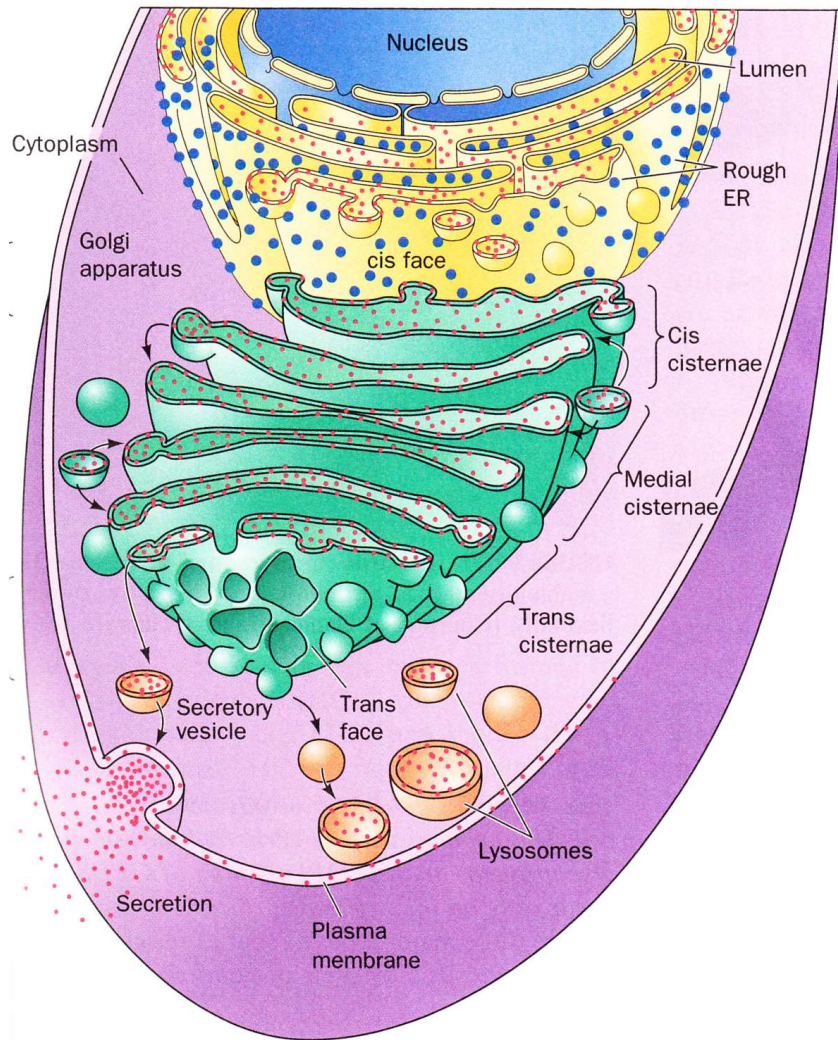
Liposome = closed, self-sealing, solvent-filled single-layer vesicle

Nils Walter: Chem 451

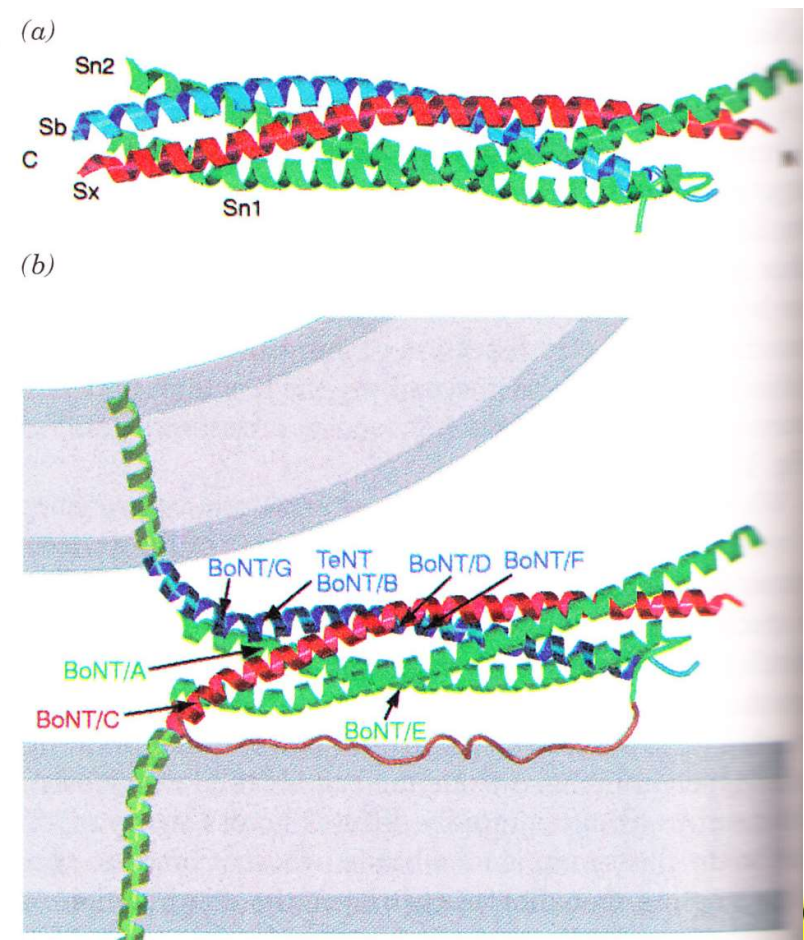


Membranes Profoundly Determine How Our Human Cells Work

➤ Compartmentalization



➤ Membrane fusion through SNARE-SNAP complexes



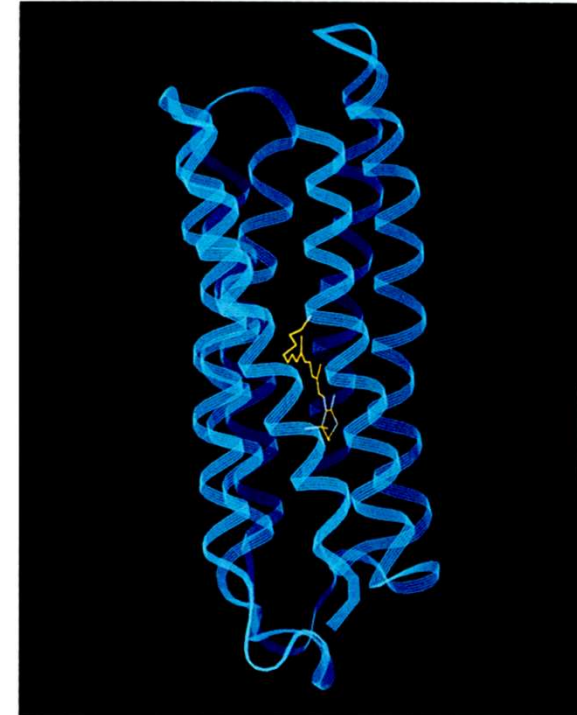
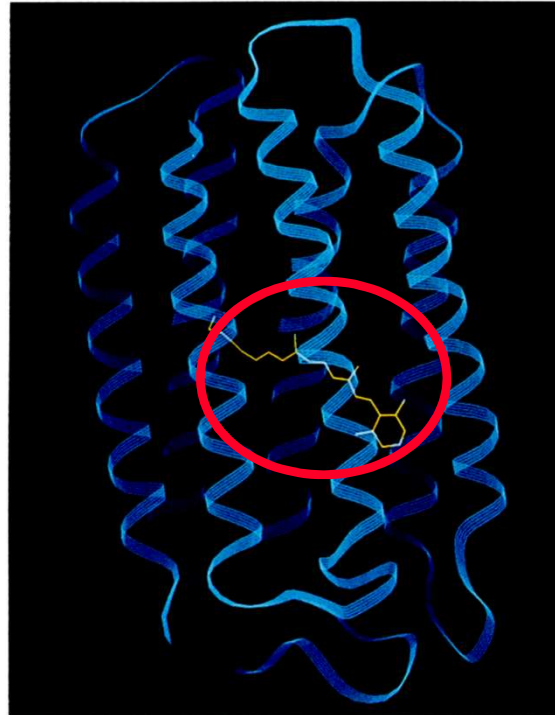
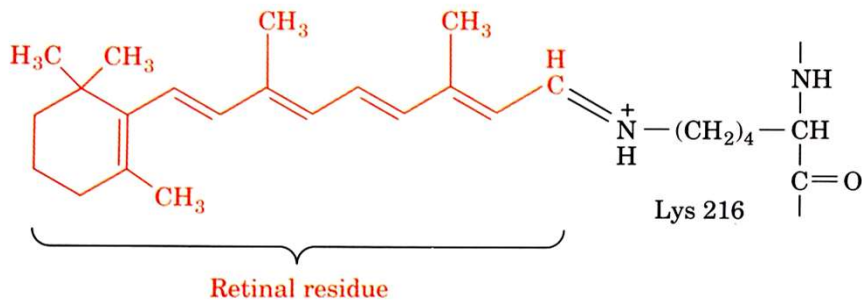
<https://www.youtube.com/watch?v=0KRB6fmqvnc>
https://www.youtube.com/watch?v=B_zD3NxSsD8
<http://www.youtube.com/watch?v=KxTYyNEbVU4>

Nils Walter: Chem 451



Membrane Protein Example: *Halobacterium halobium* Bacteriorhodopsin

- Extreme halophile - non-viable below 2 M NaCl (seawater has only 0.6 M NaCl!)
- At low [O₂] cell membrane develops 0.5-μm wide patches of purple membrane with bacteriorhodopsin as the only component ⇒ light-driven proton pump



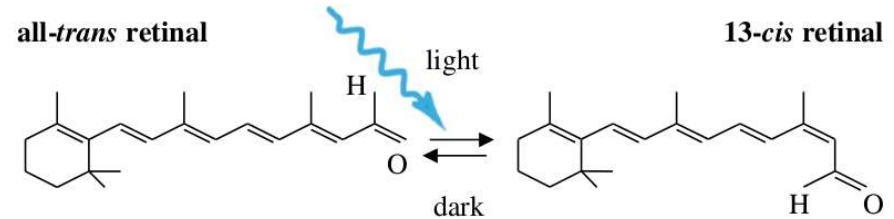
Bacteriorhodopsin : How it Works In the Membrane

a Ground state
all-*trans* retinal protonated

b L intermediate
13-*cis* retinal protonated

all-*trans* retinal

13-*cis* retinal



Vision in our eyes
works similarly:

- Rods and cones are photoreceptor cells with rhodopsin, composed of a protein (opsin) and retinal
- Trans→Cis isomerization drives a cyclic cascade that amplifies the signal, generating a nerve cell impulse (Na^+ , Ca^{2+} influx channel closed)



Nils Walter: Chem 451