

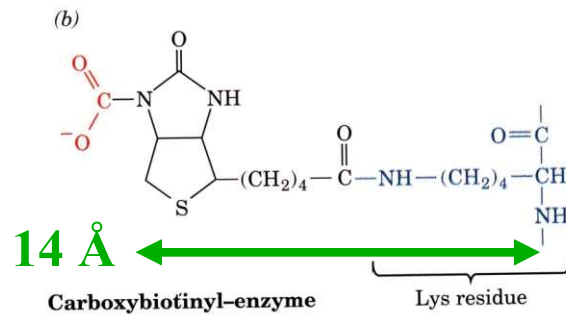
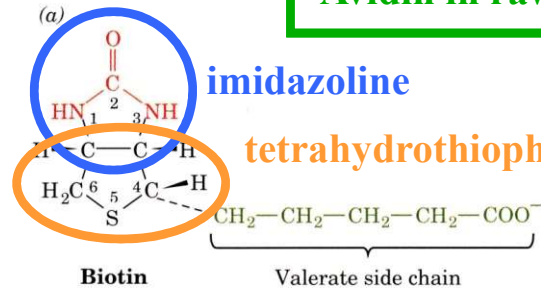
Pyruvate Carboxylase Has a Biotin Arm

ureido group

Avidin in raw egg white causes biotin (vitamin H) deficiency (and is cytostatic)

imidazoline

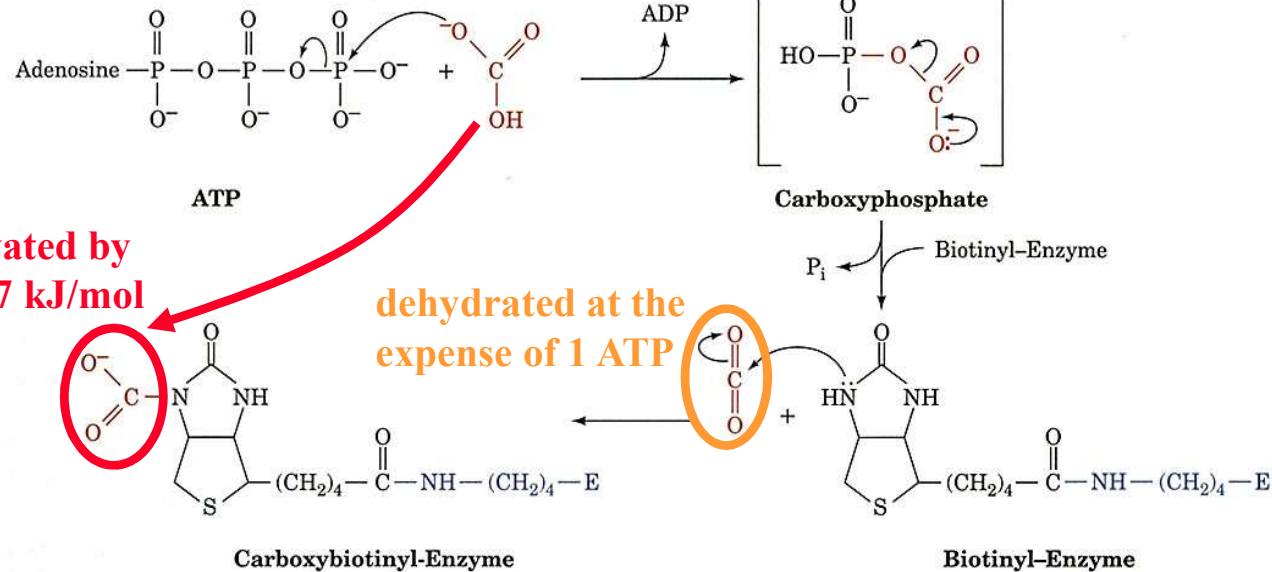
tetrahydrothiophene



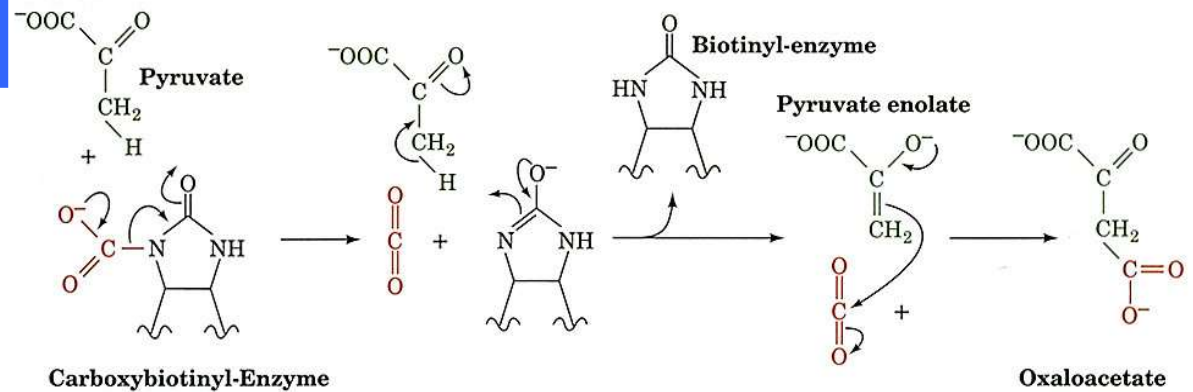
Phase I

activated by
+19.7 kJ/mol

dehydrated at the
expense of 1 ATP



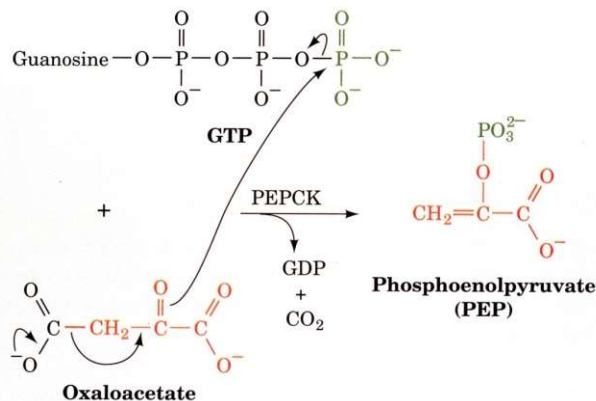
Phase II : arm swings to different subsite



BUT WAIT: Pyruvate Carboxylase is mitochondrial, PEPCK mitochondrial and/or cytosolic, other gluconeogenesis cytosolic

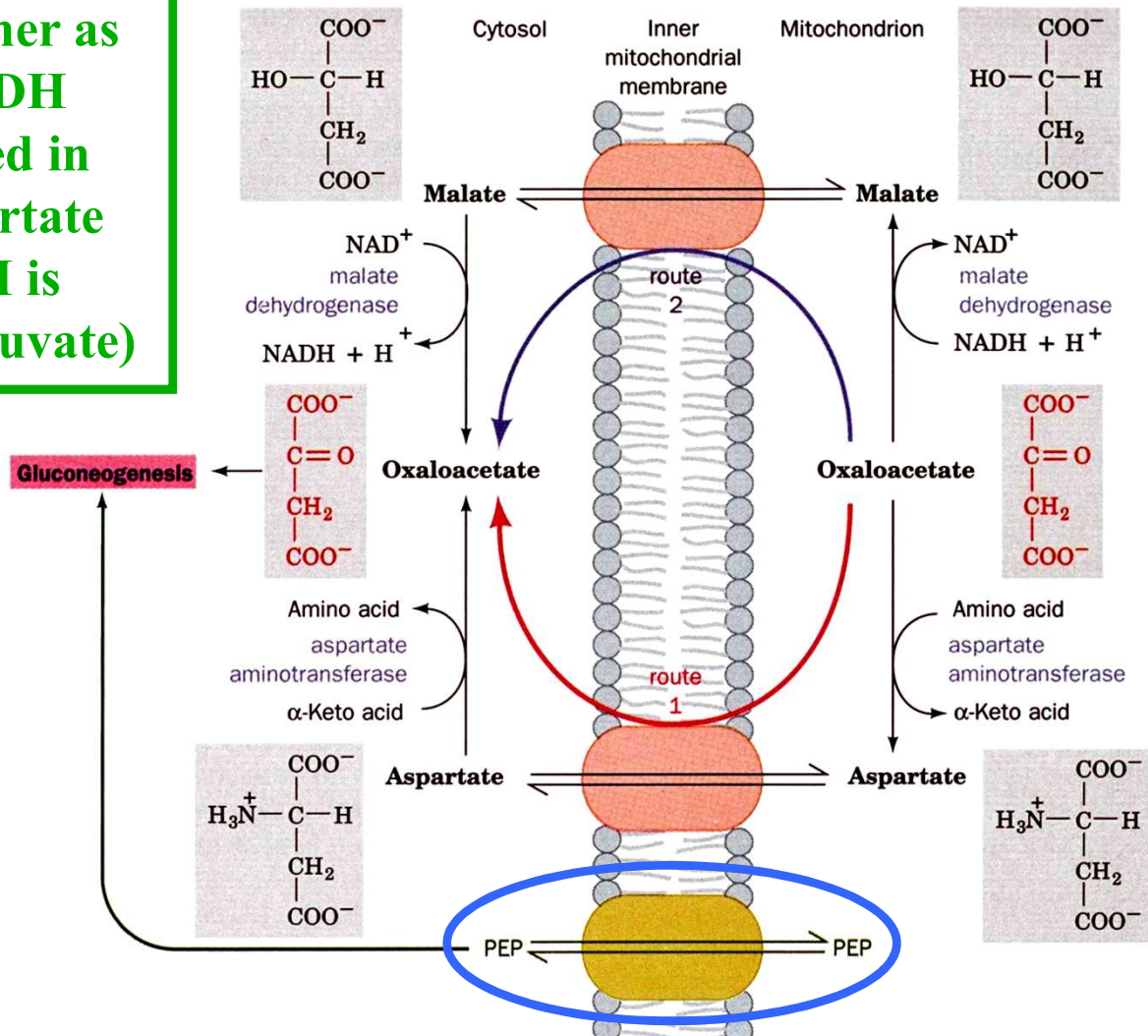


PEP Carboxykinase uses “activated” CO₂ and GTP to make high-energy PEP



PEP & Oxaloacetate Have to Leave & Enter the Mitochondrial Matrix

Oxaloacetate is shuttled either as malate (carrying also NADH reducing equivalents needed in gluconeogenesis) or as aspartate (without NADH, if NADH is available from lactate → pyruvate)



PEP is transported directly and specifically

Nils Walter: Chem 451



Glycolysis and Gluconeogenesis as Two Opposing Pathways Have to be Regulated



Glycolysis:



Gluconeogenesis:



As a "futile cycle" :

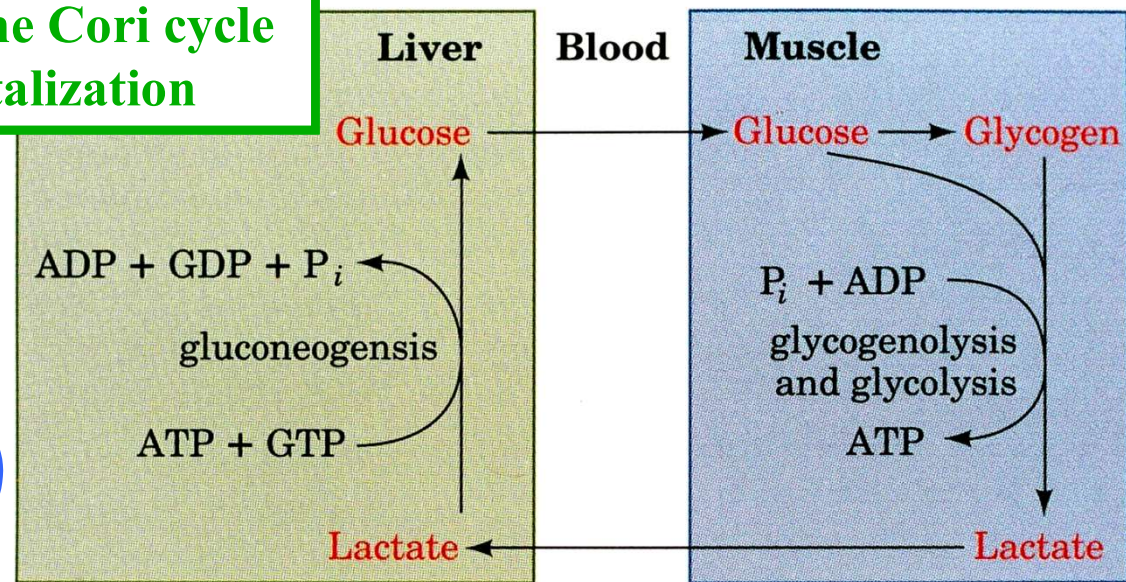


⇒ Must be prevented! ⇒ Regulation!

The liver, the altruistic organ: The Cori cycle allows for tissue compartmentalization

TABLE 21-1. REGULATORS OF GLUCONEOGENIC ENZYME ACTIVITY

| Enzyme | Allosteric Inhibitors | Allosteric Activators | Enzyme Phosphorylation | Protein Synthesis |
|----------------------|-----------------------|--------------------------|------------------------|------------------------|
| PFK | ATP, citrate | AMP, F2,6P | | |
| FBPase | AMP, F2,6P | | | |
| PK | Alanine | F1,6P | Inactivates | |
| Pyruvate carboxylase | | Acetyl-CoA | | |
| PEPCK | | | | Stimulated by glucagon |
| PFK-2 | Citrate | AMP, F6P, P _i | Inactivates | |
| FBPase-2 | F6P | Glycerol-3-P | Activates | |



Chapter 23: What have we learned?

- ☺ **Gluconeogenesis = Glycolysis reversed: The three key steps that have to be bypassed**
- ☺ **Oxaloacetate as a key metabolite**
- ☺ **The pyruvate carboxylase mechanism**
- ☺ **How to shuttle oxaloacetate out of the mitochondrion**
- ☺ **Energetics and Regulation of gluconeogenesis; the Cori cycle; ~~Lance Armstrong's story~~**



Review: Lipids and Membranes

Greek: lipos = fat

Voet & Voet, Chapter 12

➤ soluble in organic solvents

(CH₃OH, CHCl₃), not in water ⇒ easy fractionation

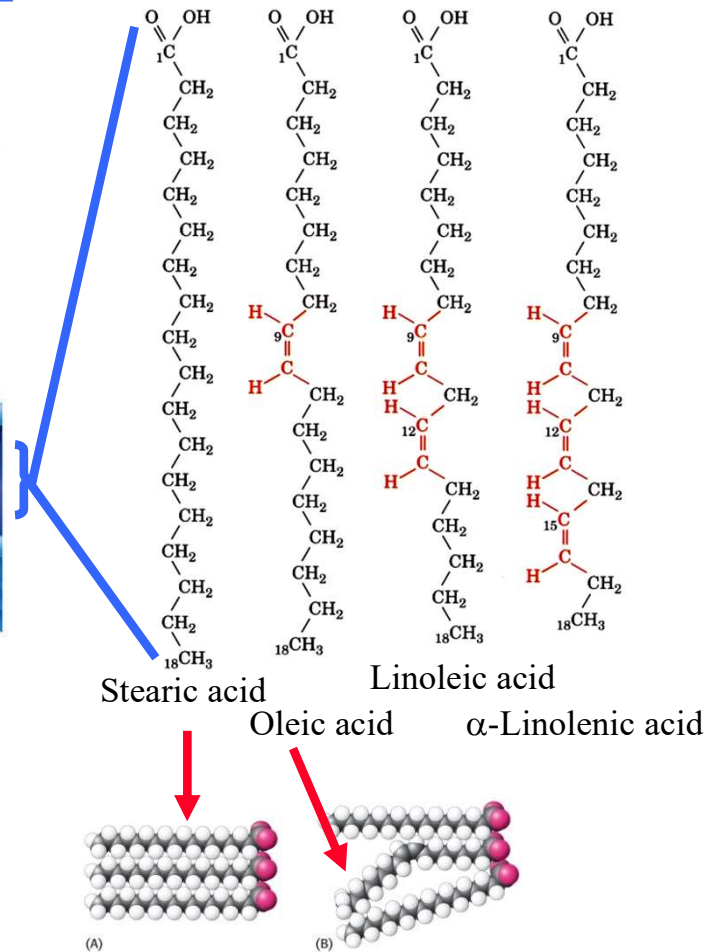
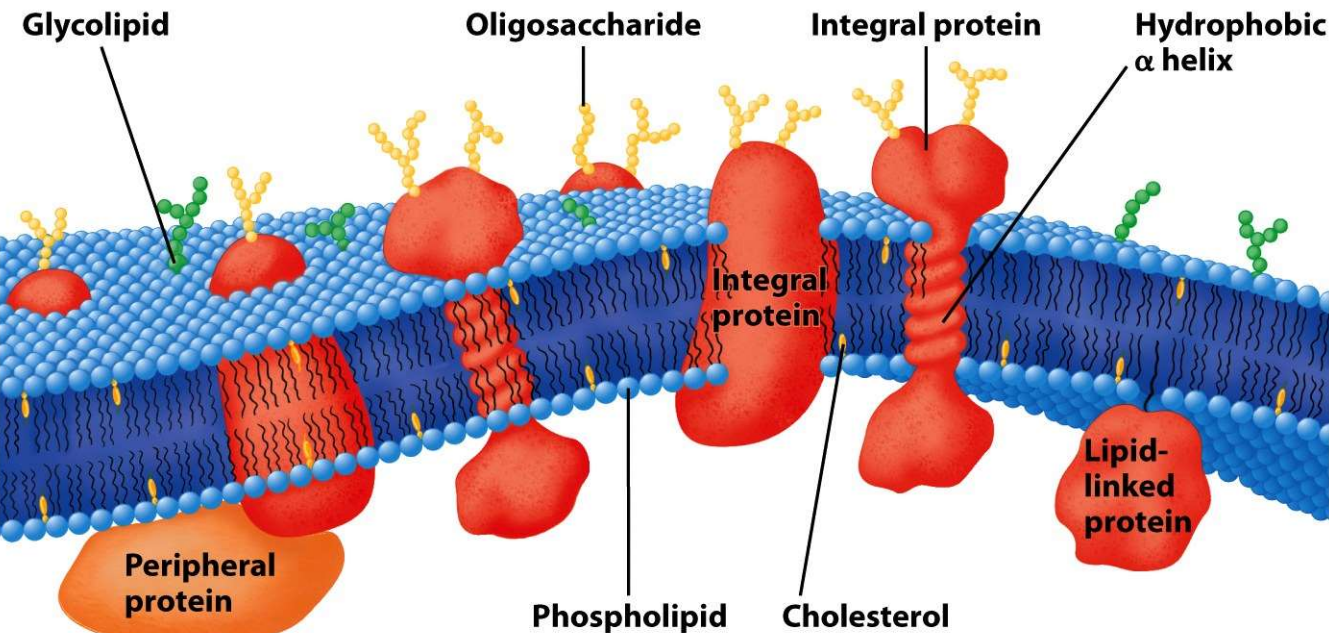


TABLE 11-1. THE COMMON BIOLOGICAL FATTY ACIDS

| Symbol ^a | Common Name | Systematic Name | Structure | mp (°C) |
|---|------------------|-----------------------------------|---|---------|
| Saturated fatty acids | | | | |
| 12:0 | Lauric acid | Dodecanoic acid | CH ₃ (CH ₂) ₁₀ COOH | 44.2 |
| 14:0 | Myristic acid | Tetradecanoic acid | CH ₃ (CH ₂) ₁₂ COOH | 52 |
| 16:0 | Palmitic acid | Hexadecanoic acid | CH ₃ (CH ₂) ₁₄ COOH | 63.1 |
| 18:0 | Stearic acid | Octadecanoic acid | CH ₃ (CH ₂) ₁₆ COOH | 69.6 |
| 20:0 | Arachidic acid | Eicosanoic acid | CH ₃ (CH ₂) ₁₈ COOH | 75.4 |
| 22:0 | Behenic acid | Docosanoic acid | CH ₃ (CH ₂) ₂₀ COOH | 81 |
| 24:0 | Lignoceric acid | Tetracosanoic acid | CH ₃ (CH ₂) ₂₂ COOH | 84.2 |
| Unsaturated fatty acids (all double bonds are cis) | | | | |
| 16:1 | Palmitoleic acid | 9-Hexadecenoic acid | CH ₃ (CH ₂) ₅ CH=CH(CH ₂) ₇ COOH | -0.5 |
| 18:1 | Oleic acid | 9-Octadecenoic acid | CH ₃ (CH ₂) ₇ CH=CH(CH ₂) ₇ COOH | 13.4 |
| 18:2 | Linoleic acid | 9,12-Octadecadienoic acid | CH ₃ (CH ₂) ₄ (CH=CHCH ₂) ₂ (CH ₂) ₆ COOH | -9 |
| 18:3 | α-Linolenic acid | 9,12,15-Octadecatrienoic acid | CH ₃ CH ₂ (CH=CHCH ₂) ₃ (CH ₂) ₆ COOH | -17 |
| 18:3 | γ-Linolenic acid | 6,9,12-Octadecatrienoic acid | CH ₃ (CH ₂) ₄ (CH=CHCH ₂) ₃ (CH ₂) ₃ COOH | -49.5 |
| 20:4 | Arachidonic acid | 5,8,11,14-Eicosatetraenoic acid | CH ₃ (CH ₂) ₄ (CH=CHCH ₂) ₄ (CH ₂) ₂ COOH | -54 |
| 20:5 | EPA | 5,8,11,14,17-Eicosapentanoic acid | CH ₃ CH ₂ (CH=CHCH ₂) ₅ (CH ₂) ₂ COOH | -49.5 |
| 24:1 | Nervonic acid | 15-Tetracosenoic acid | CH ₃ (CH ₂) ₇ CH=CH(CH ₂) ₁₃ COOH | 39 |

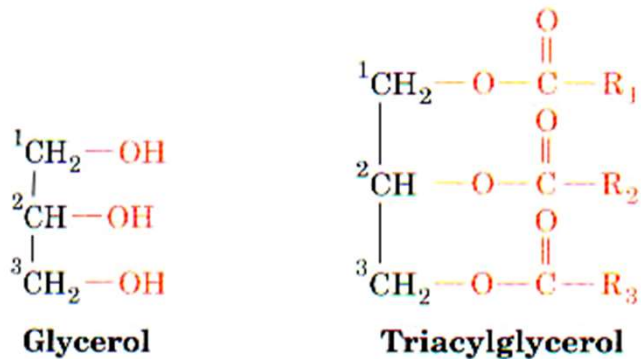
^a Number of carbon atoms : Number of double bonds.

Source: Dawson, R.M.C., Elliott, D.C., Elliott, W.H., and Jones, K.M., Data for Biochemical Research (3rd ed.), Chapter 8, Clarendon Press (1986).

- fatty acids of <C₁₄ or >C₂₀ are uncommon
- most have even # of Cs (biosynthesized from C₂)
- >50% of natural fatty acids are unsaturated
- unconjugated cis configuration
- melting point (mp) decreases with unsaturation

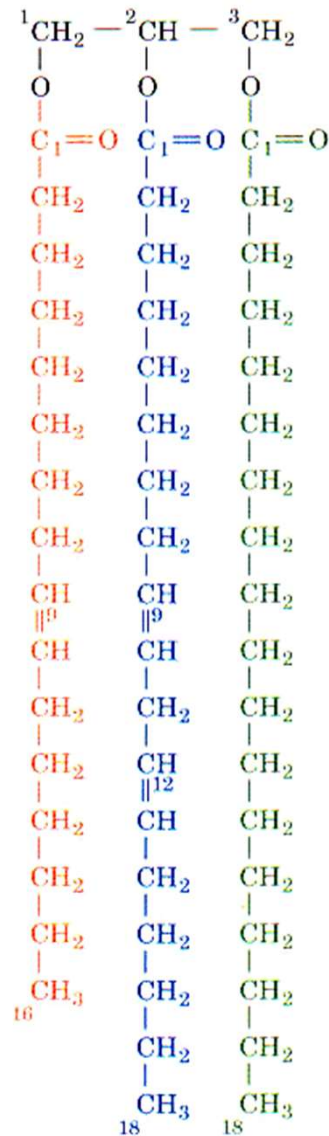
Triacylglycerols, the Fatty Ones

- **Triacylglycerols = triglycerides = neutral fats = fats and oils in plants and animal**

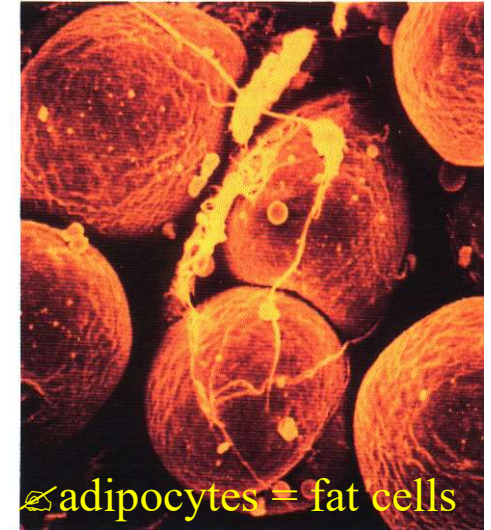


- **Simple triacylglycerols contain only 1 type of fatty acids**
- **Mixed triacylglycerols are more common:**

➤ Triacylglycerols are a highly efficient form of stored energy (less oxidized than carbohydrates or amino acids, and anhydrous \Rightarrow 6-fold higher metabolic energy than glycogen which binds 2-fold its weight of water)



1-Palmitoleoyl-2-linoleoyl-3-stearoyl-glycerol



- store 2-3 months (!) of energy supply (21%/26% of body weight!)
- insulation (whales, seals, penguins!)

