Note

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1 Atomic orbitals

pyscf.pbc.gto.Cell.pbc_eval_ao("GTOval_sph", position, kpts)
Hydrogen.pbc_eval_ao(position)

1.1 Gaussian type orbitals

We use Gaussian type orbitals as basis functions. The basis we used here, as well as in PYSCF, is a little bit different from CP2K. Take gth-dzvp as an example.

```
H DZVP-GTH
1
2
3
                          0
                             4 2
4
                             8.3744350009
                                            -0.0283380461
                                                             0.0000000000
5
                             1.8058681460
                                            -0.1333810052
                                                             0.0000000000
6
                                                             0.0000000000
                             0.4852528328
                                            -0.3995676063
7
                             0.1658236932
                                            -0.5531027541
                                                             1.0000000000
8
                    2
                       1
                         1
                             1 1
9
                             0.7270000000
                                             1.0000000000
```

The file-format used in the .dat file

$$\chi_i(\mathbf{r}) = R_i(r) \cdot Y_{l_i, m_i}(\theta, \phi)$$
$$R_i(r) = r^{l_i} \sum_{j=1}^{N} c_{ij} \exp(-\alpha_j \cdot r^2)$$

$$\chi_i(\mathbf{r}) = \sqrt{4\pi} R_i(r) \cdot Y_{l_i, m_i}(\theta, \phi) \tag{1}$$

$$R_i(r) = r^{l_i} \sum_{j=1}^{N} c_{ij} \left(\frac{2\alpha_j}{\pi}\right)^{\frac{3}{4}} \exp(-\alpha_j \cdot r^2)$$
(2)

1.2 PBC orbitals

Crystalline orbitals Gaussian basis function ϕ is a lattice sum over local Gaussians χ

$$\phi_{\mathbf{k},i}(\mathbf{r}) = \sum_{\mathbf{T}} e^{i\mathbf{k}\cdot\mathbf{T}} \chi_i(\mathbf{r} - \mathbf{T})$$
(3)

where \mathbf{k} is a vector in the first Brillouin zone and \mathbf{T} is a lattice translational vector.

2 PBC Hartree Fock

Here we use index μ, ν to specify atoms, p, q, r, s to specify orbitals of each atom, i, j, k, l to specify coefficients of each orbital, c, d to specify cells. Then the p th pbc gaussian type orbital of μ th atom can be described as

$$|\mathbf{R}_{\mu,p}\rangle = \sum_{i,c} c_{pi} |\alpha_i \mathbf{R}_{\mu,c}\rangle$$
 (4)

Here are some commonly used shorthand symbols.

$$\alpha_{ij} = \frac{\alpha_i \alpha_j}{\alpha_i + \alpha_j}$$

$$\alpha_{ij,kl} = \frac{(\alpha_i + \alpha_j)(\alpha_k + \alpha_l)}{\alpha_i + \alpha_j + \alpha_k + \alpha_l}$$

$$\mathbf{R}_{\mu i,\nu j,c} = \frac{\alpha_i \mathbf{R}_{\mu} + \alpha_j \mathbf{R}_{\nu,c}}{\alpha_i + \alpha_j}$$

2.1 Integtals

General integral on PBC basis is given by

$$A_{ij} = \sum_{m,n} \int_0^T a(x)\chi_i(x+mT)\chi_j(x+nT)dx$$

$$= \sum_{m,n} \int_{mT}^{(m+1)T} a(x'-mT)\chi_i(x')\chi_j[x'+(n-m)T]dx'$$

$$= \sum_{m,n-m} \int_{mT}^{(m+1)T} a(x')\chi_i(x')\chi_j[x'+(n-m)T]dx'$$

$$= \sum_{n'} \int_{-\infty}^{\infty} a(x')\chi_i(x')\chi_j(x'+n'T)dx'$$

2.1.1 Overlap

The overlap matrix element is given by

$$O_{\mu p, \nu q} = \langle \mathbf{R}_{\mu p} | \mathbf{R}_{\nu q} \rangle = \sum_{c} \sum_{i,i} c_{pi} c_{qj} \left(\frac{2\sqrt{\alpha_i \alpha_j}}{\alpha_i + \alpha_j} \right)^{\frac{3}{2}} \exp\left[-\alpha_{ij} (\mathbf{R}_{\mu} - \mathbf{R}_{\nu,c})^2 \right]$$
 (5)

$$O_{\mu p i, \nu q j, c} = c_{p i} c_{q j} \left(\frac{2\sqrt{\alpha_{i} \alpha_{j}}}{\alpha_{i} + \alpha_{j}}\right)^{\frac{3}{2}} \exp\left[-\alpha_{i j} (\mathbf{R}_{\mu} - \mathbf{R}_{\nu, c})^{2}\right]$$

$$(6)$$

$$O_{\mu p,\nu q} = \sum_{c} \sum_{ij} O_{\mu pi,\nu qj,c} \tag{7}$$

2.1.2 Kinetic

The kinetic matrix element

$$T_{\mu p, \nu q} = \langle \mathbf{R}_{\mu p} | -\frac{1}{2} \nabla^2 | \mathbf{R}_{\nu q} \rangle = \sum_c \sum_{ij} c_{pi} c_{qj} \left(\frac{2\sqrt{\alpha_i \alpha_j}}{\alpha_i + \alpha_j} \right)^{\frac{3}{2}} \exp \left[-\alpha_{ij} (\mathbf{R}_{\mu} - \mathbf{R}_{\nu,c})^2 \right] \alpha_{ij} \left[3 - 2\alpha_{ij} (\mathbf{R}_{\mu} - \mathbf{R}_{\nu,c})^2 \right]$$
(8)

$$T_{\mu p,\nu q} = \sum_{c} \sum_{ij} O_{\mu p i,\nu q j,c} \alpha_{ij} \left[3 - 2\alpha_{ij} (\mathbf{R}_{\mu} - \mathbf{R}_{\nu,c})^2 \right]$$

$$(9)$$

2.1.3 Potential

The potential matrix element

$$V_{\mu p, \nu q, N} = \langle \mathbf{R}_{\mu p} | \sum_{d} \frac{1}{|r - R_{N,d}|} | \mathbf{R}_{\nu q} \rangle = 2 \sum_{cd} \sum_{ij} O_{\mu p i, \nu q j, c} \sqrt{\frac{\alpha_i + \alpha_j}{\pi}} F_0 \left[(\alpha_i + \alpha_j) (\mathbf{R}_{N,d} - \mathbf{R}_{\mu i, \nu j, c})^2 \right]$$
(10)

$$V_{\mu p,\nu q} = \sum_{\mathbf{G} \neq 0} \sum_{N} \sum_{c} \sum_{ij} V(\mathbf{G}) O_{\mu p i,\nu q j,c} \exp\left[-i\mathbf{G} \cdot (\mathbf{R}_N - \mathbf{R}_{\mu i,\nu j,c})\right] \exp\left[-\frac{G^2}{4(\alpha_i + \alpha_j)}\right]$$
(11)

where

$$V(\mathbf{G}) = \frac{4\pi}{L^3} \frac{1}{G^2}$$
 (12)

2.1.4 Electron interaction

The interaction matrix element

$$E_{p\mu q\nu,r\gamma s\eta} = \langle \alpha_p \mathbf{R}_{\mu} \alpha_q \mathbf{R}_{\nu} | \frac{1}{|r - r'|} | \alpha_r \mathbf{R}_{\gamma} \alpha_s \mathbf{R}_{\eta} \rangle = 2O_{p\mu,r\gamma} O_{q\nu,s\eta} \sqrt{\frac{\alpha_{pr,qs}}{\pi}} F_0 \left[\alpha_{pr,qs} (\mathbf{R}_{p\mu,r\gamma} - \mathbf{R}_{q\nu,r\eta})^2 \right]$$
(13)

Where

$$F_0[x] = \frac{\sqrt{\pi}erf(\sqrt{x})}{2\sqrt{x}} \& F_0[0] = 1$$
 (14)

2.1.5 Hcore

Hamiltonian matrix element without interaction

$$h_{p\mu,q\nu} = T_{p\mu,q\nu} - \sum_{N} V_{p\mu,q\nu,N}$$
 (15)

2.1.6 Fock matrix

Fock operator matrix element

$$F_{p\mu,q\nu} = \langle \alpha_p \mathbf{R}_{\mu} | \hat{H} | \alpha_q \mathbf{R}_{\nu} \rangle = h_{p\mu,q\nu} + \sum_k \sum_{r\gamma,s\eta} (2E_{p\mu r\gamma,q\nu s\eta} - E_{p\mu r\gamma,s\eta q\nu}) C_{r\gamma,k}^* C_{s\eta,k}$$
(16)

2.2 Solution of Roothaan equation

Roothaan equation

$$\mathbf{FC}_k = \epsilon_k \mathbf{SC}_k \tag{17}$$

Energy

$$E_{g} = 2\sum_{k}\sum_{p\mu,q\nu}h_{p\mu,q\nu}C_{p\mu,k}^{*}C_{q\nu,k} + \sum_{k}\sum_{p\mu,q\nu,r\gamma,s\eta}(2E_{p\mu r\gamma,q\nu s\eta} - E_{p\mu r\gamma,s\eta q\nu})C_{p\mu,k}^{*}C_{q\nu,k}C_{r\gamma,k}^{*}C_{s\eta,k}$$
(18)

- 3 Molecule orbitals
- 4 Wave function
- 5 Density
- 6 Gradient logpsi
- 7 Laplacian logpsi