A1 Assignment - Zino Meyer

1.1)

For this exercise, we work our way backwards from the the SVD results to the matrix A. We use the SVD equation $A=UDV^t$.

To focus on the first column, we will use the first column of the identity matrix i_1 . That way, we can compute $A_1=UDV^t\cdot i_1.$

Starting from the right:

$$V^T \cdot e_1 = egin{pmatrix} -1 & 0 \ 0 & -1 \end{pmatrix} egin{pmatrix} 1 \ 0 \end{pmatrix} = egin{pmatrix} -1 \ 0 \end{pmatrix}$$

Next, we multiply by D:

$$D \cdot (V^T \cdot i_1) = egin{pmatrix} 1.414214 & 0 \ 0 & 1.414214 \end{pmatrix} egin{pmatrix} -1 \ 0 \end{pmatrix} = egin{pmatrix} -1.414214 \ 0 \end{pmatrix}$$

Finally, we calculate $U \cdot (D \cdot V^T \cdot i_1)$:

$$U \cdot (D \cdot V^T \cdot i_1) = egin{pmatrix} -0.7071068 & -0.7071068 \ -0.7071068 & 0.7071068 \end{pmatrix} egin{pmatrix} -1.414214 \ 0 \end{pmatrix} = egin{pmatrix} 1 \ 1 \end{pmatrix}$$

Successfully recovered the first column: $A_1 = egin{pmatrix} 1 \\ 1 \end{pmatrix}$

1.10)

Given: we need to balance the equation $C_3H_8+xO_2 o yCO_2+zH_2O$

Solution:

Carbon:
$$1 \times 3 = y \Rightarrow y = 3$$

$$\text{Hydrogen:}\quad 1\times 8=2z\Rightarrow z=4$$

Oxygen:
$$2x = 2y + z$$

Substituting C and H into O:

$$\Rightarrow 2x = 2 \times 3 + 4$$

$$\Rightarrow 2x = 10$$

$$\Rightarrow x = 5$$

Thus, the balanced equation is: $C_3H_8+5O_2
ightarrow 3CO_2+4H_2O$

2.4 a)

To proof, I will express one row as a linear combination of the others explicitly. For the matrix

$$A = egin{bmatrix} 1 & 2 & 3 \ 4 & 5 & 6 \ 7 & 8 & 9 \end{bmatrix}$$

the third row can be expressed as a combination of the first two:

$$aegin{bmatrix}1\\2\\3\end{bmatrix}+begin{bmatrix}4\\5\\6\end{bmatrix}=egin{bmatrix}7\\8\\9\end{bmatrix}$$

This equation has a solution for a=-1 and b=2, showing that row 3 (vector 3) is a linear combination of row 1 & 2 (vectors 1 & 2), thus, the rows are linearly dependant

b)

Since row 3 can be expressed by row one and two, we reduce the system of equations by row three and end up with:

1.
$$x_1 + 2x_2 + 3x_3 = 0$$

2.
$$4x_1 + 5x_2 + 6x_3 = 0$$

Treating x3 as arbitrary value, we can rewrite the system like this:

1.
$$x_1 + 2x_2 = -3x_3$$

2.
$$4x_1 + 5x_2 = -6x_3$$

where x_3 is our arbitrary value, and x_1 and x_2 are variables.

c)

Solve the two equations for x_1 and x_2 and express them in terms of x_3 :

1.
$$x_1 + 2x_2 = -3x_3$$

2.
$$4x_1 + 5x_2 = -6x_3$$

Use elimination to solve for x_2 :

$$4 \times 1. - 2.$$

$$\Rightarrow 3x_2 = -6x_3$$

$$\Rightarrow x_2 = -2x_3$$

Substitude x_2 into 1.:

$$x_1 + 2(-2x_3) = -3x_3$$

$$\Rightarrow x_1 - 4x_3 = -3x_3$$

$$\Rightarrow x_1 = x_3$$

So the solutions are:

$$x_1 = x_3$$

$$x_2=-2x_3$$

$$x_3 = x_3$$

2.5)

Given: we need to balance the equation $2C_2H_6+xO_2 o yCO_2+zH_2O$

Solution:

Carbon:
$$2 \times 2 = y \Rightarrow 4 = y$$

Hydrogen:
$$2 \times 6 = 2z \Rightarrow 6 = z$$

Oxygen:
$$2x = 2y + z$$

Substituting ${\cal C}$ and ${\cal H}$ into the Oxygen equation:

$$\Rightarrow 2x = 2(2 \times 2) + \frac{1}{2}(2 \times 6)$$

$$\Rightarrow 2x = 8 + 6$$

$$\Rightarrow x = 7$$

The balanced equation: $2C_2H_6+7O_2
ightarrow 4CO_2+6H_2O$

2.7)

Given matrix & vector:

$$A = egin{bmatrix} 0 & 4 \ -2 & -7 \end{bmatrix}, u = egin{bmatrix} 1 \ 1 \end{bmatrix}$$

a)

$$\mathbf{Au} = egin{bmatrix} 0 & 4 \ -2 & -7 \end{bmatrix} egin{bmatrix} 1 \ 1 \end{bmatrix} = egin{bmatrix} 0 imes 1 + 4 imes 1 \ -2 imes 1 + (-7) imes 1 \end{bmatrix} = egin{bmatrix} 4 \ -9 \end{bmatrix}$$

Since $\mathbf{A}\mathbf{u} \neq 4 \times \mathbf{u}$, where $4 \times \mathbf{u} = \begin{bmatrix} 4 \\ 4 \end{bmatrix}$, it shows that \mathbf{u} is not an eigenvector of \mathbf{A} . The vector is not only scaled, but also changes its direction.