



IDX G10 Biology H
Study Guide Issue S1 M2
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9.1 Cellular Respiration: An Overview

- a calorie: amount of energy needed to raise 1 degree celsius of 1 gram of water
- a Calorie: kilocalorie(1000 calorie), used on food label
- energy obtained from food
- cellular respiration: releases energy by breaking down food in prescence of oxygen
 - $6 \text{ O}_2 + \text{C}_6\text{H}_{12}\text{O}_6 \rightarrow 6 \text{ CO}_2 + 6 \text{ H}_2\text{O} + \text{Energy}$
- Glycolysis
 - cytoplasm, only a small amount of ATP produced
 - 90% of glucose's energy locked in the chemical bonds of pyruvic acid
- Krebs Cycle
 - matrix of mitochondrian
 - a little more ATP is generated from pyruvic acid
- Electron transport chain
 - inner mitochondrial membrane
 - produces great amount of ATP by using O_2 as the final electron acceptor
- Aerobic: requirie oxygen, Krebs Cycle and electron transport chain
- Anaerobic: not requirie oxygen, glycolysis

9.2 Process of Cellular Respiration

Glycolysis: process in which 1 molecule of glycolysis is split into two, producing 2 molecules of pyruvic acid, a 3 carbon compound

- happen in cytosol--the aqueous part of cytoplasm
- ATP Production
 - $2\text{ATP} \rightarrow 2\text{ADP}$
 - $4\text{ADP} \rightarrow 4\text{ATP}$
 - net gain of 2 ATP

- NADH production
 - electron carrier $\text{NAD}^+ \rightarrow \text{NADH}$
 - help pass energy from glucose to other pathways
- glycolysis does not require oxygen. Supply chemical energy when oxygen is not available
- product: 2 pyruvic acid enters krebs cycle
- Overall: $\text{glucose} + 2\text{NAD}^+ + 2\text{ATP} \rightarrow 2 \text{ pyruvate} + 2 \text{ NADH} + 4\text{ATP}$
- Formation of Acetyl-CoA
 - 2 pyruvates are transported to the mitochondria matrix
 - pyruvate is converted into acetyl-CoA, which enters the Krebs Cycle
 - electrons released from pyruvate are transferred to NAD^+ , which becomes NADH
 - NADH shuttles electrons to ETC

Krebs Cycle(Citric Acid Cycle)

- mitochondrial matrix
- product: carbon dioxide from organic immediate, ATP, NADH carrying electrons, FADH_2 carrying electrons
- Acetyl-CoA combines with oxaloacetate to form citrate
- harvest high energy electrons from carbon bond--electrons captured by NAD^+ , FAD
- citric acid- $\text{CO}_2 \rightarrow 5\text{C}$ compound- $\text{CO}_2 \rightarrow 4\text{C}$ compound
- a total of 3 carbon dioxide molecules are released in one cycle
- $\text{citrate} + \text{NAD}^+ \rightarrow 4\text{C compound} + \text{CO}_2 + \text{NADH}$
- $5\text{C compound} + \text{NAD}^+ \rightarrow 4\text{C compound} + \text{CO}_2 + \text{NADH}$
- $4\text{C compound} + \text{FAD} + \text{NAD}^+ + \text{ADP} + \text{P}_i \rightarrow \text{OXAL} + \text{FADH}_2 + \text{NADH} + \text{ATP}$
- for each pyruvate, 1 ATP produced

Oxidative Phosphorylation

- happen in inner membrane of mitochondria in eukaryote
- electrons extracted in glycolysis and krebs cycle are transferred by NADH and FADH_2 to the ETC
- ETC transfers electron in a series of oxidation reduction reaction

- energy released and pump hydrogen ions against the concentration gradient, from matrix into intermembrane space
- Oxygen is the final electron acceptor and is reduced into water molecule
- folding of inner membrane increases the surface area for ETC to happen on, allowing more ATP synthesis
- As hydrogen ions are pumped, an electrochemical gradient of hydrogen ions established
 - higher concentration (more positively charged) in the intermembrane space than matrix
- during the chemiosmosis, hydrogen ions flow from intermembrane space back to the matrix through ATP synthase, moving down the electrochemical gradient. The movement provides energy to form ATP

Conclusion

- 38% of energy by 1 glucose from 36(38)ATP
- 62% of glucose's energy is released as heat to keep warm
- net 2 ATP from glycolysis
- krebs cycle: $2 \text{ pyruvate} \rightarrow 8\text{NADH} + 2\text{FADH}_2 + 2\text{ATP} + 6\text{CO}_2$
- electron transport: $10\text{NADH} + 2\text{FADH}_2 \rightarrow 10\text{NAD}^+ + 2\text{FAD} + 32 \text{ or } 34 \text{ ATP}$
- for prokaryotes, the ETC locates on cell membrane (hydrogen ions pumped from cytoplasm to outside)

9.3 Cellular Energies: Fermentation

- fermentation: release energy from food molecules in absence of oxygen--allow glycolysis to continue supply of ATP
- Alcoholic fermentation
 - produce ethyl alcohol and carbon dioxide
 - in cytoplasm of yeast and some microorganisms
 - $\text{glucose} \rightarrow 2\text{ATP} + 2\text{Alcohol} + 2 \text{CO}_2$
 - two steps: glycolysis, alcoholic fermentation
 - $\text{glucose} + 2 \text{NAD}^+ + 2\text{ATP} \rightarrow 2 \text{pyruvate} + 2\text{NADH} + 4\text{ATP}$
 - $2 \text{pyruvic acid} + 2\text{NADH} \rightarrow 2 \text{Alcohol} + 2\text{CO}_2 + 2\text{NAD}^+$

- Lactic acid fermentation
 - lactic acid as waste
 - $2 \text{ pyruvic acid} + 2\text{NADH} \rightarrow 2 \text{ lactic acid} + 2\text{NAD}^+$
 - produced during rapid exercise where there is not enough oxygen and thus not enough ATP produced
- both kinds of fermentation regenerate NAD^+ so that glycolysis can continue
- muscles use stored ATP, lactic acid fermentation, cellular respiration
- Quick energy
 - stored ATP: contained by cell from glycolysis and cellular respiration, only for a few seconds of intense activity
 - lactic acid fermentation: after stored ATP used up, enough ATP to last 90s
- Long term energy
 - cellular respiration--release energy slower than fermentation
 - use energy stored in muscle--glycogen(only for 15-20min)
 - after 15-20min, body break down other stored molecules(eg. fats)